

## **Bis - EDOT end capped by n-hexyl or n-hexylsulfanyl groups: Effect of the substituents on the stability of the oxidized states**

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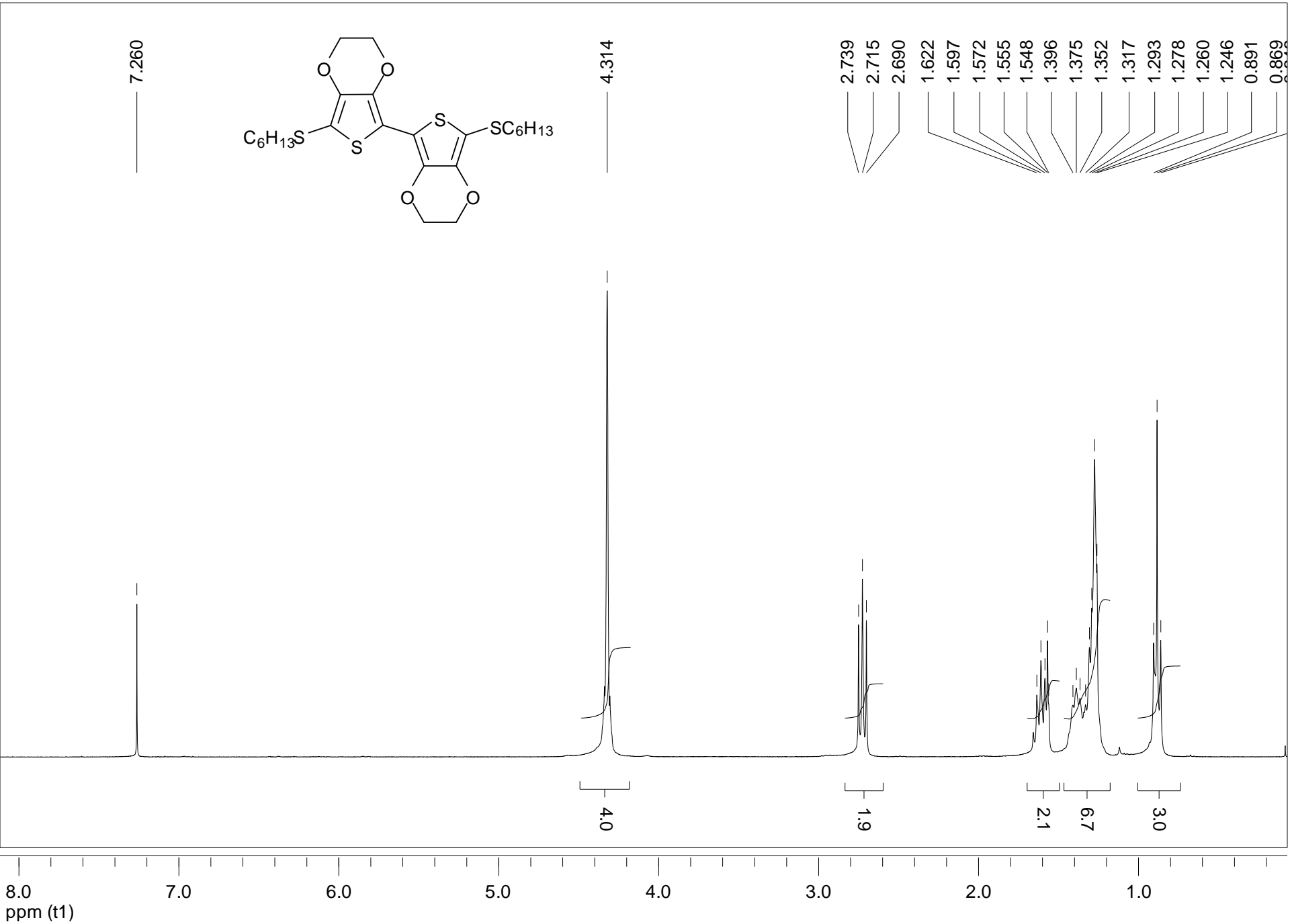
