

Electronic Supplementary Material (ESI) for New Journal of Chemistry
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[†Electronic supplementary information \(ESI\)](#)

**Up/down conversion luminescence properties of
 $\text{Na}_{0.5}\text{Gd}_{0.5}\text{MoO}_4:\text{Ln}^{3+}$ (Ln= Eu, Tb, Dy, Yb/Er, Yb/Tm, Yb/Ho)
microstructures: Synthesis, morphology and structural investigation**

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Note added after first publication: This Supplementary Information file replaces the one originally published on 20th May 2014 due to an error in Fig. S5(c).

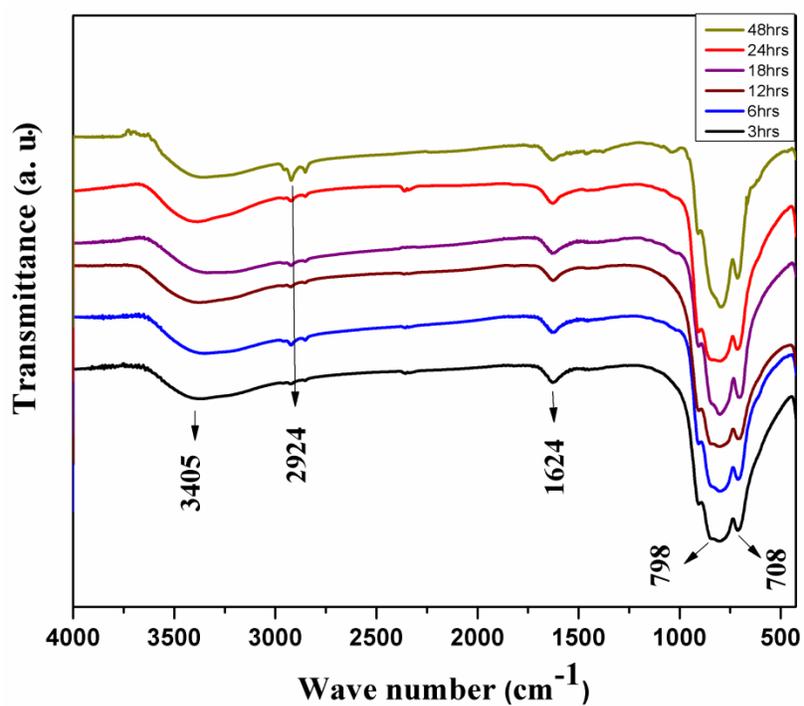


Fig. S1 FTIR spectra of $\text{Na}_{0.5}\text{Gd}_{0.5}\text{MoO}_4:\text{Eu}^{3+}$ synthesized with different reaction time intervals.

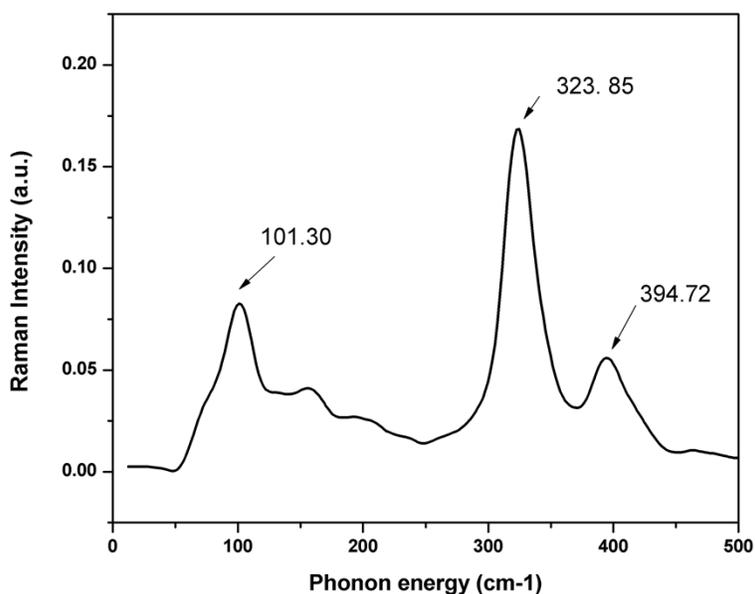


Fig. S2 Raman spectra of microstructures focusing on the Na_{0.5}Gd_{0.5}MoO₄ phonon region.

