Supporting Information

p-Sulfonatocalix[4]arene as carrier for curcumin

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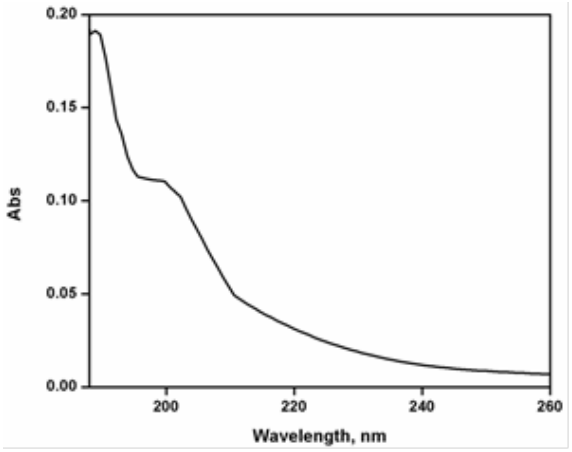

Fig S1. Absorption spectrum of *p*-SC4 (1×10^{-4} M)

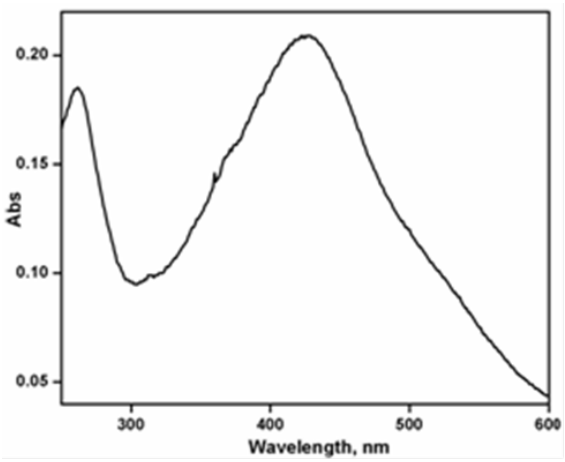

Fig S2. Absorption spectrum of curcumin (1×10^{-6} M).

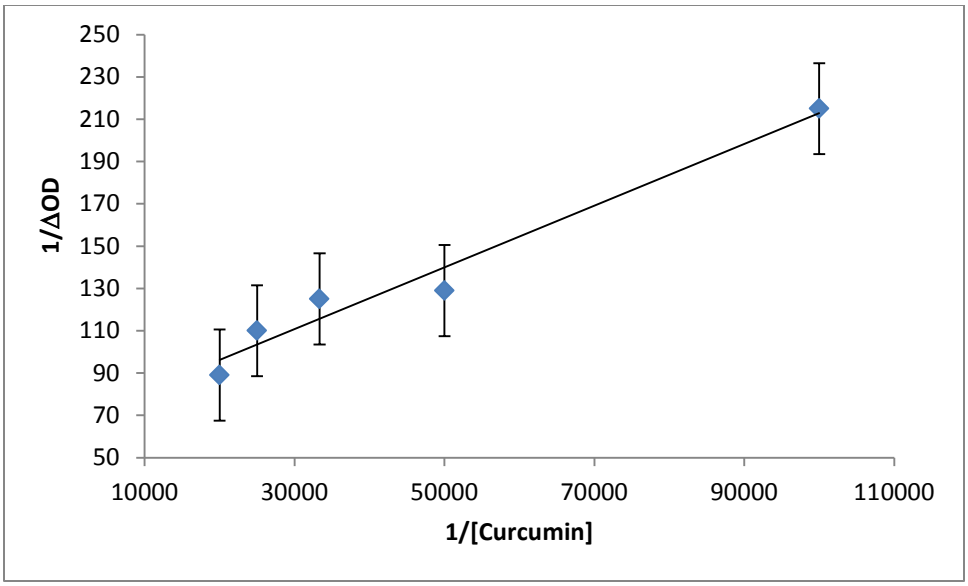

Fig S3. Benesi-Hildebrand plot of curcumin-*p*-SC4 titration by UV-visible spectrum.

