Electronic Supplementary Information for

Pyrrolizidine and cyclobutane bridged double-caged fullerene derivatives

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2. NMR spectra of the synthesized compounds

Figure S 1. ¹H NMR spectrum of compound 2a.



Figure S 2. ¹³C NMR spectrum of compound 2a.



Figure S 3. $^{1}H-^{13}C$ HSQC spectrum of compound 2a.



Figure S 4. $^{1}H^{-13}C$ HMBC spectrum of compound **2a**.



Figure S 5. ¹H NMR spectrum of compound 2b.



Figure S 6. ¹³C NMR spectrum of compound **2b**.



Figure S 7. ¹H–¹³C HSQC spectrum of compound 2b.



Figure S 8. ¹H–¹³C HMBC spectrum of compound 2b.







Figure S 10. ¹³C NMR spectrum of compound 2c.



Figure S 11. $^{1}H^{-13}C$ HSQC spectrum of compound **2c**.



Figure S 12. $^{1}H^{-13}C$ HMBC spectrum of compound **2c**.



Figure S 13. ¹H NMR spectrum of compound 3a.



Figure S 14. ¹³C NMR spectrum of compound 3a.



Figure S 15. ¹H NMR spectrum of compound 3b.



Figure S 16. ¹³C NMR spectrum of compound 3b.









Figure S 19. $^{1}H^{-13}C$ HMBC spectrum of compound 1a.



Figure S 20. ¹H NMR spectrum of compound 1b.







Figure S 22. ¹H–¹³C HSQC spectrum of compound 1b.



Figure S 23. ¹H–¹³C HMBC spectrum of compound 1b.

3. Mass spectra of the synthesized compounds



Figure S 24. MALDI mass spectrum of the reaction mixture after synthesis of 1a. Inset: mass range 1594–1600.



Figure S 25. MALDI mass spectrum of compound 2a. Inset: PSD spectrum of M⁻, M= 2a.



Figure S 26. MALDI mass-spectrum of compound 2b. Inset: mass range 1566–1572.



Figure S 27. MALDI mass-spectrum of compound 2c. Inset: mass range 1678–1684.



Figure S 28. HRMS spectrum (ESI) of compound 1a.



Figure S 29. HRMS spectrum (ESI) of compound 1b.



Figure S 30. HRMS spectrum (ESI) of compound 3a.



Figure S 31. HRMS spectrum (ESI) of compound 3b.

4. UV/Vis spectra of compounds 1a-c and 2a-c



Figure S 32. UV/Vis spectra of fulleropyrrolidines **1a–1c** and double-caged derivatives **2a–2c** (solvent toluene, 290–950 nm range).

5. Quantum chemical calculations



Figure S 33. Side views of two configurations of the DFT-optimized double-caged derivatives 2a–c (the best conformation are shown)

2 (without cyclobutane bridge)

double-caged derivatives $2a-2c$ (structures are shown on figures 34 at page S29)							
Configuration #	2a	2b	2c				
1 (<i>cis</i> -1)	0	0	0				

28.4

28.8

Table S 1. Relative DFT energy values (PBE/TZ2P, $kJ \cdot mol^{-1}$) of various configurations of double-caged derivatives **2a**–**2c** (structures are shown on figures 34 at page *S*29)

33.5

Table S 2. Experimental and DFT-calculated (PBE/TZ2P) ¹³C chemical shifts for **2a**, **2b**, **2c** and their C_s -symmetrical configurations #1 and #2 (for structures see figures 34 at page S29)

