

Supplemental Material for:

**A Novel Cu₁₂(pz)₁₂ Loop-based Coordination Polymer
Templated by Double-Keggin Anions†**

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Table S1. Summarization of known coordination compounds assembled from non-coordinating double-Keggin anions as templates.

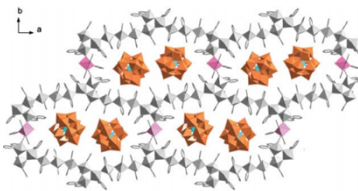
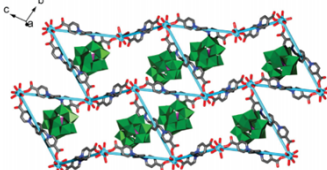
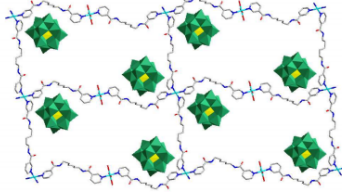
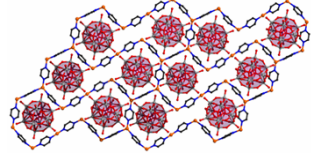
Compounds	References
 $[\{\text{Mn}(\text{bpy})(\text{py})(\text{H}_2\text{O})_2\}\{\text{Mo}_{12}\text{O}_{34}(\text{bpy})_{12}\}][\text{PMo}_{12}\text{O}_{40}]_2 \cdot 2\text{H}_2\text{O}$	<i>Enbo Wang et al., Chem. Commun.</i> 2007, 770
 $[\text{Ln}(\text{L})_{1.5}(\text{H}_2\text{O})_5][\text{PMo}_{12}\text{O}_{40}] \cdot 1.5\text{CH}_3\text{CN} \cdot 2\text{H}_2\text{O}$	<i>Enbo Wang et al., Dalton Trans.</i> 2011, 40, 5971
 $[\text{Cu}_2(\text{L}_3)(\text{SiMo}_{12}\text{O}_{40})(\text{H}_2\text{O})_6] \cdot 4\text{H}_2\text{O}$ $[\text{Cu}_2(\text{L}_3)(\text{SiW}_{12}\text{O}_{40})(\text{H}_2\text{O})_6] \cdot 4\text{H}_2\text{O}$	<i>Xiuli Wang et al., CrystEngComm.</i> 2012, 14, 5836
 $[\text{Cu}_5(\text{pz})_6(\text{H}_2\text{O})_4][\text{PW}^{\text{VI}}_{10}\text{W}^{\text{V}}_2\text{O}_{40}]$	This work



Fig. S1. The coordination modes of three crystallographically independent Cu cations.

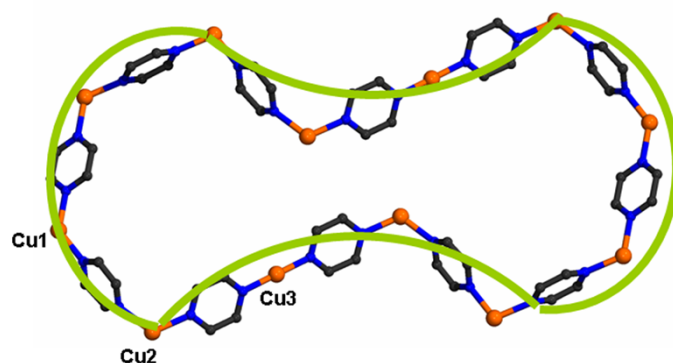


Fig. S2. The $\text{Cu}_{12}(\text{pz})_{12}$ loop in compound 1.

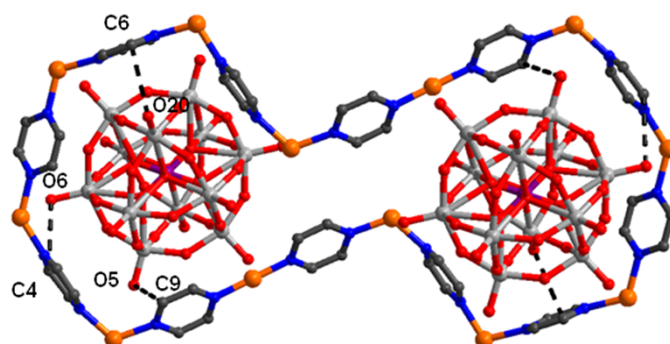


Fig. S3. The hydrogen bonding interactions between dumbbell $\text{Cu}_{12}(\text{pz})_{12}$ loop and Keggin PW_{12} anions.

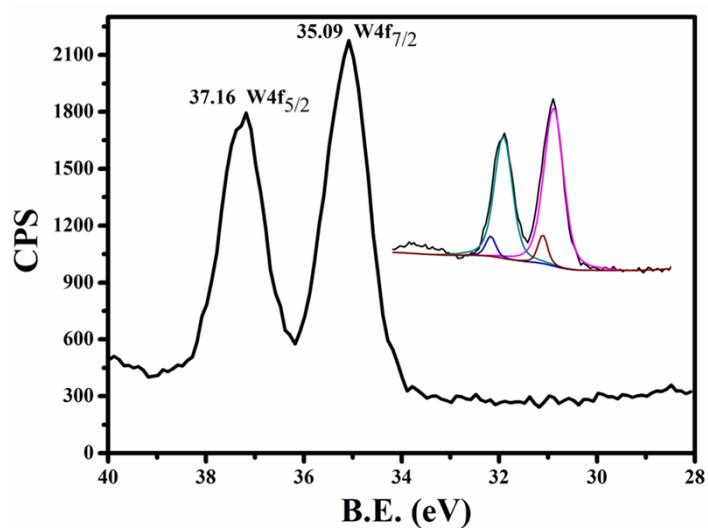


Fig. S4. The XPS spectrum of compound 1.

