Supporting Information for "Synthesis of a Novel Hybrid Metal Organic Salt and its Solid-State Transformation."

Javier Martí-Rujas,* Massimo Cametti.

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Fig. S1. ORTEP view of the crystal structure of **1** after manually removing the water molecule in order to show the void space left (i.e., pocket). Using a 1.2 Å probe radius, grid spacing of 0.5 Å and considering the contact surface area the void volume corresponds to 1.9 % of the total unit cell volume.

Table. S1 List of all the symmetry operators observed in 1.

2. Single crystal X-ray structure of 2.

Fig. S2. (a) Crystal structure of salt **2** viewed along the *b*-axis; (b) detailed view of the dication showing the trifurcated weak intramolecular N–H···O hydrogen bonding (only one is showed); (c) view of two $[ZnI_4]^{2-}$ dianions surrounded by K222–H₂²⁺ dications. (d) C–H···I interactions observed in **2**. Color code: Carbon: green; Nitrogen: blue; Oxygen: red; Nitrogen: grey; Hydrogen: light grey. Weak hydrogen bonding shown as dashed lines.

3. Single crystal X-ray structure of 3.

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Fig. S10. Figure showing the μ -O-based and μ -S clusters based on *ab initio* calculations on the adamantanoid dianion of formula $[Zn_4(MeO)_6I_4]^{2-}$ and $[Zn_4(MeS)_6I_4]^{2-}$ at the HF-3-21G level. The volumes and distances between opposite μ -O(S) atoms 382.5 Å3 – 4.32 Å and 446.5 Å3 – 5.44 Å for the μ -O and μ -S clusters, respectively). An additional carbon on the alkoxide chain would probably lead to significant sterical clash with the apical I atoms.

Fig. S11. ¹H-NMR spectrum of a 2.5 mM solution of **1** in DMSO-d6:acetone-d6 2:1 mixture at 25°C.

Fig. S12. Portion of the ¹H-NMR spectrum of a 2.5 mM solution of **1** in DMSO-d6:acetoned6 2:1 mixture at 25°C. Two groups of signals (labeled • and •) can be identified (at ppm δ = 3.60 s, 3.54 t and 2.58 t; δ = 3.79 t, 3.78 t and 3.69 s) and they can be assigned to the cationic K222–H₂²⁺ of **1** in both its associated ion-pair and its solvated forms. **Fig. S13**. ¹H-NMR spectrum of a 2.5 mM solution of **2** in DMSO-d6:acetone-d6 2:1 mixture at 25°C and magnification of the 4.0-3.6 ppm region.

Fig. S14. Spectral comparison between an equimolar solution of 1 (top) and 2 (bottom) in DMSO-d6:acetone-d6 2:1 mixture at 25°C. Shift of the K222–H $_2^{2+}$ signals are highlighted by the black arrows.

1. Experimental

1.2 X-ray Diffraction Data

Single crystal X-ray diffraction data for 1, 2 and 3 were recorded with a Bruker X8 Prospector APEX-II/CCD diffractometer equipped with an Incoatec I μ S_{Cu} microphocus X-ray source (CuK_{α} radiation, $\lambda = 1.54056$ Å) and with a CCD detector. The low temperature experiments were collected using a Bruker Oxford Cryosystem device. SHELXL¹ was used for structure solution and refinement. X-ray powder diffraction experiments were recorded on a Bruker D2 Phaser diffractometer ($\lambda = 1.54056$ Å). The 2θ scan range was $4.7^{\circ} - 40^{\circ}$ with a step size 0.02°. The exposure time was 2s. Simulated PXRD patterns were generated using Mercury Version 3.1 software program. Figures depicting the crystal packing were created using PyMol software.



Fig. S1. ORTEP view of the crystal structure of **1** after manually removing the water molecule in order to show the void space left (i.e., pocket). Using a 1.2 Å probe radius, grid spacing of 0.5 Å and considering the contact surface area the void volume corresponds to 1.9 % of the total unit cell volume.

