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Synthesis and mechanism of novel fluorescent coumarindihydropyrimidinone dyads obtained by Biginelli multicomponent reaction.

Felipe Vitório^{*a,b*}, Thiago Moreira Pereira^{*a*}, Rosane Nora Castro^{*b*}, Guilherme Pereira Guedes^{*b*}, Cedric Stephan Graebin^{*a,b*} and Arthur Eugen Kummerle^{*,*a,b*}

^a Laboratório de Diversidade Molecular e Química Medicinal (LaDMol-QM, Molecular Diversity and Medicinal Chemistry Laboratory), Departament of Chemistry, Universidade Federal Rural do Rio de Janeiro, Seropédica, Rio de Janeiro, 239897-000, Brazil.

^b Programa de Pós-Gradução em Química (PPGQ), Universidade Federal Rural do Rio de Janeiro, Seropédica, Rio de Janeiro, 239897-000, Brazil.

Email: akummerle@ufrrj.br

Supporting Information

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Fig. S1 ¹H NMR (500 MHz), ¹³C NMR (125 MHz) spectra of **3a** in DMSO-*d6*.



Fig. S2 ESI spectra of 3a.



Fig. S3 ¹H NMR (500 MHz), ¹³C NMR (125 MHz) spectra of **3b** in DMSO-*d6*.



Fig. S4 ESI spectra of 3b.



Fig. S5 ¹H NMR (400 MHz), ¹³C NMR (100 MHz) spectra of **4a** in DMSO-*d6*.



Fig. S6 ESI spectra of 4a.



Fig. S7 ¹H NMR (400 MHz), ¹³C NMR (100 MHz) spectra of **4b** in DMSO-*d6*.



Fig. S8 ESI spectra of 4b.



Fig. S9 ¹H NMR (400 MHz), ¹³C NMR (100 MHz) spectra of **4c** in DMSO-*d6*.



Fig. S10 ESI spectra of 4c.



Fig. S11 1 H NMR (400 MHz), 13 C NMR (100 MHz) spectra of **4d** in DMSO-*d6*.



Fig. S12 ESI spectra of 4d.



Fig. S13 ¹H NMR (500 MHz), ¹³C NMR (100 MHz) spectra of **4e** in DMSO-*d6*.



Fig. S14 ESI spectra of 4e.



Fig. S15 ¹H NMR (500 MHz), ¹³C NMR (125 MHz) spectra of **4f** in DMSO-*d6*.



Fig. S16 ESI spectra of 4f.



Fig. S17 ¹H NMR (500 MHz), ¹³C NMR (125 MHz) spectra of **4g** in DMSO-*d6*.



Fig. S18 ESI spectra of 4g.



Fig. S19 1H NMR (500 MHz), 13C NMR (125 MHz) spectra of 4h in DMSO-d6.



Fig. S20 ESI spectra of 4h.



Fig. 21 ¹H NMR (400 MHz), ¹³C NMR (100 MHz) spectra of **4i** in DMSO-*d6*.



Fig. S22 ESI spectra of 4i.



Fig. 23 ¹H NMR (500 MHz), ¹³C NMR (100 MHz) spectra of **4j** in DMSO-*d6*.



Fig. S24 ESI spectra of 4j.



Fig. 25 ¹H NMR (500 MHz), ¹³C NMR (100 MHz) spectra of **4k** in DMSO-*d6*.



Fig. S26 ESI spectra of 4k.



Fig. S27 ¹H NMR (400 MHz), ¹³C NMR (100 MHz) spectra of **4I** in DMSO-*d6*.



Fig. S28 ESI spectra of 4l.



Fig. S29 1 H NMR (400 MHz), 13 C NMR (100 MHz) spectra of **4m** in DMSO-*d6*.



Fig. S30 ESI spectra of 4m.



Fig. S31 1 H NMR (500 MHz), 13 C NMR (125 MHz) spectra of **4n** in DMSO-*d6*.



Fig. S32 ESI spectra of 4n.



Fig. S33 1 H NMR (500 MHz), 13 C NMR (125 MHz) spectra of **40** in DMSO-*d6*.



Fig. S34 ESI spectra of 40.



Fig. S35 1 H NMR (500 MHz), 13 C NMR (125 MHz) spectra of **4p** in DMSO-*d6*.



Fig. S36 ESI spectra of 4p.



Fig. S37 1 H NMR (500 MHz), 13 C NMR (125 MHz) spectra of 4q in DMSO-d6.



Fig. S38 ESI spectra of 4q.



Fig. S39 1 H NMR (500 MHz), 13 C NMR (125 MHz) spectra of **4r** in DMSO-*d6*.



Fig. S40 ESI spectra of 4r.



Fig. S41 1 H NMR (500 MHz), 13 C NMR(125 MHz) spectra of **5** in DMSO-d6.



Fig. S42 ESI spectra of 5.