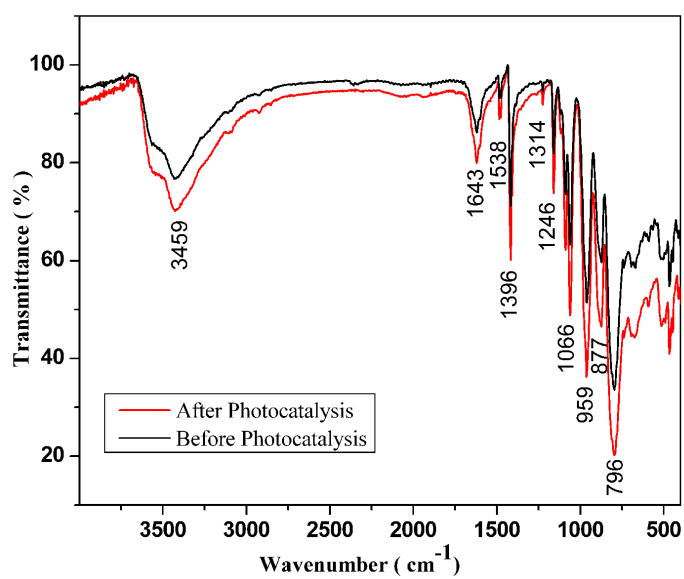


Fig. S5. The IR spectra of compound 1 (black) and the sample after the third run of the RhB degradation test (red).

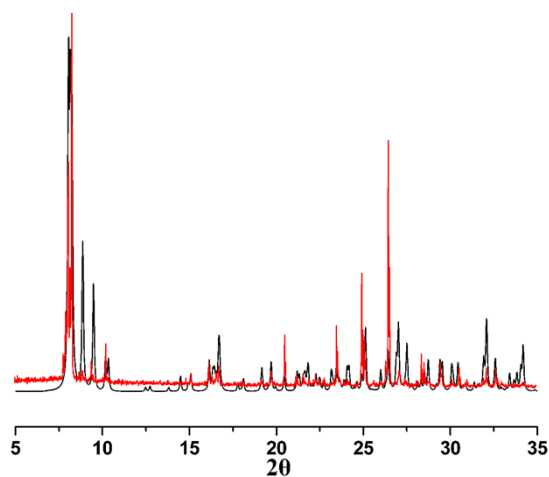


Fig. S6. The simulative (black) and experimental (red) powder X-ray diffraction patterns for compound **1**.

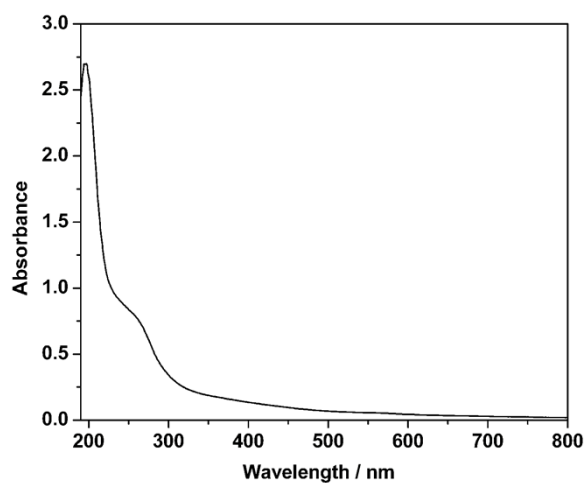


Fig. S7. The UV-vis absorption spectrum of compound **1** in aqueous solution.

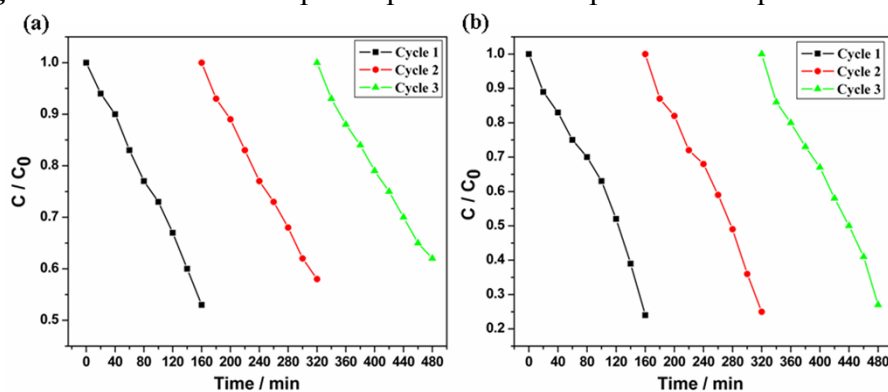


Fig. S8. Three time of the RhB degradation test by (a) NBu₄)₃[PW₁₂O₄₀], and (b) compound **1**.