

## Electronic Support Information (ESI) Figures & Tables

**Table S1: Abbreviations and main components of the test media and the numerically estimated ionic strengths.** The data were numerically estimated using the USGS software PHREEQCi 2.18 using the database *minteq.v4* omitting the following components: amino acids, proteins, vitamins and glucose.

Medium abbreviation	Test system	Main components	Concentration [mM]	Ionic strength [mM]
Elendt	<i>Daphnia magna</i>	CaCl <sub>2</sub> NaHCO <sub>3</sub> MgSO <sub>4</sub>	2.0 0.8 0.5	8.3
DMEM+FCS	OLN-93 cells	NaCl NaHCO <sub>3</sub> KCl CaCl <sub>2</sub> NaH <sub>2</sub> PO <sub>4</sub> MgSO <sub>4</sub> proteins, amino acids, glucose	109.5 44.6 5.4 1.8 0.9 0.8	167.8
DSM	<i>Arthrobacter globiformis</i>	NaCl glucose, proteins	28.6	28.6
AS	Activated sludge	Sodium acetate Na <sub>2</sub> HPO <sub>4</sub> KH <sub>2</sub> PO <sub>4</sub> NaNO <sub>3</sub> CaCl <sub>2</sub> MgSO <sub>4</sub>	52.1 49.0 27.6 9.8 0.25 0.09	207.9
AchE	Enzyme inhibition for acetylcholine esterase	Na <sub>2</sub> HPO <sub>4</sub> NaHCO <sub>3</sub>	20.0 0.6	52.3
GR	Enzyme inhibition for glutathione reductase	KH <sub>2</sub> PO <sub>4</sub> / K <sub>2</sub> HPO <sub>4</sub>	100.0	205.9

**Table S2a: Application scenario A: “Polyvinylpyrrolidone coated iron oxide (Fe<sub>3</sub>O<sub>4</sub>; magnetite) nanoparticles as contrast agent for magnetic resonance imaging (MRI)” (tested with the “precautionary matrix” [Höck et al. 2010]).**

Scenarios	Precautionary need			
	Workers, general [WG]	Workers, worst case [WWC]	Consumers [C]	Environment [E]
<u>MRI: production, liquid [PL]</u> <i>Life cycle stage:</i> production of the IONP <i>Medium:</i> IONP in suspension	Precautionary need for workers in general  <i>Workers:</i> all persons getting potentially into contact with the IONP during production	Precautionary need for workers in the worst case  <i>Workers:</i> all persons getting potentially into contact with the IONP during production	[Not applicable]	Precautionary need for the environment
<u>MRI: production, dry [PD]</u> <i>Life cycle stage:</i> production of the IONP <i>Medium:</i> IONP as dry powder	Assumptions refer to normal production process	Assumptions refer to worst case		
<u>MRI: use, liquid [UL]</u> <i>Life cycle stage:</i> use of the IONP as contrast agent in clinical settings <i>Medium:</i> IONP in suspension	Precautionary need for workers in general  <i>Workers:</i> all clinical personnel getting potentially into contact with the IONP at the clinic	Precautionary need for workers in worst case  <i>Workers:</i> all clinical personnel getting potentially into contact with the IONP at the clinic	Precautionary need for consumers  <i>Consumers:</i> patients that are administered the IONP as contrast agent for MRI scanning purposes	Precautionary need for the environment
<u>MRI: use, dry [UD]</u> <i>Life cycle stage:</i> use of the PVP-IO-NP as contrast agent in clinical settings <i>Medium:</i> IONP in suspension	Assumptions refer to normal handling of pharmaceuticals and medical products	Assumptions refer to worst case handling of pharmaceuticals and medical products		

**Table S2b: Application scenario B: “Iron oxide (Fe<sub>3</sub>O<sub>4</sub>; magnetite) or zero valent iron (Fe<sup>0</sup>; nZVI) nanoparticles as adsorbent or reductive agent for soil or groundwater remediation purposes” (tested with the “precautionary matrix” [Höck et al. 2010]).**

