

Electronic Supplementary Information

Highly photoactive SnO₂ nanostructures engineered by electrochemically active biofilm

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Schematic representation of modification process of pure SnO₂ (*p*-SnO₂) nanostructures by EAB

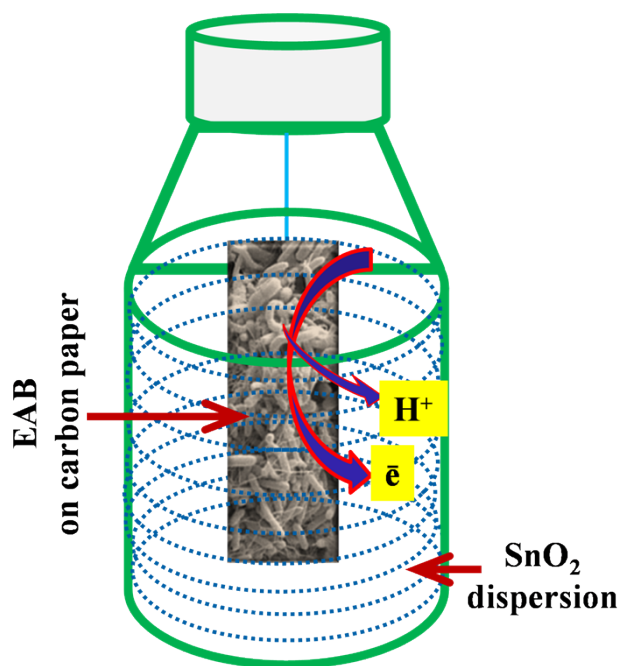


Fig. S1. Schematic representation of the SnO₂ nanostructures modification by electrochemically active biofilm.

Adsorption-desorption equilibrium spectra of 4-NP only, 4-NP with p -SnO₂ and 4-NP with m -SnO₂ nanostructures

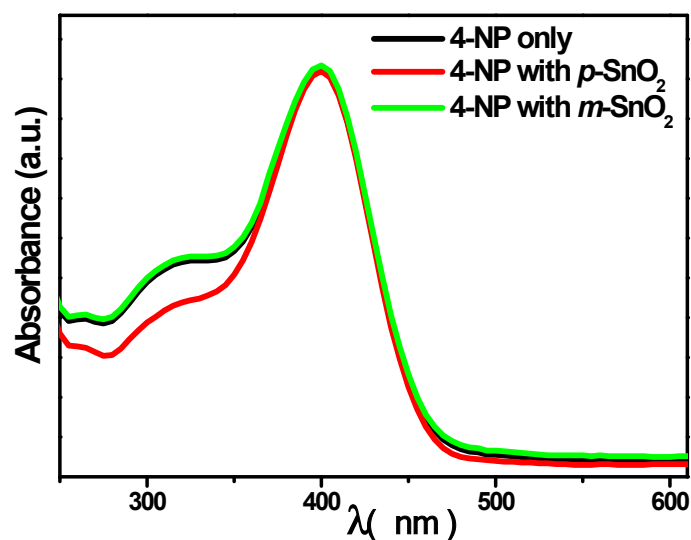


Fig. S2. Adsorption-desorption equilibrium spectra of 4-NP only, 4-NP with p -SnO₂ and 4-NP with m -SnO₂ nanostructures.

Adsorption-desorption equilibrium spectra of MB only, MB with p -SnO₂ and MB with m -SnO₂ nanostructures

