Electronic Supplementary Information:

The self-assembly effect in NLO polymers containing isolation chromophores: enhanced NLO coefficient and stability

Wenbo Wu,^a Zhen Xu,^a Ying Xiong,^{*a} Shaohui Xin,^a Hongding Tang,^a Cheng Ye,^b Guofu Qiu,^c Jingui Qin,^a and Zhen Li^{*a}



Chart S1 The structure of NLO polymer PS1 containing isolation chromophore.



Chart S2 The structure of dendron-substituted NLO chromophores containing perfluoroaromatic rings.



Chart S3 The structures of NLO polymers PS2-PS5 with different types of isolation groups.



Fig. S1 The ¹⁹F NMR spectra of P2 (up), P3 (middle) and P4 (down).



Fig. S2 TGA thermograms of P1-P4, measured in nitrogen at a heating rate of 10 °C/min.



Fig. S3 UV-vis spectra of polymer P1 in different solvents (0.02 mg/mL).



Fig. S4 UV-vis spectra of polymer P2 in different solvents (0.02 mg/mL).



Fig. S5 UV-vis spectra of polymer P3 in different solvents (0.02 mg/mL).



Fig. S6 UV-vis spectra of polymer P4 in different solvents (0.02 mg/mL).



Fig. S7 Graphical illustration of the alignment formation of self-assembled polymers.



Fig. S8 Graphical illustration of the alignment formation of self-assembled chromophore **3** by Ar-ArF interactions: (a) locked random dipoles (shown as arrows) before poling; (b) unlocked random dipoles before poling; (c) locked acentric dipoles after poling followed by cooling.

Table S1. The maximum absorption of hyperbranched polymers (λ_{max} , nm).

	THF	1,4-dioxane	chloroform	dichloromethane	DMF	DMSO
P1	454	447	447	449	465	473
P2	456	446	447	455	468	473
P3	456	452	454	454	470	475
P4	453	443	447	448	464	470



Chart S4 Some NLO polymers and their d_{33} values.



Fig. S9 Absorption spectra of the film of P3 before and after poling.



Fig. S10 Absorption spectra of the film of P4 before and after poling.