

## Supporting Information

# Size Effect of Endohedral Cluster on Fullerene Cage: Preparation and Structural Studies of



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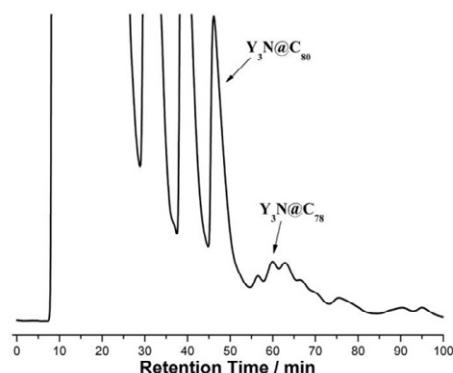
### Experimental Section:

#### 1. The synthesis and purification of $\text{Y}_3\text{N}@\text{C}_{78}$

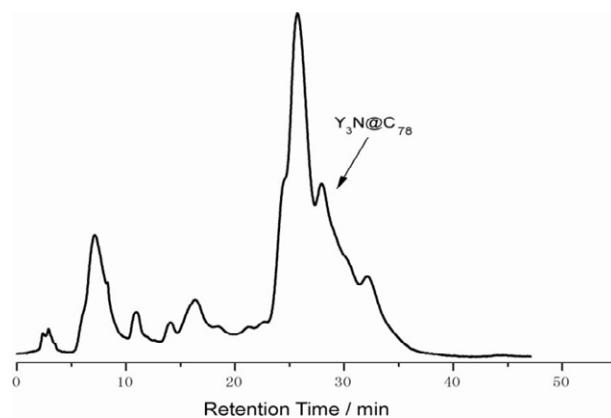
Graphite rods were core-drilled and subsequently packed with a mixture of Y/Ni<sub>2</sub> alloy and graphite powder in a weight ratio of 2:1. These rods were then vaporized in a Krätschmer-Huffman generator at 192 Torr He and 8 Torr N<sub>2</sub>. The resulting soot was Soxlet-extracted with toluene for 12 h.  $\text{Y}_3\text{N}@\text{C}_{78}$  was isolated from various empty fullerenes and other yttrium metallofullerenes by multi-stage HPLC.

#### 2. HPLC data of purified $\text{Y}_3\text{N}@\text{C}_{78}$

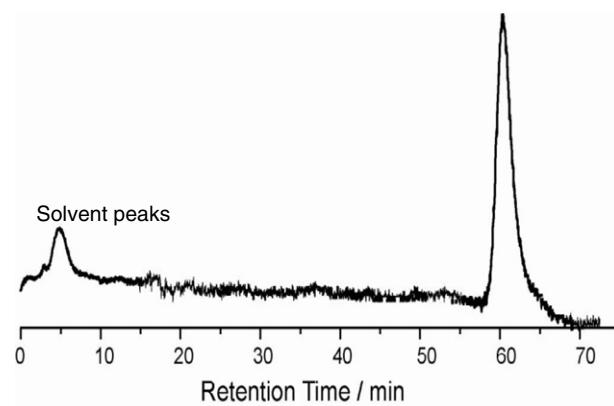
The stage 1 and 2 separations were repeated several times to obtain purified  $\text{Y}_3\text{N}@\text{C}_{78}$ . Figure S3 and Figure S4 show the HPLC data of purified  $\text{Y}_3\text{N}@\text{C}_{78}$  with Buckyprep and Buckyprep-M columns. These HPLC analyses also confirm the purity of  $\text{Y}_3\text{N}@\text{C}_{78}$ .



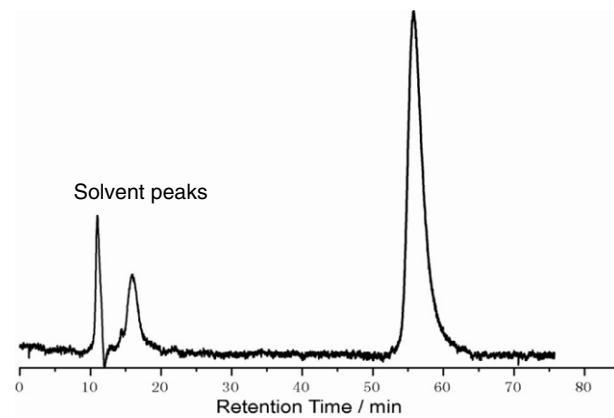
**Figure S1.** The first stage HPLC profile of toluene extract of the soot containing mostly yttrium endohedral metallofullerenes (20×250 mm Buckyprep column; flow rate 12 mL/min; toluene as eluent).