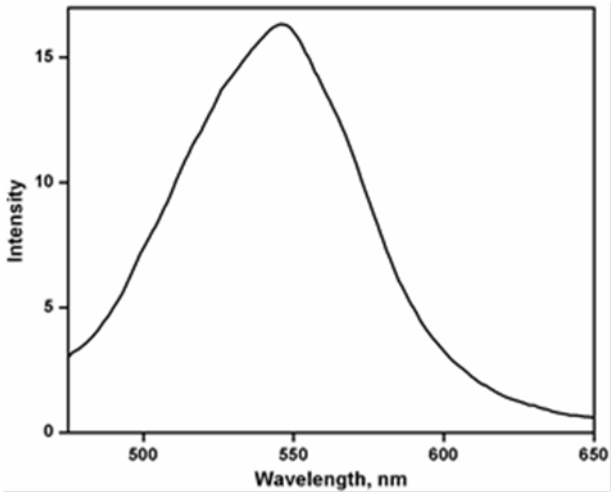

Fig S4. Emission spectrum of curcumin (1×10^{-6} M).

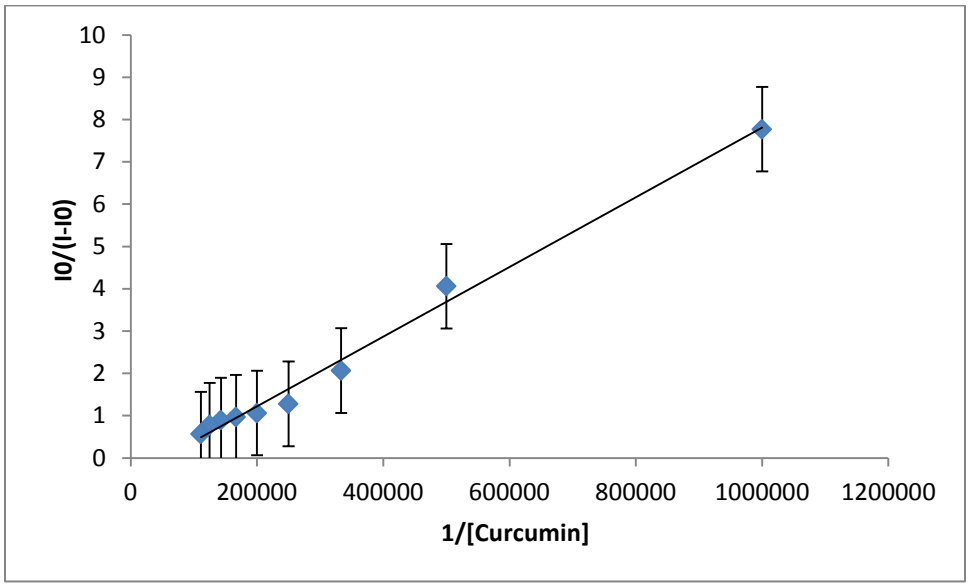

Fig S5. Modified Benesi-Hildebrand plot of curcumin-*p*-SC4 titration by emission spectrum.

