

Supplementary Information for

**A Conceptual Translation of Homogeneous Catalysis into
Heterogeneous Catalysis: Homogeneous-Like Heterogeneous Gold
Nanoparticle Catalyst Induced by Ceria Supporter**

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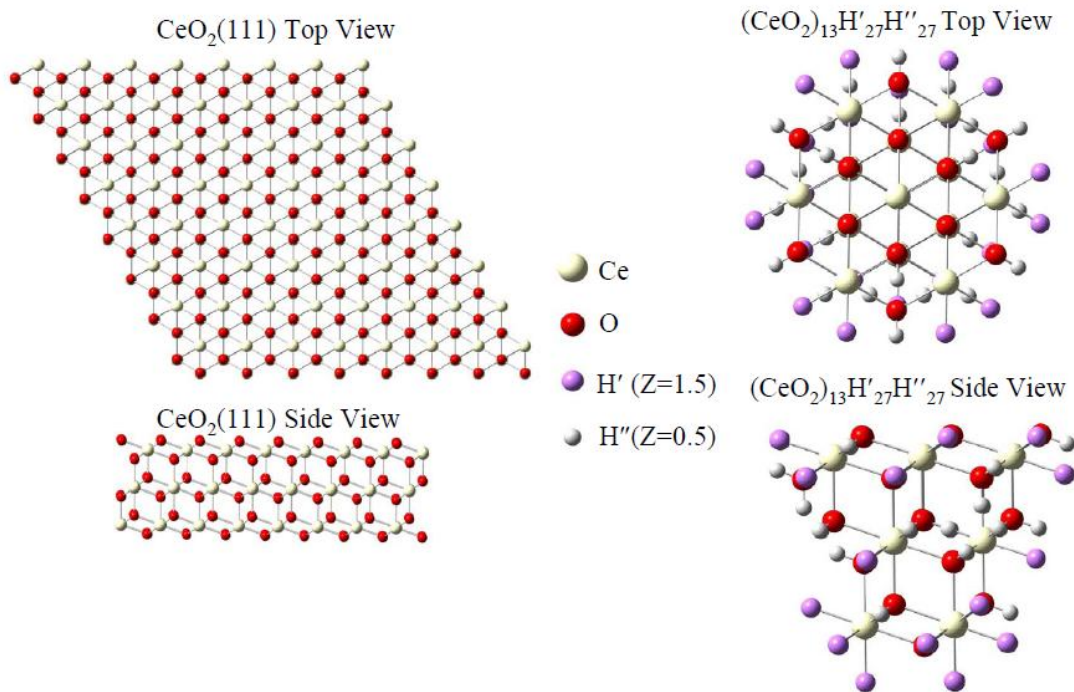


Figure S1. To mimic the properties of $\text{CeO}_2(111)$ surface using $(\text{CeO}_2)_{13}\text{H}'_{27}\text{H}''_{27}$ cluster model, in which H' and H'' are the pseudo-hydrogens with the corresponding fractional charge in parentheses.

