Electronic Supplementary Material (ESI) for New Journal of Chemistry.

This journal is © The Royal Society of Chemistry and the Centre National de la Recherche Scientifique 2014

## New Journal of Chemistry

**RSCPublishing** 

**ARTICLE** 

## Revisiting urea-based gelators: strong solvent- and castingmicrostructuration dependencies and organogel processing with an alumina template

Thanh-Loan Lai,<sup>a</sup> David Canevet,<sup>a\*</sup> Yasser Almohamed,<sup>a</sup> Jean-Yves Mévellec,<sup>b</sup> Régis Barillé,<sup>a</sup> Narcis Avarvari<sup>a\*</sup> and Marc Sallé<sup>a\*</sup>

Fig. S1 Teas diagram including the set of solvents under consideration	p 2
Fig. S2 <sup>1</sup> H NMR spectrum of compound 1 in DMSO-D <sub>6</sub> at 393 K	p 2
Fig. S3 <sup>1</sup> H NMR spectrum of compound 2 in DMSO-D <sub>6</sub> at 393 K	p 3
<b>Table S1</b> . Images of <b>1</b> -based xerogels by optical microscopy depending on the solvent of preparation and the casting method (the corresponding gels were prepared at a concentration $C = 1.25 \times CGC$ )	p 4
<b>Table S2</b> . Images of <b>2</b> -based xerogels by optical microscopy depending on the solvent of preparation (the corresponding gels were prepared at a concentration $C = 1.25 \times CGC$ and deposited as such)	p 5
<b>Table S3</b> . Images of <b>2</b> -based xerogels by optical microscopy depending on the solvent of preparation (the corresponding gels were prepared at a concentration $C = 1.25 \times CGC$ and drop-casted).	p 6
<b>Table S4</b> . Images of the slice, upper and lower faces of the alumina membranes by SEM microscopy depending on the method of deposition.	p 7
Fig. S4 NMR spectra of a sample of compound 1 before and after a DSC cycle $(30^{\circ}\text{C} \rightarrow 120^{\circ}\text{C} \rightarrow 30^{\circ}\text{C})$ .	p 9
<b>Fig. S5</b> SEM micrographs of the <b>2</b> -based xerogel prepared from $o$ DCB after one week in a 5M NaOH aqueous solution at 80°C	p 9