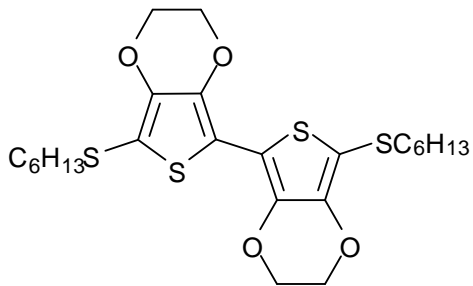
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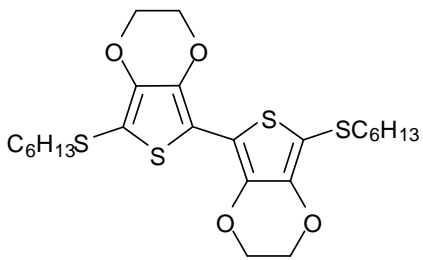
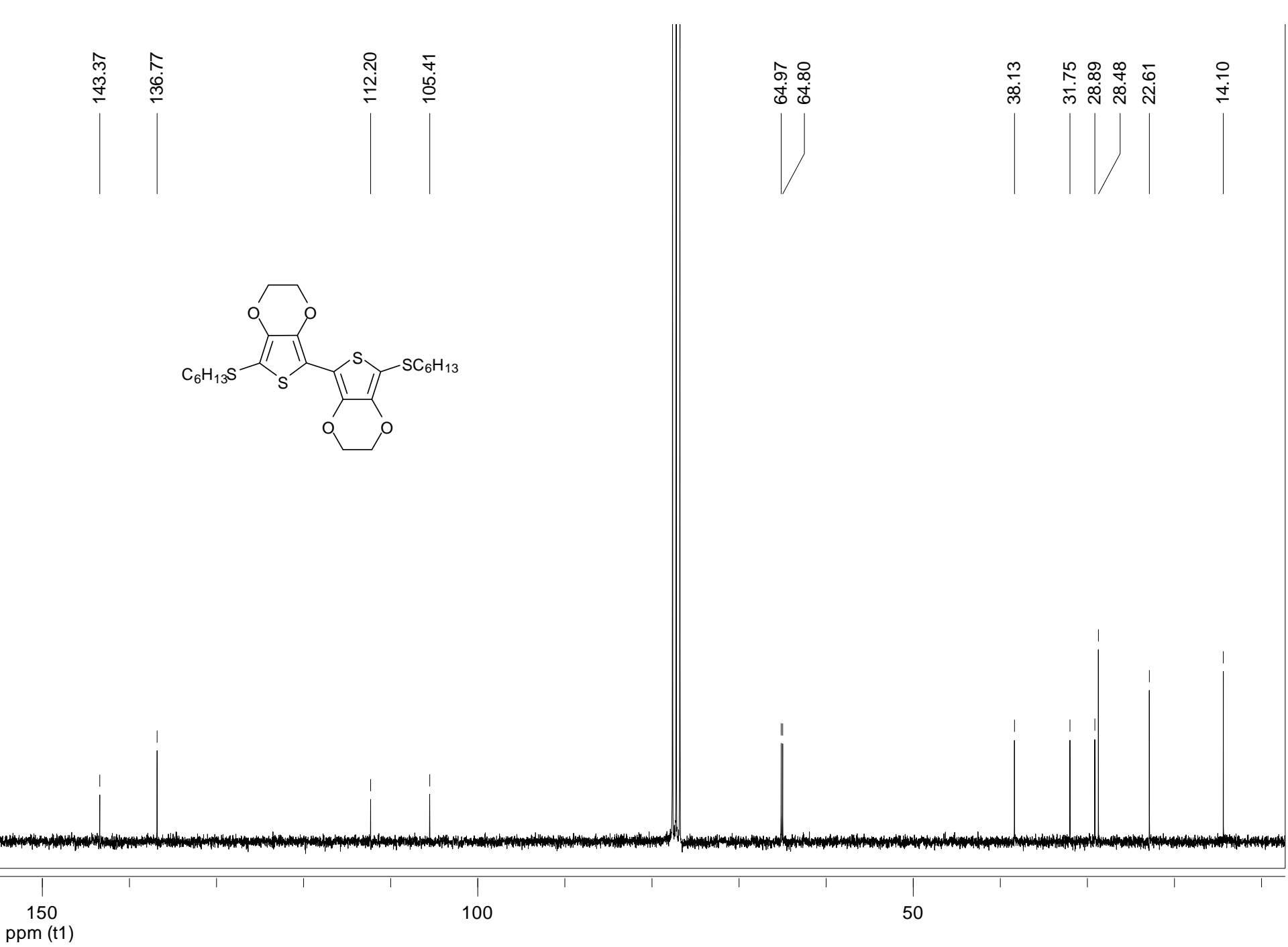
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### **Supplementary information**