Table. S1	List of all	the symmetry	operators	observed in 1	1.
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Numb	Symm.			Ord	Тур
er	Op.	Description	Detailed Description	er	e
1	x,y,z	Identity Rotation axis (3-	Identity	1	1
2	z,x,y	fold) Rotation axis (3-	3-fold rotation axis with direction [1, 1, 1] at x, x, x	3	3
3	y,z,x	fold)	3-fold rotation axis with direction [1, 1, 1] at x, x, x 3-fold scraw axis with direction [1, 1, 1] at $x+5/6$, $x+1/3$, x with scraw	3	3
4	z, 1/2+x	fold)	component [1/3, -1/3, 1/3]	3	3
5	1/2+2,1/2- x,-y	fold)	3-fold rotation axis with direction $[1, -1, 1]$ at x+1/2, -x, x	3	3
6	y,1/2+z,1/2 -x	Screw axis (3- fold)	3-fold screw axis with direction $[1, -1, -1]$ at x+1/6, -x+1/6, -x with screw component $[-1/3, 1/3, 1/3]$	3	3
7	1/2-z,- x,1/2+y	Rotation axis (3- fold)	3-fold rotation axis with direction $[1, -1, -1]$ at $x+1/2$, $-x+1/2$, $-x$	3	3
8	z,1/2+x,1/2 -y 1/2+y 1/2-	Rotation axis (3- fold)	3-fold rotation axis with direction $[1, 1, -1]$ at x, x+1/2, -x 3-fold screw axis with direction $[1, 1, -1]$ at x+1/3, x+1/6, -x with screw	3	3
9	Z,-X	fold)	component $[1/3, 1/3, -1/3]$	3	3
10	1/2+x,1/2- y,-z	Screw axis (2- fold)	2-fold screw axis with direction $[1, 0, 0]$ at x, 1/4, 0 with screw component $[1/2, 0, 0]$	2	2
11	x,1/2+y,1/2 -z	Screw axis (2- fold)	2-fold screw axis with direction $[0, 1, 0]$ at 0, y, 1/4 with screw component $[0, 1/2, 0]$	2	2
12	1/2-x,- y,1/2+z	Screw axis (2- fold)	2-rota screw axis with direction $[0, 0, 1]$ at $1/4, 0, z$ with screw component $[0, 0, 1/2]$	2	2

2. Single crystal X-ray structure of 2.

In crystal **2**, the K222 is doubly-protonated and adopts the endo-endo type conformation where the N atoms are directed towards the inner cavity of the macrocycle (Figure S1). The O atoms also point inwards the cavity where weak hydrogen bond interaction with the endo N+–H···O group is observed. In **2**, the nonbonding N···N separation (6.340(1) Å) is shorter than that of the empty cryptand (6.871(4) Å). The $[ZnI_4]^{2-}$ anions lying in the 3-fold crystallographic axis have tetrahedral coordination. Two $[ZnI_4]^{2-}$ anions are positioned in a "head-to-head" fashion with intermolecular I···I distance of 3.679(1) Å (Figure S1a and S1c). There is a disordered water molecule that is located between two symmetry related dications. Out of the mother liquors, the crystals of **2** are stable at room temperature in contact with air for several days.



Fig. S2. (a) Crystal structure of salt **2** viewed along the *b*-axis; (b) detailed view of the dication showing the trifurcated weak intramolecular N–H···O hydrogen bonding (only one is showed); (c) view of two $[ZnI_4]^{2-}$ dianions surrounded by K222–H₂²⁺ dications. (d) C–H···I interactions observed in **2**. Color code: Carbon: green; Nitrogen: blue; Oxygen: red; Nitrogen: grey; Hydrogen: light grey. Weak hydrogen bonding shown as dashed lines.