**Figure S2.** The second stage HPLC profile of  $\text{Y}_3\text{N}@\text{C}_{78}$  fullerenes (20×250 mm Buckyprep-M column; flow rate 12 mL/min; toluene as eluent).



**Figure S3.** Chromatogram of the isolated  $\text{Y}_3\text{N}@\text{C}_{78}$  (20×250 mm Buckyprep column; flow rate 12 mL/min; toluene as eluent).

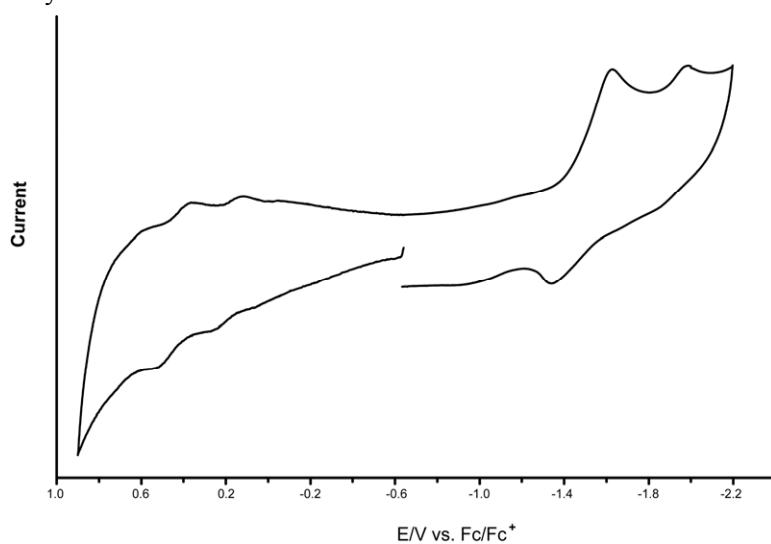


**Figure S4.** Chromatogram of the isolated  $\text{Y}_3\text{N}@\text{C}_{78}$  (20×250 mm Buckyprep-M column; flow rate 6 mL/min; toluene as eluent).

### 3. CV spectrum of purified Y<sub>3</sub>N@C<sub>78</sub>

Electrochemistry experiments were carried out in *o*-DCB solvent with glassy carbon as the working, Pt flake and a SCE as the counter and reference electrodes, respectively. The potentials were referred to the E value of the Fc/Fc<sup>+</sup> redox couple measured in the sample solution.

The first and second oxidation potentials, <sub>ox</sub>E<sub>1</sub> and <sub>ox</sub>E<sub>2</sub>, appear at 0.25 V and 0.53 V, respectively. While the first and second reduction potentials, <sub>red</sub>E<sub>1</sub> and <sub>red</sub>E<sub>2</sub>, appear at -1.62 V and -1.99 V, respectively.



**Figure S5.** CV spectrum of Y<sub>3</sub>N@C<sub>78</sub> in *o*-DCB. 0.05 M (*n*-Bu)<sub>4</sub>NPF<sub>6</sub>; scan rate, 100 mV s<sup>-1</sup>.

**Calculation Section:**

**1. DFT calculated results of  $\text{Sc}_x\text{Y}_{3-x}\text{N}@\text{C}_{78}$  ( $x=0-3$ ) isomers**

Density functional theory (DFT) calculations were preformed to study the relative energies and geometric structures of  $\text{Sc}_x\text{Y}_{3-x}\text{N}@\text{C}_{78}$  ( $x=0-3$ ) isomers at GGA-PBE/DNP<sup>S1</sup> level by using the Dmol<sup>3</sup> code<sup>S2</sup>.  $\text{C}_{78}$  has in total 24 109 isomers, and five of them conform with the IPR<sup>S3</sup>. As showed in reference S4, the IPR isomer  $D_{3h}$ : 24109, was found to be the most stable isomer of  $\text{C}_{78}$ <sup>6</sup>. The second most stable isomer is the non-IPR  $C_2$ : 22010 cage, which contains two pairs of adjacent pentagons.<sup>S4</sup> Therefore, we executed our calculations based on these two  $\text{C}_{78}$  isomers. The geometry optimizations were carried out with no symmetry and spin constraints in Cartesian coordinates and with analytically constructed energy gradient.

