

## Supporting Information

### *New J. Chem.*

#### **Investigation of the copper(I) catalysed azide-alkyne cycloaddition reactions (CuAAC) in molten PEG<sub>2000</sub>**

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## Results of the CuAAC between azides **1** and **2** and alkynes **3b-d**

The reactions involving **3c** required more diluted conditions since using 0.3 g PEG<sub>2000</sub> per mmol of substrate gave a very viscous reacting mixture and, under these conditions, the conversion did not exceed 90%, even after a prolonged reaction time (12 h), either with 3-azidopropanol (**1**) or its acetate derivatives **2**. Using 0.4 g PEG<sub>2000</sub> per mmol allowed complete conversions with all catalysts and the isolated yields were between 91 and 97% (Table).

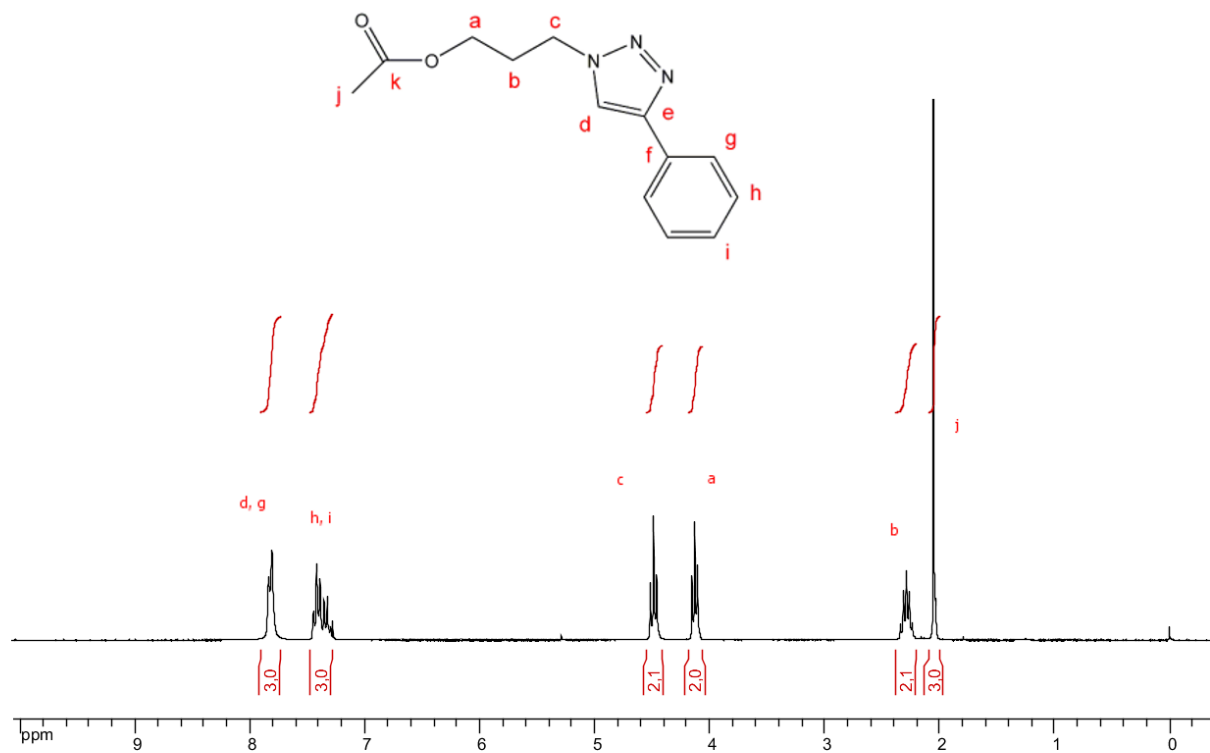
Yields<sup>a</sup> for the CuAAC between azides **1** and **2** and alkynes **3b-d** using CuI, Cu nanopowder (CuNP), or Cu turnings (CuT) in PEG<sub>2000</sub> at 70 °C.

Entry	Azide	Acetylene	Catalyst (wt %)	Time (h)	Product	Yield (%)
1	<b>1</b>	<b>3b</b>	CuI (1%)	2	<b>4b</b>	96
2			CuNP (1%)	2		97
3			CuT (40%)	8		94
4	<b>2</b>		CuI (1%)	4	<b>5b</b>	100
5			CuNP (1%)	4		100
6			CuT (40%)	4		100
7	<b>1</b>	<b>3c<sup>a</sup></b>	CuI (1%)	2	<b>4c<sup>b</sup></b>	91
8			CuNP (1%)	2		92
9			CuT (40%)	4		93
10	<b>2</b>		CuI (1%)	2	<b>5c<sup>b</sup></b>	96
11			CuNP (1%)	6		97
12			CuT (40%)	17		96
13	<b>1</b>	<b>3d</b>	CuI (1%)	0.5	<b>4d</b>	98
14			CuNP (1%)	0.5		97
15			CuT (40%)	6		96
16	<b>2</b>		CuI (1%)	0.5	<b>5d</b>	97
17			CuNP (1%)	2		96
18			CuT (40%)	4		94

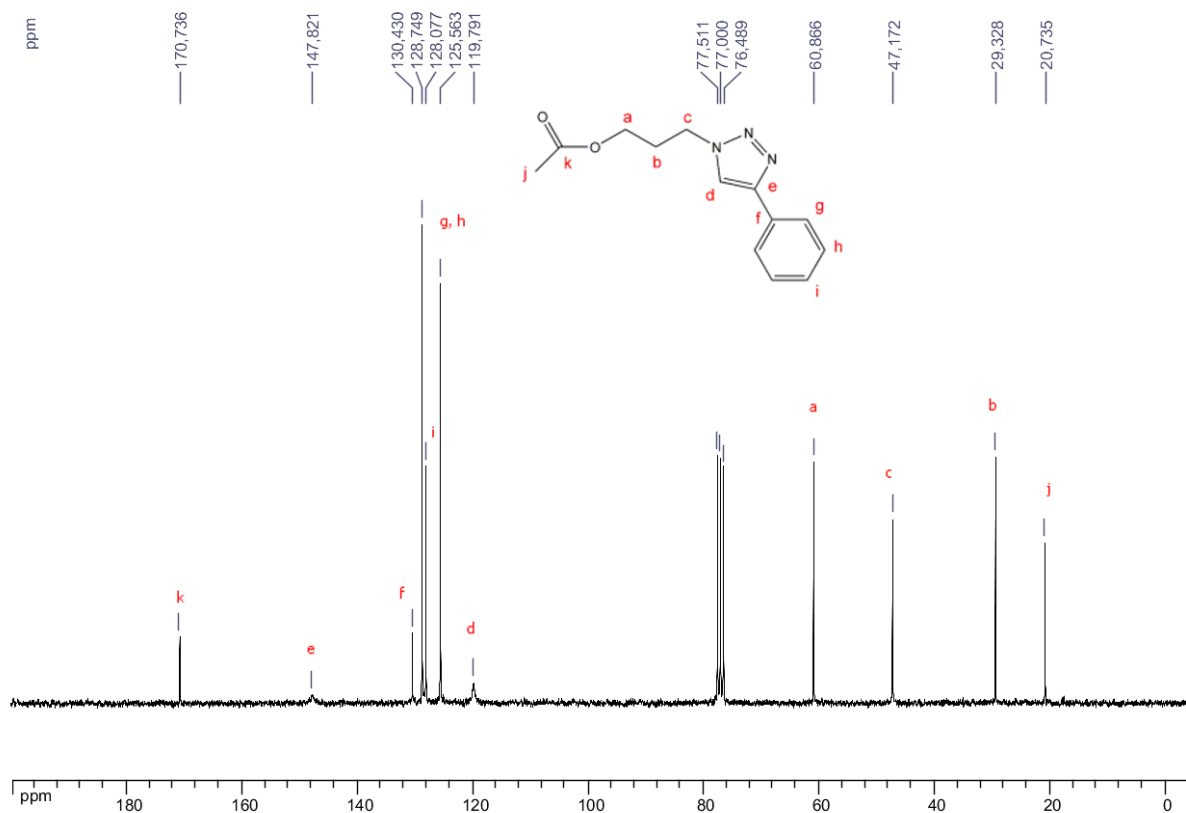
<sup>a</sup> 0.4 g PEG<sub>2000</sub> per mmol of substrate; <sup>b</sup> EtOH was used instead of CH<sub>2</sub>Cl<sub>2</sub> for the workup.