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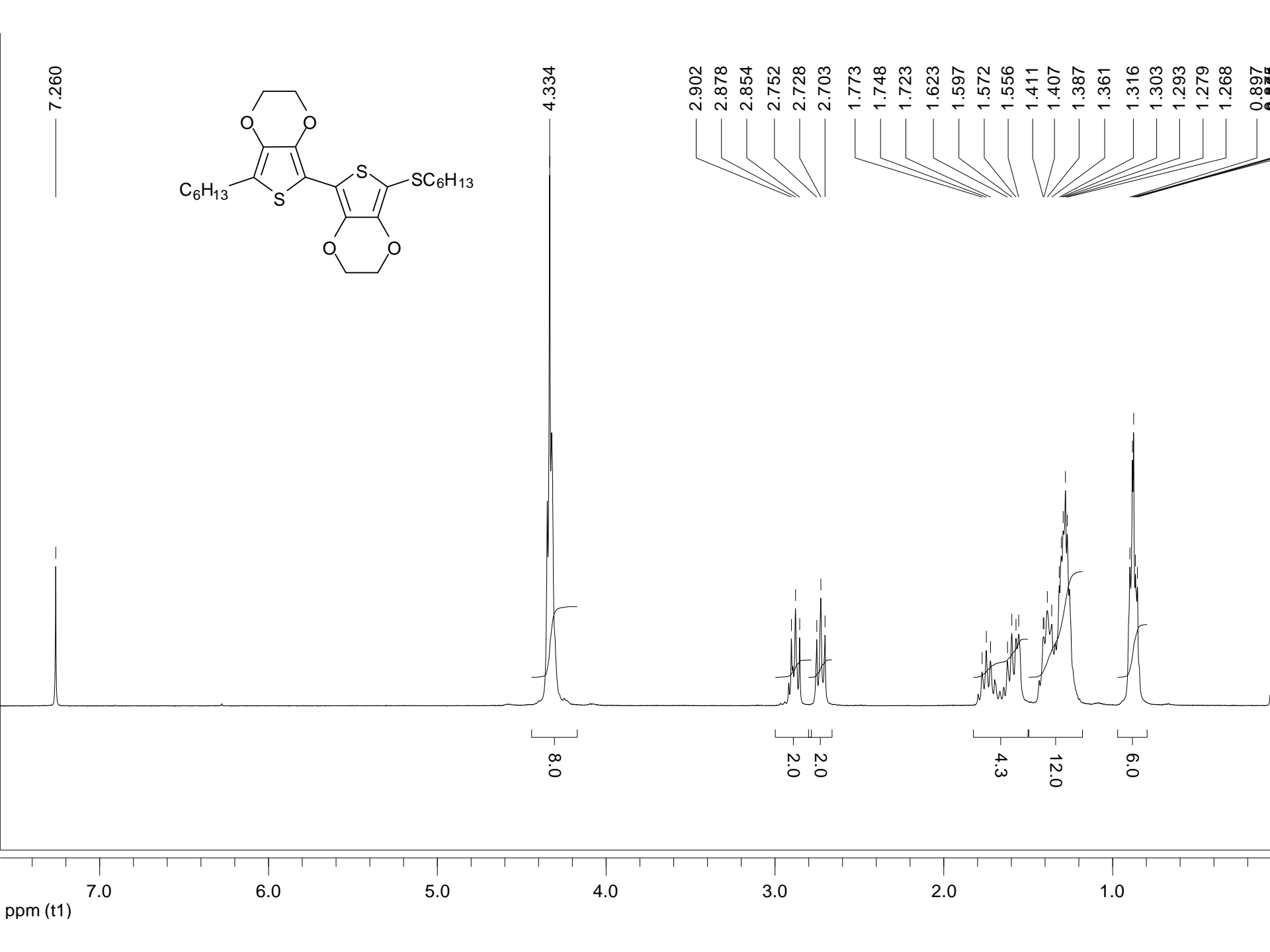