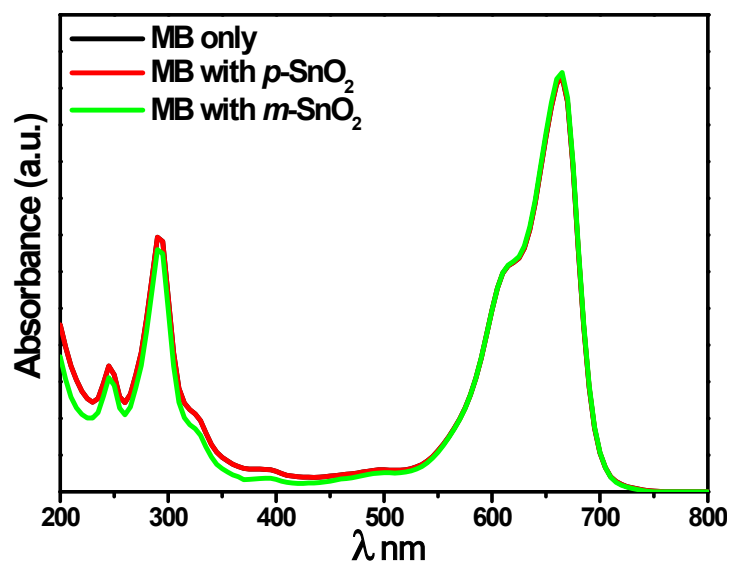


Fig. S3. Adsorption-desorption equilibrium spectra of MB only, MB with p -SnO₂ and MB with m -SnO₂ nanostructures.

Dark reaction (with catalyst, 3 h) spectra of 4-NP only, 4-NP with p -SnO₂ and 4-NP with m -SnO₂ nanostructures

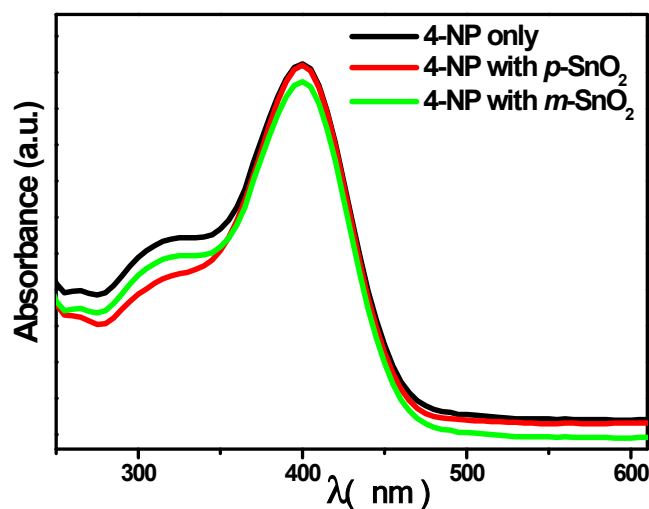


Fig. S4. Dark reaction (with catalyst, 3 h) spectra of 4-NP only, 4-NP with p -SnO₂ and 4-NP with m -SnO₂ nanostructures.

Dark reaction (with catalyst, 3 h) spectra of MB only, MB with p -SnO₂ and MB with m -SnO₂ nanostructures

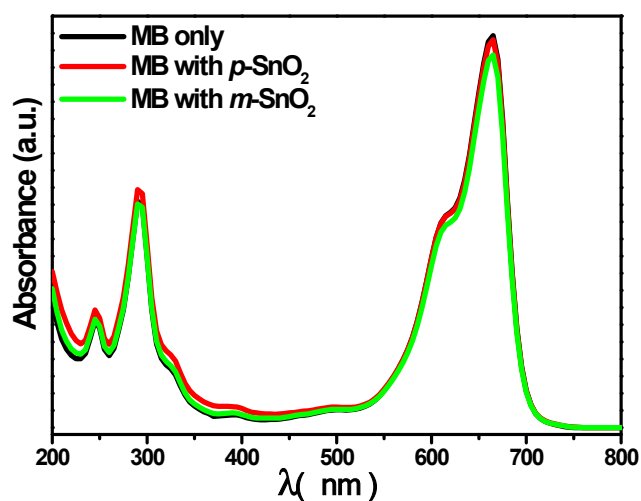


Fig. S5. Dark reaction (with catalyst, 3 h) spectra of MB only, MB with p -SnO₂ and MB with m -SnO₂ nanostructures.

Light reaction (without catalyst, 3 h) spectra of 4-NP only, 4-NP with p -SnO₂ and 4-NP with m -SnO₂ nanostructures

