Electronic Supporting Information

Dual Colorimetric Receptor with Logic Gate Operations: Anion Induced Solvatochromism

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Fig. S2. ¹H NMR spectra of R2.



Fig. S3. ¹H NMR spectra of R3.

¹³C NMR Spectra



Fig. S4.¹³C NMR spectra of R1.



Fig. S6.¹³C NMR spectra of R3.



Fig. S7 The ORTEP diagrams (50% probability) of receptor R1 and R3

Table S1 Crystallographic data of receptor R1 and R3

Parameters	Receptor R1	Receptor R3	
Chemical formula	$C_{13}H_{10}N_2O_3$	$C_{13}H_{10}N_2O_2$	
Formula weight	242.23	226.23	
Crystal System	Monoclinic	Monoclinic	
Space group	P2(1)/c	P2(1)/n	
a (Å)	11.9767(8)	14.6480(2)	
b (Å)	5.9302(4)	10.8265(2)	
c (Å)	16.1269(10)	14.7278(2)	
S β (°)	96.635(2)	101.940 (2)	
γ (°)	90.00	90.00	
$V(Å)^3$	1137.73(13)	2285.10 (6)	
Z	4	8	
Crystal size	$0.47 \times 0.30 \times 0.27$	$0.47 \times 0.33 \times 0.26$	
F (000)	504	944	
R-factor (%)	5.23	6.85	

Photographs:



Fig. S8. Change in colour of R1 (5 \times 10⁻⁵M) in dry ACN with the addition of 1 equiv. of tetrabutylammonium anions. (a) Free Receptor R1, (b) F⁻, (c) Cl⁻, (d) Br⁻ (e) I⁻, (f) NO₃⁻, (g) HSO₄⁻, (h) H₂PO₄⁻ and (i) AcO⁻.



Fig. S9. Change in colour of R2 $(5 \times 10^{-5} \text{M})$ in dry DMSO with the addition of 1 equiv. of tetrabutylammonium anions. (a) Free Receptor R2, (b) F⁻, (c) Cl⁻, (d) Br⁻ (e) I⁻, (f) NO₃⁻, (g) HSO₄⁻, (h) H₂PO₄⁻ and (i) AcO⁻.



Fig. S10. Change in colour of R2 $(5 \times 10^{-5} \text{M})$ in dry ACN with the addition of 1 equiv. of tetrabutylammonium anions. (a) Free Receptor R2, (b) F⁻, (c) Cl⁻, (d) Br⁻ (e) I⁻, (f) NO₃⁻, (g) HSO₄⁻, (h) H₂PO₄⁻ and (i) AcO⁻.



Fig. S11. Change in colour of R3 $(5 \times 10^{-5} \text{M})$ in dry DMSO with the addition of 1 equiv. of tetrabutylammonium anions. (a) Free Receptor R3, (b) F⁻, (c) Cl⁻, (d) Br⁻ (e) I⁻, (f) NO₃⁻, (g) HSO₄⁻, (h) H₂PO₄⁻ and (i) AcO⁻.



Fig. S12. Receptor R3 (5 \times 10⁻⁵M) in dry ACN after the addition of 1 equiv. of tetrabutylammonium anions. (a) Free Receptor R3, (b) F⁻, (c) Cl⁻, (d) Br⁻ (e) I⁻, (f) NO₃⁻, (g) HSO₄⁻, (h) H₂PO₄⁻ and (i) AcO⁻.



Fig. S13. Change in color after addition of 3 equiv. of different cations (as nitrate salts) to the receptor solution in ACN (5×10^{-5} M). (a) Receptor R2, (b) Mg²⁺, (c) Ca²⁺, (d) Co²⁺, (e) Ni²⁺, (f) Cu²⁺, (g) Zn²⁺, (h) Cd²⁺, (i) Hg²⁺ and (j) Pb²⁺.



Fig. S14. Change in color after addition of 3 equiv. of different cations (as nitrate salts) to the receptor solution in ACN (5×10^{-5} M). (a) Receptor R3, (b) Mg²⁺, (c) Ca²⁺, (d) Co²⁺, (e) Ni²⁺, (f) Cu²⁺, (g) Zn²⁺, (h) Cd²⁺, (i) Hg²⁺ and (j) Pb²⁺.



Fig. S15. Competitive study of receptor R1 (5×10^{-5} M) in DMSO by adding 1 equiv. F⁻ ion and 1 equiv. of other anions. (a) Receptor R1, (b) R1 + F⁻, (c) R1 + Cl⁻ + F⁻, (d) R1 + Br⁻ + F⁻, (e) R1 + I⁻ + F⁻, (f) R1 + NO₃⁻ + F⁻, (g) R1 + HSO₄⁻ + F⁻ and (h) R1 + H₂PO₄⁻ + F⁻.



