

Novel organic–inorganic hybrid soft xerogels with lanthanide complexes through an ionic liquid linkage

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Supporting Information

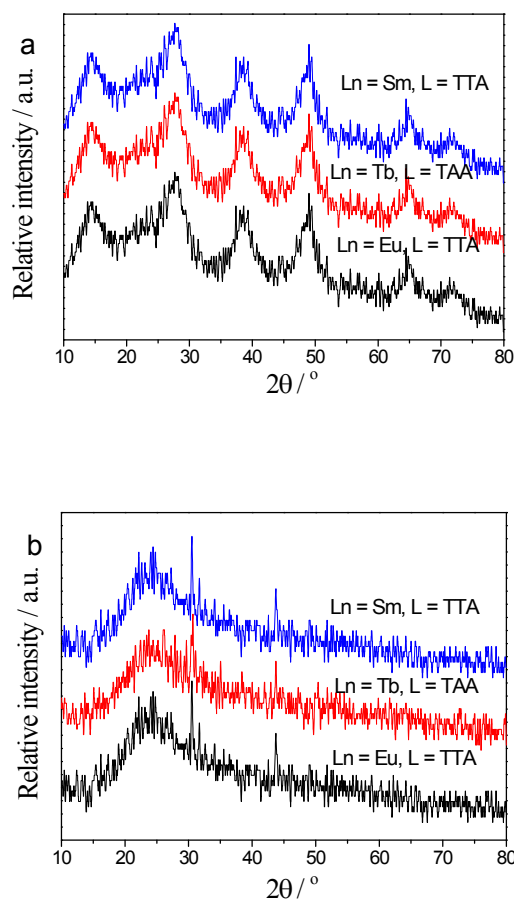


Figure S1 XRD patterns of lanthanide soft hybrid xerogel materials: (a) $[\text{Ln}(\text{L})_4]^-$ $\text{IM}^+\text{-Al}$, (b) $[\text{Ln}(\text{L})_4]^-$ $\text{IM}^+\text{-Ti1}$ ($\text{Ln} = \text{Eu}, \text{Tb}, \text{Sm}$, $\text{L} = \text{TTA}, \text{TAA}$)

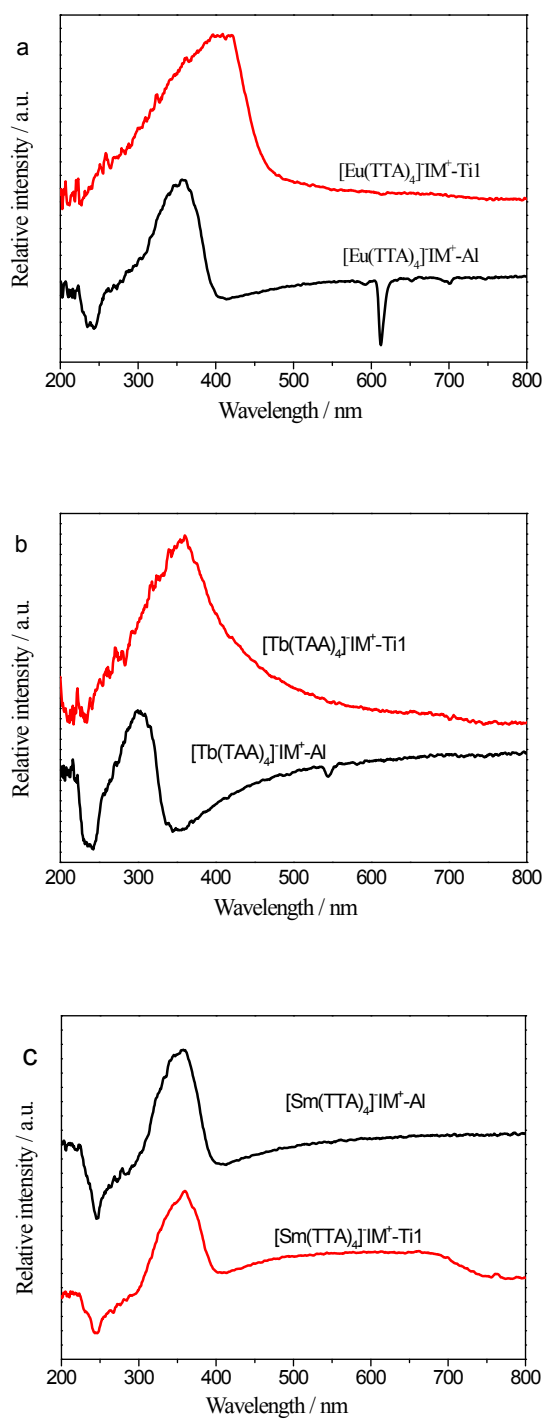


Figure S2 Ultraviolet-visible diffuse reflection absorption spectra of lanthanide soft hybrid xerogel materials: (a) containing Eu^{3+} , (b) containing Tb^{3+} , (c) containing Sm^{3+}