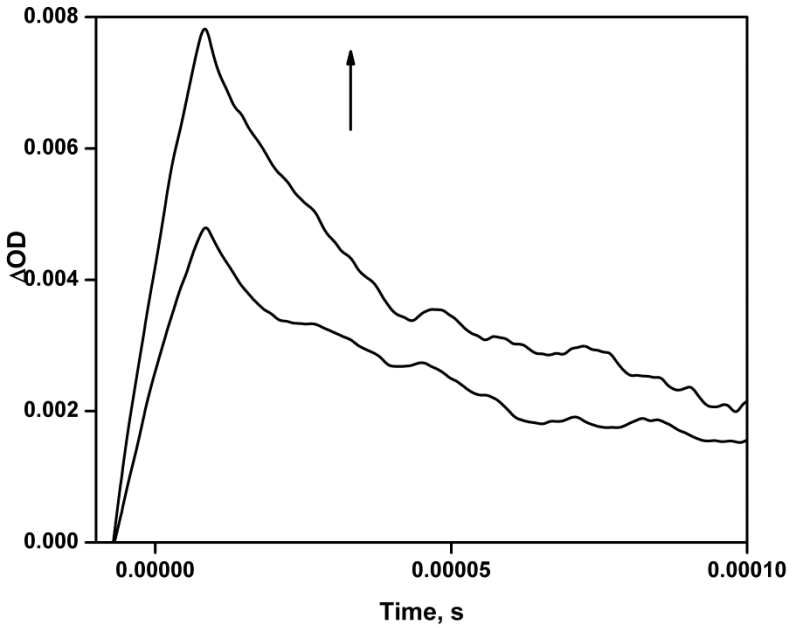

Fig. S6. Kinetics of decay of curcumin in the absence and presence of *p*-SC4 at 500 nm.

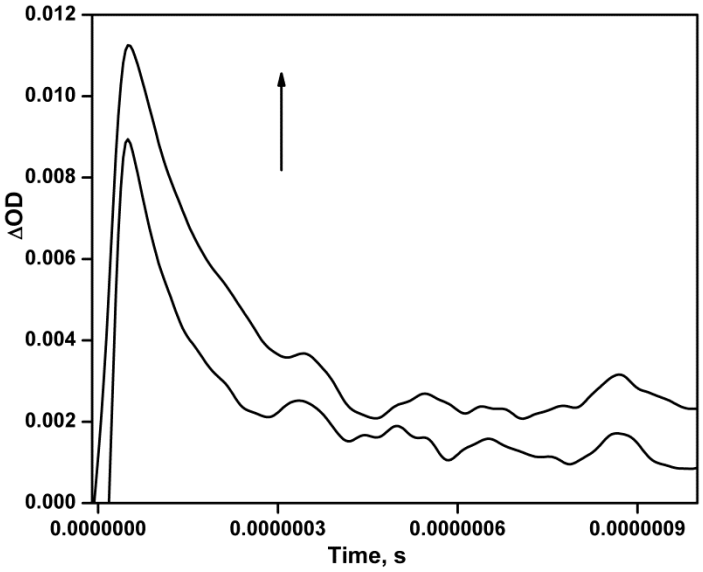

Fig. S7. Kinetics of decay of curcumin in the absence and presence of *p*-SC4 at 680 nm.

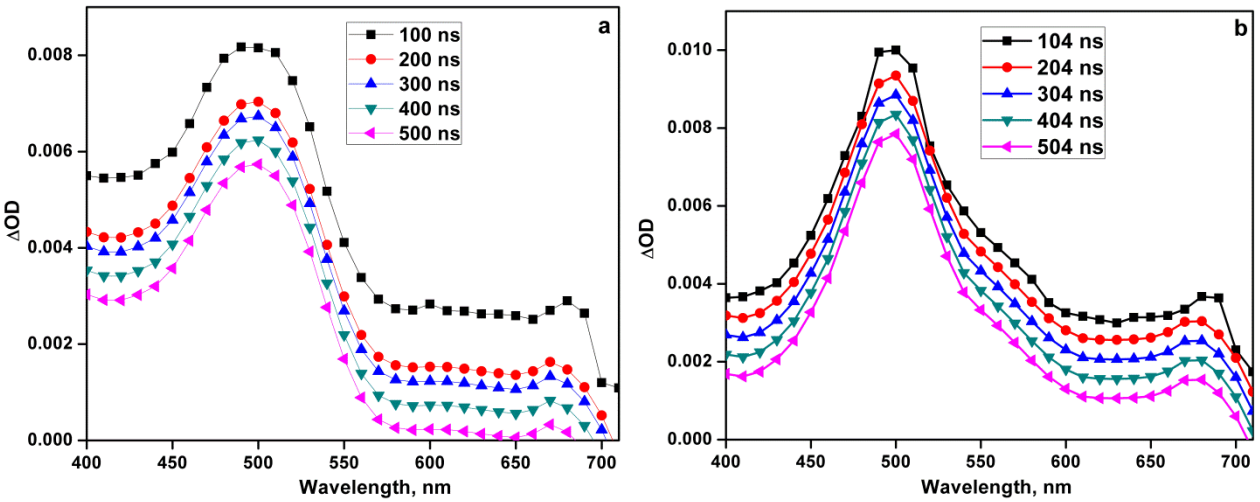

Fig. S8. Transient absorption spectra of (a) curcumin at various time scales, and (b) curcumin-*p*-SC4 at various time scales.