Scenarios	Precautionary need			
	Workers, general [WG]	Workers, worst case [WWC]	Consumers [C]	Environment [E]
<u>Remediation: magnetite, liquid [ML]</u> <i>Particles:</i> IONP (magnetite) <i>Medium:</i> IONP in suspension	Precautionary need for workers in general  <i>Workers:</i> all persons getting potentially into contact with the IONP during production	Precautionary need for workers in the worst case  <i>Workers:</i> all persons getting potentially into contact with the IONP during production	Precautionary need for consumers  <i>Consumers:</i> all persons getting potentially into contact with the IONP during use at remediation site	Precautionary need for the environment
<u>Remediation: magnetite, dry [MD]</u> <i>Particles:</i> IONP (magnetite) <i>Medium:</i> IONP as dry powder	Assumptions refer to normal production process	Assumptions refer to worst case	Assumptions refer to worst case	Assumptions refer to worst case
<u>Remediation: nZVI, liquid [ZVL]</u> <i>Particles:</i> nZVI <i>Medium:</i> nZVI in suspension	Precautionary need for workers in general  <i>Workers:</i> all persons getting potentially into contact with the nZVI during production	Precautionary need for workers in the worst case  <i>Workers:</i> all persons getting potentially into contact with the nZVI during production	Precautionary need for consumers  <i>Consumers:</i> all persons getting potentially into contact with the nZVI during use at remediation site	Precautionary need for the environment
<u>Remediation: nZVI, dry [ZVD]</u> <i>Particles:</i> nZVI <i>Medium:</i> nZVI as dry powder	Assumptions refer to normal production process	Assumptions refer to worst case	Assumptions refer to worst case	Assumptions refer to worst case

**Table S3: Hydrodynamic diameter (Hd), pH and zeta potential (Zp) of the PVP-coated IONP stock solution and the used samples in medium.** Stock: IONP [1 g Fe/L] in water Sample: IONP [100 mg Fe/L] in medium Blank: pure medium. See Table S1 for media abbreviations.

Medium	Day	Sample / Stock			Blank		
		Hd	pH	Zp	Hd	pH	Zp
H <sub>2</sub> O Stock	0	26.5 ± 2.3	5.2	1.17 ± 2.06			
	1	23.4 ± 1.5	5.6	0.8 ± 0.97			
	3	25.7 ± 3.1	5.8	1.74 ± 0.83			
	6	23.6 ± 1.1	5.7	-2.25 ± 2.34			
H <sub>2</sub> O Sample	0	22.9 ± 2.1	5.2	0.13 ± 0.36		5.6	-0.28 ± 0.28
	1	23.6 ± 3.1	5.1	2.3 ± 1.31		5.8	-0.68 ± 0.3
	3	26.7 ± 3.7	5.1	6.25 ± 4.64		5.6	-0.62 ± 0.28
	6	29.8 ± 4.8	5.0	3.2 ± 3.6		5.8	-0.03 ± 0.14
Eldent M7	0	46.6 ± 2	7.4	-0.96 ± 1.13		7.6	-0.16 ± 0.18
	1	82.3 ± 3.4	7.4	-1.01 ± 0.84		7.7	-0.59 ± 1.07
	3	132.5 ± 15.6	7.4	-1.41 ± 0.34		7.6	-0.35 ± 1.35
	6	232.7 ± 50.4	7.5	-1.2 ± 0.68		7.6	-0.88 ± 0.75
DMEM-FCS	0	28.5 ± 5.1	7.9	-5.42 ± 1.57	25.4 ± 4.3	7.8	-6.76 ± 1.44
	1	32.5 ± 3.2	8.2	-7.12 ± 0.98	29.3 ± 0.6	8.0	-7.89 ± 1.09
	3	33.8 ± 1.5	8.2	-7.99 ± 0.52	26.6 ± 7.5	8.0	-7.05 ± 1.67
	6	36.5 ± 1.9	8.2	-7.2 ± 0.97	35.1 ± 7.9	8.1	-7.8 ± 0.59
AchE	0	21.5 ± 2.7	8.0	-0.36 ± 0.3		8.0	-0.33 ± 0.41
	1	23.7 ± 2.2	8.0	-0.49 ± 0.42		8.0	0.07 ± 0.92
GR	0	23.5 ± 2.8	7.6	0.18 ± 0.55		7.6	-0.75 ± 1.84
	1	26.6 ± 3.7	7.6	-0.14 ± 1.71		7.6	-2.68 ± 2.15
AS	0	28.3 ± 1.2	7.6	-0.92 ± 0.39		7.6	-1.39 ± 1.24
	1	36.5 ± 2.4	7.7	3.69 ± 1.36		7.6	-1.46 ± 2.6
DSM	0	27.2 ± 1.4	6.9	-1.36 ± 1.77		6.9	0 ± 0.74
	1	26.3 ± 1.6	6.9	-0.77 ± 0.81		6.9	-0.53 ± 0.71

**Table S4: List of physical constants for DLVO theory calculations** (according to Lide et al. 1997.)

Name	Symbol	Value	Unit
Boltzmann constant	$k_B$	$1.38066 \cdot 10^{-23}$	J/K
Elementary charge	$e_0$	$1.60218 \cdot 10^{-19}$	As
Avogadro number	$N_A$	$6.02214 \cdot 10^{23}$	1/mol
Permittivity of the vacuum	$\epsilon_0$	$8.85419 \cdot 10^{-12}$	F/m
Permeability of free space	$\mu_0$	$1.25664 \cdot 10^{-6}$	N/A <sup>2</sup>