HPLC data analysis



PDA Ch1 370nm 4nm			
Name	Ret. Time	Area %	Height %
4a	4.834	59.303	60.119
3a	5.660	28.032	29.618
nd	6.578	1.338	1.428
nd	8.007	3.399	3.462
nd	11.005	2.075	1.775
5	12.847	5.852	3.598
		100.000	100.000



PDA Ch1 370nm 4nm			
Name	Ret. Time	Area %	Height %
nd	4.438	1.602	3.106
4a	4.864	62.448	68.499
3a	5.695	18.934	17.237
nd	7.653	1.937	1.495
nd	9.710	2.315	1.889
5	12.953	12.763	7.774
		100.000	100.000



DA Ch1 370nm 4nm			
Name	Ret. Time	Area %	Height %
nd	4.412	4.922	7.927
4a	4.822	19.095	19.651
3a	5.649	65.719	66.310
nd	6.488	0.796	0.731
5	12.798	9.467	5.382
		100.000	100.000



DA Ch1 370nm 4nm			
Name	Ret. Time	Area %	Height %
4a	4.843	86.818	86.924
3a	5.696	4.098	4.432
nd	5.995	6.416	5.692
nd	6.561	0.681	1.077
nd	7.354	1.988	1.875
		100.000	100.000

Fig. S43 Comparison of reaction catalyzed conditions. A – H_2SO_4 (25 µL); B – H_2SO_4 (12.5 µL); C – HCl (25 µL); D – HCl (12.5 µL). nd – not determined product.



Fig. S44 Reaction conditions: non-catalyzed. A - 2 hours time reaction; B - 4 hours time reaction; C - 10 hours time reaction; nd - not determinate.



Fig. S45 Reaction conditions: acetic acid as catalyst. A – 2 hours time reaction; B – 4 hours time reaction; C – 10 hours time reaction and D – 24 hours time reaction; nd – not determinated.



Fig. S46 Reaction conditions: Lewis acid as catalyst (CaF₂): A - 2 hours time reaction; B - 4 hours time reaction; C - 10 hours time reaction and D - 24 hours time reaction; nd - not determinated.



Fig. S47 Reaction conditions: HCl as catalyst (12.5μL): A – 2 hours time reaction; B – 4 hours time reaction; C – 10 hours time reaction and D – 24 hours time reaction; nd – not determinated.

X-ray diffraction

Single crystal X-ray diffraction data for compound **4c** were collected on an Bruker D8 Venture diffractometer at room temperature, using graphite monochromatic MoK α radiation ($\lambda = 0.71069$ Å). Data collection and cell refinement were performed with Bruker Instrument Service v4.2.2 and APEX2 [ⁱ], respectively. Data reduction was carried out using SAINT [ⁱⁱ]. Empirical multiscan absorption correction using equivalent reflections was performed with the SADABS program [ⁱⁱⁱ]. The structure solutions and full-matrix least-squares refinements based on F^2 were performed with the SHELXS-97 and SHELXL-97 program packages [iv]. All atoms except hydrogen were refined anisotropically. Hydrogen atoms were treated by a mixture of independent and constrained refinement. The structure was drawn by Mercury program [v]. Details of data collection and refinement are listed in Table S1.

Empirical formula	$C_{49}H_{36}N_4O_{14}\\$	
Formula weight	904.82	
Temperature	293(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P -1	
Unit cell dimensions	a = 9.2503(5) Å	$\alpha = 76.141(2)^{\circ}$
	b = 10.6790(6) Å	β= 77.031(2)°
	c = 13.4211(6) Å	$\gamma = 64.664(2)^{\circ}$
Volume	1152.10(10) $Å^3$	
Z	1	
Density (calculated)	1.304 Mg/m ³	
Absorption coefficient	0.10 mm ⁻¹	
F(000)	470	
Crystal size	0.28 x 0.17 x 0.05 mm	3

Table S1: Summary of crystal data and structure refinement of compound 4c.

Theta range for data collection	2.1 to 25.0°
Index ranges	-10<=h<=10,
	-12<=k<=12,
	-15<=l<=15
Reflections collected	32147
Independent reflections	4053 [R(int) = 0.075]
Completeness to theta = 25.06°	100 %
Max. and min. transmission	0.980 and 0.995
Refinement method	Full-matrix least-squares on F^2
Data / restraints / parameters	4048 / 0 / 312
Goodness-of-fit on F^2	1.06
Final R indices [I>2sigma(I)]	$R_1 = 0.062, wR_2 = 0.199$
R indices (all data)	$R_1 = 0.099, wR_2 = 0.171$
Largest diff. peak and hole	0.52 and -0.23 e.Å ⁻³



Fig. S48 Molecular structure of compound **4c**. Thermal ellipsoids are drawn at 40 % of probability. Crystallization solvent molecule was omitted for sake of clarity.

^{[&}lt;sup>i</sup>] Bruker (2007). APEX2 v2014.5-0. Bruker AXS Inc., Madison, Wisconsin, USA.

^{[&}lt;sup>ii</sup>] Bruker (2013). SAINT v8.34A. Bruker AXS Inc., Madison, Wisconsin, USA.

^{[&}lt;sup>iii</sup>] Sheldrick, G.M. SADABS, Program for Empirical Absorption Correction of Area Detector Data, University of Göttingen, Germany, 1996.

^{[&}lt;sup>iv</sup>] Sheldrick, G.M. Acta Cryst. 2008, A64, 112-122.

[^v]Macrae, C. F., Edgington, P. R., McCabe, P., Pidcock, E., Shields, G. P., Taylor, R., Towler, M. & van de Streek, J., *J. Appl. Cryst.* **2006**, 39, 453-457.