3. Single crystal X-ray structure of 3.

Using the same protocol, we set up crystallization experiments using $ZnBr_2$ and $ZnCl_2$ instead of ZnI_2 . Colorless plate-like crystals precipitated after 1 hour, however the solution was left over a week to allow crystallization to proceed. Single crystal X-ray diffraction showed that an isostructural $[ZnBr_4]^{2-}$ complex (3) to the reported $ZnCl_2$ version (4),² had been obtained (R-3c)

and therefore, the molecular packing has been maintained in this class of hybrid metal-organic salts.

Crystal	C-H····X ^a	Zn…Zn	<u>N…N (Å)</u>
•	(Å)	(Å)	
1	4.009(9)	-	6.124(9)
2	3.95(1)	8.823(2)	6.34(1)
3	3.705(5)	8.317(1)	6.380(2)
4	3.770(2)	8.0269(8)	6.376(2)

Table. S2Relevant distances in molecular salts 1–4.

^a C····X = I, Br, C	71.
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Fig. S3. Ball and stick model viewed along the *a*-axis of isostructural crystal structures **2**, **3** and **4** (a), (b) and (c) respectively.



Fig. S4. Simulated PXRD pattern of 2 and experimental PXRD pattern of a sample obtained after grinding K222 and ZnI_2 (1:2 molar ratio) in the presence of a drop of methanol.



Fig. S5. Experimental vs simulated from single crystal PXRD patterns of **3** measured at rt. The experimental pattern of the $ZnBr_4^{2-}$ was crystallized from methanol. Clearly, both diffraction patterns show the same crystalline structure.



Fig. S6. Experimental PXRD of obtained after crystallization of K222 and ZnI_2 (1:2) in ethanol and simulated of 2 crystallized from methanol. Clearly, both diffraction patterns show the same crystalline structure.



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Fig. S8. Experimental PXRD corresponding to the solid obtained when $ZnBr_2$ was added into a solution of K222 (MeOH) cooled at 0°C using ice bath. The diffraction pattern corresponds to that of **3**.



Fig. S9. Experimental PXRD corresponding to the solid obtained when $ZnCl_2$ was added into a solution of K222 (MeOH) cooled at 0°C using ice bath. The diffraction pattern corresponds to that of 4.



Fig. S10. Figure showing the μ -O-based and μ -S clusters based on *ab initio* calculations on the adamantanoid dianion of formula $[Zn_4(MeO)_6I_4]^{2-}$ and $[Zn_4(MeS)_6I_4]^{2-}$ at the HF-3-21G level. The volumes and distances between opposite μ -O(S) atoms 382.5 Å3 – 4.32 Å and 446.5 Å3 – 5.44 Å for the μ -O and μ -S clusters, respectively). An additional carbon on the alkoxide chain would probably lead to significant sterical clash with the apical I atoms.



Fig. S11. ¹H-NMR spectrum of a 2.5 mM solution of **1** in DMSO-d6:acetone-d6 2:1 mixture at 25°C.



Fig. S12. Portion of the ¹H-NMR spectrum of a 2.5 mM solution of **1** in DMSO-d6:acetone-d6 2:1 mixture at 25°C. Two groups of signals (labeled • and •) can be identified (at ppm δ = 3.60 s, 3.54 t and 2.58 t; δ = 3.79 t, 3.78 t and 3.69 s) and they can be assigned to the cationic K222–H₂²⁺ of **1** in both its associated ion-pair and its solvated forms.



Fig. S13. ¹H-NMR spectrum of a 2.5 mM solution of **2** in DMSO-d6:acetone-d6 2:1 mixture at 25°C and magnification of the 4.0-3.6 ppm region.



Fig. S14. Spectral comparison between an equimolar solution of **1** (top) and **2** (bottom) in DMSO-d6:acetone-d6 2:1 mixture at 25°C. Shift of the K222– H_2^{2+} signals are highlighted by the black arrows.

checkCIF/PLATON (standard)

You have not supplied any structure factors. As a result the full set of tests cannot be run.

