

Electronic Supplementary Information (ESI)

## **Ag Nanocrystal as a Promoter for Carbon Nanotube-based Room-temperature Gas Sensors**

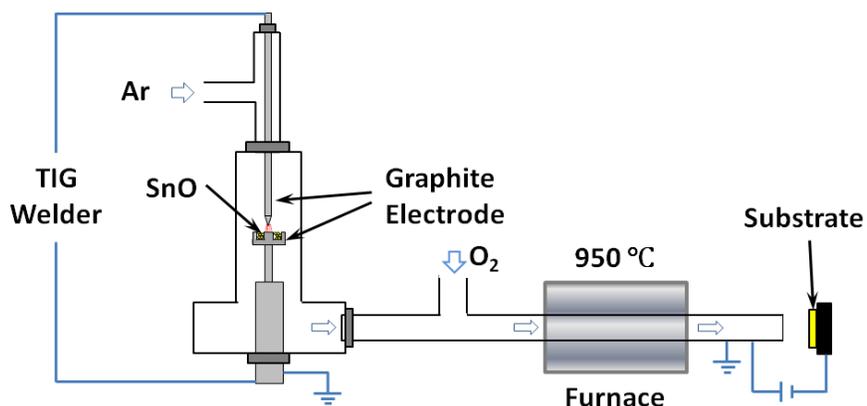
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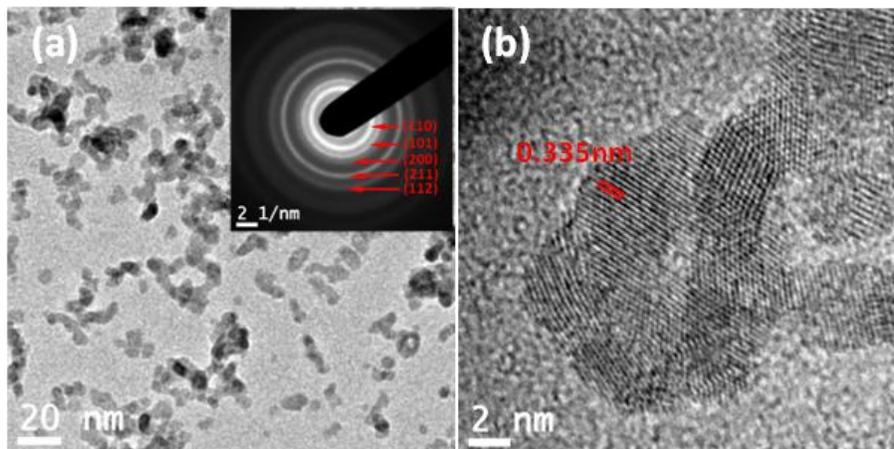
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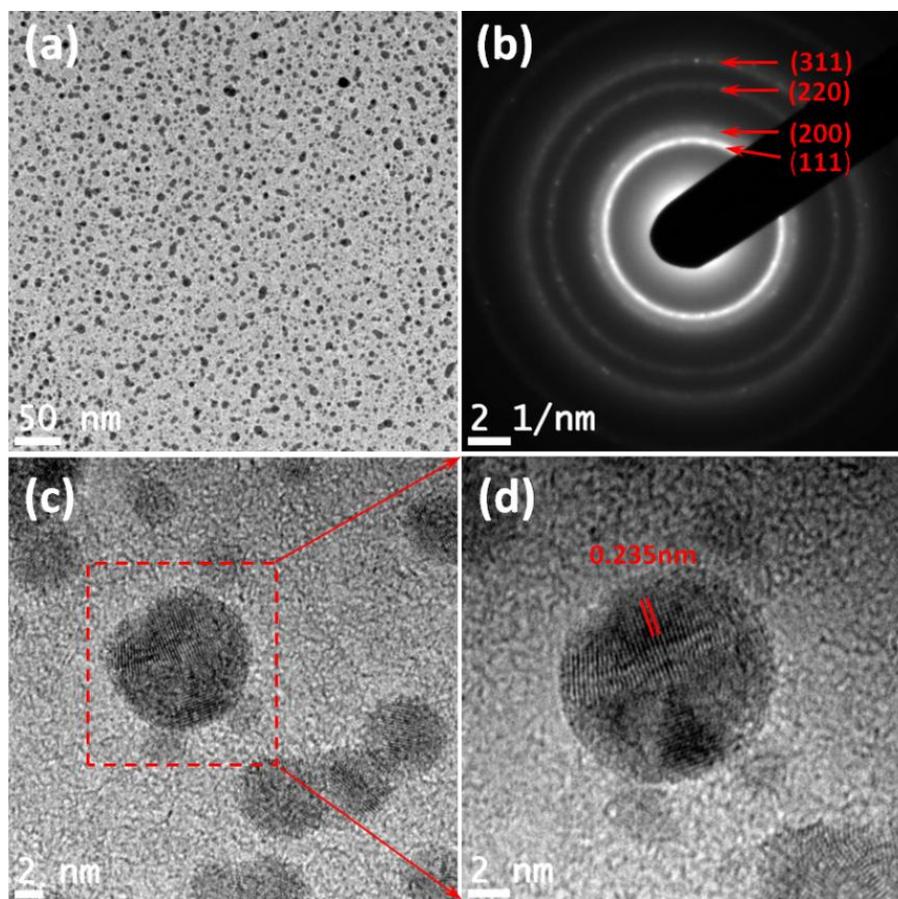
<sup>§</sup> These authors contributed equally.