# 1-(3-Acetoxypropyl)-4-phenyl-1,2,3-triazole (5a)

$^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )

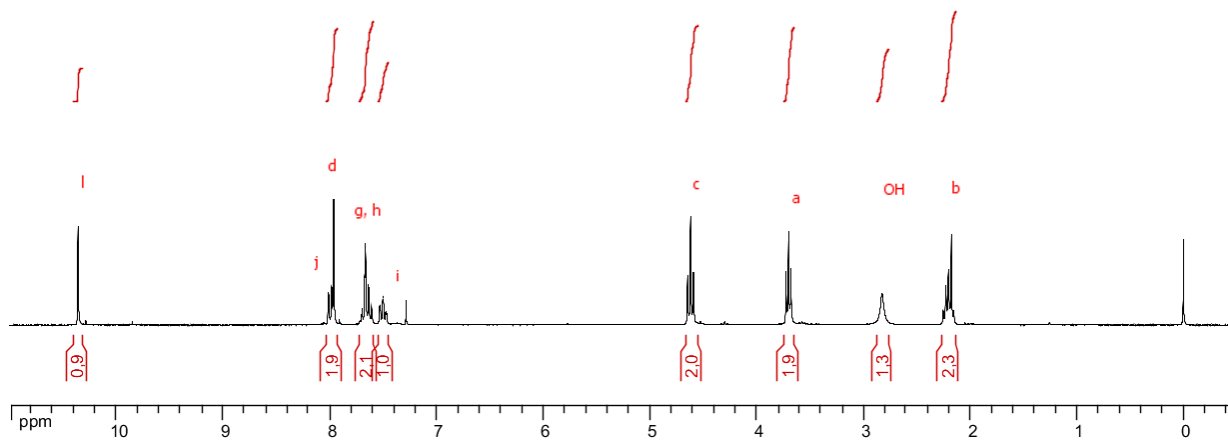
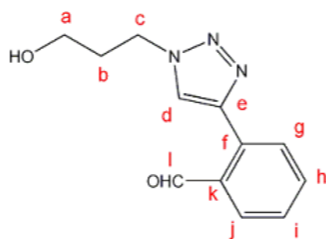


$^{13}\text{C}$  NMR (62.9 MHz,  $\text{CDCl}_3$ )

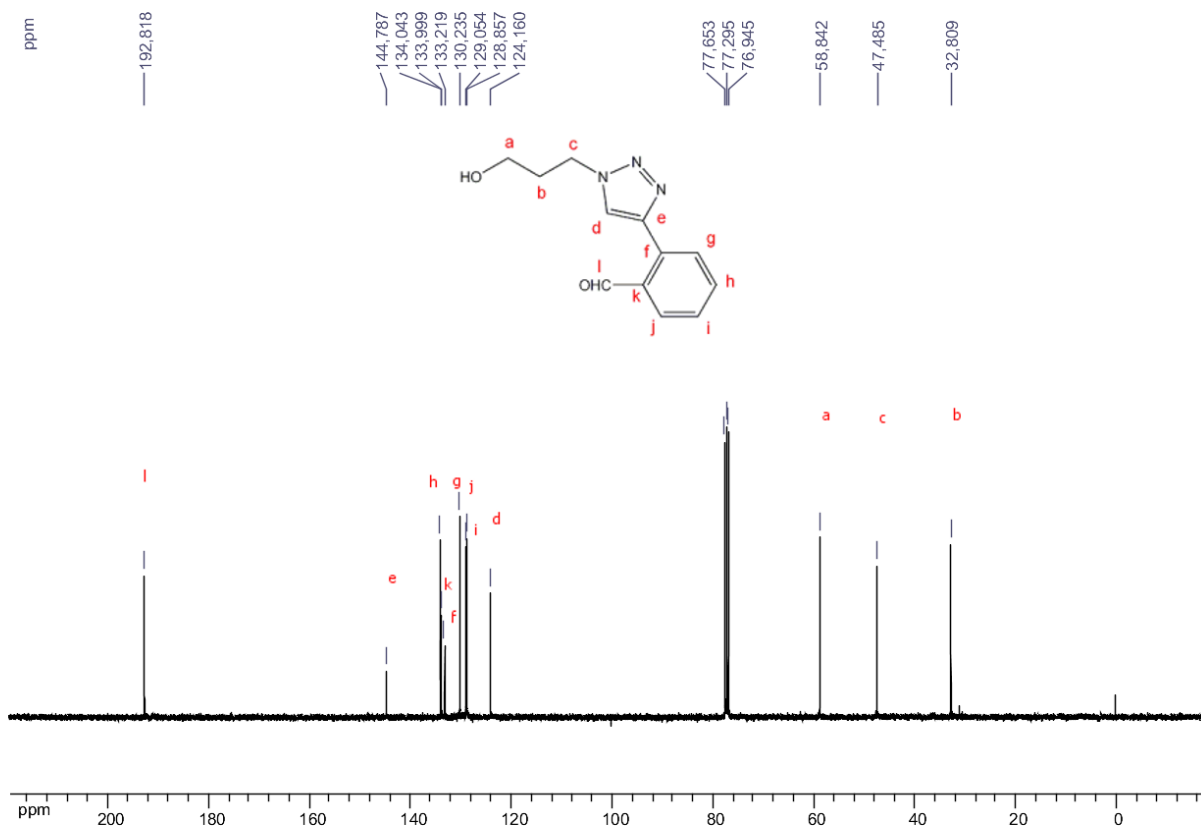


# 4-(2-Formylphenyl)-1-(3-hydroxypropyl)-1,2,3-triazole (4b)

$^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )

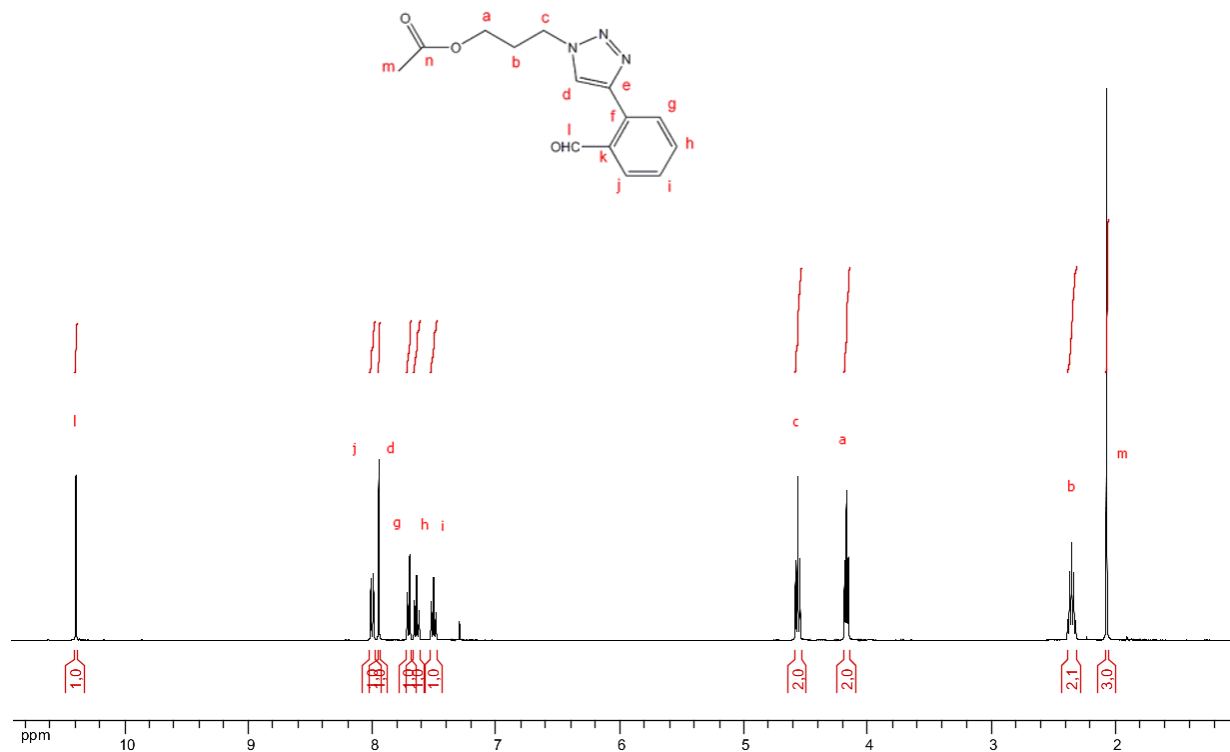


$^{13}\text{C}$  NMR (90.6 MHz,  $\text{CDCl}_3$ )

