Electronic Supplementary Material (ESI) for New Journal of Chemistry This journal is © The Royal Society of Chemistry and The Centre National de la Recherche Scientifique 2014

## **ELECTRONIC SUUPLEMENTRAY INFORMATION**

## Synthesis and Application of a Novel Bio-based Polyol for Preparation of Polyurethane Foams

Xiaolin Li, <sup>a</sup> Zheng Fang, <sup>a</sup> Xin Li, <sup>b</sup> Shigui Tang, <sup>b</sup> Kai Zhang <sup>c</sup> and Kai Guo, <sup>\*b</sup>

<sup>a</sup> School of Pharmaceutical Sciences, Nanjing University of Technology, Nanjing 211816, China
<sup>b</sup> College of Biotechnology and Pharmaceutical Engineering, Nanjing University of Technology, Nanjing 211816,
China

<sup>c</sup> School of Mechanical and Power Engineering of NJUT, Nanjing University of Technology, Nanjing 211816, China

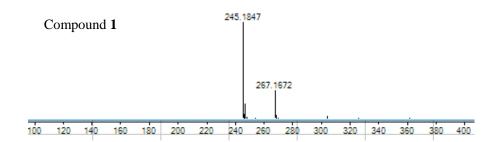
## **Contents**

- 1. Synthesis and characterization of  $\alpha$ -amino- $\epsilon$ -caprolactam (ACL)
- 2. **Figure SI1.** The mass of the structures **1** and **3**.
- 3. **Figure SI2.** The LC-MS of the ACL-OL at 24 h.
- 4. **Figure SI3.** The LC-MS of the ACL-OL at 72 h.
- 5. **Figure SI4.** DMTA date of foams.
- 6. **Figure SI5.** DSC date of ACL-RF.
- 7. **Figure SI6.** DSC date of PB-RF.

## Synthesis and characterization of α-amino-ε-caprolactam (ACL)

A stirred mixture of L-lysine hydrochloride (110 g, 600 mmol) and NaOH (24 g, 600 mmol) in hexanol (2.4 L) was heated to reflux with a Dean-Stark trap used to remove H<sub>2</sub>O. The suspension was refluxed for 8 hours until all starting material was consumed (which was determined by TLC). The suspension was then cooled and filtered to remove byproduct NaCl. The filtrate was concentrated and the resulting crude α-amino-ε-caprolactam was dissolved in water. After acidification to pH 6 with addition of concentrated HCl and partial concentration, crystal was formed at room temperature to afford ACL hydrochloride (74 g) in 75 % yield. The ACL hydrochloride was dissolved in methanol and NaOH (18 g, 450 mmol) was added.

After stir for 2 h, the suspension was then filtered and the filtrate was evaporated under reduced pressure conditions to afford ACL (57 g) as a white solid. Mp: 97 - 101 °C. Elemental analysis (%) found: C: 56.36, H: 9.50, N: 21.71;  $C_6H_{12}N_2O$ ; C: 56.22, H: 9.44, N: 21.86. HRMS: found m/z = 129.1046 ([M + H]<sup>+</sup>), calculated 129.1028. <sup>1</sup>H-NMR (500 MHz, CD<sub>3</sub>OD):  $\delta$  (ppm) = 7.19 (br s, 2 H), 5.12 (d, J = 11.1 Hz, 1 H), 3.58 (s, 1 H), 3.15 (m, 2 H), 1.49-1.14 (m, 6 H). <sup>13</sup>C-NMR (125 MHz, CD<sub>3</sub>OD):  $\delta$  (ppm) = 178.26, 52.82, 41.01, 33.72, 28.17, 27.94.