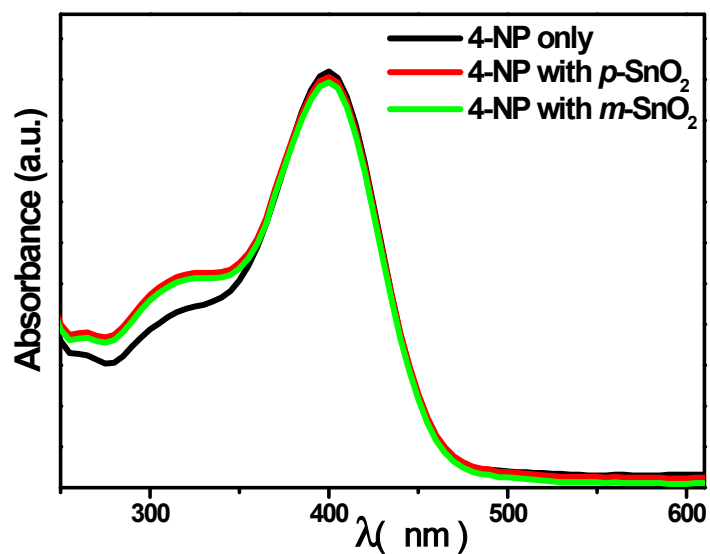


Fig. S6. Light reaction (without catalyst, 3 h) spectra of 4-NP only, 4-NP with p -SnO₂ and 4-NP with m -SnO₂ nanostructures.

Light reaction (without catalyst, 3 h) spectra of MB only, MB with p -SnO₂ and MB with m -SnO₂ nanostructures

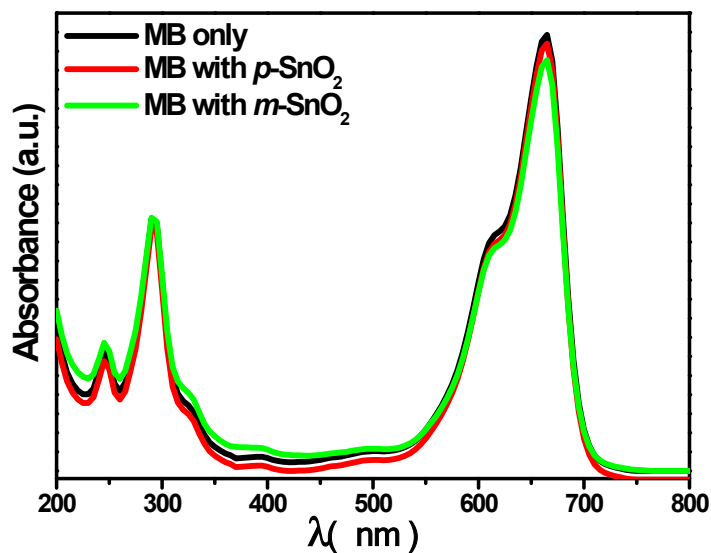


Fig. S7. Light reaction (without catalyst, 3 h) spectra of MB only, MB with p -SnO₂ and MB with m -SnO₂ nanostructures.

UV-vis diffuse reflectance spectra of the p -SnO₂ and m -SnO₂ nanostructures

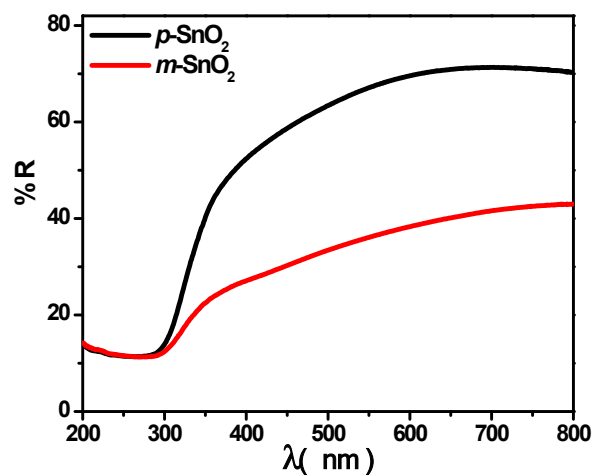


Fig. S8. UV-vis diffuse reflectance spectra of the p -SnO₂ and m -SnO₂ nanostructures.

XPS survey spectra of the p -SnO₂ and m -SnO₂ nanostructures

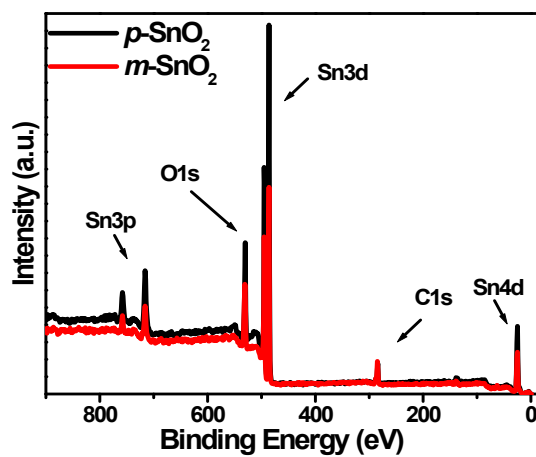


Fig. S9. XPS survey spectra of the p -SnO₂ and m -SnO₂ nanostructures.