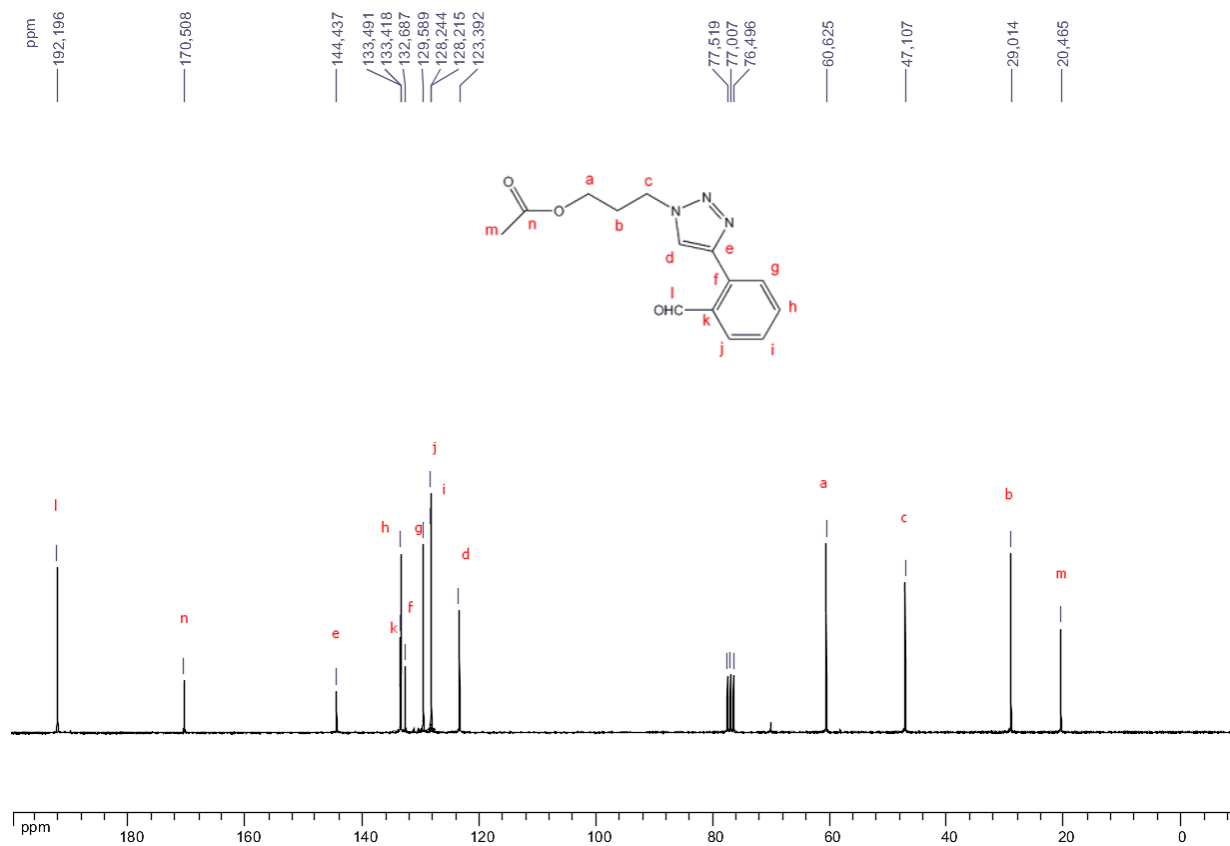


# 1-(3-Acetoxypropyl)-4-(2-formylphenyl)-1,2,3-triazole (5b)

$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )

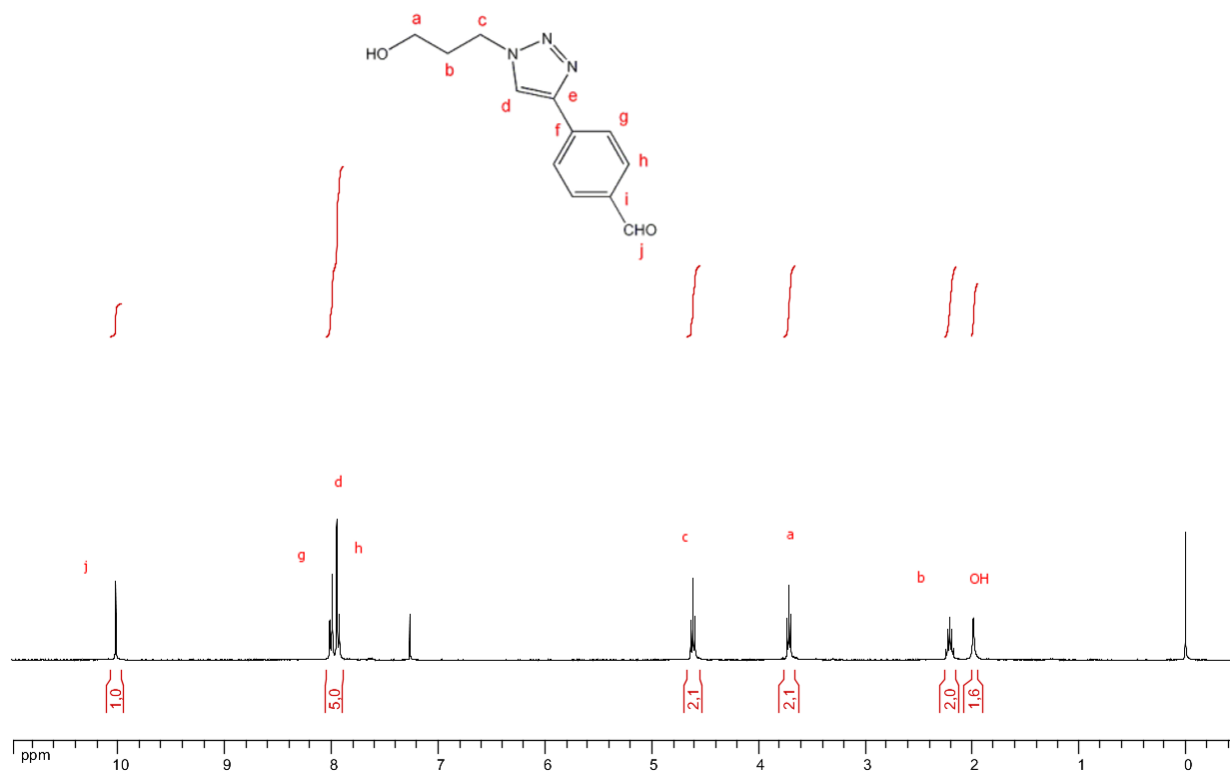


$^{13}\text{C}$  NMR (62.9 MHz,  $\text{CDCl}_3$ )

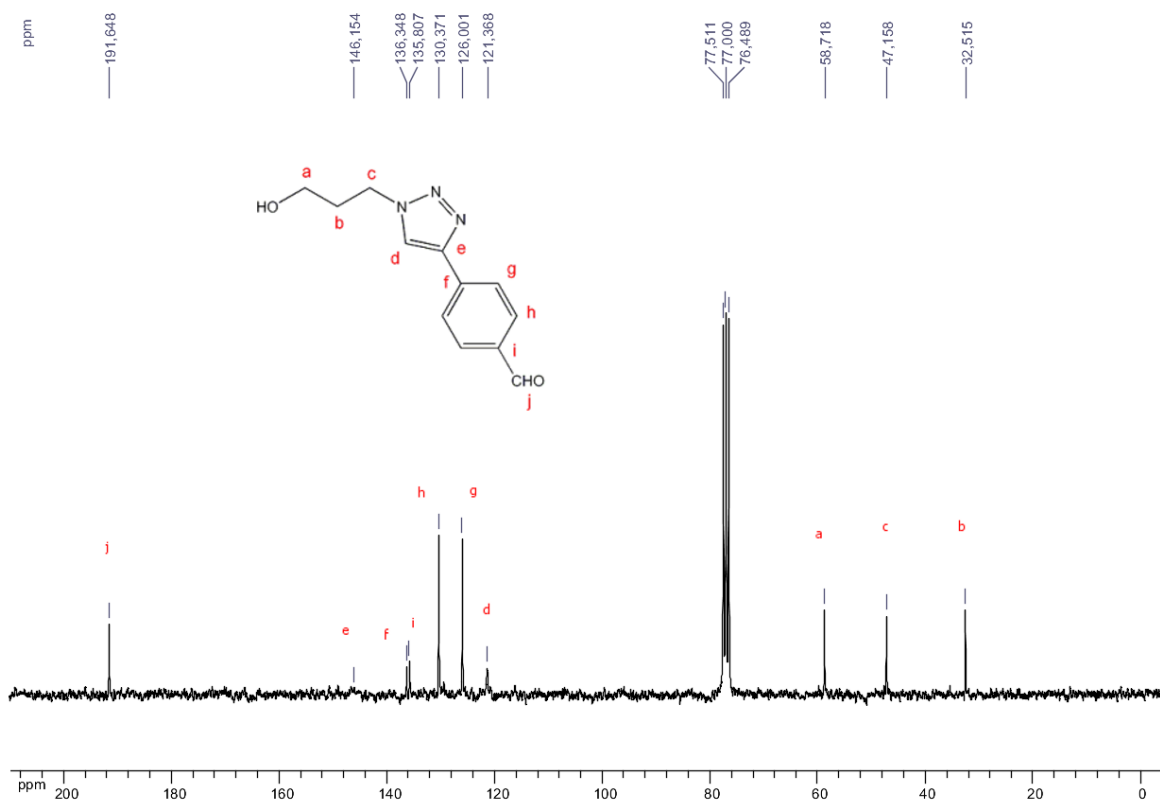


### 4-(4-Formylphenyl)-1-(3-hydroxypropyl)-1,2,3-triazole (4c)

$^1\text{H}$  NMR (360 MHz,  $\text{CDCl}_3$ )