**Table S1.** Relative Energies ( $\Delta E$ , kJ/mol) of  $\text{Sc}_2\text{YN}@\text{C}_{78}$  isomers

Species	$\Delta E$
$\text{Sc}_2\text{YN}@\text{C}_{78}-D_{3h}$	0
$\text{Sc}_2\text{YN}@\text{C}_{78}-C_2$	5.798376
$\text{Sc}_2\text{YN}@\text{C}_{78}-C_2-1^a$	9.146516

<sup>a</sup> One isomer of  $\text{Sc}_2\text{YN}@\text{C}_{78}$ , whose two Sc atoms are all coordinated to two pairs of fused pentagons.

**Table S2.** Relative Energies ( $\Delta E$ , kJ/mol) of  $\text{ScY}_2\text{N}@\text{C}_{78}$  Isomers

Species	$\Delta E$
$\text{ScY}_2\text{N}@\text{C}_{78}-D_{3h}$	9.713157
$\text{ScY}_2\text{N}@\text{C}_{78}-C_2$	0
$\text{ScY}_2\text{N}@\text{C}_{78}-C_2-1^b$	4.363889

<sup>b</sup> One isomer of  $\text{ScY}_2\text{N}@\text{C}_{78}$ , whose one Sc atom is coordinated to one pair of fused pentagons.

**Table S3.** Relative Energies ( $\Delta E$ , kJ/mol) of  $\text{Sc}_3\text{N}@\text{C}_{78}$  Isomers

Species	$\Delta E$
$\text{Sc}_3\text{N}@\text{C}_{78}-D_{3h}$	0
$\text{Sc}_3\text{N}@\text{C}_{78}-C_2$	18.43717

**Table S4.** Relative Energies ( $\Delta E$ , kJ/mol) of  $\text{Y}_3\text{N}@\text{C}_{78}$  Isomers

Species	$\Delta E$
$\text{Y}_3\text{N}@\text{C}_{78}-D_{3h}$	18.00902
$\text{Y}_3\text{N}@\text{C}_{78}-C_2$	0

**Table S5.** Selected bonding length (Å) in optimized structures of  $\text{Sc}_3\text{N}@\text{C}_{78}-D_{3h}$ ,  $\text{Y}_3\text{N}@\text{C}_{78}-C_2$ ,  $\text{Sc}_2\text{YN}@\text{C}_{78}-D_{3h}$ , and  $\text{ScY}_2\text{N}@\text{C}_{78}-C_2$ .

$\text{Sc}_3\text{N}@\text{C}_{78}-D_{3h}$	Sc-cage	2.25	Sc-N	2.01
$\text{Y}_3\text{N}@\text{C}_{78}-C_2$	Y <sub>1</sub> -cage	2.43	Y <sub>1</sub> -N	2.07
	Y <sub>2</sub> -cage	2.49	Y <sub>2</sub> -N	2.10
	Y <sub>3</sub> -cage		Y <sub>3</sub> -N	
$\text{Sc}_2\text{YN}@\text{C}_{78}-D_{3h}$	Y <sub>1</sub> -cage	2.39	Y <sub>1</sub> -N	2.12
	Sc <sub>2</sub> -cage	2.22	Sc <sub>2</sub> -N	1.96
	Sc <sub>3</sub> -cage		Sc <sub>3</sub> -N	
$\text{ScY}_2\text{N}@\text{C}_{78}-C_2$	Sc <sub>1</sub> -cage	2.30	Sc <sub>1</sub> -N	1.94
	Y <sub>2</sub> -cage	2.51	Y <sub>2</sub> -N	2.16
	Y <sub>3</sub> -cage		Y <sub>3</sub> -N	

## References

- S1. J. P. Perdew, K. Burke, M. Ernzerhof , *Phys. Rev. Lett.*, 1996, **77**, 3865.
- S2. (a) Delley, *J. Phys. Chem. B*, 1990, **92**, 508. (b) Delley, *J. Phys. Chem. B*, **113**, 7756.  
Dmol<sup>3</sup> is available as part of Material Studio and Cerius2 by Accelrys Inc.
- S3. P. W. Fowler, D. E. Manolopoulos, *An Atlas of Fullerenes*:Clarendon Press: Oxford, U.K., 1995.
- S4. A. A. Popov, M. Krause, S. F. Yang, J. Wong and L. Dunsch, *J. Phys. Chem. B*, 2007, **111**, 3363-3369.