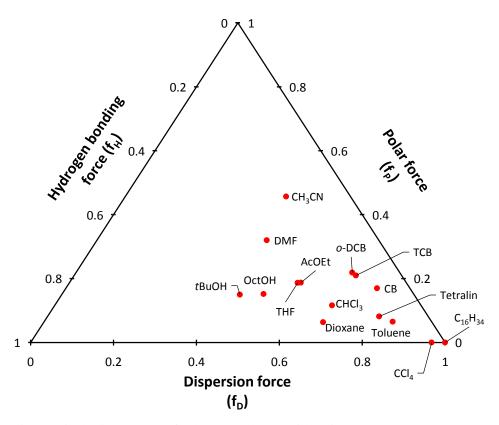


Fig. S1 Teas diagram including the set of solvents under consideration

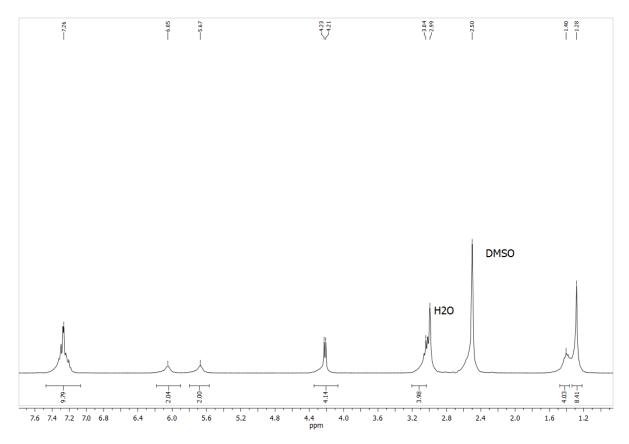


Fig. S2  $^1\text{H}$  NMR spectrum of compound 1 in DMSO-D6 at 393 K

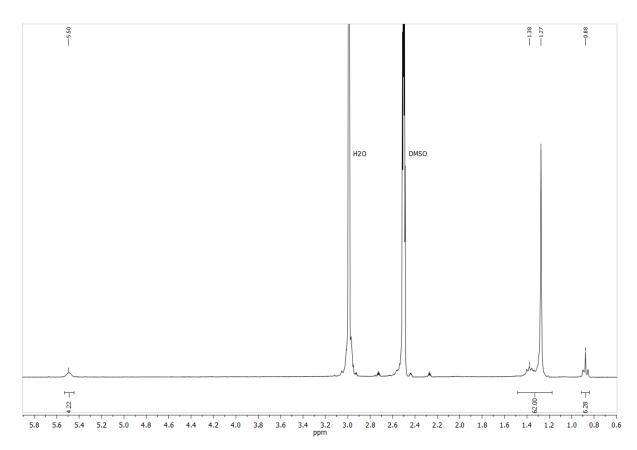
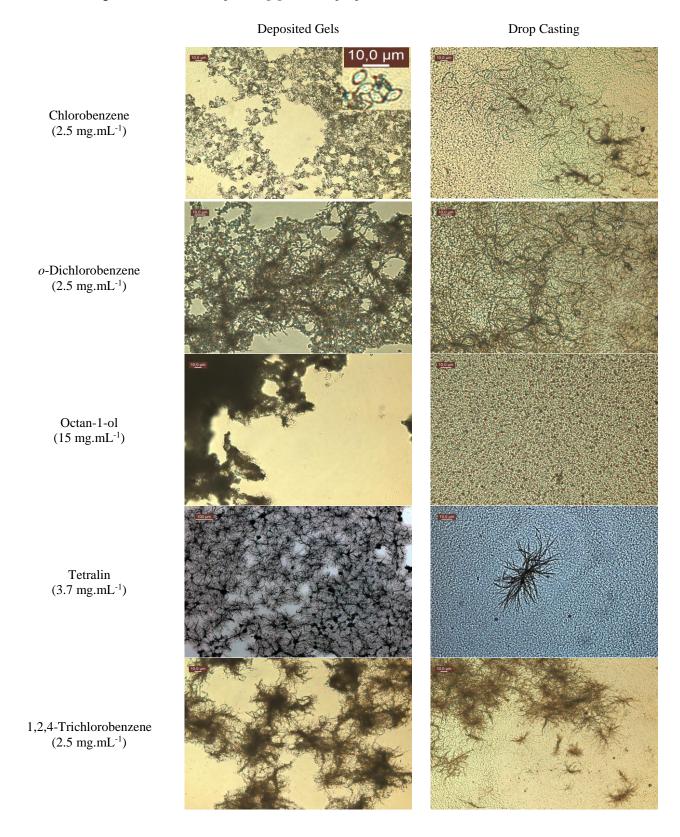
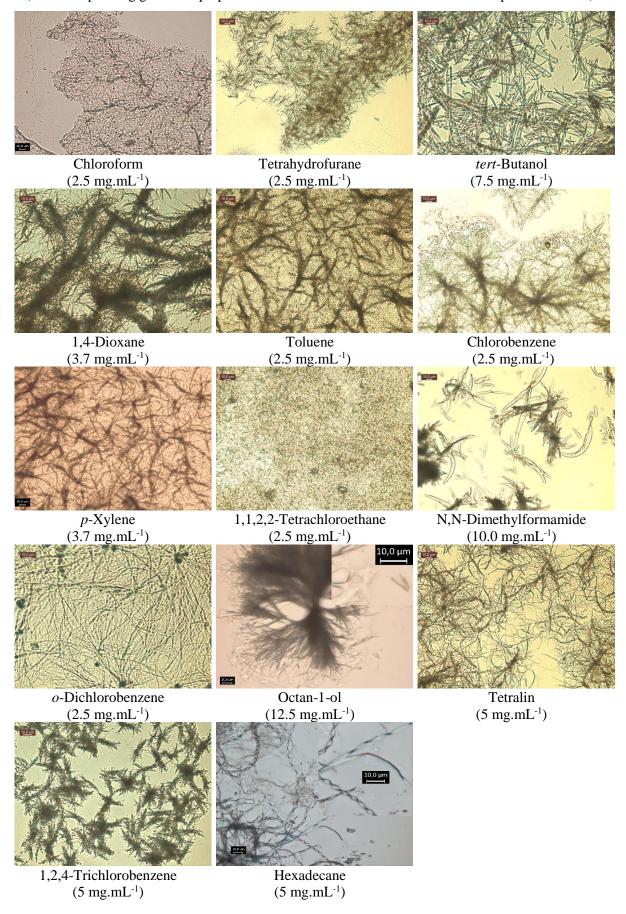