SAED patterns of the $p\text{-SnO}_2$ nanostructures

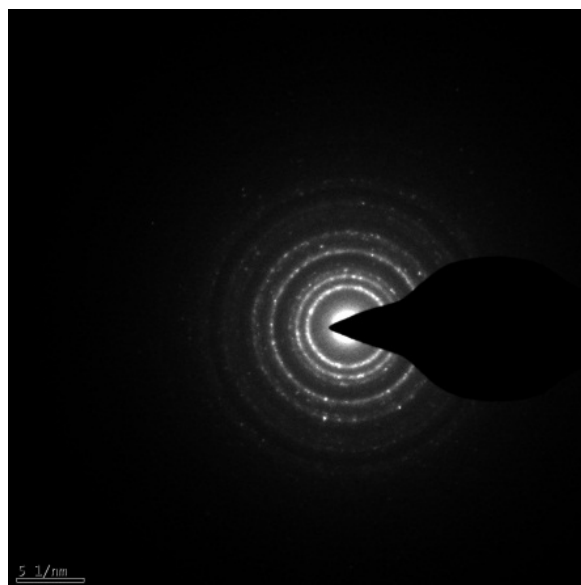


Fig. S10. SAED patterns of the $p\text{-SnO}_2$ nanostructures.

SAED patterns of the $m\text{-SnO}_2$ nanostructures

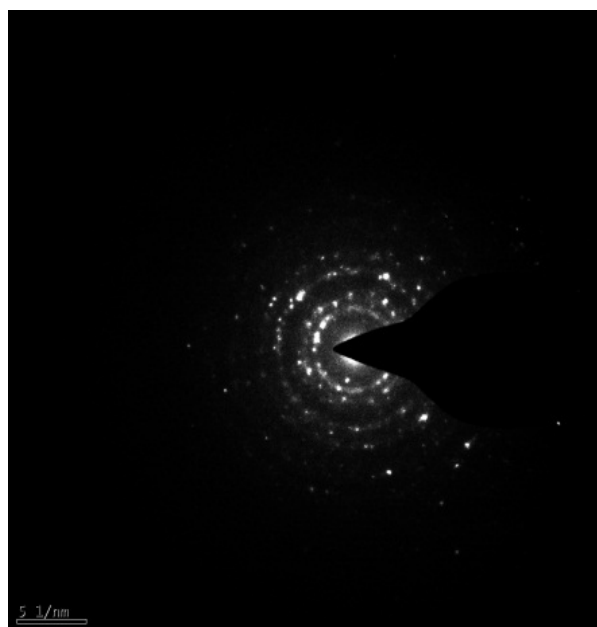


Fig. S11. SAED patterns of the $m\text{-SnO}_2$ nanostructures.

Acquire HAADF of the p -SnO₂ nanostructures

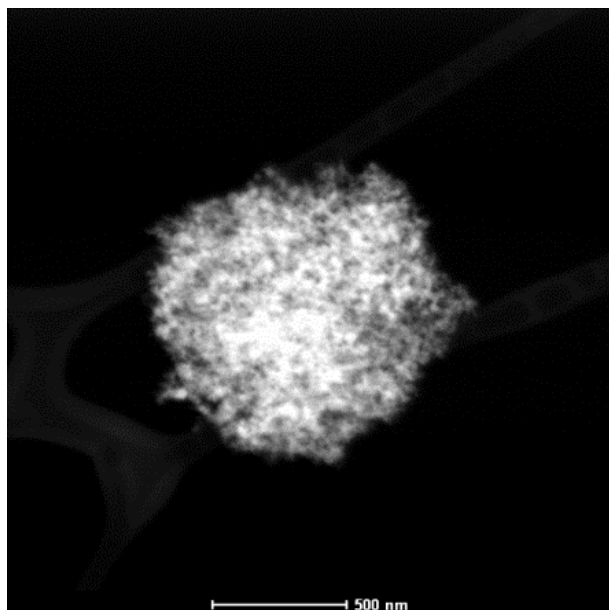


Fig. S12. Representative HAADF FE-TEM image of the p -SnO₂ nanostructures.