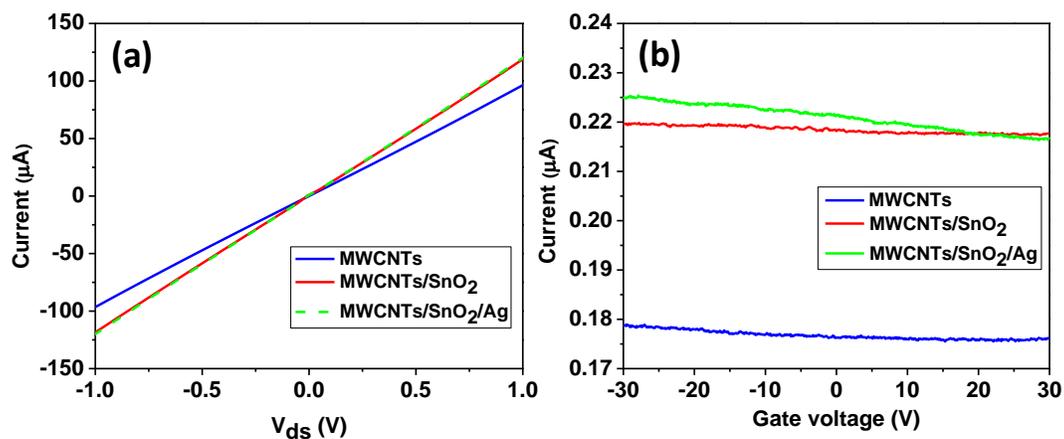
**Fig. S1** Schematic setup of SnO<sub>2</sub> nanoparticle synthesis system.



