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Supporting information

for

A highly selective turn-on fluorescent sensor for fluoride and its application in

imaging of living cells

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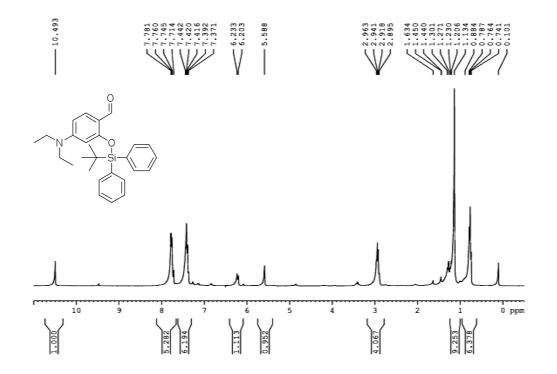


Figure S1. ¹H NMR spectra (300 MHz) of 1 in CDCl₃.

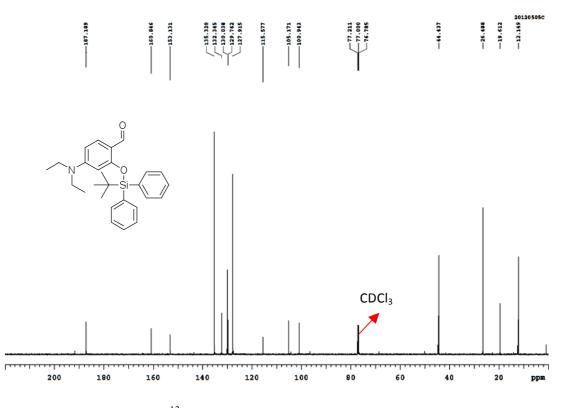


Figure S2. ¹³C NMR spectra (125 MHz) of 1 in CDCl₃.

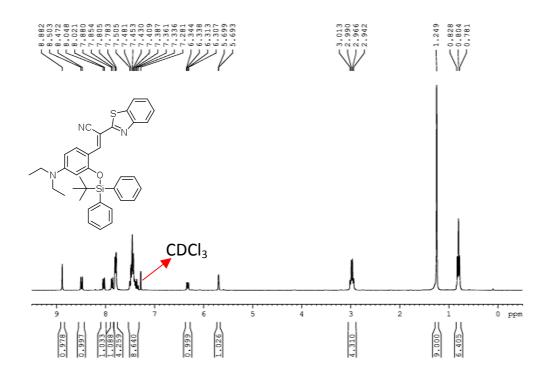


Figure S3. ¹H NMR spectra (300 MHz) of FS in CDCl₃.

PS1/CDC13 120303C

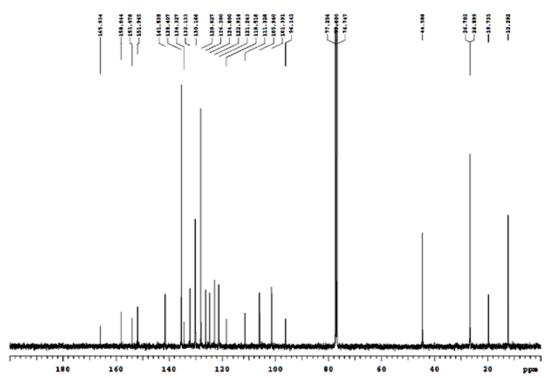


Figure S4. ¹³C NMR spectra (125 MHz) of FS in CDCl₃.

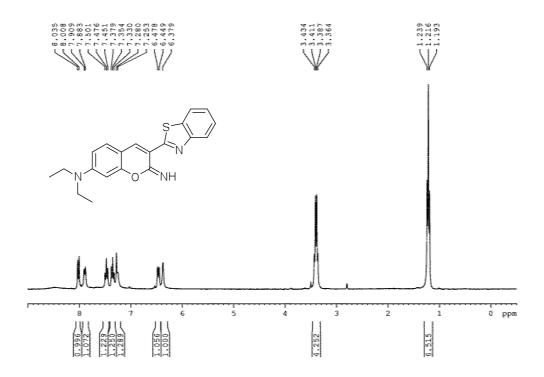


Figure S5. ¹H NMR (300 MHz, CDCl₃) spectra of the product of FS react with F⁻

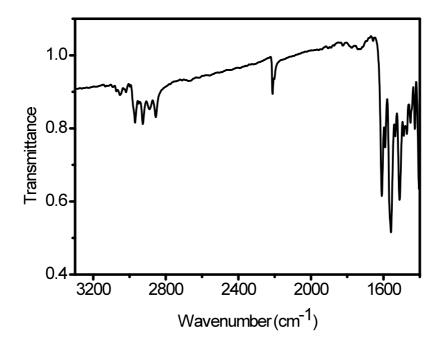


Figure S6. FTIR spectrum of FS.

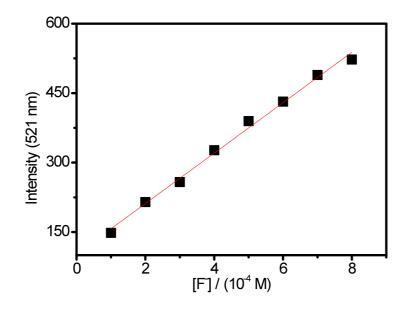


Figure S7. Calibration curve of chemosesor FS (10 μ M) with F⁻. The excitation wavelength was 465 nm.

Linear Regression Data:

Y = A + S * X						
Parameter	Value	Error	R	SD	N	Р
А	102.52857	8.29516	0.99728	10.64583	8	< 0.0001
S	544630.95238	16426.8647	7			

The detection limit (DL) of F^- ions using chemosensor **FS** was determined from the following equation:

 $DL = K * S_b / S$

Where K = 3; S_b is the standard deviation of the blank solution; S is the slope of the calibration curve. DL = $3 * 0.263039 / 544630.95238 = 1.45 * 10^{-6} M$