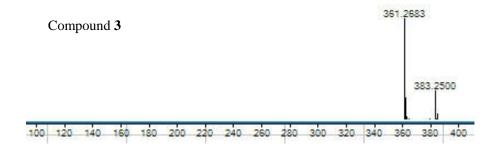
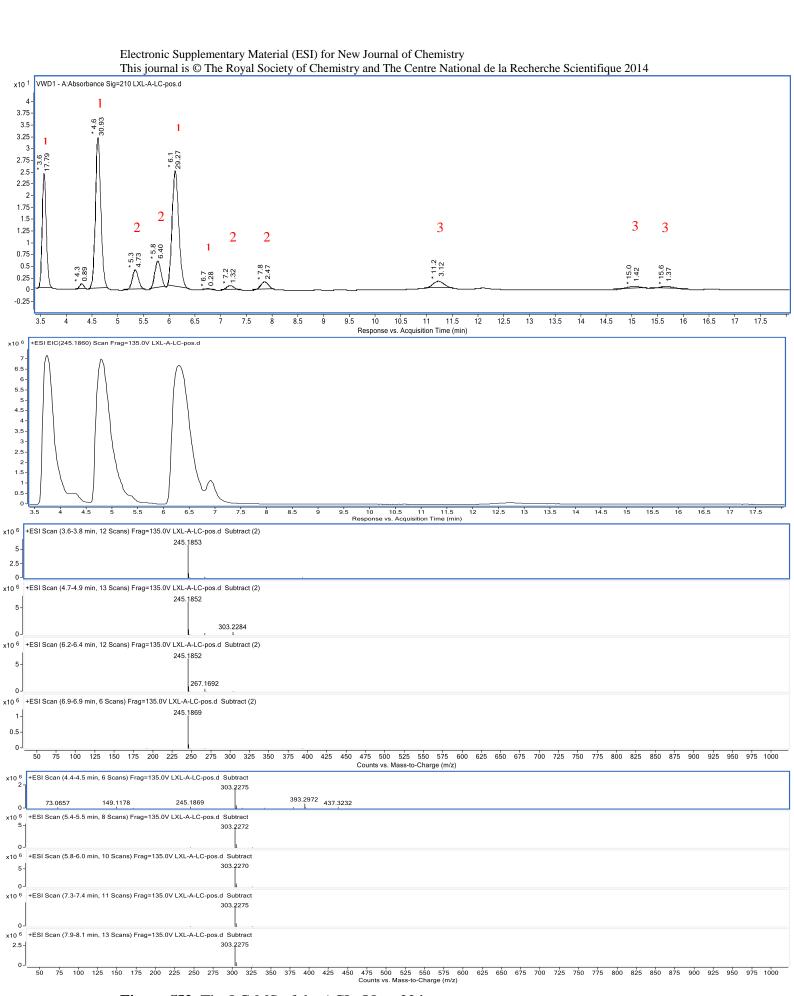
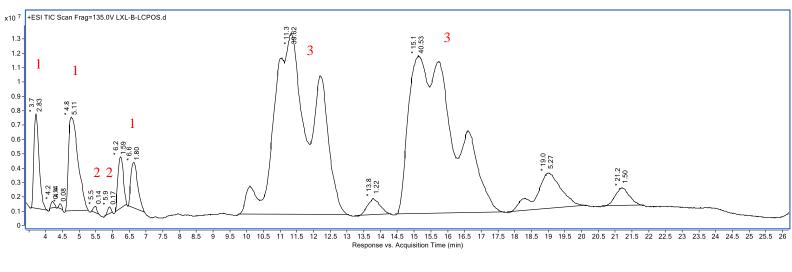
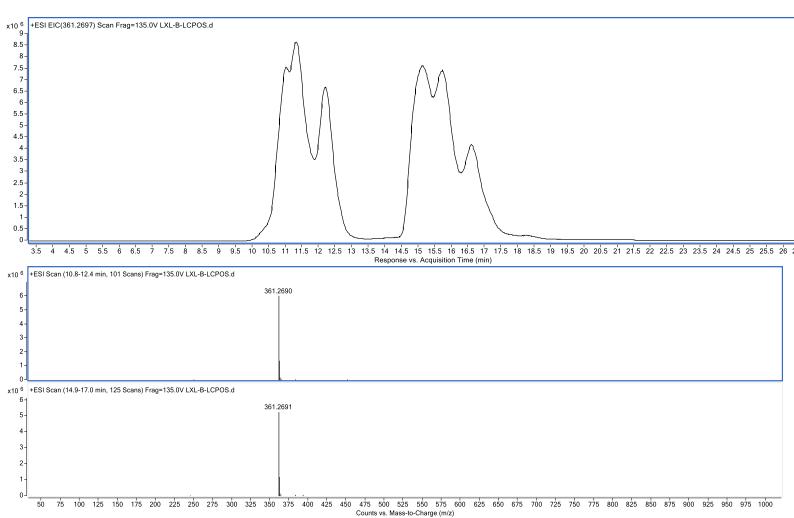


Figure SI1. The mass of the structures 1 and 3.