**Table S5: List of parameters for DLVO theory calculations**

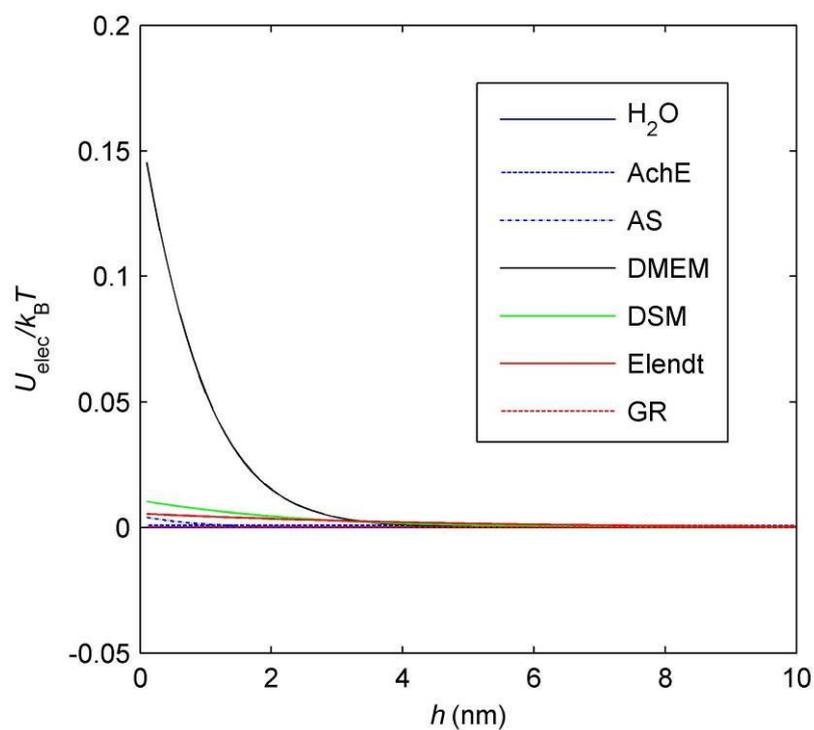
Name	Symbol	Value	Unit	Source
Temperature	$T$	298.15	K	Temperature during dynamic light scattering (DLS) measurements
Radius of the particles	$r$	various	m	
Separation distance between particles (surface to surface)	$h$	various	m	
Separation distance between particles (centre to centre)	$s$	various	m	
Relative dielectric constant	$\epsilon_r$	78.36		Water, 298.15 K (Lide et al. 1997)
Surface potential	$\zeta$	various	V	According to Table S6
Inverse Debye length	$\kappa$	various	1/m	
Ionic strength of the solution	$I$	various	mol/m <sup>3</sup>	According to Table S6
Hamaker constant for Fe <sub>3</sub> O <sub>4</sub> /H <sub>2</sub> O / Fe <sub>3</sub> O <sub>4</sub>	$A$	33	zJ (1 zJ = 10 <sup>-21</sup> J)	(Fauré 2011)
Saturation magnetization (volume)	$M_s$	various	emu/cm <sup>3</sup> = 10 <sup>3</sup> A/m	
Thickness of polymer layer	$L$	various	m	
Segment length of PVP	$l$	0.269	nm	estimated by Hyperchem 7.5
Surface density of adsorbed chains	$\sigma_p$	1	nm <sup>-2</sup>	(Lim et al. 2009) , estimated
Number of segments in polymer chains	$N_p$	43	-	See ESI text

Density of Fe <sub>3</sub> O <sub>4</sub>	$\rho$	5.17	g/cm <sup>3</sup>	at 298.15 K (Lide et al. 1997)
Density of PVP	$\rho$	1.23	g/cm <sup>3</sup>	MSDS (ScienceLab.com)
Saturation magnetization of Fe <sub>3</sub> O <sub>4</sub> (bulk) (mass)	$M_s$	92	emu/g = Am <sup>2</sup> /kg	at 300 K Utech et al. (2010)
Thickness of non magnetic oxide layer	$l_m$	1.0	nm	Kim et al. (2009)