For representative analysis, the phonon bands of the Na_{0.5}Gd_{0.5}MoO₄:Eu³⁺ were identified by using Raman spectroscopy and shown in fig. S2. The phonon energy of Eu³⁺ doped Na_{0.5}Gd_{0.5}MoO₄ is found to be maximum at 323 cm⁻¹ which is assigned due to the symmetric bending vibrations within the [MoO₄]²⁻ clusters.

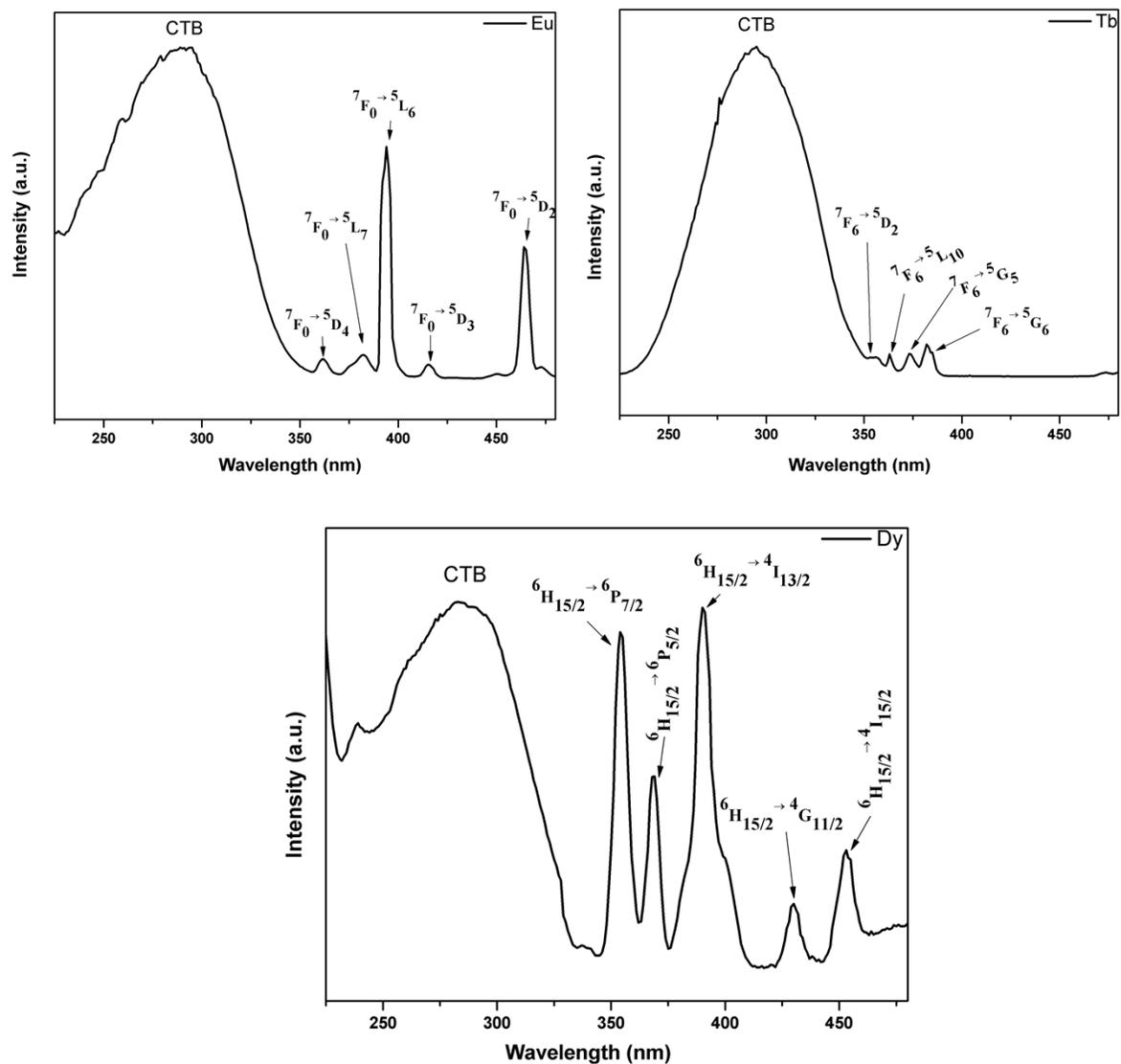


Fig. S3 Excitation spectrum of $(Na_{0.5}Gd_{0.5})MoO_4$ doped with Ln^{3+} (Eu, Tb, Dy).

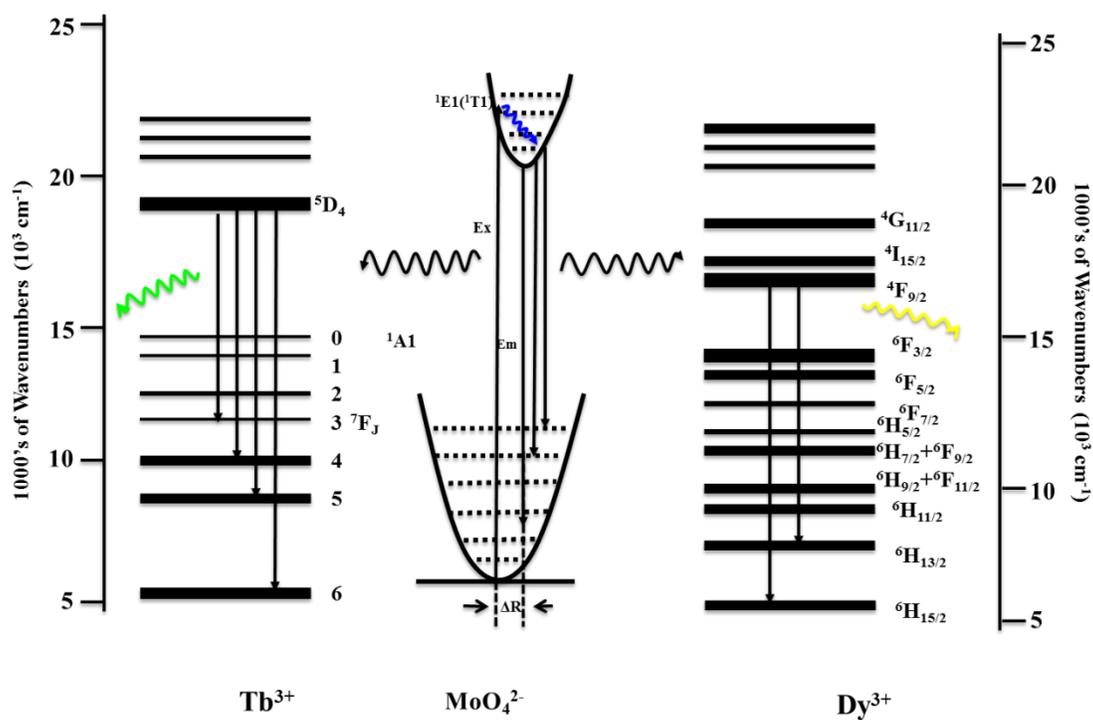


Fig. S4: The energy level diagram and the energy transfer process in the $\text{Na}_{0.5}\text{Gd}_{0.5}\text{MoO}_4:\text{Ln}^{3+}$ ($\text{Ln} = \text{Tb}^{3+}, \text{Dy}^{3+}$).

