

Supporting Materials

6.5% Efficient Perovskite Quantum-Dot-Sensitized Solar Cell

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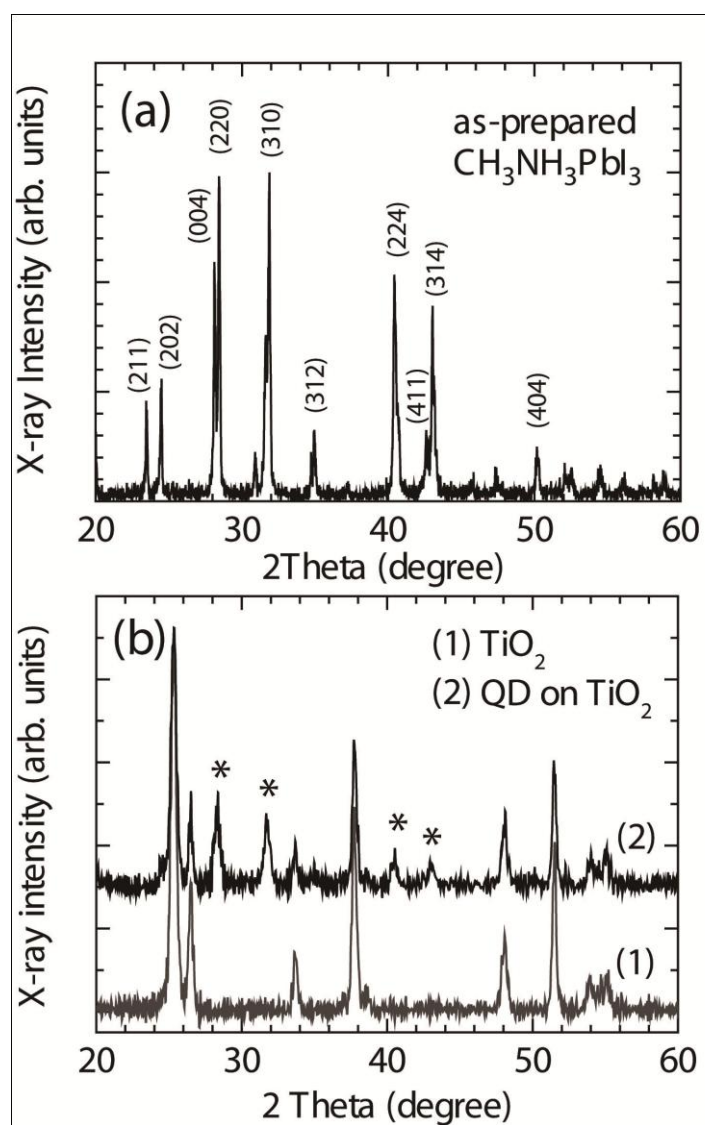


Figure S1. X-ray diffraction patterns for (a) as-prepared $\text{CH}_3\text{NH}_3\text{PbI}_3$ powder (not deposited on TiO_2 surface) and (b) nanoparticulate TiO_2 film on FTO glass with and without deposition of $\text{CH}_3\text{NH}_3\text{PbI}_3$ on TiO_2 surface. The powder in (a) was obtained by drying the γ -butyrolactone solution contained $\text{CH}_3\text{NH}_3\text{I}$ and PbI_2 .