150  
ppm (t1)

100

50

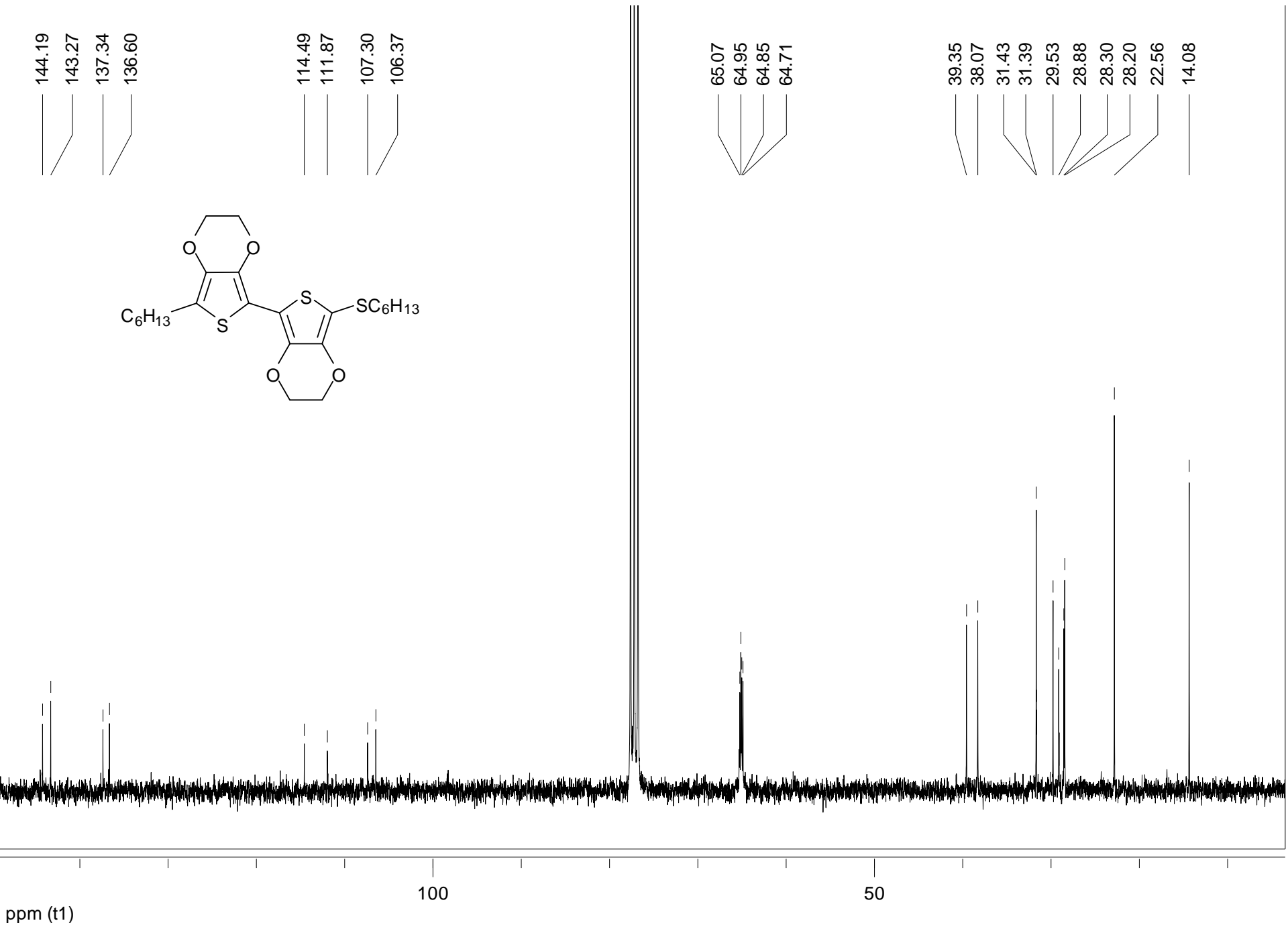
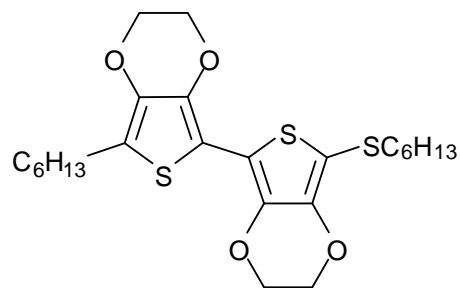


