

Electronic Supplementary Information (ESI)

General synthesis of carbon nanocages and their adsorption of toxic compounds from cigarette smoke

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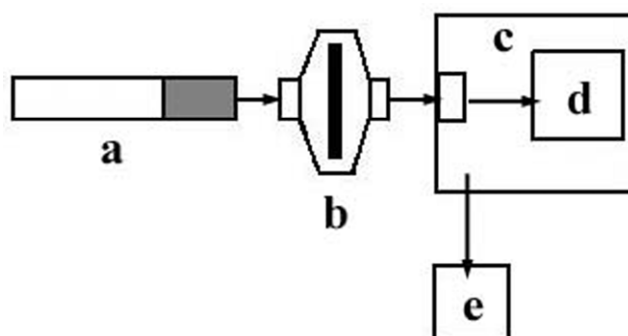


Fig. S1 Schematic of experimental system. (a) cigarette, (b) Cambridge filter pad, (c) smoking machine, (d) gas collector, (e) computer control system.

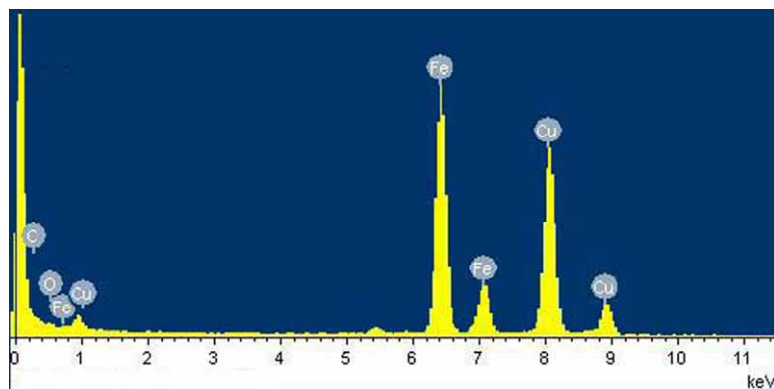


Fig. S2 EDS from the $\text{Fe}_3\text{O}_4/\text{C}$ composites. The result of EDS indicates that the chemical compositions of the samples are C, Fe and O. The peaks of Cu signals in the spectrum originate from the copper grid that supports the specimen.

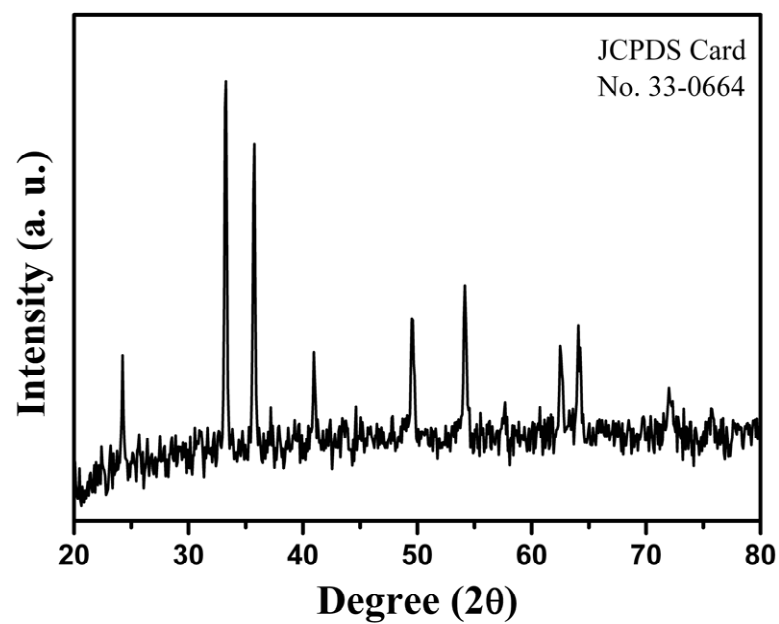


Fig. S3 XRD pattern of the Fe₃O₄/CNCs calcined at 600 °C for 10 h.

Table S1 Comparison of the different CNCs in previous reports.

References	Synthesis method	Temperature (°C)	Length of the side (nm)	Wall thickness (nm)
[9]	In suit template approach	600~900	30~50	~5
[11]	Arc evaporation method		20~100	~10
[13]	Template (gold) method		20~50	~5
[16]	Solvothermal reaction	500~700	10~100	~5
This work	Pyrolytic process	550	200~350	10~15

[9] J. N. Wang, L. Zhang, J. J. Niu, F. Yu, Z. M. Sheng and Y. Z. Zhao, *Chem. Mater.*, 2007, **19**, 453.

[11] Y. Saito and T. Matsumoto, *Nature*, 1998, **392**, 237.

[13] R. Y. Zhang, M. Hummelgard and H. Olin, *Carbon*, 2010, **48**, 424.

[16] J. Liu, L. Xu, W. Zhang, W. Lin, X. Chen, Z. Wang and Y. Qian, *J. Phys. Chem. B*, 2004, **108**, 20090.

Table S2 Fe/C ratios obtained by adjust the quantity of the alcohols.

Alcohols	Volume (ml)	Fe ₃ O ₄ /C (g)	Fe ₂ O ₃ (g)	Fe ₃ O ₄ (g)	Fe ₃ O ₄ (%)	C (g)	C (%)	Fe/C (mol)
Ethanol	12	0.1677	0.1471	0.1422	84.8	0.0255	15.2	0.87
	14	0.1478	0.1234	0.1193	80.7	0.0285	19.3	0.65
	16	0.1673	0.1345	0.1300	77.7	0.0373	22.3	0.54
EG	12	0.1503	0.1270	0.1228	81.7	0.0275	18.3	0.69
	14	0.1584	0.1285	0.1242	78.4	0.0342	21.6	0.56
	16	0.1549	0.1182	0.1143	73.8	0.0406	26.2	0.44
PEG-400	12	0.1500	0.1322	0.1278	85.2	0.0222	14.8	0.89
	14	0.1517	0.1244	0.1203	79.3	0.0314	20.7	0.59
	16	0.1634	0.1297	0.1254	76.7	0.0380	23.3	0.51