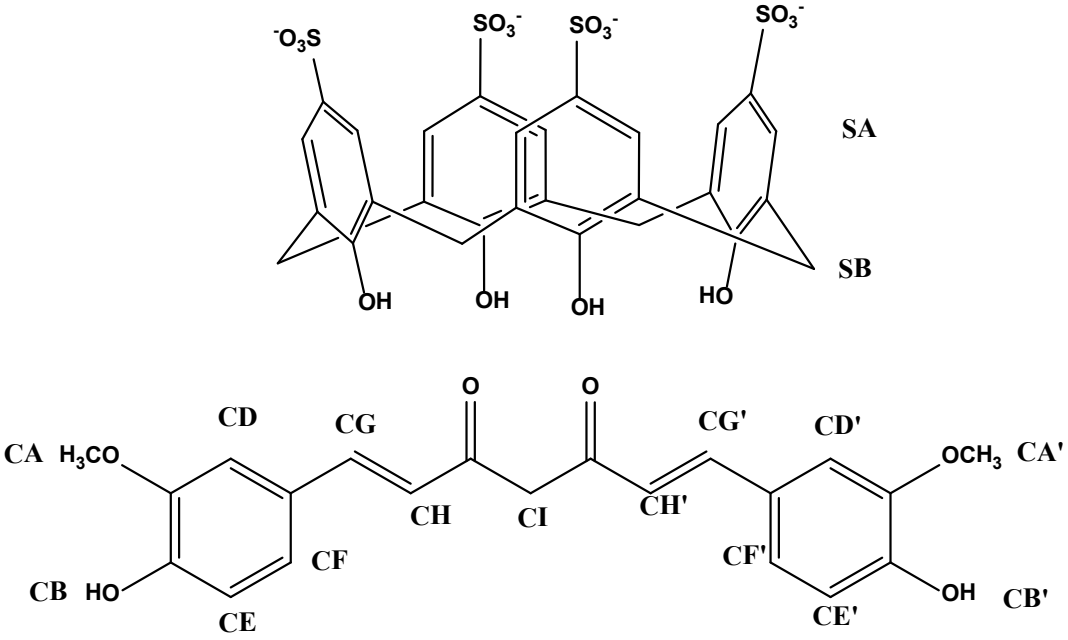

Fig. S9. Naming for proton position in the structure of p-SC4 and curcumin

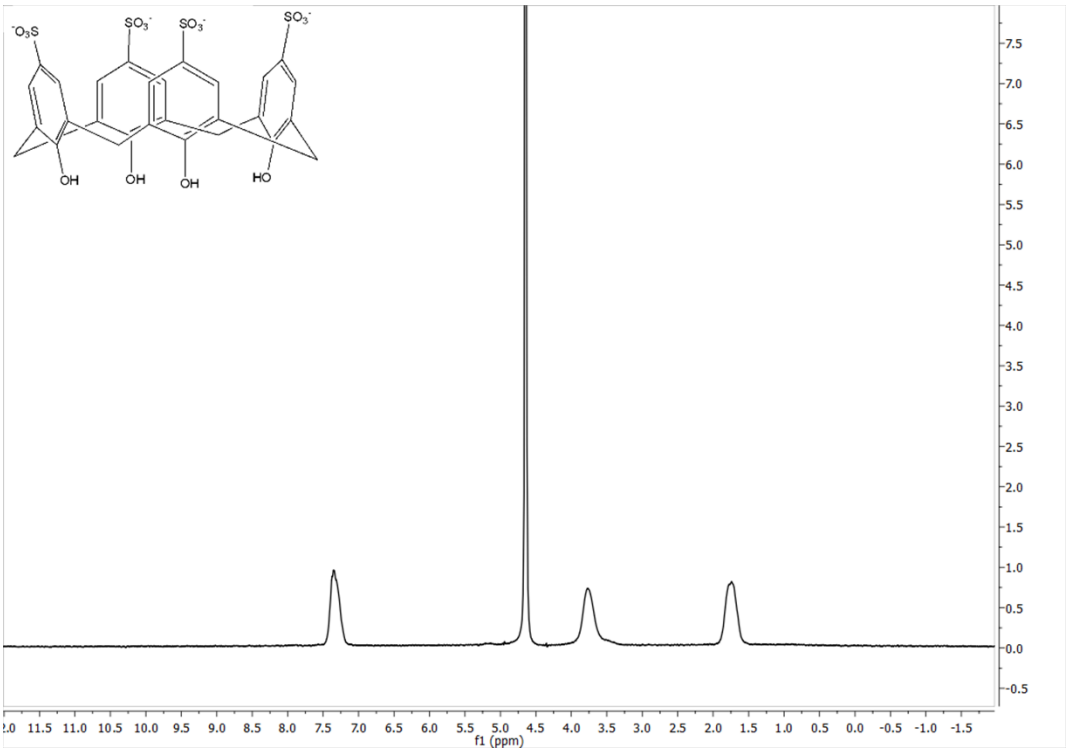

Fig. S10. ^{1}H NMR spectrum of p-SC4 in D_2O/CD_3CN (70/30 %) with few drops of CD_3COOD.

Fig. S11. ^{1}H NMR spectrum of curcumin in D_2O/CD_3CN (70/30 %) with few drops of CD_3COOD.