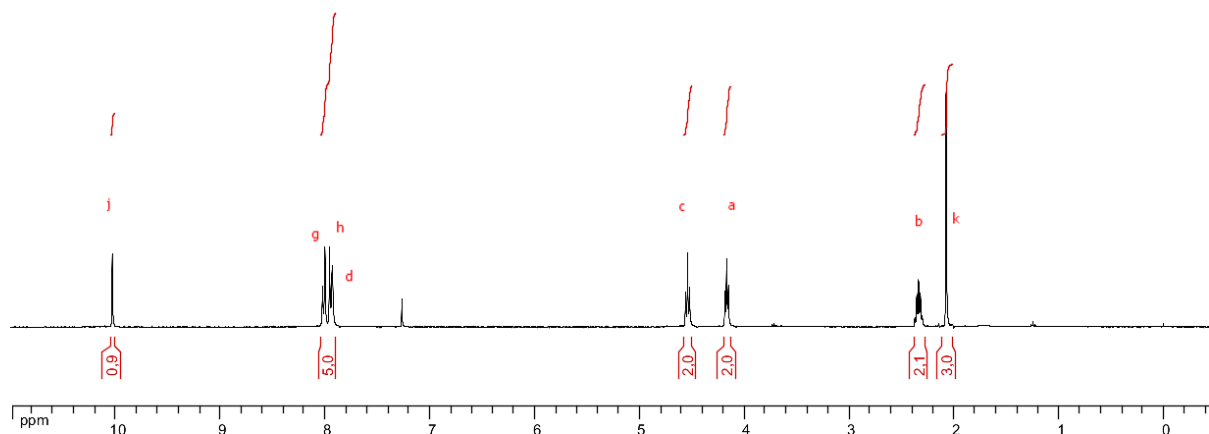
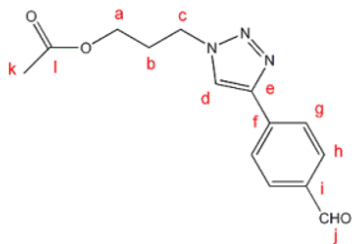


$^{13}\text{C}$  NMR (62.9 MHz,  $\text{CDCl}_3$ )

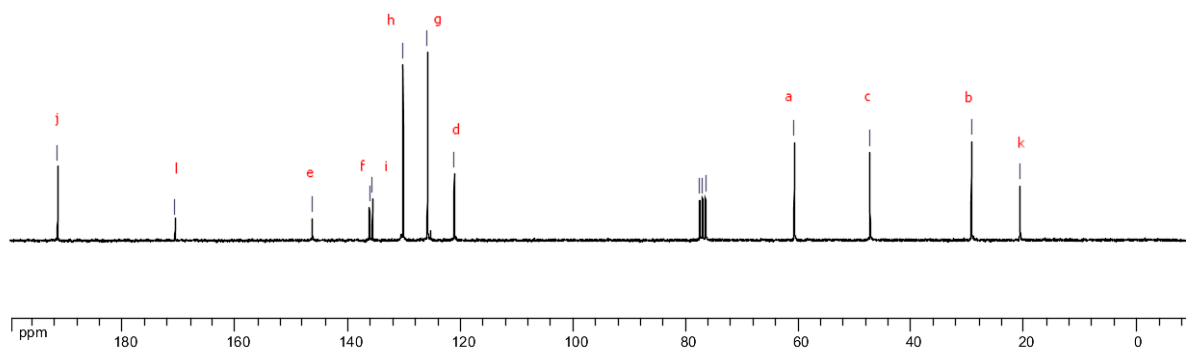
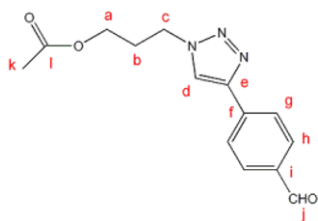


# 1-(3-Acetoxypropyl)-4-(4-formylphenyl)-1,2,3-triazole (5c)

$^1\text{H}$  NMR (360 MHz,  $\text{CDCl}_3$ )

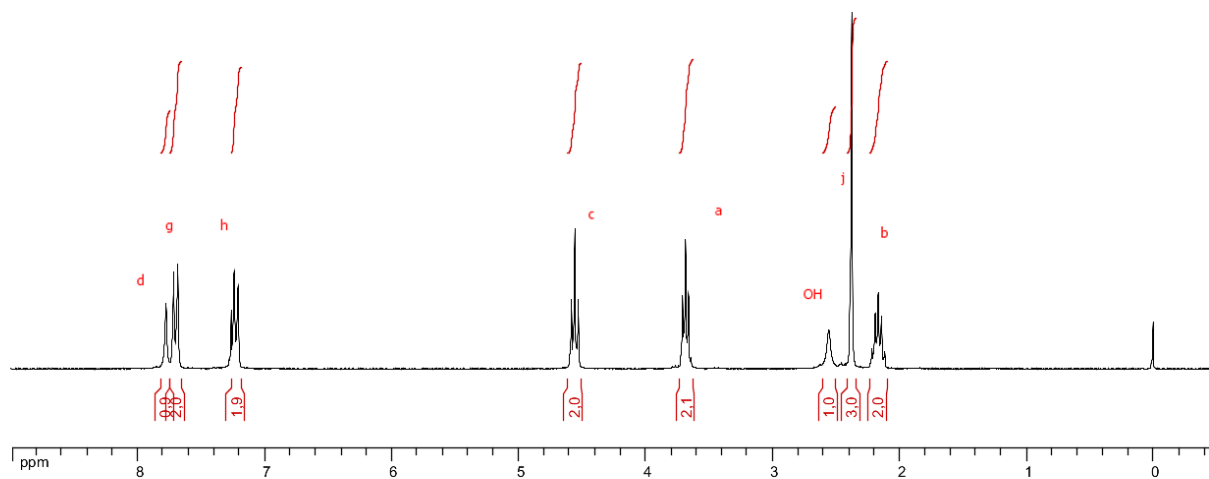
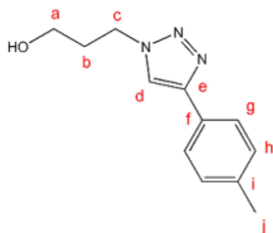


$^{13}\text{C}$  NMR (62.9 MHz,  $\text{CDCl}_3$ )