No syntax errors found. Please wait while processing <u>CIF dictionary</u> <u>Interpreting this report</u>

Datablock: 1

Bond precision:		C-C =	0.0127 A	Γ	Navelength=1.54178
Cell:	a=16.3	228(18)	b=16.3228(18)	c=16.32	28(18)
	alpha=	90	beta=90	gamma=9	0
Temperature:	100 K				
		Calculat	ed		Reported
Volume		4349.0(1	.4)		4348.9(8)
Space group		P 21 3			P213
Hall group		P 2ac 2a	ıb 3		?
Moiety formu	ıla	C18 H38 Zn4, 0.3	N2 06, C6 H18 I 8(0)	4 06	?
Sum formula		C24 H56	I4 N2 012.30 Zm	14	C24 H58 I4 N2 O12 Zn4
Mr		1338.67			1335.80
Dx,g cm-3		2.045			2.040
Ζ		4			4
Mu (mm-1)		25.223			25.215
F000		2577.6			2576.0
F000'		2554.58			
h,k,lmax		19,19,19)		14,18,18
Nref		2502[13	93]		2127
Tmin,Tmax		0.180,0.	220		0.141,0.299

Tmin' 0.025 Correction method= MULTI-SCAN Data completeness= 1.53/0.85 Theta(max) = 65.460 R(reflections) = 0.0316(1972) wR2(reflections) = 0.0718(2127) S = 1.072 Npar= 148

The following ALERTS were generated. Each ALERT has the format **test-name_ALERT_alert-type_alert-level**. Click on the hyperlinks for more details of the test.

Alert level B

<u>PLAT222 ALERT 3 B</u> Large Non-Solvent H Uiso(max)/Uiso(min) .. 8.0 Ratio

Alert level C

ABSTY02_ALERT_1_C An _exptl_absorpt_correction_type has been given without a literature citation. This should be contained in the _exptl_absorpt_process_details field. Absorption correction given as Multi-scan PLAT029 ALERT 3 C _____diffrn_measured_fraction_theta_full Low 0.969 PLAT041 ALERT 1 C Calc. and Reported SumFormula Strings Differ Please Check PLAT068 ALERT 1 C Reported F000 Differs from Calcd (or Missing)... Please Check PLAT090 ALERT 3 C Poor Data / Parameter Ratio (Zmax > 18) 9.41 PLAT230_ALERT_2_C Hirshfeld Test Diff for C3 -- C4 5.3 ... SU PLAT245 ALERT 2 C U(iso) H101 Smaller than U(eq) N1 by ... 0.023 AngSq

PLAT342 ALERT 3 C Low Bond Precision on C-C Bonds 0.0127 Ang.

Alert level G

 $\underline{\mbox{FORMU01}}$ ALERT 2 $\underline{\mbox{G}}$ There is a discrepancy between the atom counts in the

_chemical_formula_sum and the formula from the _atom_site* data.

Atom count from _chemical_formula_sum:C24 H58 I4 N2 O12 Zn4

Atom count from the _atom_site data: C24 H56 I4 N2 O12.3 Zn4 <u>CELLZ01 ALERT 1 G</u> Difference between formula and atom_site contents detected.

CELLZ01 ALERT 1 G ALERT: Large difference may be due to a

symmetry error - see SYMMG tests From the CIF: _cell_formula_units_Z 4

From the CIF: __cen_ionida_units_2 4

TEST: Compare cell contents of formula and atom_site data

 atom
 Z*formula
 cif
 sites
 diff

 C
 96.00
 96.00
 0.00
 0.00

 H
 232.00
 224.00
 8.00

 I
 16.00
 16.00
 0.00

 N
 8.00
 8.00
 0.00

49.20 -1.20 0 48.00 Zn 16.00 16.00 0.00 PLAT002_ALERT_2_G Number of Distance or Angle Restraints on AtSite 4 Note PLAT005_ALERT_5_G No _iucr_refine_instructions_details in the CIF Please Do ! PLAT152 ALERT 1 G The Supplied and Calc. Volume s.u. Differ by ... 6 Units PLAT302 ALERT 4 G Anion/Solvent Disorder Percentage = 100 Note PLAT311 ALERT 2 G Isolated Disordered Oxygen Atom (No H's ?) 01S Check PLAT860 ALERT 3 G Number of Least-Squares Restraints 2 Note PLAT950 ALERT 5 G Reported and Calculated Hmax Values Differ by ... 5 0 ALERT level A = Most likely a serious problem - resolve or explain 1 ALERT level B = A potentially serious problem, consider carefully 8 ALERT level C = Check. Ensure it is not caused by an omission or oversight 10 ALERT level G = General information/check it is not something unexpected