Fig. S16 Competitive study of receptor R1 (5×10⁻⁵M) in ACN by adding 3 equiv. Cu^{2+} ion and 3 equiv. of other cations. (a) Receptor R1, (b) R1 + Mg²⁺ + Cu²⁺, (c) R1 + Ca²⁺ + Cu²⁺, (d) R1 + Co²⁺ +

 Cu^{2+} , (e) R1 + Ni²⁺ + Cu²⁺, (f) R1 + Cu²⁺, (g) R1 + Zn²⁺ + Cu²⁺, (h) R1 + Cd²⁺ + Cu²⁺, (i) R1 + Hg²⁺ + Cu²⁺ and (j) R1 + Pb²⁺ + Cu²⁺



Fig. S17: Solvatochromic effect of the receptor R2 upon addition of 1 equiv. of F^- ions in different solvents. Top row: R2 solution (5 × 10⁻⁵M) in different solvents. Bottom row: R2+F⁻ ions; (a) 1,4-Dioxane, (b) THF, (c) DCM, (d) Acetone, (e) ACN and (f) DMSO.

UV-vis titrations:



Fig. S18: UV-vis titration spectra of R1 (5×10^{-5} M) with the increasing concentration of TBAOH (0–25 equiv.) in dry DMSO.



Fig. S19: UV-vis titration spectra of R2 $(5 \times 10^{-5} \text{M})$ with the increasing concentration of TBAF (0–25 equiv.) in dry DMSO.



Fig. S20: UV-vis titration spectra of R2 $(5 \times 10^{-5} \text{M})$ with the increasing concentration of TBAF (0–10 equiv.) in dry ACN.



Fig. S21: UV-vis spectra of R3 (5 \times 10⁻⁵M) with the addition TBAF (20 equiv.) in dry DMSO.



Fig. S22: UV-vis spectra of R3 (5×10^{-5} M) with the addition TBAF (20 equiv.) in dry ACN.



Fig. S23: UV-vis titration spectra of R2 (5×10^{-5} M) with the increasing concentration of Cu²⁺ ions (0–4 equiv.) in ACN.

The receptor did not show any new peak even after addition of 20 equiv. of Cu^{2+} solution. However, the peak corresponding to –OH disappeared in UV-vis spectra as shown in Fig. S23 (showed addition of Cu^{2+} ions only upto 4 equiv. for clarity).



Fig. S24: UV-vis titration spectra of R3 (5×10^{-5} M) with 20 equiv. of Cu²⁺ ions in ACN.

UV-Vis changes with different anions:



Fig. S25: UV–Vis changes of R1 in DMSO (5 \times 10⁻⁵M) after addition of 20 equiv. of (a) F⁻ ion, (b) AcO⁻ ion and (c) Cl⁻, Br⁻, I⁻, NO₃⁻, HSO₄⁻ and H₂PO₄⁻ ions in the form of TBA salts.



Fig. S26: UV–Vis changes of R2 in DMSO (5 \times 10⁻⁵M) after addition of 20 equiv. of (a) F⁻ ion, (b) AcO⁻ ion and (c) Cl⁻, Br⁻, I⁻, NO₃⁻, HSO₄⁻ and H₂PO₄⁻ ions in the form of TBA salts.





Fig. S27: UV–Vis changes of R1 in ACN (10×10^{-5} M) after addition of 40 equiv. of metal nitrates (a) Free receptor R1 (b) R1+ Cu²⁺ and (c) R1+ other metal salts (Mg²⁺, Ca²⁺, Co²⁺, Cd²⁺, Hg²⁺). Ni²⁺ and Pb²⁺ are omitted for clarity.



Fig. S28: UV–Vis changes of R2 in ACN (10×10^{-5} M) after addition of 40 equiv. of metal nitrates (a) Free receptor R2 (b) R2+ Cu²⁺ and (c) R2+ other metal salts (Mg²⁺, Ca²⁺, Co²⁺, Zn²⁺, Cd²⁺, Hg²⁺). Ni²⁺, Zn²⁺ and Pb²⁺ are omitted for clarity.





Fig. S29: Jobs plot for R2 with F^- ion at 500 nm in dry ACN.



Fig. S30: Jobs plot for R2 with F^- ion at 558 nm in dry DMSO.

Binding constant:

Binding constant was calculated using equation (1).

Where, A_0 , A, A_{max} are the absorption considered in the absence of F⁻, at an intermediate, and at a concentration of saturation. K is binding constant, $[F^-]$ is concentration of F⁻ ion and n is the stoichiometric ratio.