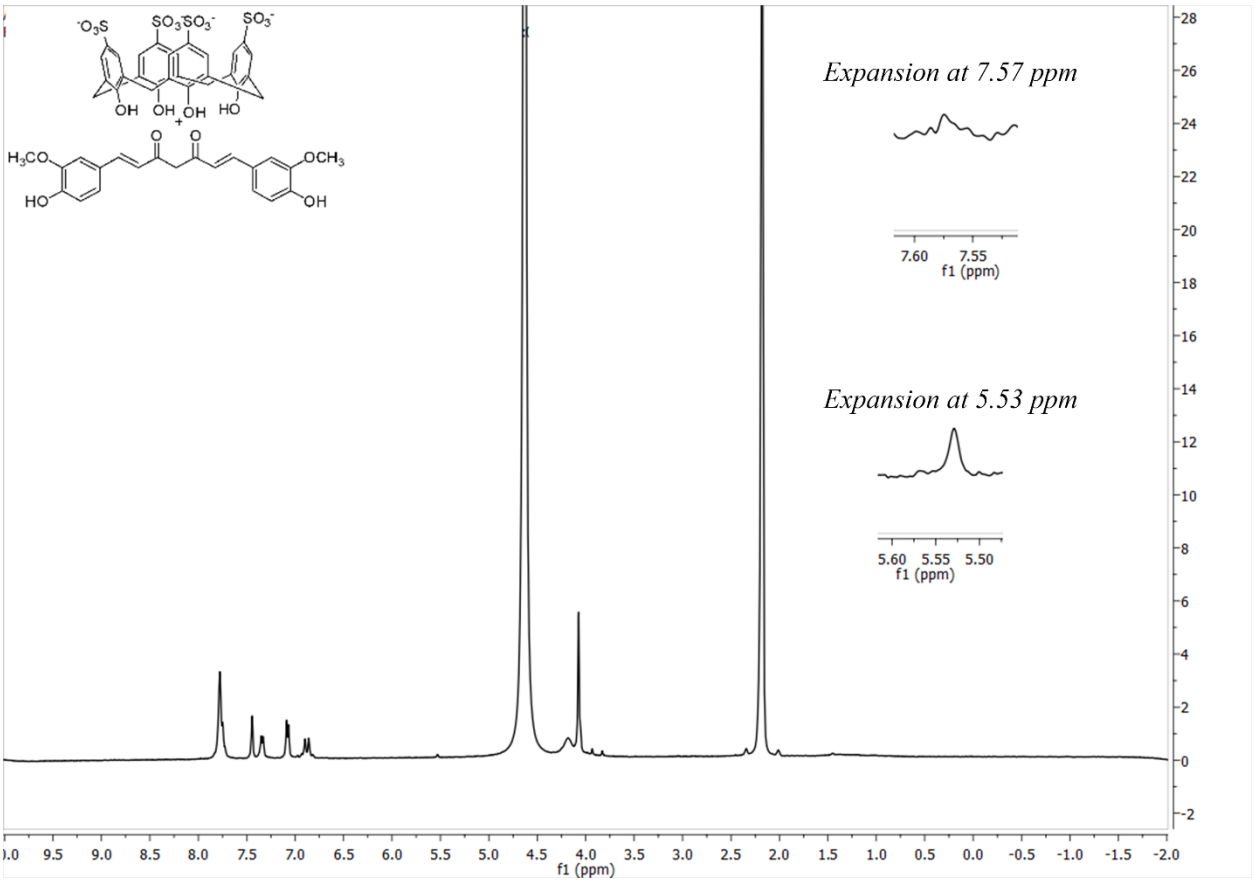

Fig. S12. ^{1}H NMR spectrum of p-SC4-curcumin in D_2O/CD_3CN (70/30 %) with few drops of CD_3COOD.