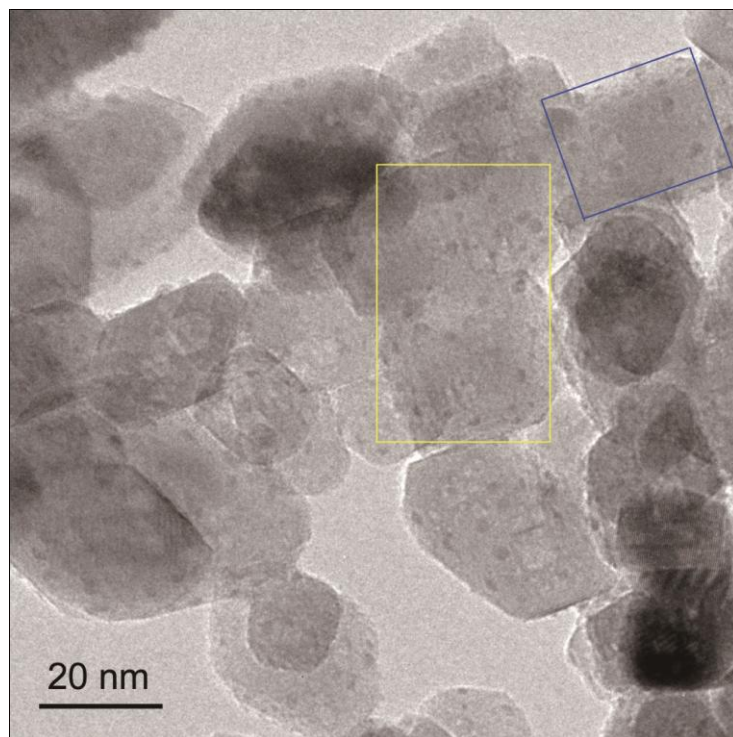


Figure S2. Number of perovskite QDs in a Gaussian box (yellow rectangle) with dimension of $28.07 \text{ nm} \times 44.91 \text{ nm}$ was estimated to be $40/1260.6 \text{ nm}^2$, corresponding to $3.2 \times 10^4/\mu\text{m}^2$, and $3.8 \times 10^4/\mu\text{m}^2$ for a Gaussian box (blue rectangle) with dimension of $19.6 \text{ nm} \times 20 \text{ nm}$. Sample for TEM image was prepared using a 40.26 wt% perovskite precursor solution.

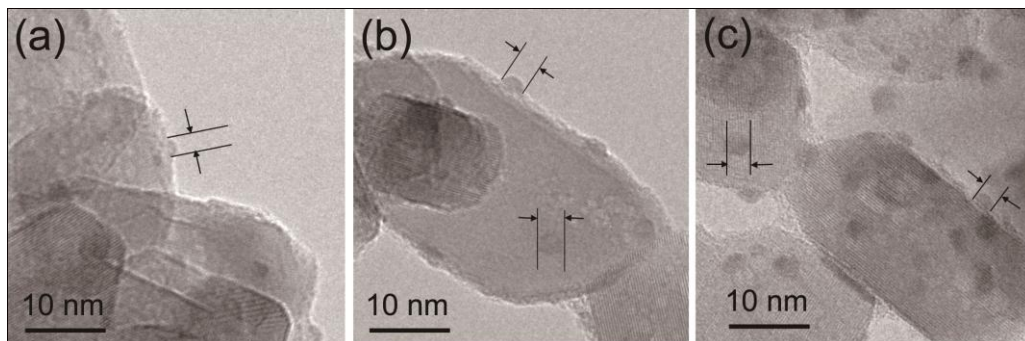


Figure S3. TEM images of the perovskite $(\text{CH}_3\text{NH}_3)\text{PbI}_3$ QD-deposited TiO_2 particles from (a) 10.05 wt%, (b) 21.13 wt% and (c) 30.18 wt% equimolar mixture of $\text{CH}_3\text{NH}_3\text{I}$ and PbI_2 in γ -butyrolactone. Average diameter of perovskite QD was determined to be about 2.5 nm, regardless of perovskite precursor solution concentration.