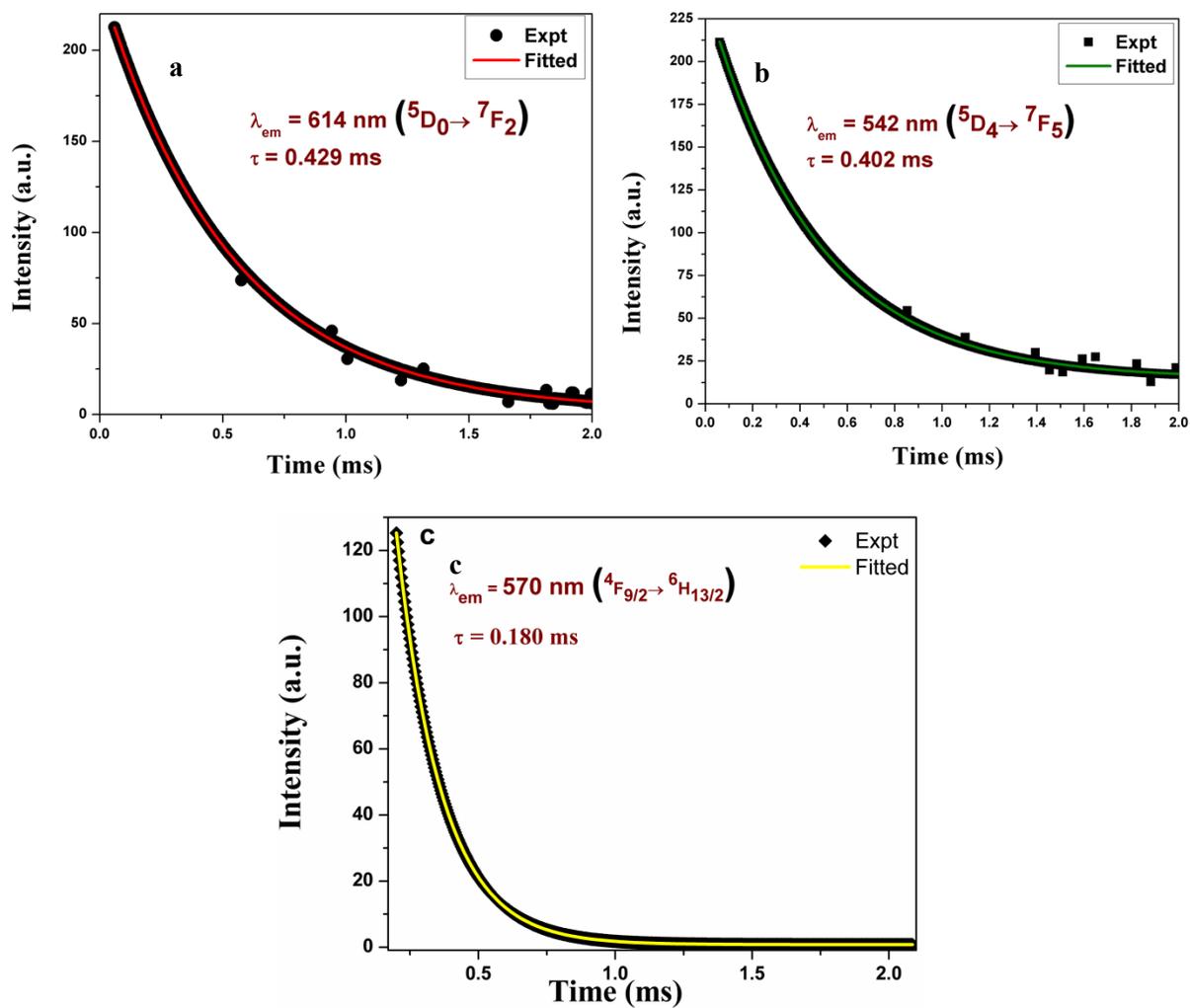


Fig. S5: Fluorescent life time of (a) ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2$ transition in Eu^{3+} , (b) ${}^5\text{D}_4 \rightarrow {}^7\text{F}_5$ transition in Tb^{3+} , (c) ${}^4\text{F}_{9/2} \rightarrow {}^6\text{H}_{13/2}$ transition in Dy^{3+} .