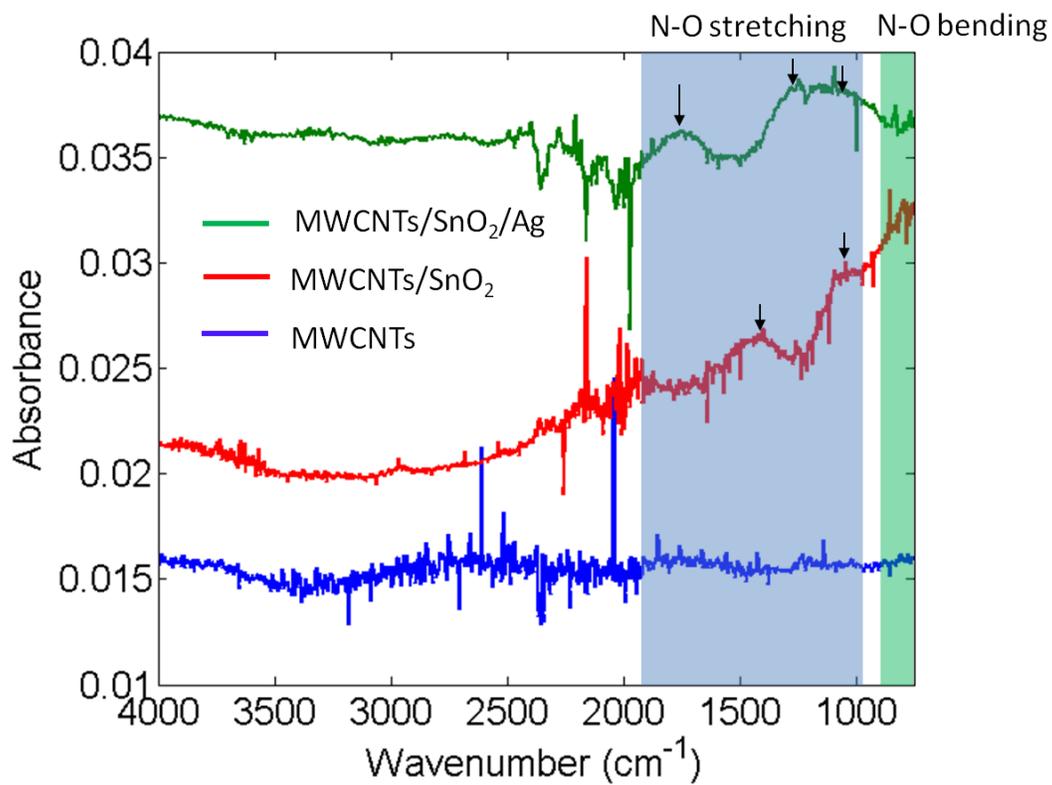
**Fig. S2** (a) Low magnification TEM image and SAED pattern (inset) of SnO<sub>2</sub> nanoparticles synthesized at 950 °C. (b) HRTEM image of SnO<sub>2</sub> nanoparticles. 0.335 nm lattice spacing is indexed to (110) plane of rutile SnO<sub>2</sub>.



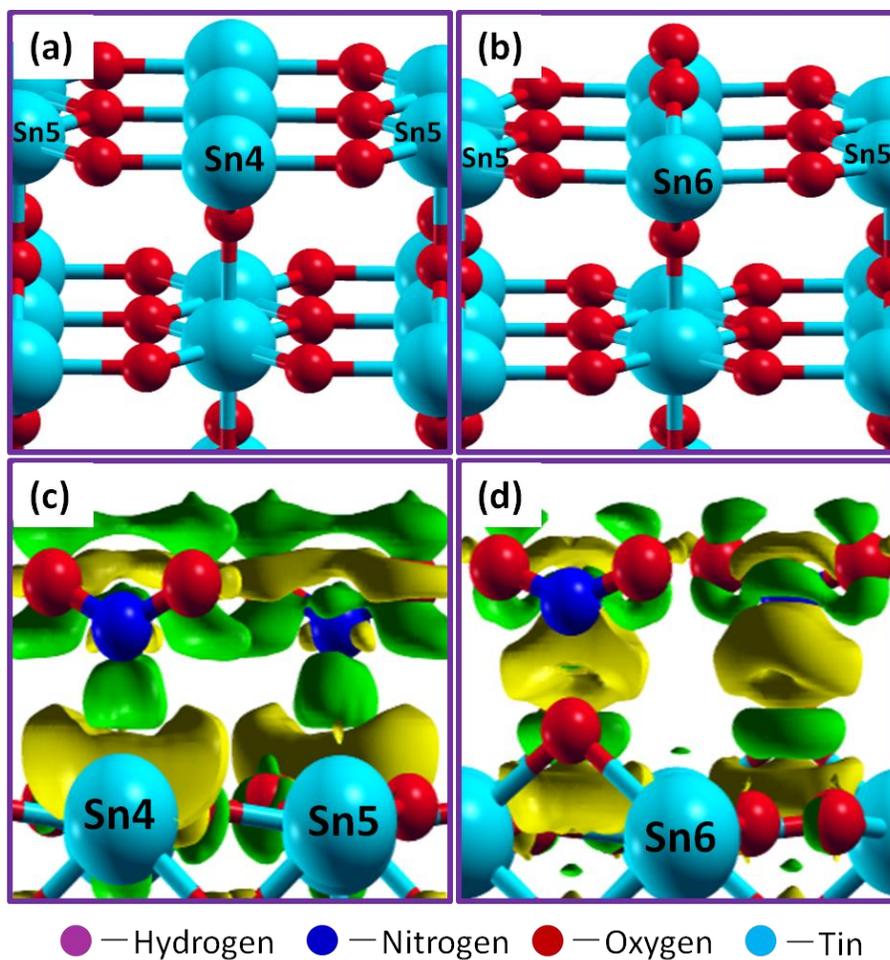
**Fig. S3** Low magnification TEM image (a) and SAED pattern (b) of as-produced Ag NPs. (c), and (d) HRTEM images of Ag NPs, and 0.235 nm lattice spacing is indexed to (111) plane of Ag.