 $\,$ 6 ALERT type 1 CIF construction/syntax error, inconsistent or missing data

5 ALERT type 2 Indicator that the structure model may be wrong or deficient

5 ALERT type 3 Indicator that the structure quality may be low

1 ALERT type 4 Improvement, methodology, query or suggestion

2 ALERT type 5 Informative message, check

Validation response form

Please find below a validation response form (VRF) that can be filled in and pasted into your CIF.

start Validation Reply Form
_vrf_PLAT222_test
;
PROBLEM: Large Non-Solvent H Uiso(max)/Uiso(min) .. 8.0 Ratio
RESPONSE: ...
;

end Validation Reply Form

It is advisable to attempt to resolve as many as possible of the alerts in all categories. Often the minor alerts point to easily fixed oversights, errors and omissions in your CIF or refinement strategy, so attention to these fine details can be worthwhile. In order to resolve some of the more serious problems it may be necessary to carry out additional measurements or structure refinements. However, the purpose of your study may justify the reported deviations and the more serious of these should normally be commented upon in the discussion or experimental section of a paper or in the "special_details" fields of the CIF. checkCIF was carefully designed to identify outliers and unusual parameters, but every test has its limitations and alerts that are not important in a particular case may appear. Conversely, the absence of alerts does not guarantee there are no aspects of the results needing attention. It is up to the individual to critically assess their own results and, if necessary, seek expert advice.

Publication of your CIF in IUCr journals

A basic structural check has been run on your CIF. These basic checks will be run on all CIFs submitted for publication in IUCr journals (*Acta Crystallographica, Journal of Applied*

Crystallography, *Journal of Synchrotron Radiation*); however, if you intend to submit to *Acta Crystallographica Section C* or *E*, you should make sure that <u>full publication checks</u> are run on the final version of your CIF prior to submission.

Publication of your CIF in other journals

Please refer to the *Notes for Authors* of the relevant journal for any special instructions relating to CIF submission.

PLATON version of 18/09/2013; check.def file version of 12/09/2013

Datablock 1 - ellipsoid plot



checkCIF/PLATON (standard)

You have not supplied any structure factors. As a result the full set of tests cannot be run.

No syntax errors found. Please wait while processing <u>CIF dictionary</u> <u>Interpreting this report</u>

Datablock: 2

Bond precision:		C-C = 0.0157 A			Wavelength=1.54178	
Cell:	a=11.4	410(7)	b=11.4410(7)	c=81.31	3(6)	
	alpha=	=90	beta=90	gamma=1	20	
Temperature:	296 K					
		Calculat	ed		Reported	
Volume		9217.6(1	.5)		9217.6(11)	
Space group		R -3 c			R-3c	
Hall group		-R 3 2"c	2		?	
Moiety formu	la	C18 H38	N2 06, I4 Zn, ().15(0)	?	
Sum formula		C18 H38	I4 N2 O6.15 Zn		C18 H40 I4 N2 O6.17 Zn	
Mr		953.89			956.16	
Dx,g cm-3		2.062			2.067	
Ζ		12			12	
Mu (mm-1)		32.930			32.931	
F000		5414.4			5440.0	
F000'		5398.73				
h,k,lmax		13,13,96)		13,13,95	
Nref		1804			1797	
Tmin,Tmax		0.035,0.	518		0.100,0.518	

Tmin'	0.006				
Correction method=	MULTI-SCAN				
Data completeness=	0.996	Theta(max) =	65.980		
R(reflections) = 0.0	0617(1620)	wR2(refle	ections)=	0.1760(1797)
S = 1.127	Npar= 100				

The following ALERTS were generated. Each ALERT has the format **test-name_ALERT_alert-type_alert-level**. Click on the hyperlinks for more details of the test.