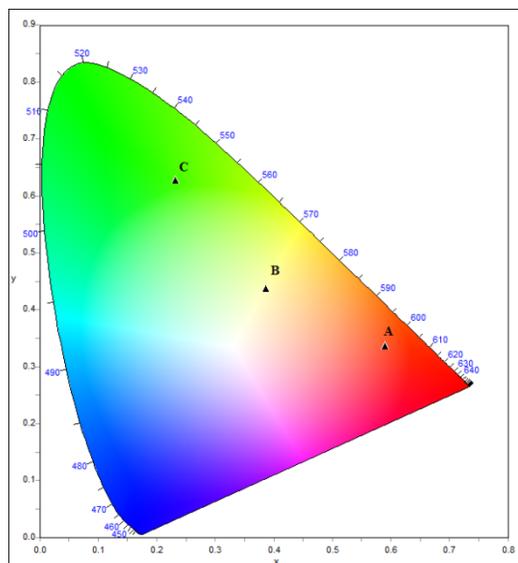


Fig. S6 CIE diagram of Na_{0.5}Gd_{0.5}MoO₄ microstructures doped with Eu³⁺, Tb³⁺, Dy³⁺ (A, B, C, respectively).

Table S1: Photometric characteristics of $(\text{Na}_{0.5}\text{Gd}_{0.5-x})\text{MoO}_4:\text{xEu}^{3+}$

Phosphor	Eu^{3+} concentration in moles	CCT (K)	CRI	Colour coordinates		LER (lm W^{-1})
				x	y	
$(\text{Na}_{0.5}\text{Gd}_{0.5-x})\text{MoO}_4:\text{xEu}^{3+}$	x = 0.02	1534	57	0.511	0.318	178
	x = 0.04	1489	53	0.521	0.321	182
	x = 0.06	1204	47	0.589	0.334	198
	x = 0.08	1326	51	0.555	0.326	192
	x = 0.10	1447	52	0.528	0.321	189

Table S2: Photometric characteristics of $(\text{Na}_{0.5}\text{Gd}_{0.5-x-y})\text{MoO}_4:\text{xYb}^{3+}/\text{yEr}^{3+}$

Phosphor	Er^{3+} Concentration in moles	CCT (K)	CRI	Colour coordinates		LER (lm W^{-1})
				x	y	
$(\text{Na}_{0.5}\text{Gd}_{0.5-x-y})\text{MoO}_4:\text{xYb}^{3+}/\text{yEr}^{3+}$ ($x=0.05$ moles of Yb^{3+})	y = 0.01	Undefined	24	0.185	0.780	447
	y = 0.02	Undefined	19	0.182	0.775	457
	y = 0.03	Undefined	12	0.178	0.766	482
	y = 0.04	Undefined	16	0.181	0.732	475
	y = 0.05	Undefined	14	0.183	0.756	473

Table S3: Photometric characteristics of $(\text{Na}_{0.5}\text{Gd}_{0.5})\text{MoO}_4:\text{Ln}^{3+}$ ($\text{Ln}=\text{Tb}, \text{Dy}, \text{Yb}/\text{Tm}, \text{Yb}/\text{Ho}$) under optimized condition (200°C for 24 h)

Phosphor	CCT (K)	CRI	Colour coordinates		LER (lm W ⁻¹)
			x	y	
$(\text{Na}_{0.5}\text{Gd}_{0.5})\text{MoO}_4:\text{Eu}^{3+}$	1204	47	0.589	0.334	198
$(\text{Na}_{0.5}\text{Gd}_{0.5})\text{MoO}_4:\text{Tb}^{3+}$	Undefined	8	0.232	0.625	278
$(\text{Na}_{0.5}\text{Gd}_{0.5})\text{MoO}_4:\text{Dy}^{3+}$	4208	7	0.386	0.433	283
$(\text{Na}_{0.5}\text{Gd}_{0.5})\text{MoO}_4:0.05\text{Yb}^{3+}/0.03\text{Er}^{3+}$	Undefined	12	0.178	0.766	482
$(\text{Na}_{0.5}\text{Gd}_{0.5})\text{MoO}_4:0.05\text{Yb}^{3+}/0.03\text{Tm}^{3+}$	>100000	Undefined	0.153	0.145	96
$(\text{Na}_{0.5}\text{Gd}_{0.5})\text{MoO}_4:0.05\text{Yb}^{3+}/0.03\text{Ho}^{3+}$	Undefined	33	0.394	0.510	281