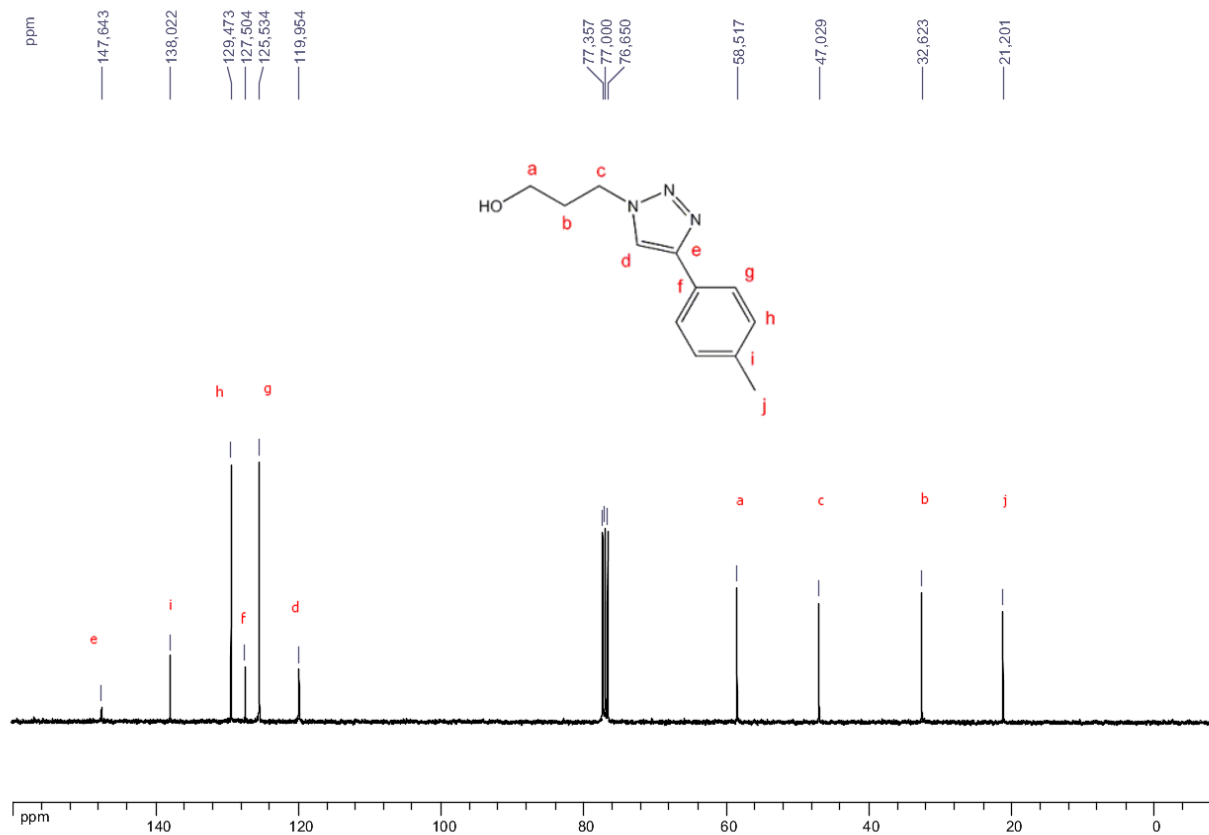


# 1-(3-Hydroxypropyl)-4-(4-methylphenyl)-1,2,3-triazole (4d)

$^1\text{H}$  NMR (250 MHz,  $\text{CDCl}_3$ )



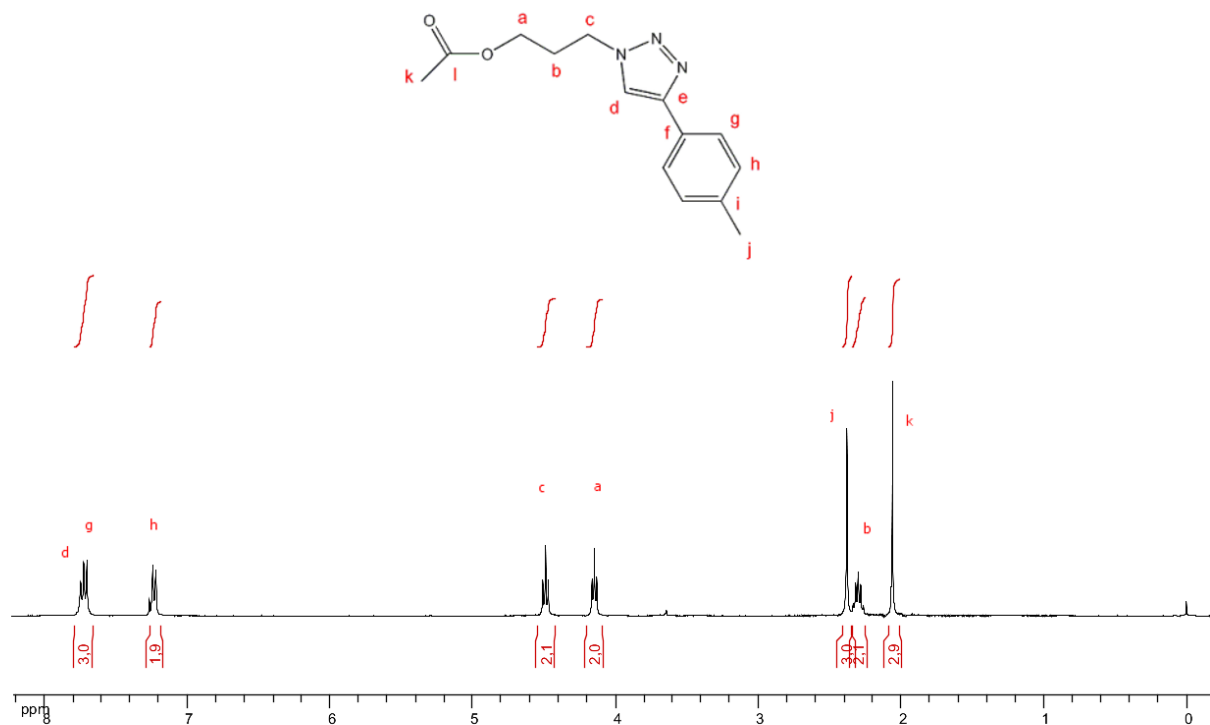
$^{13}\text{C}$  NMR (90.6 MHz,  $\text{CDCl}_3$ )



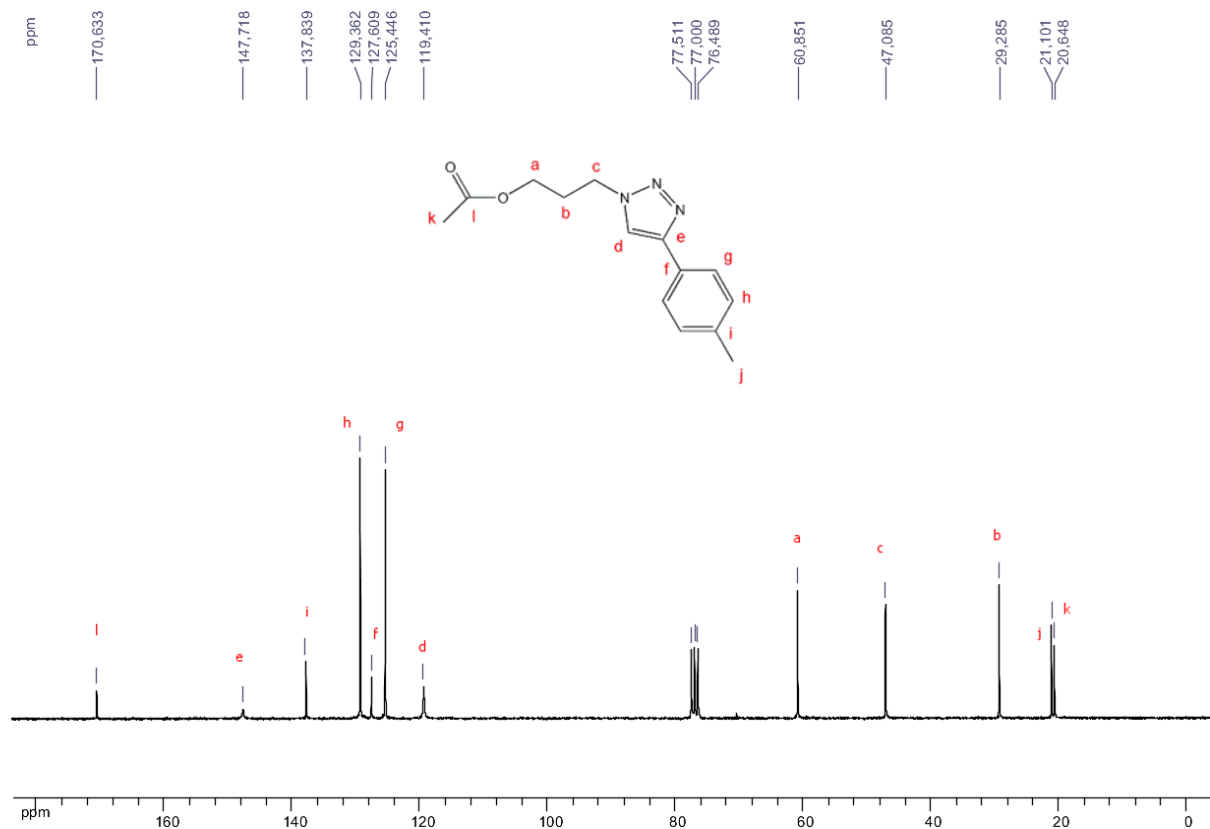


# 1-(3-Acetoxypropyl)-4-(4-methylphenyl)-1,2,3-triazole (5d)

$^1\text{H}$  NMR (360 MHz,  $\text{CDCl}_3$ )