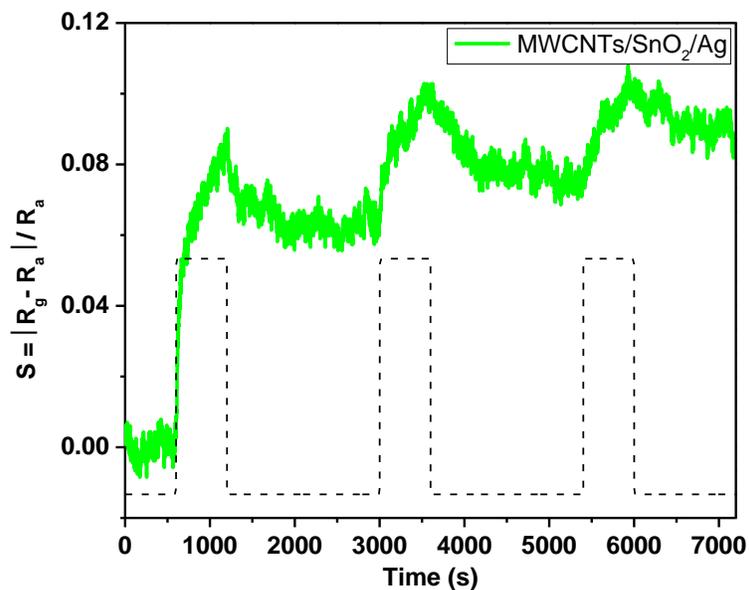
**Fig. S4** (a)  $I$ - $V$  curve evolution of MWCNTs at different conditions (with and without NP coating). (b) The dependence of current on gate voltage.



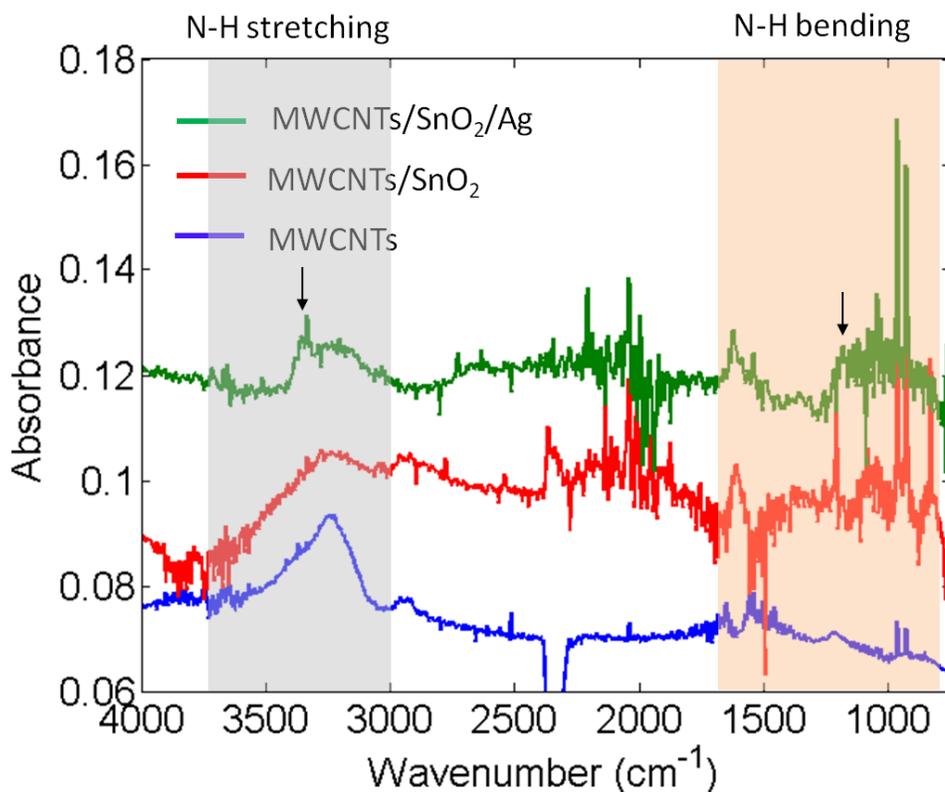
**Fig. S5** Differential IR absorbance spectra of MWCNT hybrid sensors exposed to 100 ppm NO<sub>2</sub> diluted in Ar.