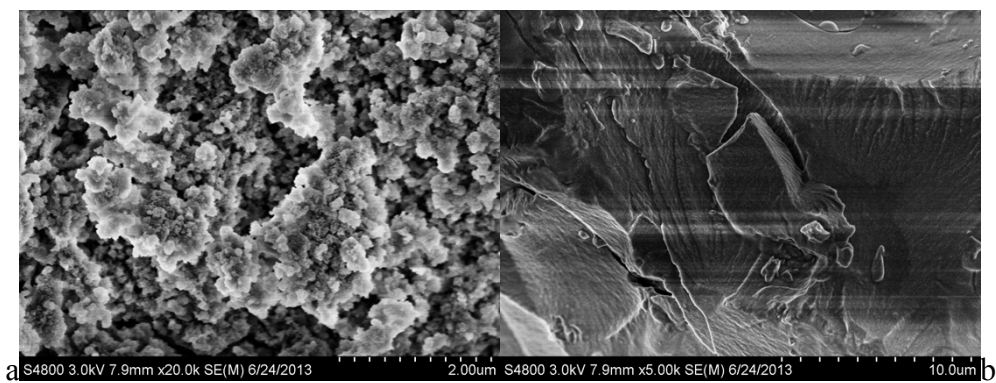


Figure S3 SEM images of lanthanide soft hybrid xerogel materials: (a) $[\text{Eu}(\text{TTA})_4]^-$ IM^+-Al and (b) $[\text{Eu}(\text{TTA})_4]^-$ $\text{IM}^+-\text{Ti1}$

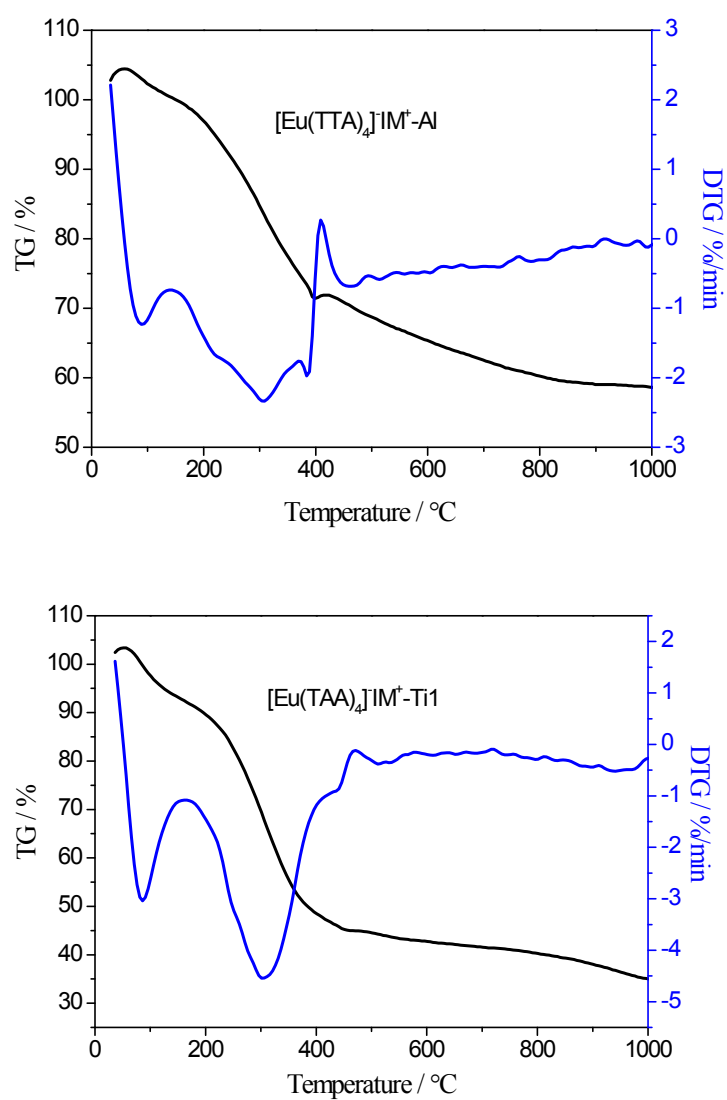


Figure S4 Differential scanning calorimetry (DTG) and thermogravimetric (TG) curves of lanthanide soft hybrid xerogel materials: $[\text{Eu}(\text{TTA})_4] \cdot \text{IM}^+ \cdot \text{Al}$ (a) and $[\text{Eu}(\text{TTA})_4] \cdot \text{IM}^+ \cdot \text{Ti1}$ (b)