		¹³ C chemical shift (ppm)		shift	Scheme of carbon atom numeration		
	Carbon atom				5 5'		
C		Predicted for		ted for			
Comp.		T	configuration				
		Exp.					
			1	2			
	C(2)	108.95	109.0	101.5			
	C(3)	84.65	87.5	80.3			
	C(4)	69.67	72.8	75.8			
	C(5)	71.86	67.2	54.7			
2a	C(6)	77.27	78.2	148.0			
	C(10)	70.99	74.3	133.5			
	C(O)	168.11	164.4	163.1			
	O–CMe ₃	84.13	84.0	87.0			
	CH ₃	28.03	20.7	21.6			
	C(2)	108.35	110.0	97.8			
	C(3)	84.48	87.8	79.4			
	C(4)	69.11	72.3	75.8			
	C(5)	71.73	68.7	54.1			
2b	C(6)	77.31	78.3	148.3			
	C(10)	74.32	74.1	133.4			
	C(O)	169.76	165.1	162.7			
	O–CH ₂	61.55	60.3	58.6			
	CH ₃	14.09	7.1	5.3			
	C(2)	109.53	109.4	97.9			
	C(3)	84.64	87.6	79.3			
	C(4)	69.08	72.6	75.8			
	C(5)	71.86	67.6	54.1			
2c	C(6)	-	78.3	148.3			
	C(10)	_	74.1	133.4			
	C(0)	169.93	164.5	162.4			
	O-CH ₂	66.13	64.8	63.6			
	CH ₃	14.41	8.5	7.4			



shielding values (PBE/TZ2P) for related fullerene derivatives (see Table S4 for details).

Table S3. Experimental ¹³C chemical shifts, DFT-calculated isotropic shielding values (PBE/TZ2P) and predicted ¹³C chemical shifts for related fullerene derivatives.

		¹³ C chemical shift (ppm)		
Commond	Carbon		DFT predicted	
Compound	atom	Exp.	isotropic	Fitted
		-	shielding	value
	C(2)	75.49	-94.2	75.49
Fpr-OtBu, 1a	C(3)	76.55	-89.0	76.7
O II	C(4)	75.81	-92.5	75.5
H	C(5)	62.62	-107.5	62.62
	C(0)	168.5	-2.9	168.5
5´ `2´ `O—tBu	C–Me ₃	83.73	-87.7	83.73
	CH ₃	28.27	-152.9	28.27
	C(6)	150.53	-20.6	150.53
	C(7)	153.53	-13.7	153.53
	C(8)	155.73	-17.9	155.73
	C(9)	153.85	-13.9	153.85
	C(2)	75.42	-94.8	75.42
Fpr-OEt, 1b	C(3)	77.39	-88.7	77.39
0 	C(4)	78.29	-92.6	78
H II	C(5)	63.3	-107.2	63.3
	C(O)	169.55	-2.8	169.55
5^{-}_{1} 2^{-}_{2} $O^{}Et$	OCH ₂	61.92	-112.4	61.92
\backslash /	CH ₃	14.51	-166.7	14.51
	C(6)	151.02	-20.8	151.02
	C(7)	153.39	-13.9	153.39
	C(8)	155.01	-17.8	155.01
	C(9)	153.79	-14.2	153.79
MeFP [S1]	C(2)	69.97	-101.1	69.97
N	C(3)	71.07	-90.1	71.07
2	C(6)	154.72	-13.6	154.72
3 7	C(7)	147.2	-17.4	147.2
	CH ₃	41.48	-135.1	41.48
	C(2)	72.1	-94.7	72.1
Me ₂ FPr-OEt [S2]	C(3)	78.2	-87.9	78.2
	C(4)	76.5	-85.8	76.5
н	C(5)	70.6	-93.6	70.6
\mathbb{N}	C(O)	169.2	-2.6	169.2
5 2 0—Et	OCH ₂	61.9	-112.4	61.9
- \ /	CH ₃ (Et)	14.4	-166.7	14.4
4-3	C(6)	151.3	-20.6	151.3
	C(7)	153.4	-13.0	153.4
	C(8)	153.3	-17.9	153.3
	C(9)	154	-16.1	154
	$CH_3(1)$	28.2	-149.3	28.2

	CH ₃ (2)	29.4	-147.0	29.4
[6,6]-C ₁₂₀ [S3]	C(1)	76.22	-91.1	79.8
	C(2)	151.42	-17.7	150.6
C ₁₂₀ CH ₂ [S4]	C(1)	82.4	-85.1	85.5
	C(2)	68.4	-100.6	70.6
	C(3), CH ₂	59.9	-112.2	59.5
$C_{120}O[S4]$	C(1)	78.9		82.4
	C(2)	99	-69.7	100.4
$C_{120}(CH_2)_2[S5]$	C(1)	74.93	-93.2	77.8
	C(2)	63.52	-104.2	67.2
	C(3), CH ₂	57.35	-113.7	58.0

6. References

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