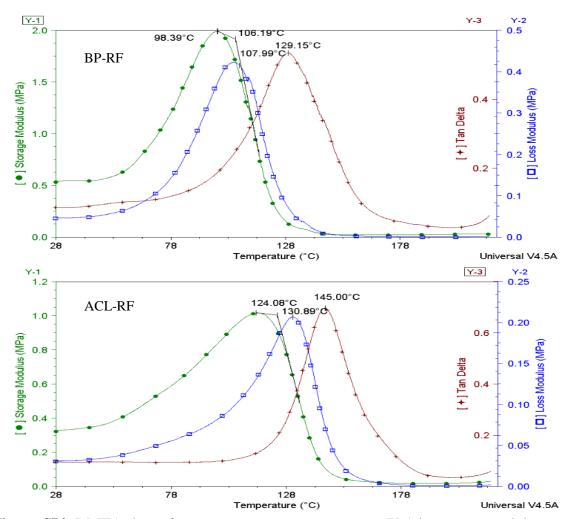


**Figure SI2.** The LC-MS of the ACL-OL at 22 h.





**Figure SI3.** The LC-MS of the ACL-OL at 72 h. For the different position of -NH<sub>2</sub> attacking the PO, the compounds 1, 2 and 3 have several different configuration. We do not talk about it in this article.



**Figure SI4.** DMTA date of foams showing the thermostability. (Y-1 is storage modulus; Y-2 is loss modulus; Y-3 is  $\delta$ )

Details about DMTA experiment: sample dimensions, cylinders of  $\Phi15 \times 10$  mm; frequency,  $0.01 \sim 200$  Hz; force,  $0.0001 \sim 18$  N; compression mode,  $10^3 \sim 3 \times 10^{12}$  Pa; temperature, -150 $\sim$ 600 °C. Tg was determined by the maximum peak at Tan  $\delta$ .

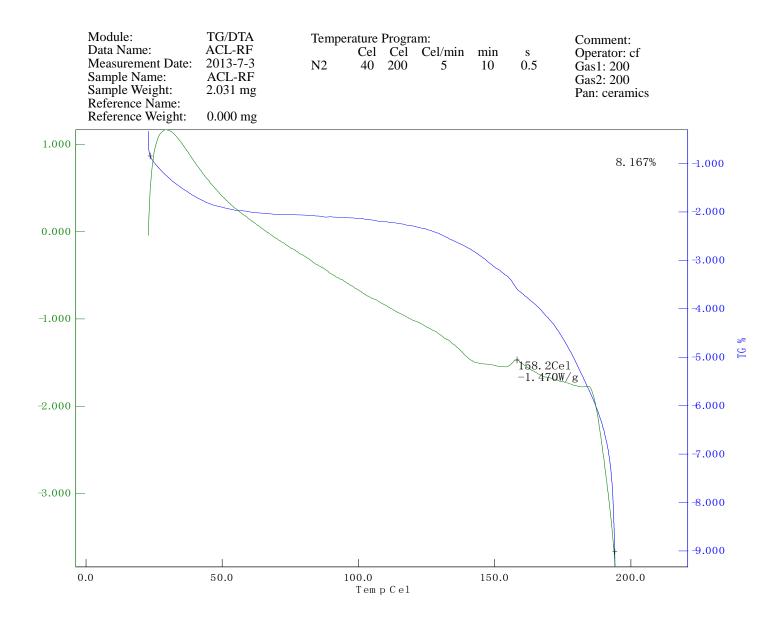


Figure SI5. DSC date of ACL-RF

Comment: Module: TG/DTA Temperature Program: Operator: cf Gas1: 200 PB-RF Data Name: Cel Cel 40 150 Cel/min min Measurement Date: 2013-7-3 0.5 N2 5 10 Gas2: 200 Sample Name: PB-RF Sample Weight: Reference Name: Pan: ceramics 2.678 mg

Reference Weight: 0.000 mg

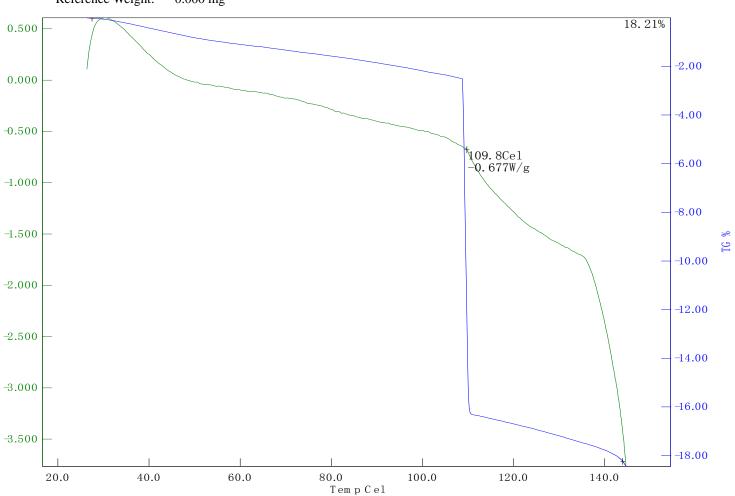


Figure SI6. DSC date of PB-RF.