Alert level C

ABSTY02_ALERT_1_C_An_exptl_absorpt_correction_type has been given without a literature citation. This should be contained in the _exptl_absorpt_process_details field. Absorption correction given as Multi-scan PLAT041_ALERT_1_C_Calc. and Reported SumFormula Strings Differ Please Check PLAT068_ALERT_1_C_Reported F000 Differs from Calcd (or Missing)... Please Check PLAT094_ALERT_2_C_Ratio of Maximum / Minimum Residual Density 2.40 PLAT222_ALERT_3_C_Large Non-Solvent H_Uiso(max)/Uiso(min) ... 6.5 Ratio PLAT342_ALERT_3_C_Low Bond Precision on C-C Bonds 0.0157 Ang.

Alert level G

 $\underline{\mbox{FORMU01}}$ ALERT 2 $\underline{\mbox{G}}$ There is a discrepancy between the atom counts in the

_chemical_formula_sum and the formula from the _atom_site* data.

Atom count from _chemical_formula_sum:C18 H40 I4 N2 O6.17 Zn1

Atom count from the _atom_site data: C18 H38 I4 N2 O6.15 Zn1 <u>CELLZ01 ALERT 1 G</u> Difference between formula and atom_site contents detected.

 $\label{eq:cell201_ALERT 1_G} \mbox{WARNING: H atoms missing from atom site list. Is this intentional?}$

From the CIF: _cell_formula_units_Z 12 From the CIF: _chemical_formula_sum C18 H40 I4 N2 O6.17 Zn TEST: Compare cell contents of formula and atom_site data

atom	Z*form	ula cif si	tes diff
С	216.00	216.00	0.00
Н	480.00	456.00	24.00
I	48.00	48.00	0.00
N	24.00	24.00	0.00
0	74.04	73.80	0.24
Zn	12.00	12.00	0.00

PLAT002 ALERT 2 G Number of Distance or Angle Restraints on AtSite 4 Note

<u>PLAT005 ALERT 5 G</u> No _iucr_refine_instructions_details in the CIF Please Do !

<u>PLAT072 ALERT 2 G</u> SHELXL First Parameter in WGHT Unusually Large. 0.10

PLAT083_ALERT_2_G SHELXL Second Parameter in WGHT Unusually 170.68 Large. PLAT152_ALERT_1_G The Supplied and Calc. Volume s.u. Differ by ... 4 Units PLAT302_ALERT_4_G Anion/Solvent Disorder Percentage = 6 Note PLAT311 ALERT 2 G Isolated Disordered Oxygen Atom (No H's ?) 07 Check PLAT764 ALERT 4 G Overcomplete CIF Bond List Detected (Rep/Expd) . 1.13 Ratio PLAT794 ALERT 5 G Tentative Bond Valency for Zn1 (II) 2.02 Note PLAT860 ALERT 3 G Number of Least-Squares Restraints 2 Note 0 ALERT level A = Most likely a serious problem - resolve or explain 0 ALERT level B = A potentially serious problem, consider carefully 6 ALERT level C = Check. Ensure it is not caused by an omission or oversight 13 ALERT level G = General information/check it is not something unexpected 6 ALERT type 1 CIF construction/syntax error, inconsistent or missing data 6 ALERT type 2 Indicator that the structure model may be wrong or deficient 3 ALERT type 3 Indicator that the structure quality may be low

2 ALERT type 4 Improvement, methodology, query or suggestion

2 ALERT type 5 Informative message, check

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A basic structural check has been run on your CIF. These basic checks will be run on all CIFs submitted for publication in IUCr journals (*Acta Crystallographica, Journal of Applied Crystallography, Journal of Synchrotron Radiation*); however, if you intend to submit to *Acta Crystallographica Section C* or *E*, you should make sure that <u>full publication checks</u> are run on the final version of your CIF prior to submission.