Fig. S3  $^1H$  NMR spectrum of compound 2 in DMSO-D6 at 393 K

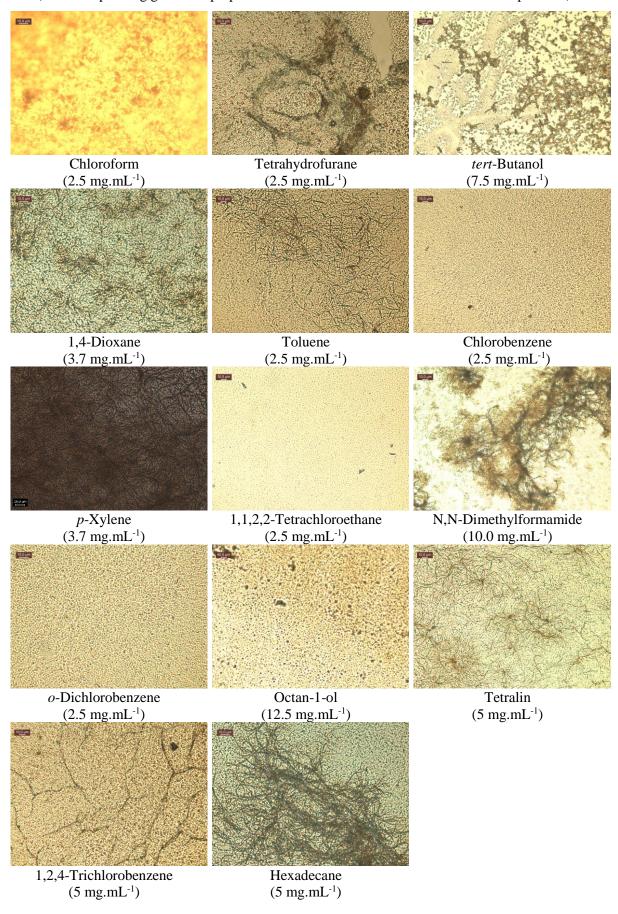
**Table S1.** Images of **1**-based xerogels by optical microscopy depending on the solvent of preparation and on the casting method (*the corresponding gels were prepared at a concentration*  $C = 1.25 \times CGC$ )



**Table S2**. Images of **2**-based xerogels by optical microscopy depending on the solvent of preparation (the corresponding gels were prepared at a concentration  $C = 1.25 \times CGC$  and deposited as such)

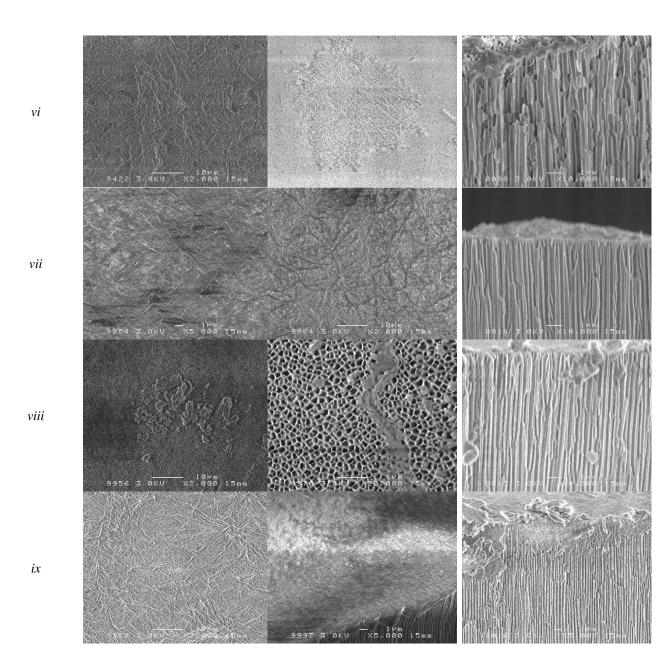


**Table S3**. Images of **2**-based xerogels by optical microscopy depending on the solvent of preparation (the corresponding gels were prepared at a concentration  $C = 1.25 \times CGC$  and drop-casted).

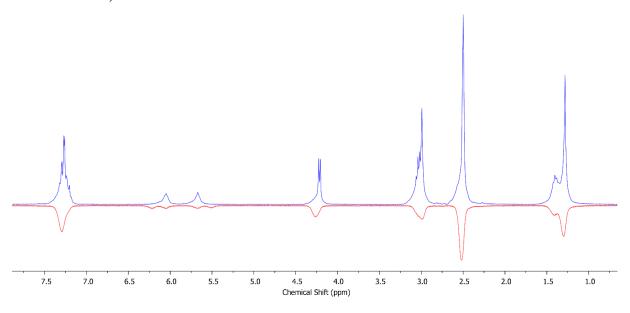


**Table S4**. Images of the slice, upper and lower faces of the alumina membranes by SEM microscopy depending on the method of deposition.

Upper face Slice Lower face Method i ii iii iv



**Figure S4.** <sup>1</sup>H NMR spectra of a sample of compound **1** before and after a DSC cycle (30°C  $\rightarrow$ 120°C $\rightarrow$ 30°C).



**Figure S5**. SEM micrographs of the **2**-based xerogel prepared from oDCB after one week in a 5M NaOH aqueous solution at 80°C

