

Supporting Information

Site-specific Growth of Au Particles on ZnO Nanopyramids under Ultraviolet Illumination

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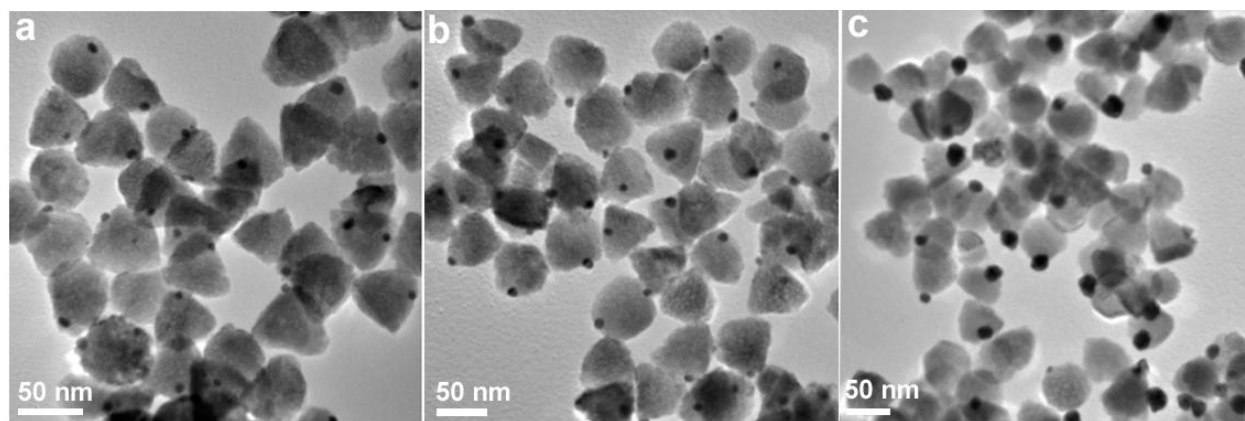


Figure S1. TEM images of Au/ZnO nanocomposites with Au nanoparticles in different sizes: 8 nm (a), 10 nm (b) and 20 nm (c).

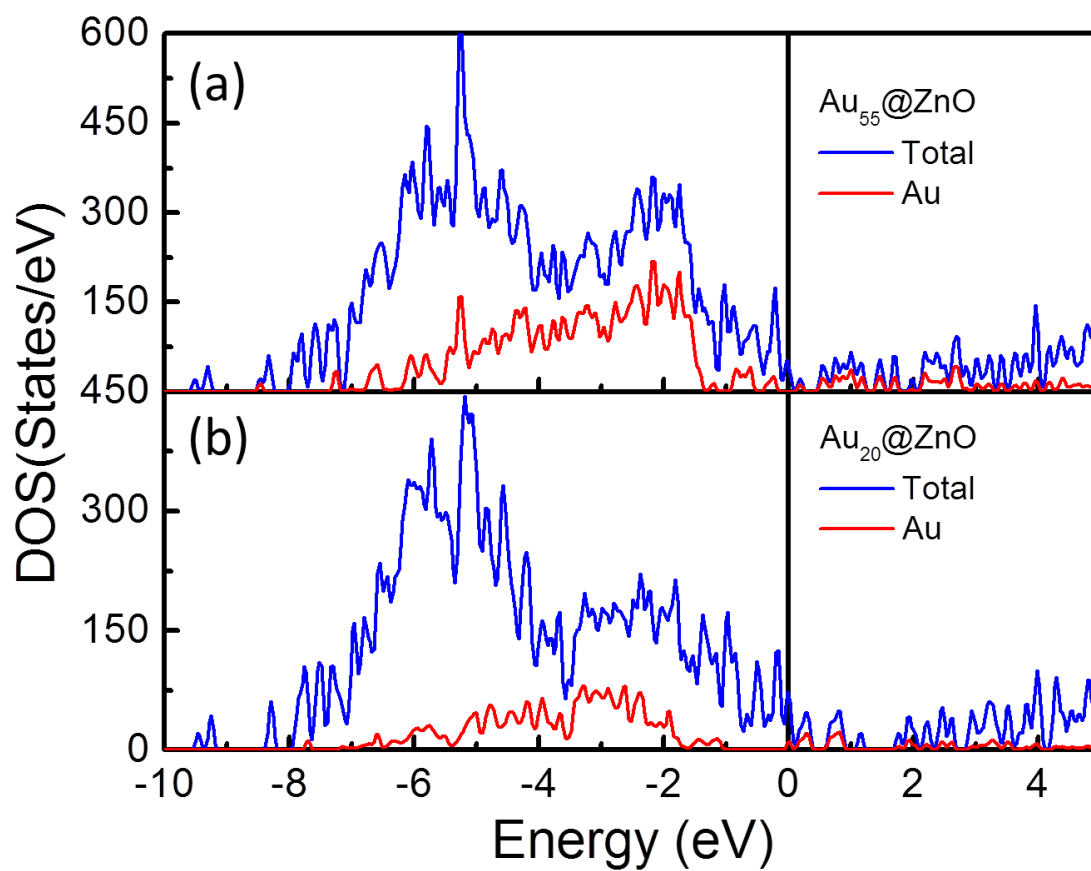


Figure S2. Projected density of states of Au₅₅ (a) and Au₂₀ (b) deposited on ZnO (001) surface.