**Fig. S6** Upper panel: Perspective views of the  $\text{SnO}_2$  (110) surface with the coordination number of different Sn atoms labeled, (a) planar-reduced surface and (b) oxidized surface with oxygen atoms sitting on the bridge sites. Lower panel: Charge density difference upon  $\text{NO}_2$  molecule adsorption on the surface. Green and yellow represent electron accumulation and depletion regions, respectively. (c)  $\text{NO}_2$  is attracted directly to the Sn4 atom with gain of  $0.0576 e^-$ . (d)  $\text{NO}_2$  forms  $\text{NO}_3^-$  complex with gain of  $0.0069 e^-$ . For  $\text{NH}_3$  adsorption, it prefers the Sn5 site instead of the bridging oxygen. However, this site is preoccupied by  $\text{O}_2$  at room temperature and leads to the reluctance of sensor to  $\text{NH}_3$ .

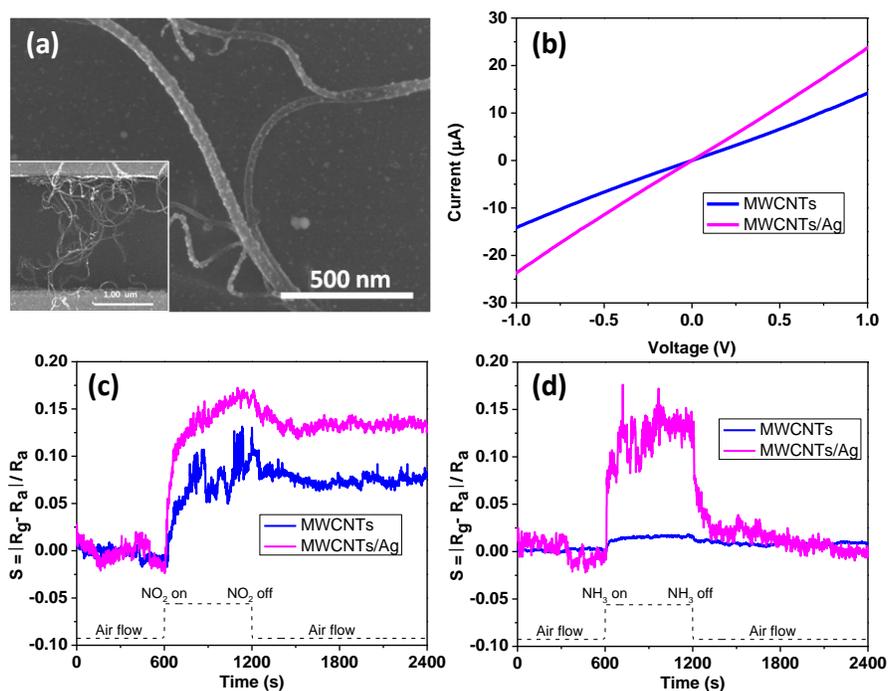


**Fig. S7** Representative sensing response of Ag/SnO<sub>2</sub>/MWCNTs hybrid structures to 100 ppm NO<sub>2</sub> at room temperature.