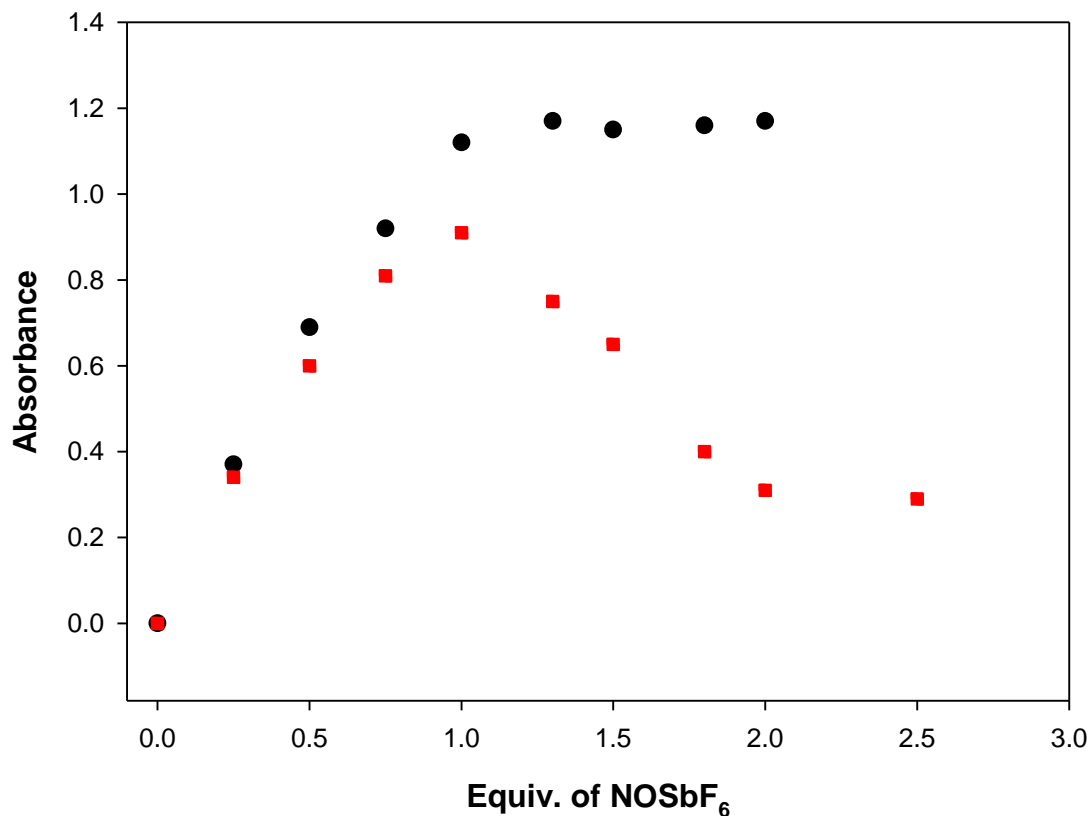
144.19  
143.27  
137.34  
136.60

114.49  
111.87  
107.30  
106.37

65.07  
64.95  
64.85  
64.71

39.35  
38.07  
31.43  
31.39  
29.53  
28.88  
28.30  
28.20  
22.56  
14.08





**Figure S1** : Evolution of absorbance at  $\lambda = 440$  nm ( $A_{440}$  black circle) for **EEa**  $10^{-4}$  M in  $\text{CH}_2\text{Cl}_2$  and at  $\lambda = 510$  nm ( $A_{510}$  red square) for **EEb**  $10^{-4}$  M in  $\text{CH}_2\text{Cl}_2$  during the stepwise addition of  $\text{NOSbF}_6$

$\text{NOSbF}_6$  ( $10^{-2}$  M in  $\text{CH}_3\text{CN}$ ) was added with a microsyringe to 5mL of **EEa** or **EEb** solution  $10^{-4}$  M in  $\text{CH}_2\text{Cl}_2$ . 1 equivalent of  $\text{NOSbF}_6$  corresponds to 50  $\mu\text{L}$  of oxidant solution.