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Datablock 2 - ellipsoid plot



Download CIF editor (publCIF) from the IUCr Download CIF editor (enCIFer) from the CCDC Test a new CIF entry

checkCIF/PLATON (standard)

You have not supplied any structure factors. As a result the full set of tests cannot be run.

No syntax errors found. Please wait while processing <u>CIF dictionary</u> <u>Interpreting this report</u>



 Bond precision:
 C-C = 0.0067 A
 Wavelength=1.54178

 Cell:
 a=11.0600(6)
 b=11.0600(6)
 c=78.221(5)

 alpha=90
 beta=90
 gamma=120

Temperature: 100 K

	Calculated	Reported		
Volume	8286.4(12)	8286.4(8)		
Space group	R -3 c	R-3c		
Hall group	-R 3 2"c	?		
Moiety formula	2(C18 H38 N2 O6), 2(Br4 Zn), O	?		
Sum formula	C36 H76 Br8 N4 O13 Zn2	C18 H40 Br4 N2 O7 Zn		
Mr	1542.99	781.53		
Dx,g cm-3	1.855	1.879		
Z	6	12		
Mu (mm-1)	8.320	8.343		
F000	4584.0	4656.0		
F000'	4541.18			
h,k,lmax	13,13,92	13,13,92		
Nref	1632	1622		
Tmin, Tmax	0.576,0.716	0.388,0.731		
Tmin'	0.296			
Correction method=	MULTI-SCAN			
Data completeness= 0.994 Theta(max)= 66.390				
R(reflections) = 0.0	0344(1439) wR2(reflections)= 0.1019(1622)		
S = 1.093	Npar= 100			

The following ALERTS were generated. Each ALERT has the format test-name_ALERT_alert-type_alert-level Click on the hyperlinks for more details of the test.

Alert level B

PLAT306 ALERT 2 B Isolated Oxygen Atom (H-atoms Missing ?) 01S Check

Alert level C

<u>ABSTY02_ALERT_1_C</u> An _exptl_absorpt_correction_type has been given without a literature citation. This should be contained in the

_exptl_absorpt_process_details field.

Absorption correction given as Multi-scan <u>CHEMW03 ALERT 2 C</u> The ratio of given/expected molecular weight as

calculated from the _atom_site* data lies outside the range 0.99 <> 1.01 From the CIF: _cell_formula_units_Z 12 From the CIF: _chemical_formula_weight 781.53 TEST: Calculate formula weight from _atom_site_* atom mass num sum 12.01 18.00 216.20 C н 1.01 38.00 38.30 Ν 14.01 2.00 28.01 0 16.00 6.50 103.99 Zn 65.39 1.00 65.39 4.00 319.62 Br 79.90 Calculated formula weight 771.52 PLAT041 ALERT 1 C Calc. and Reported SumFormula Strings Differ

Please Check <u>PLAT043 ALERT 1 C</u> Calculated and Reported Mol. Weight Differ by .. 20.07 Check

<u>PLAT068_ALERT_1_C</u> Reported F000 Differs from Calcd (or Missing)... Please Check PLAT094_ALERT_2_C Ratio of Maximum / Minimum Residual Density

2.08 PLAT341 ALERT 3 C Low Bond Precision on C-C Bonds

0.0067 Ang.

Alert level G

 $\underline{\mbox{FORMU01}\ \mbox{ALERT 2}\ \mbox{G}\ \mbox{There}}$ There is a discrepancy between the atom counts in the

_chemical_formula_sum and the formula from the _atom_site* data.

Atom count from _chemical_formula_sum:C18 H40 Br4 N2 O7 Zn1

Atom count from the _atom_site data: C18 H38 Br4 N2 O6.5 Zn1 $\underline{CELLZ01\ ALERT\ 1\ G}$ Difference between formula and atom_site contents detected.