Acquire HAADF of the m -SnO₂ nanostructures

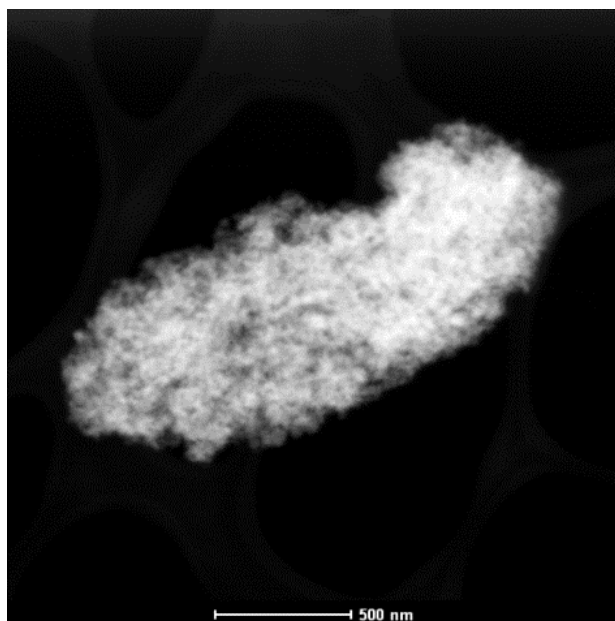


Fig. S13. Representative HAADF FE-TEM image of the m -SnO₂ nanostructures.

EDX of the $p\text{-SnO}_2$ nanostructures

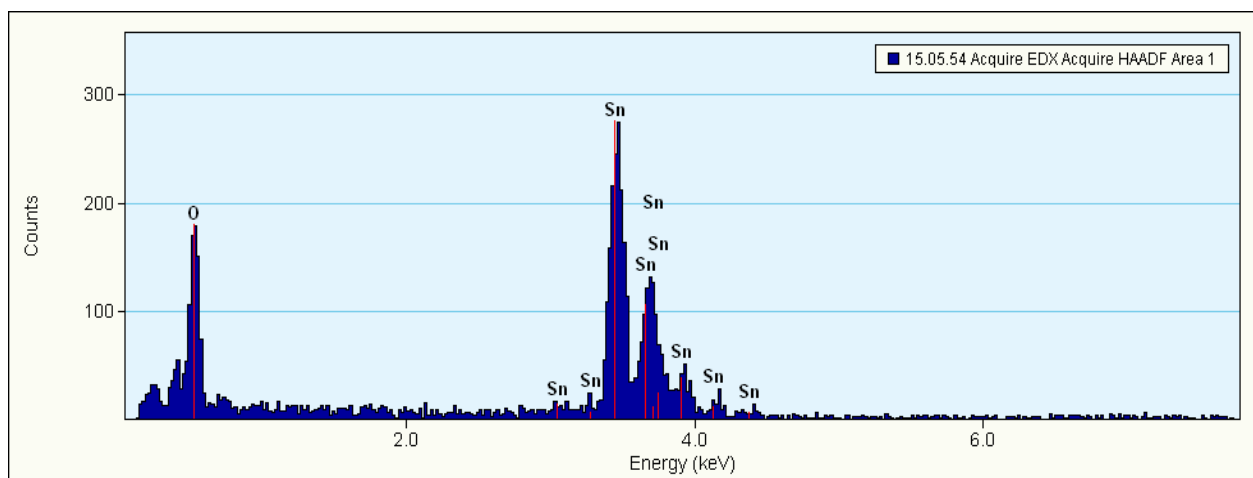


Fig. S14. EDX of the $p\text{-SnO}_2$ nanostructures.

