SUPPORTING INFORMATION

Synthesis and Fast Electron-Transfer Reactions of Fullerene-Carbazole Dendrimers with Short Linkages

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Figure captions

Figure S1. MALDI-TOF mass spectra of compound 1.

Figure S2. MALDI-TOF mass spectra of compound 2.

Figure S3. MALDI-TOF mass spectra of compound 3.

Figure S4. Rise profile of the C_{60} of 1 at 1000 nm shown in Figure 4, monitoring charge separation process.

Figure S5. Decay profile of the C_{60} at 1000 nm of **2** in PhCN, monitoring charge separation from carbazole to the singlet C_{60} .

Figure S6. Decay Profiles of the C₆₀ radical anion at 1000 nm of **1** (upper figure), **2** (middle figure), and **3** (lower figure) in deaerated DMF. The concentrations are kept at 1.0×10^{-4} M; $\lambda_{ex} = 390$ nm.

Figure S7. (a) CV and (b) DPV of **1** $(1.0 \times 10^{-4} \text{ M})$ in deaerated PhCN with TBAPF₆ (0.1 M) as support electrolyte in PhCN. Scan rate = 20 mV/s.

Figure S8. (a) CV and (b) DPV of **2** $(1.0 \times 10^{-4} \text{ M})$ in deaerated PhCN with TBAPF₆ (0.1 M) as support electrolyte in PhCN. Scan rate = 20 mV/s.

Figure S9. (a) CV and (b) DPV of **3** $(1.0 \times 10^{-4} \text{ M})$ in deaerated PhCN with TBAPF₆ (0.1 M) as support electrolyte in PhCN. Scan rate = 20 mV/s.

Figure S10. Optimized Frontier molecular orbitals of 1 obtained by using *ab initio* B3LYP/6-311G method.

Figure S11. Optimized Frontier molecular orbital of 2 obtained by using *ab initio* B3LYP/6-311G method.

Figure S12. (Upper figure) Nanosecond transient absorption spectra of **1** (1.0×10^{-4} M) in deaerated PhCN; $\lambda_{ex} = 430$ nm. (Lower figure) Decay profile of ${}^{3}C_{60}$ * at 700 nm.

Figure S13. (Upper figure) Nanosecond transient absorption spectra of **2** (1.0×10^{-4} M) in deaerated PhCN; $\lambda_{ex} = 430$ nm. (Lower figure) Decay profile of ${}^{3}C_{60}$ * at 700 nm.



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Figure S2. MALDI-TOF mass spectra of compound 2.

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Figure S3. MALDI-TOF mass spectra of compound 3.



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