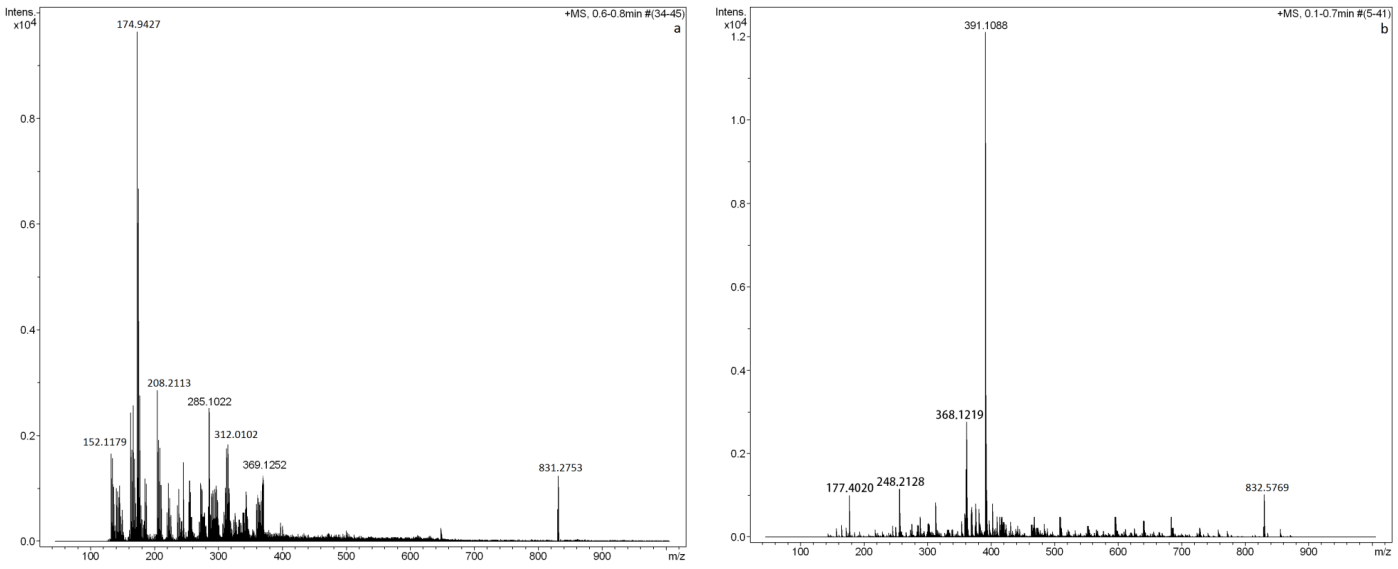

Fig. S13. ESI-MS spectra of 1:1 mixture of curcumin(10^{-3} M) and *p*-SC4 (10^{-3} M) at (a) pH 9.2 and (b) pH 3.