First-principles Calculations Details

Extensive first-principles based calculations were performed with Vienna ab initio simulation package (VASP).¹⁻⁴ The wave functions are expressed by plane waves with a cutoff energy 400.0 eV. Ionic potentials are represented by PAW pseudopotential^{5,6} with the Perdew-Burke-Ernzerhof (PBE) generalized gradient approximation (GGA) correction.⁷ The Brillouin-zone integrations were performed using Monkhorst-Pack (MP) grids. K-point sampling convergence was achieved with (7×7×4) MP grid for the bulk, (7×7×1) MP grid for the surfaces, and only Γ point sampling for nanorods, nanopylramids and the nanocomposites. A denser (11×11×7) MP grid was also tested for bulk calculation, and confirmed the convergence. All the atomic geometries were fully relaxed until the forces on all the atoms were less than 0.02 eV/Å, except for the nanocomposites.

The calculated lattice and internal parameters ($a_0 = 3.256$ Å, $c_0 = 5.240$ Å) of ZnO are in good agreement with experimental results ($a_0 = 3.25$ Å, $c_0 = 5.21$ Å).⁸⁻¹⁰ Thus, the Zn-O bond length is 2.00 Å. And, the DFT-PBE fundamental band gap at Γ point, E_g is 0.78 eV, though different from the experimental value of 3.37 eV which is due to the well-known problem of GGA functional, is still in line with the previous theoretical results.¹¹

The w-ZnO nanorods grown along c -axis with finite length were truncated to have both (001)-Zn and (00-1)-O termination. To avoid the possible influence of size effect and the interaction between surface terminations on the electronic structure analysis, these nanorods were constructed to contain 4, 8, and 12 Zn-O double atomic layers (equivalent to rod dimension along c -axis of 0.9 nm, 1.9 nm and 2.9 nm), together with vacuum layers of at least 30 Å. The resulting rectangular cells for calculation are about 4 nm, 5 nm and 6 nm in dimensions.

The w-ZnO nanopyramide of 4 Zn-O double atomic layers thickness was optimized in a $25.0 \text{ \AA} \times 25.1 \text{ \AA} \times 25.2 \text{ \AA}$ rectangle cell, with dipole corrections in all three dimensions. The calculated band gap is 1.81 eV, which is higher than the bulk gap due to the well-known confinement effect.

Au₂₀ and Au₅₅ clusters were selected to as models of Au nanoparticles to study the interfacial interaction between the particle and the (00±1) surface, as their electronic structures are well studied experimentally and theoretically, and were extensively used to study the electronic structure of nanoparticle-substrate interface.¹²⁻¹⁷ The interfaces were constructed by putting the compact surface of Au particle onto (001)-Zn and (00-1)-O surfaces the ZnO nanosubstrate. The calculation was performed in a $25.0 \text{ \AA} \times 25.1 \text{ \AA} \times 25.2 \text{ \AA}$ rectangular cell. The top 4 layers of the ZnO substrate surface and the Au particles are fully optimized until the forces on all the atoms were less than 0.02 eV/Å.

References

- (1) Kresse, G.; Hafner, J. *Phys. Rev. B* **1993**, *47*, 558.
- (2) Kresse, G.; Hafner, J. *Phys. Rev. B* **1994**, *49*, 14251.
- (3) Kresse, G.; Furthmüller, J. *Comput. Mater. Sci.* **1996**, *6*, 15.
- (4) Kresse, G.; Furthmüller, J. *Phys. Rev. B* **1996**, *54*, 11169.
- (5) Blöchl, P. E. *Phys. Rev. B* **1994**, *50*, 17953.
- (6) Kresse, G.; Joubert, D. *Phys. Rev. B* **1999**, *59*, 1758.
- (7) Perdew, J. P.; Burke, K.; Ernzerhof, M. *Phys. Rev. Lett.* **1996**, *77*, 3865.
- (8) Abrahams, S. C.; Bernstein, J. *Acta Crystallographica Section B-Structural Crystallography and Crystal Chemistry* **1969**, *B 25*, 1233.
- (9) Freeman, C. L.; Claeysens, F.; Allan, N. L.; Harding, J. H. *Phys. Rev. Lett.* **2006**, *96*, 066102.
- (10) Zhou, Z.; Li, Y.; Liu, L.; Chen, Y.; Zhang, S. B.; Chen, Z. *J. Phys. Chem. C* **2008**, *112*, 13926.
- (11) Da Silva, J. L. F.; Walsh, A.; Wei, S. H. *Phys. Rev. B* **2009**, *80*, 214118.
- (12) Li, J.; Li, X.; Zhai, H. J.; Wang, L. S. *Science* **2003**, *299*, 864.
- (13) Boyen, H. G.; Kastle, G.; Weigl, F.; Koslowski, B.; Dietrich, C.; Ziemann, P.; Spatz, J. P.; Riethmüller, S.; Hartmann, C.; Moller, M.; Schmid, G.; Garnier, M. G.; Oelhafen, P. *Science* **2002**, *297*, 1533.
- (14) Molina, L. M.; Hammer, B. *J. Catal.* **2005**, *233*, 399.

- (15) Gruene, P.; Rayner, D. M.; Redlich, B.; van der Meer, A. F. G.; Lyon, J. T.; Meijer, G.; Fielicke, A. *Science* **2008**, *321*, 674.
- (16) Yoon, B.; Landman, U. *Phys. Rev. Lett.* **2008**, *100*, 056102.
- (17) Schmid, G. *Chem. Soc. Rev.* **2008**, *37*, 1909.