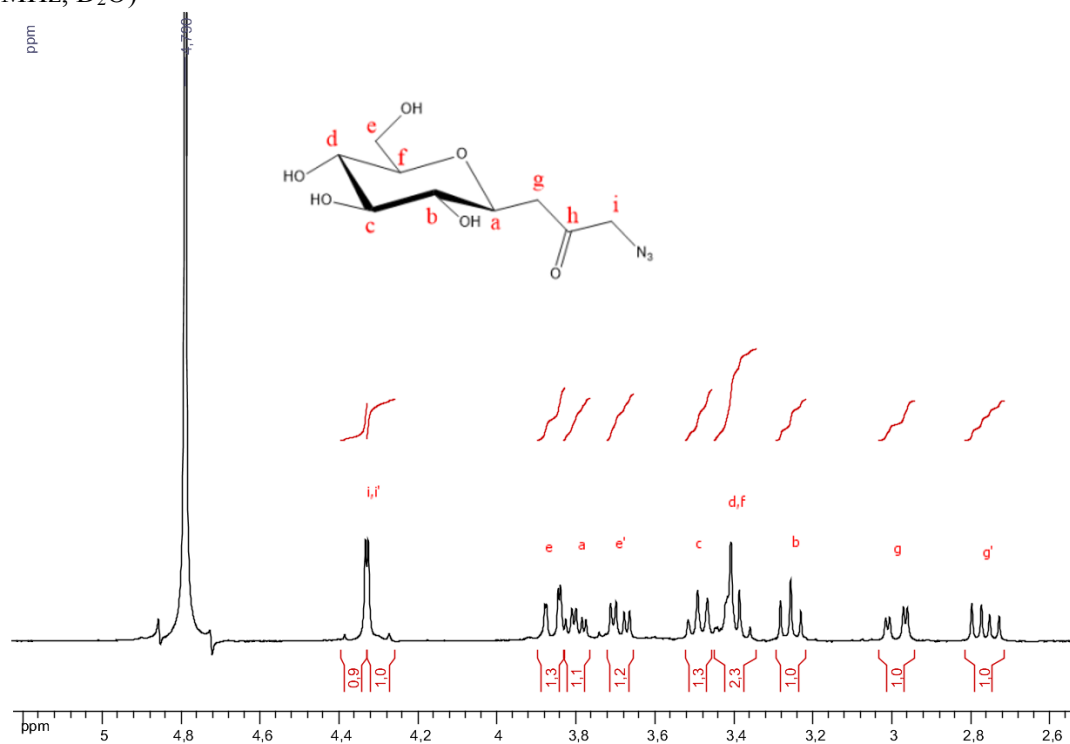


$^{13}\text{C}$  NMR (62.9 MHz,  $\text{CDCl}_3$ )

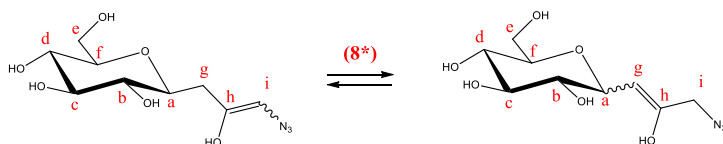
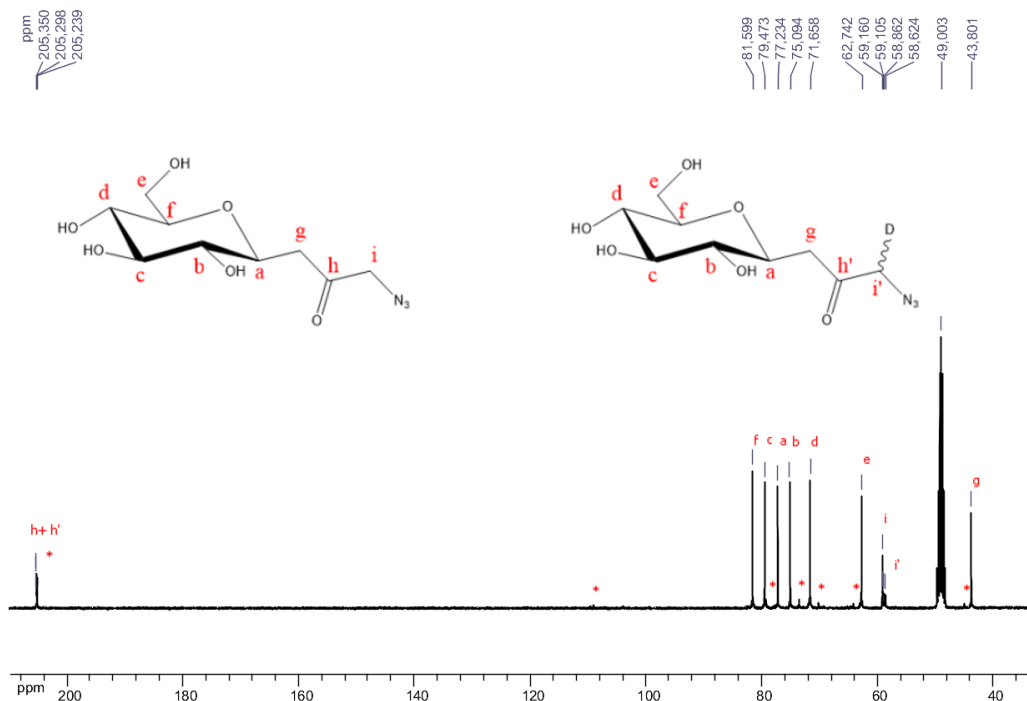


**3-Azido-1-(C- $\beta$ -D-glucopyranosyl)-propan-2-one (8)**

$^1\text{H}$  NMR (360 MHz,  $\text{D}_2\text{O}$ )



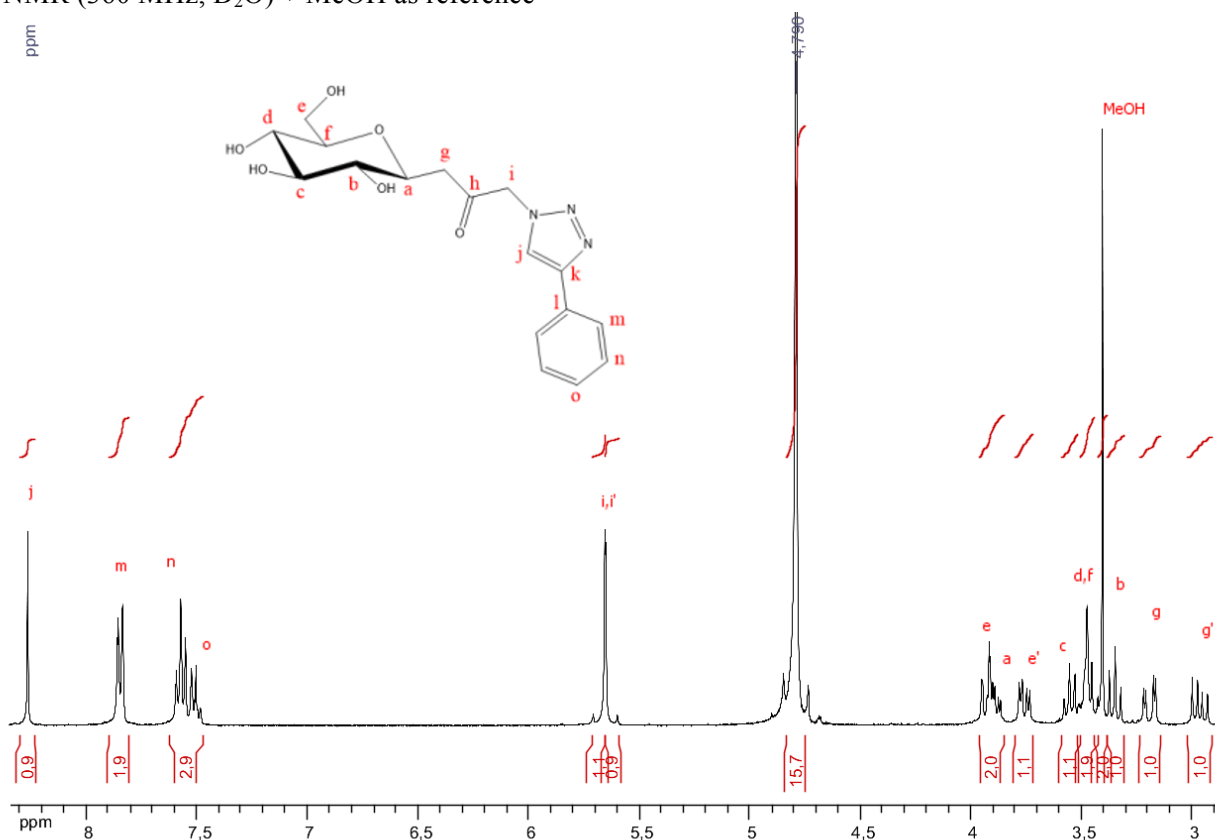
$^{13}\text{C}$  NMR (90.6 MHz,  $\text{CD}_3\text{OD}$ )