EDX of the $m\text{-SnO}_2$ nanostructures

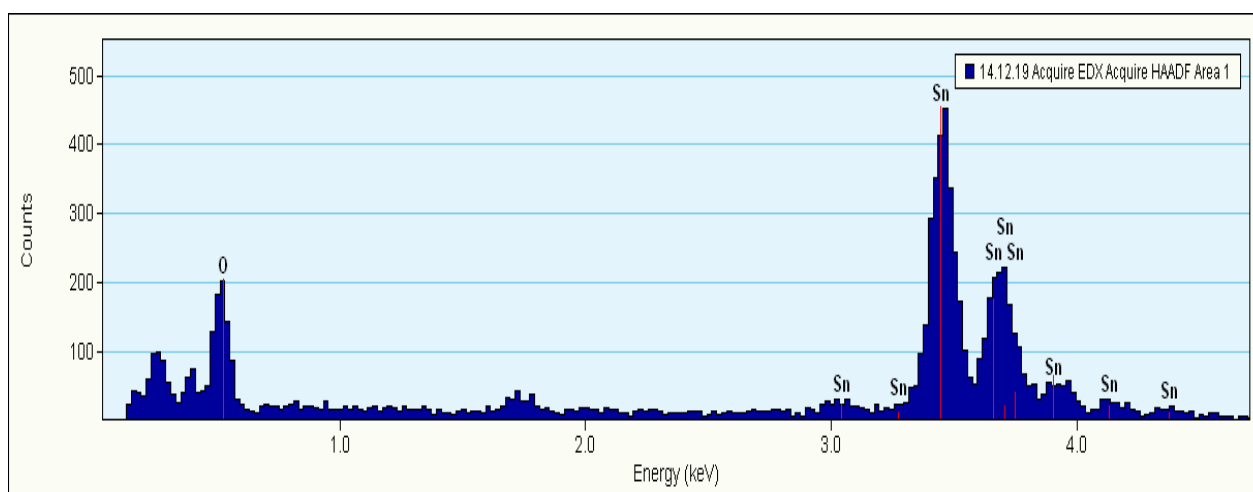


Fig. S15. EDX of the $m\text{-SnO}_2$ nanostructures.

Reaction kinetics of the degradation of 4-NP and MB

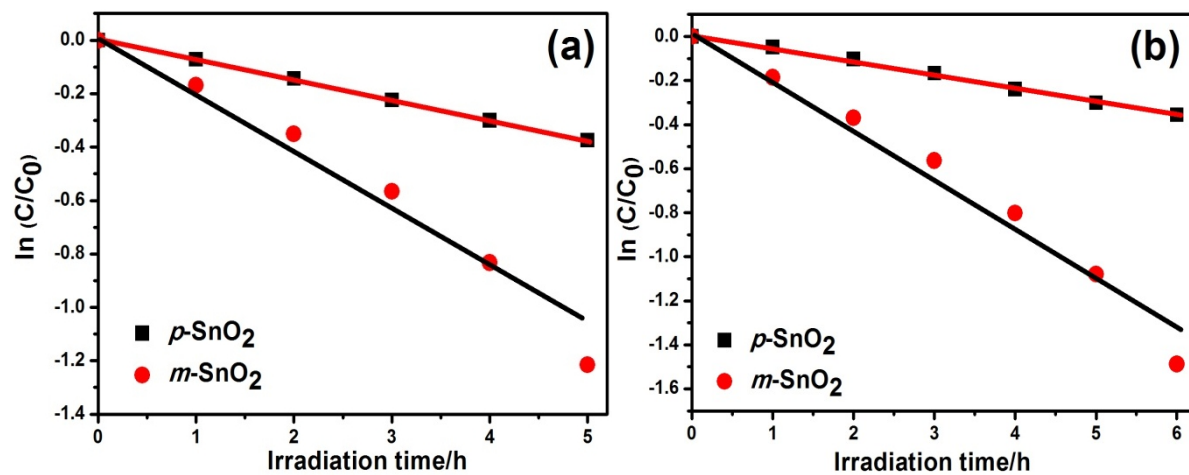


Fig. S16. Kinetics and linear fit spectra for the degradation of (a) 4-NP and, (b) MB with *p*-SnO₂ and *m*-SnO₂ nanostructures.

Table S1. Pseudo-first order rate constants

Photocatalyst	Dye	<i>k</i> (h⁻¹)	R²
<i>p</i> -SnO ₂	4-NP	0.0714	0.9995
<i>m</i> -SnO ₂	4-NP	0.2081	0.9710
<i>p</i> -SnO ₂	MB	0.0481	0.9966
<i>m</i> -SnO ₂	MB	0.2157	0.9722