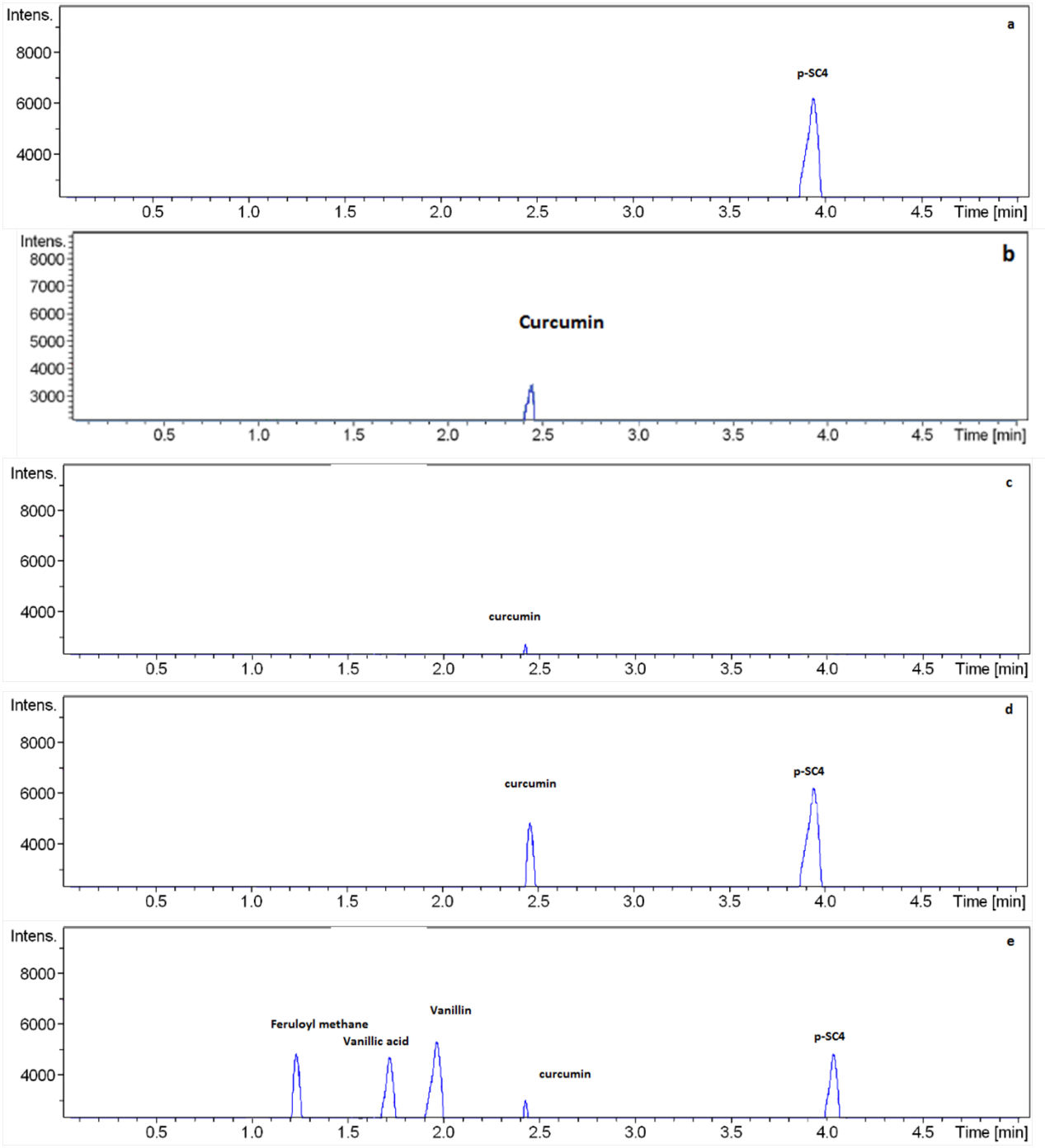

Fig. S14. HTLC chromatograms of (a) p-SC4 (10^{-3} M) at pH 3, (b) curcumin (5 mg in 10 ml water with 1% acetonitrile) at pH 3, (c) curcumin (5 mg in 10 ml), (d) curcumin in p-SC4 solution (5 mg in 10 ml of p-SC4 (10^{-3} M) solution) at pH 3 and (e) curcumin in p-SC4 solution (5 mg in 10 ml of p-SC4 (10^{-3} M) solution) at pH 9.2