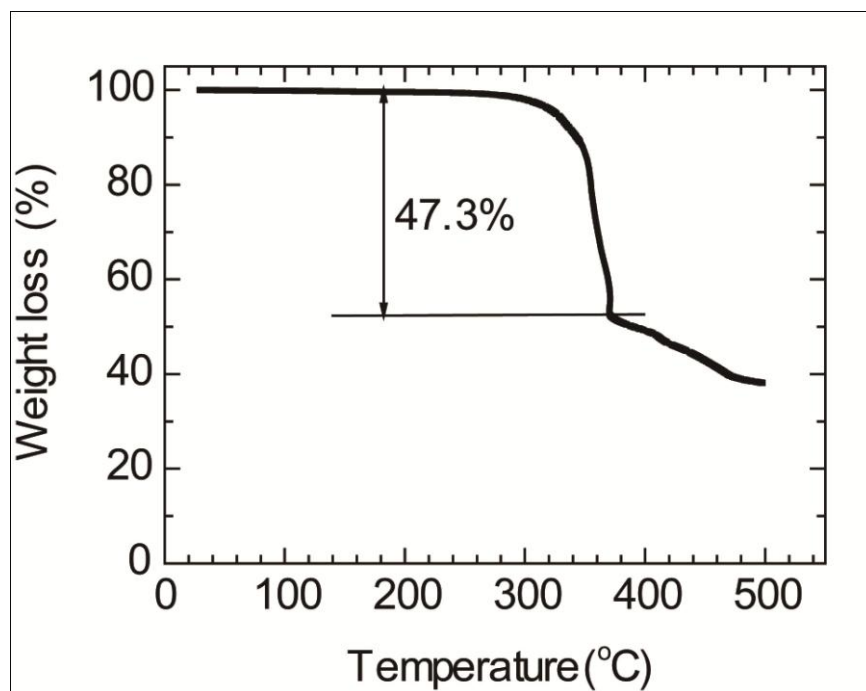


Figure S4. Weight loss as a function of temperature for the powdered perovskite $(\text{CH}_3\text{NH}_3)\text{PbI}_3$. Thermogravimetric analysis was performed at rate of $10\text{ }^\circ\text{C}/\text{min}$ under air atmosphere