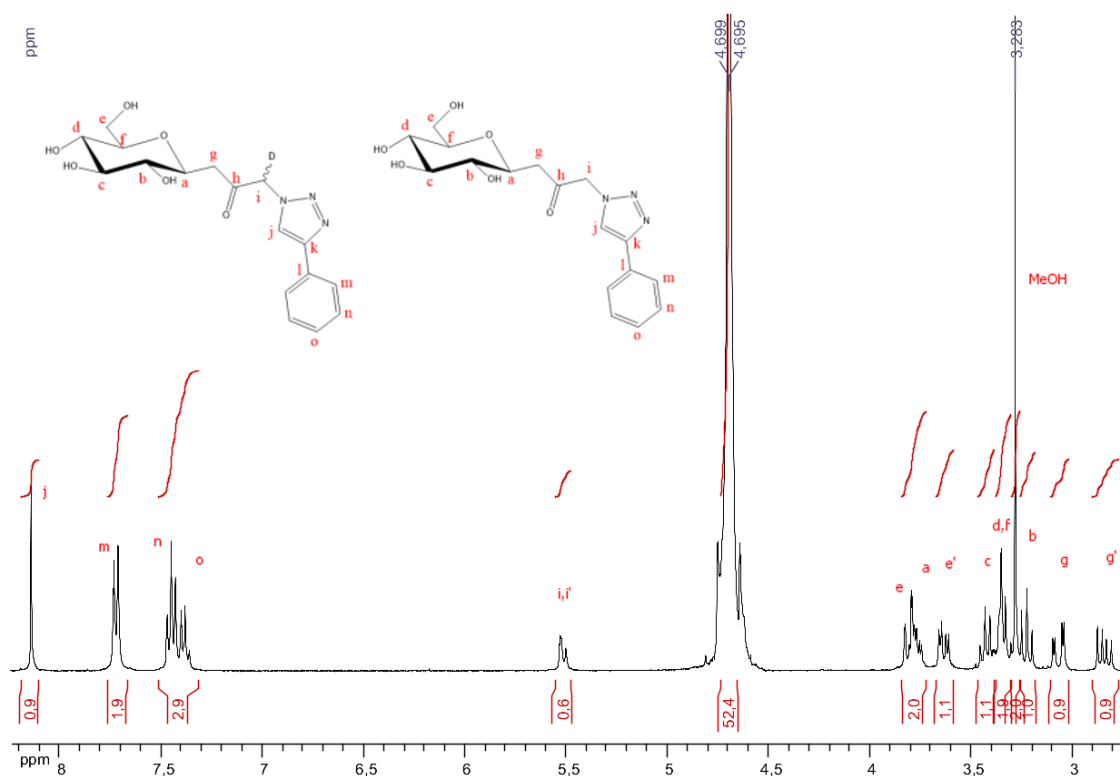
With possible trace of enol forms, indicated with an asterisk.

**1-(1-(C- $\beta$ -D-glucopyranosyl)-2-oxo-propyl)-4-phenyl-1,2,3-triazole (9)**

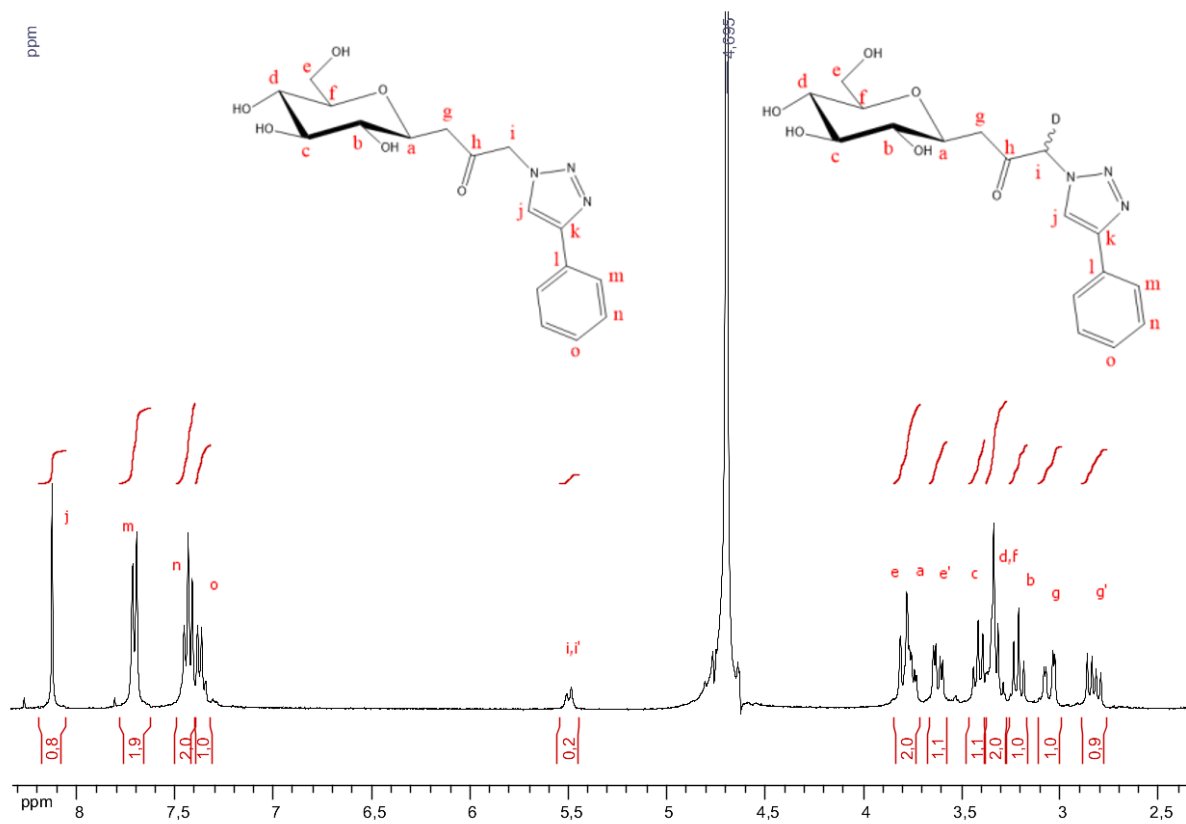
$^1\text{H}$  NMR (360 MHz,  $\text{D}_2\text{O}$ ) + MeOH as reference



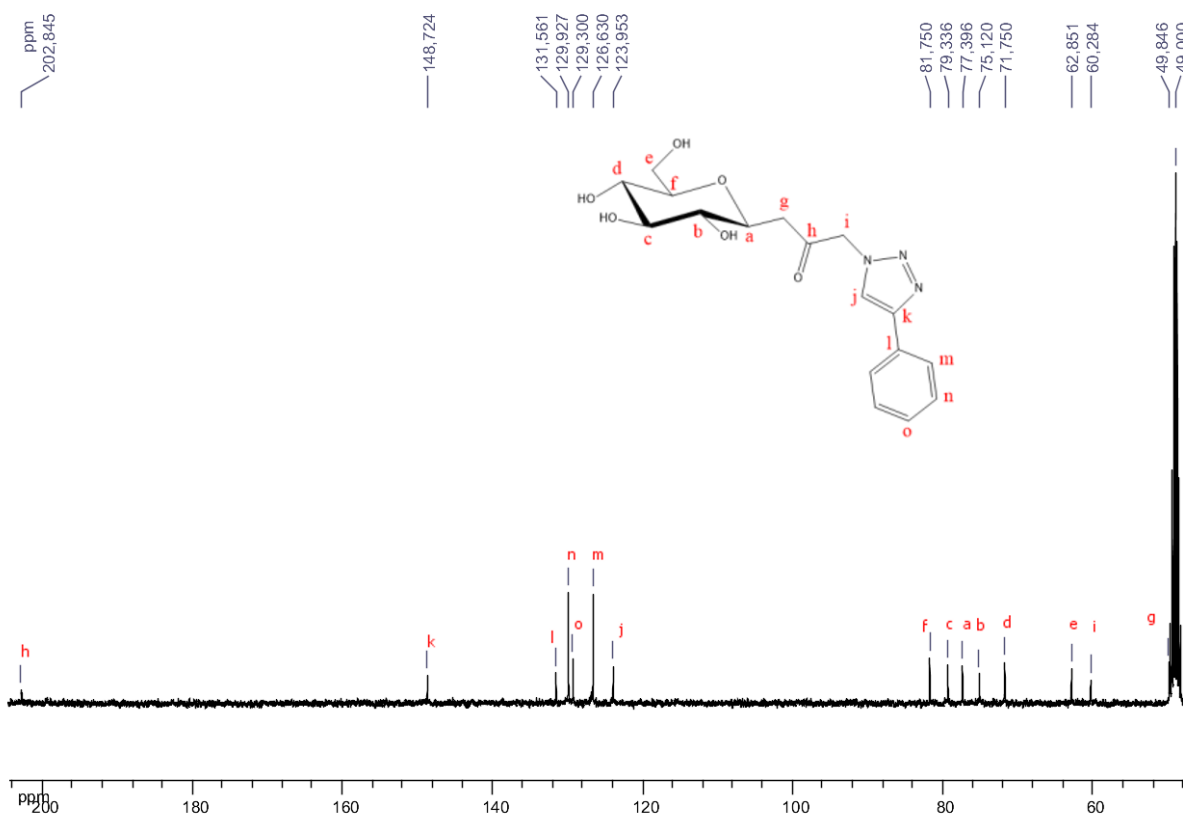
$^1\text{H}$  NMR (360 MHz,  $\text{D}_2\text{O}$ ), with exchange H/D at position i, i' (MeOH as reference)