---

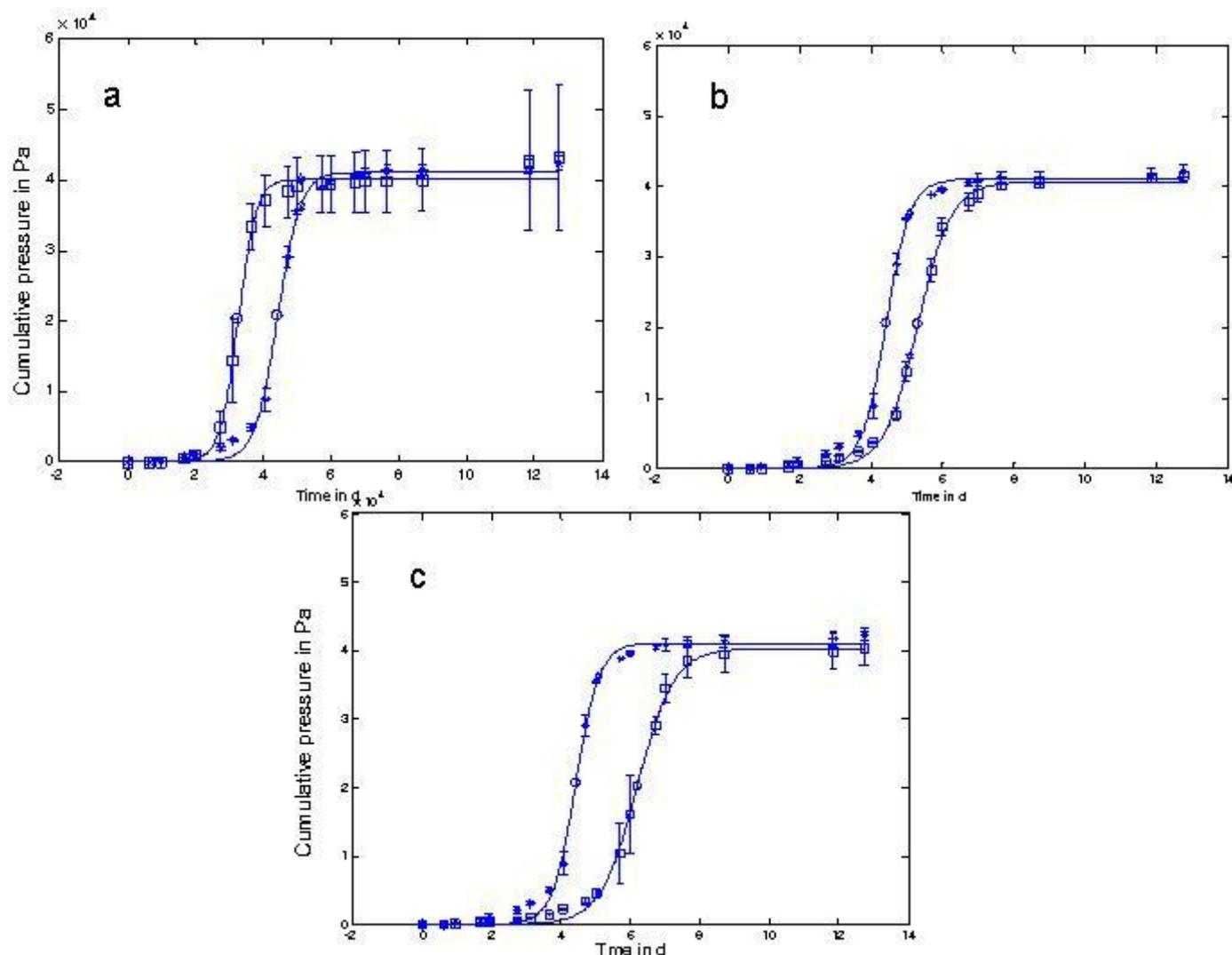
**Table S6: Experimental database for extended DLVO theory calculations.** The data were compiled from Tables S1 and S3. The calculations were done for primary particles (data of IONP in water:  $d_H = 23$  nm,  $r = 7.5$  nm,  $L = 4$  nm). Influence of media on possibly further agglomeration was considered due to different zeta potential and ionic strength in media.

<b>Medium</b>	<b>Zeta potential [mV]</b>	<b>Ionic strength/ [mol/ m<sup>3</sup>]</b>
Water	0.1	10 <sup>-4</sup>
Elendt M7	-1.0	8.3
DMEM	-5.4	167.8
AchE	-0.4	28.6
GR	0.2	207.9
AS	-0.9	52.3
DSM	-1.4	205.9



**Figure S1: Electrostatic repulsion  $U_{\text{elec}}$  of PVP-coated IONP in the test media as a function of the surface to surface distance  $h$  between the magnetite cores.**

Differences in  $U_{\text{elec}}$  resulting from the changes in ionic strength of the media and the zeta potential change are minute compared to the other interactions  $U_{\text{vdW}}$ ,  $U_{\text{steric}}$  and  $U_{\text{magn}}$  (see also Figure 5 and data in Table S6).



**Figure S2: Fitted bacterial growth curves shown as the cumulative pressure development over the incubation time of the sewage sludge samples in days.** Each data point represents the mean  $\pm$  SD of three independent replicates. Open squares: PVP-FeNP treated samples, asterisks: untreated controls. The sewage sludge samples were incubated with three different PVP-coated IONP concentrations: a.) 1.5 mg Fe/mg TS, b.) 0.16 mg Fe/mg TS and c.) 0.01 mg Fe/mg TS.