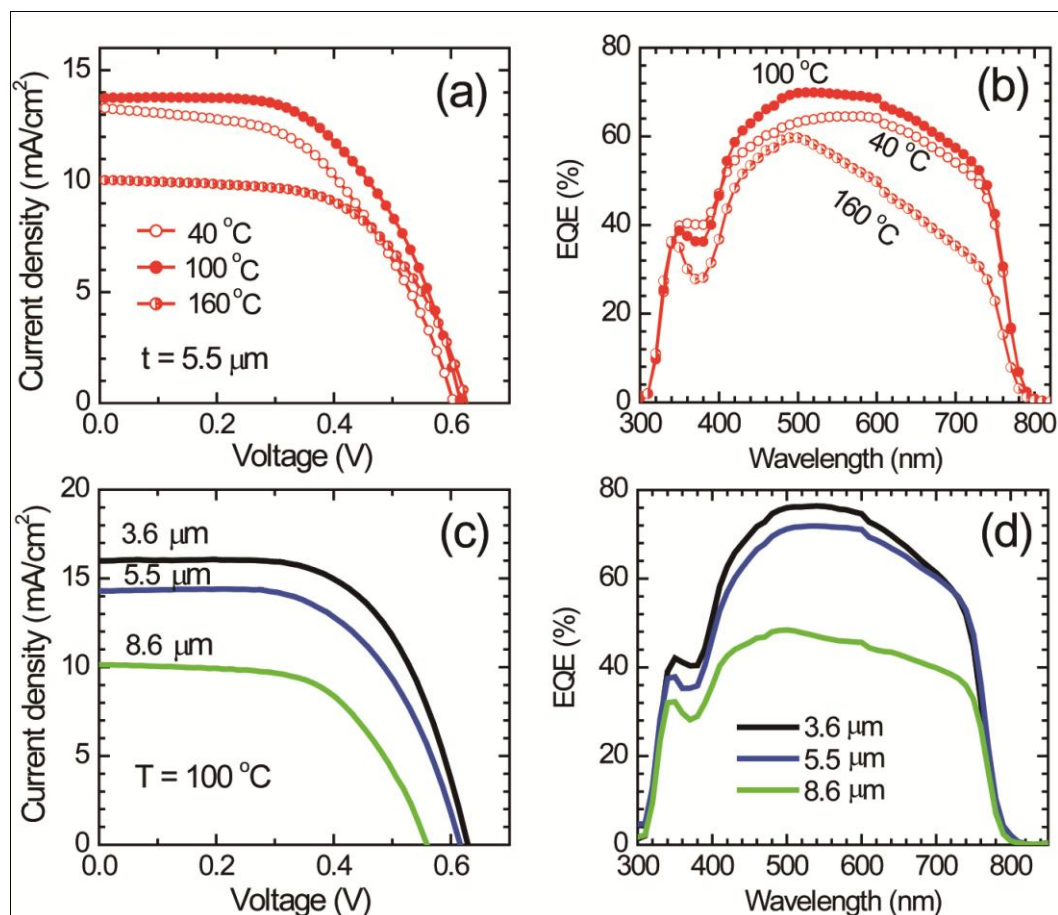


Figure S5. Effects of post annealing temperature (a and b) and film thickness (c and d) on J-V curves and EQE spectra of the perovskite (CH₃NH₃)PbI₃ QD-sensitized solar cells. For annealing temperature experiments, TiO₂ films with thickness of about 5.5 μm were used and a 40.26 wt% of the equimolar mixture of CH₃NH₃I and PbI₂ in γ-butyrolactone was used for perovskite QD deposition. For film thickness experiments, QD-sensitized TiO₂ films were annealed at 100 °C and a 41.22 wt% of the equimolar mixture of CH₃NH₃I and PbI₂ in γ-butyrolactone was used for perovskite QD deposition. J-V measurement was performed at AM 1.5 G one sun illumination (100 mW/cm²).

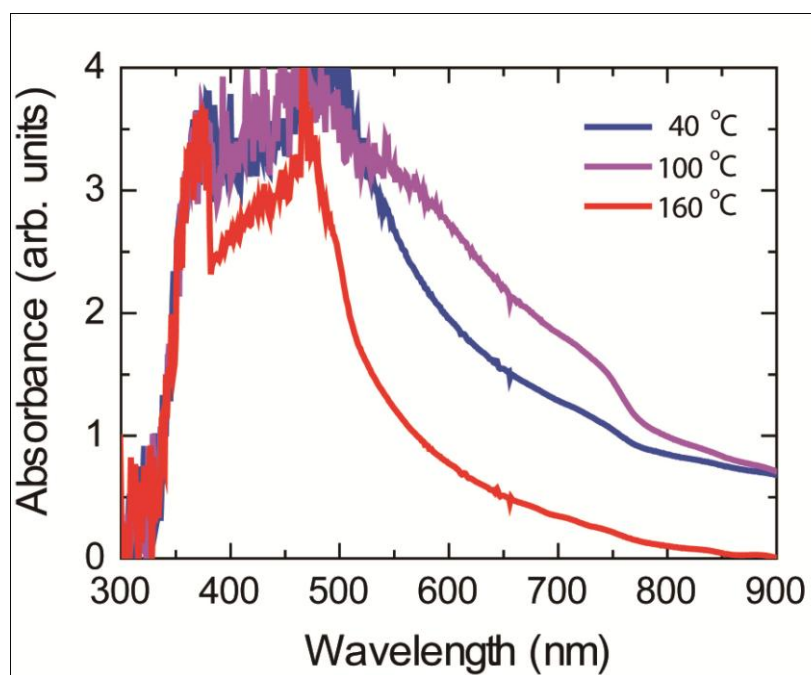


Figure S6. Absorption spectra of perovskite OD deposited TiO₂ films depending on post-annealing temperature.