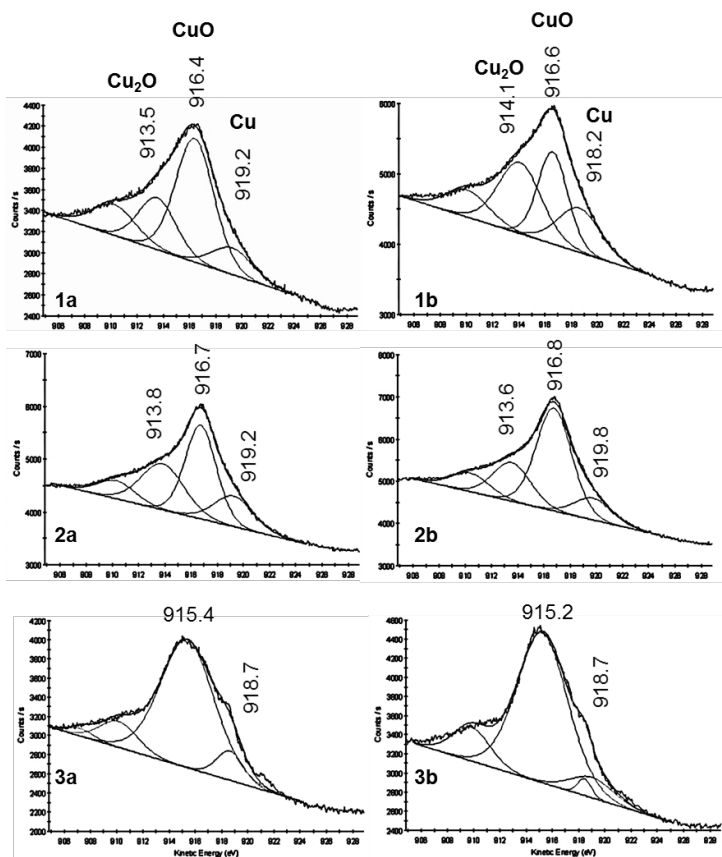
$^1\text{H}$  NMR (360 MHz,  $\text{D}_2\text{O}$ ) with exchange H/D at position i,i'



$^{13}\text{C}$  NMR (90.6 MHz,  $\text{CD}_3\text{OD}$ )

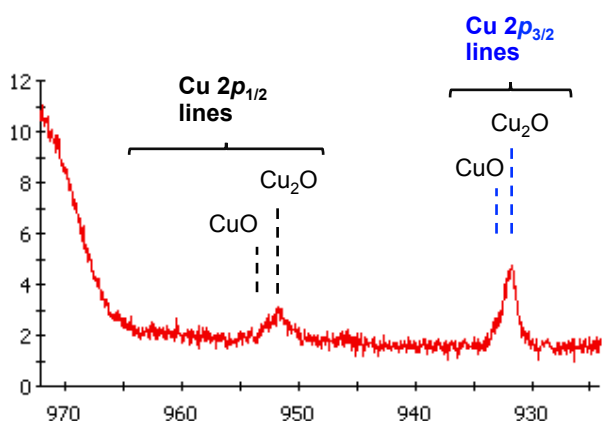


### Auger kinetic energies (Cu L<sub>3</sub>M<sub>45</sub>M<sub>45</sub>)



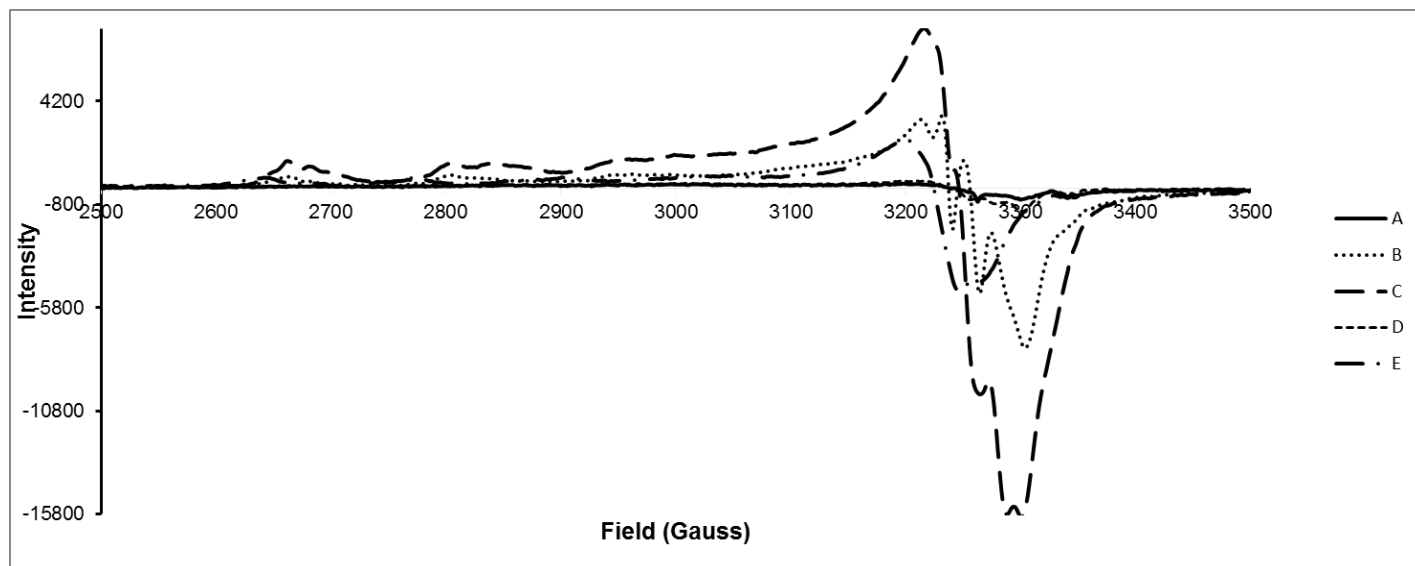
**1 a and b:** Matte and glossy surfaces of copper turnings before reaction ; **2 a and b:** Matte and glossy surfaces after reaction in the presence of azide **1** ; **3 a and b:** Matte and glossy surfaces after reaction in the presence of azide **8**.

### XPS spectra (Cu 2p<sub>3/2</sub> lines) for PEG sample after 5 recycling of PEG and CuT, [Cu] = 4502 ppm.



The ratio of CuO/Cu<sub>2</sub>O in the reaction mixture could be approximate by the XP spectra of a PEG sample after CuAAC reaction using CuT. The PEG sample with a high copper concentration (4502 ppm) gave a ratio signal/noise acceptable to identify copper species in PEG. Comparison of both signal areas shown that the ratio CuO/Cu<sub>2</sub>O was approximately 9:90, which is in agreement with the weight percentage of the Cu(II) determined by EPR spectroscopy at 5% for the same sample.

**EPR spectra of PEG samples with catalysts recovered by precipitation.**



- (A) PEG + CuNP after the first run, [Cu] = 1989 ppm
- (B) PEG + Cu T after 6 recycling in new PEG, [Cu] = 962 ppm
- (C) PEG + Cu T after 6 recycling of PEG and Cu T, [Cu] = 4502 ppm
- (D) PEG + Cu T, 6 days at 70 °C no use for CuAAC, [Cu] = 114 ppm
- (E) PEG + CuI after the first run, [Cu] = 1879 ppm.