Supplementary Information

Aggregation-induced emission of triphenylamine substituted cyanostyrene derivatives

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Table S1 Electrochemical properties of G1, G1-N and G2.

Figure S1¹H NMR (400 MHz) spectrum of G1.

Figure S2 ¹³C NMR (100 MHz) spectrum of G1.

Figure S3 MALDI/TOF MS spectrum of G1.

Figure S4 ¹H NMR (400 MHz) spectrum of G1-N.

Figure S5 ¹³C NMR (100 MHz) spectrum of G1-N.

Figure S6 MALDI/TOF MS spectrum of G1-N.

Figure S7¹H NMR (400 MHz) spectrum of G2.

Figure S8¹³C NMR (100 MHz) spectrum of G2.

Figure S9 MALDI/TOF MS spectrum of G2.

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Molecule	E HOMO [a][eV]	E ₀₋₀ ^[b] [eV]	$E_{LUMO}^{[c]} [eV]$
G1	-5.52	2.74	-2.78
G1-N	-5.52	2.42	-3.10
G2	-4.91	2.56	-2.35

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^a E_{HOMO} values were measured in CH_2Cl_2 with 0.1M tetrabutylammonium tetrafluoroborate (TBABF₄) as the electrolyte (working electrode: Pt; reference electrode: SCE, calibrated with ferrocene/ ferrocenium (Fc/Fc⁺) as an external reference; counter electrode: Pt wire).

 b E $_{0.0}$ values were estimated from the edge of the absorption spectra in THF.

 $^{\rm c}$ E $_{\rm LUMO}$ values were estimated by subtracting E $_{\rm 0-0}$ from the HOMO.



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Figure S2¹³C NMR (100 MHz) spectrum of G1.



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Figure S3 MALDI/TOF MS spectrum of G1.



Figure S4 ¹H NMR (400 MHz) spectrum of G1-N.



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Figure S5¹³C NMR (100 MHz) spectrum of G1-N.



Figure S6 MALDI/TOF MS spectrum of G1-N.



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Figure S7 ¹H NMR (400 MHz) spectrum of G2.



Figure S8¹³C NMR (100 MHz) spectrum of G2.



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Figure S9 MALDI/TOF MS spectrum of G2.