CELLZ01 ALERT 1 G ALERT: Large difference may be due to a

symmetry error - see SYMMG tests From the CIF: _cell_formula_units_Z 12 From the CIF: _chemical_formula_sum C18 H40 Br4 N2 O7 Zn TEST: Compare cell contents of formula and atom_site data

atom	Z*form	ula cif sit	es diff
С	216.00	216.00	0.00
Н	480.00	456.00	24.00
Br	48.00	48.00	0.00
Ν	24.00	24.00	0.00
0	84.00	78.00	6.00
_			

Zn 12.00 12.00 0.00 <u>PLAT002 ALERT 2 G</u> Number of Distance or Angle Restraints on AtSite 4 Note

<u>PLAT005 ALERT 5 G No _iucr_refine_instructions_details</u> in the CIF Please Do !

<u>PLAT045 ALERT 1 G</u> Calculated and Reported Z Differ by 0.50 Ratio

PLAT083 ALERT 2 G SHELXL Second Parameter in WGHT Unusually Large. 35.49

<u>PLAT152 ALERT 1 G</u> The Supplied and Calc. Volume s.u. Differ by ... 4 Units

<u>PLAT434 ALERT 2 G</u> Short Inter HL..HL Contact Br1 .. Br1 . 3.55 Ang.

PLAT764 ALERT 4 G Overcomplete CIF Bond List Detected (Rep/Expd) .

1.13 Ratio

<u>PLAT794 ALERT 5 G</u> Tentative Bond Valency for Zn3 (II) 1.94 Note

PLAT860 ALERT 3 G Number of Least-Squares Restraints 2 Note 0 ALERT level A = Most likely a serious problem - resolve or explain 1 ALERT level B = A potentially serious problem, consider carefully 7 ALERT level C = Check. Ensure it is not caused by an omission or oversight 12 ALERT level G = General information/check it is not something unexpected 8 ALERT type 1 CIF construction/syntax error, inconsistent or missing

data 7 ALERT type 2 Indicator that the structure model may be wrong or deficient

2 ALERT type 3 Indicator that the structure quality may be low

1 ALERT type 4 Improvement, methodology, query or suggestion

2 ALERT type 5 Informative message, check

Validation response form

Please find below a validation response form (VRF) that can be filled in and pasted into your CIF.

start Validation Reply Form
_vrf_PLAT306_test_trigonal
;
PROBLEM: Isolated Oxygen Atom (H-atoms Missing ?) O1S
Check
RESPONSE: ...
;
end Validation Reply Form

It is advisable to attempt to resolve as many as possible of the alerts in all categories. Often the minor alerts point to easily fixed oversights, errors and omissions in your CIF or refinement strategy, so attention to these fine details can be worthwhile. In order to resolve some of the more serious problems it may be necessary to carry out additional measurements or structure refinements. However, the purpose of your study may justify the reported deviations and the more serious of these should normally be commented upon in the discussion or experimental section of a paper or in the "special_details" fields of the CIF. checkCIF was carefully designed to identify outliers and unusual parameters, but every test has its limitations and alerts that are not important in a particular case may appear. Conversely, the absence of alerts does not guarantee there are no aspects of the results needing attention. It is up to the individual to critically assess their own results and, if necessary, seek expert advice.

Publication of your CIF in IUCr journals

A basic structural check has been run on your CIF. These basic checks will be run on all CIFs submitted for publication in IUCr journals (*Acta Crystallographica, Journal of Applied Crystallography, Journal of Synchrotron Radiation*); however, if you intend to submit to *Acta Crystallographica Section C* or *E*, you should make sure that <u>full publication checks</u> are run on the final version of your CIF prior to submission.

Publication of your CIF in other journals

Please refer to the *Notes for Authors* of the relevant journal for any special instructions relating to CIF submission.

Datablock 3 - ellipsoid plot



- G. M. Sheldrick, Acta Crystallogr., Sect. A: Fundam. Crystallogr., 2008, 64, 112.
 A. N. Chekhlov, Russ. J. Coord. Chem. 2007, 33, 932.