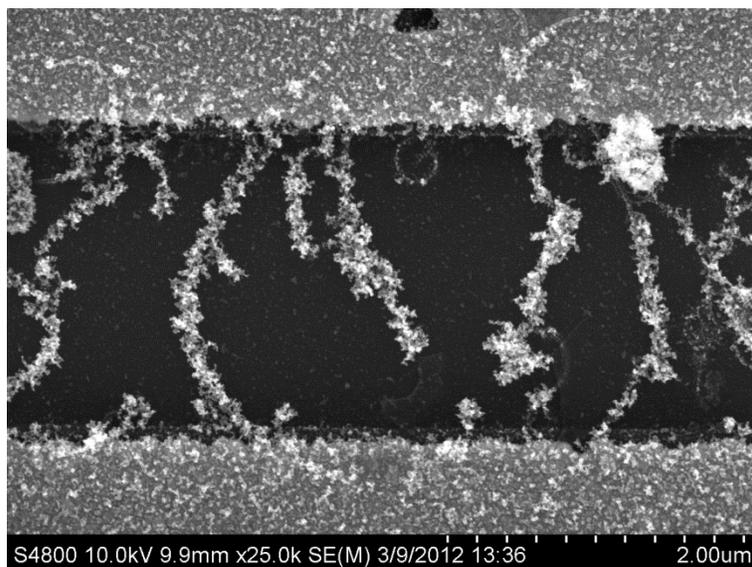


**Fig. S8** Differential IR absorbance spectra of MWCNT hybrid sensors exposed to 1% NH<sub>3</sub> diluted in Ar. Spectra are scaled and offset for clarity. Features at 2,350 cm<sup>-1</sup> and 1,400-1,800 cm<sup>-1</sup> are due to incomplete cancellation of atmospheric CO<sub>2</sub> and water vapor, respectively. Gas-phase NH<sub>3</sub> rotation-vibration bands are present in the region

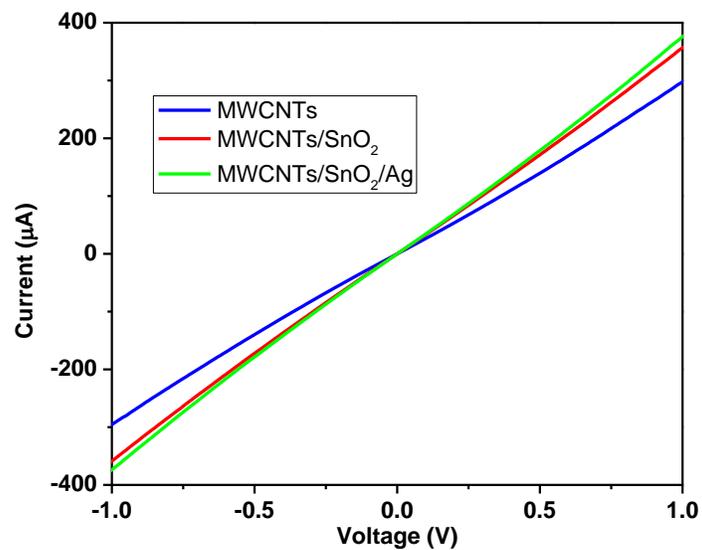
900-1,200  $\text{cm}^{-1}$ . The MWCNT and MWCNT/SnO<sub>2</sub> spectra are quite similar, with only small difference between the two in the region 900-1,100  $\text{cm}^{-1}$ .



**Fig. S9** (a) SEM image of MWCNTs coated with Ag NPs. Inset is an SEM image showing the hybrid structure bridging two gold electrodes. (b) I-V characteristics of MWCNTs and MWCNTs/Ag. (c) and (d) are sensing responses of bare MWCNTs and MWCNTs/Ag structures to 100 ppm NO<sub>2</sub> and 1% NH<sub>3</sub> at room temperature, respectively.



**Fig. S10** SEM image of MWCNTs with full surface coverage of SnO<sub>2</sub> NPs



**Fig. S11** *I-V* curve evolution of the control sample, which was obtained by first coating MWCNTs with SnO<sub>2</sub> NPs at a high coverage followed by additional coating of Ag NPs.