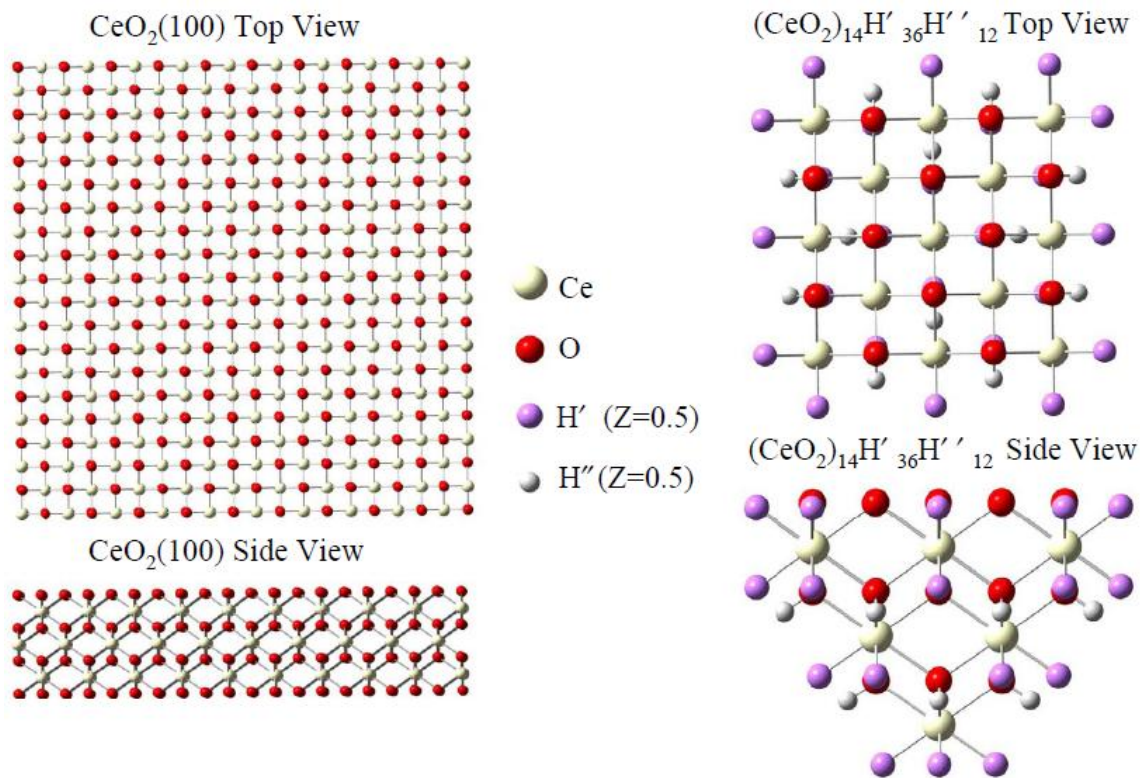


Figure S2. $(\text{CeO}_2)_{14}\text{H}'_{36}\text{H}''_{12}$ cluster model mimicking $\text{CeO}_2(100)$ surface.

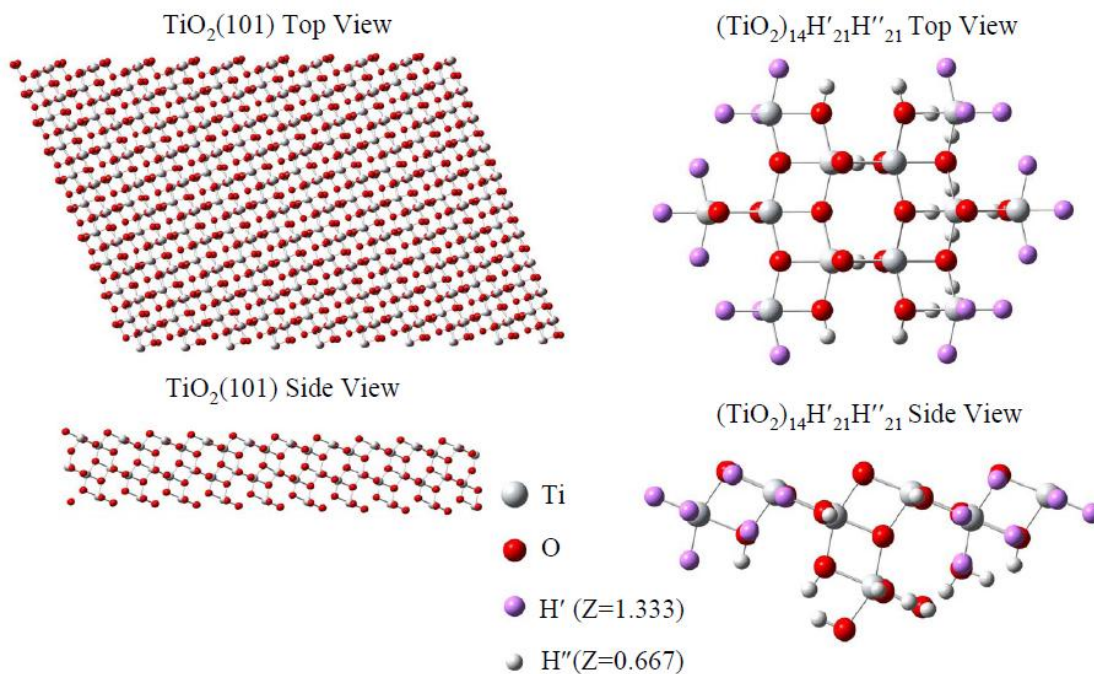


Figure S3. $(\text{TiO}_2)_{14}\text{H}'_{21}\text{H}''_{21}$ cluster model mimicking anatase(101) surface.

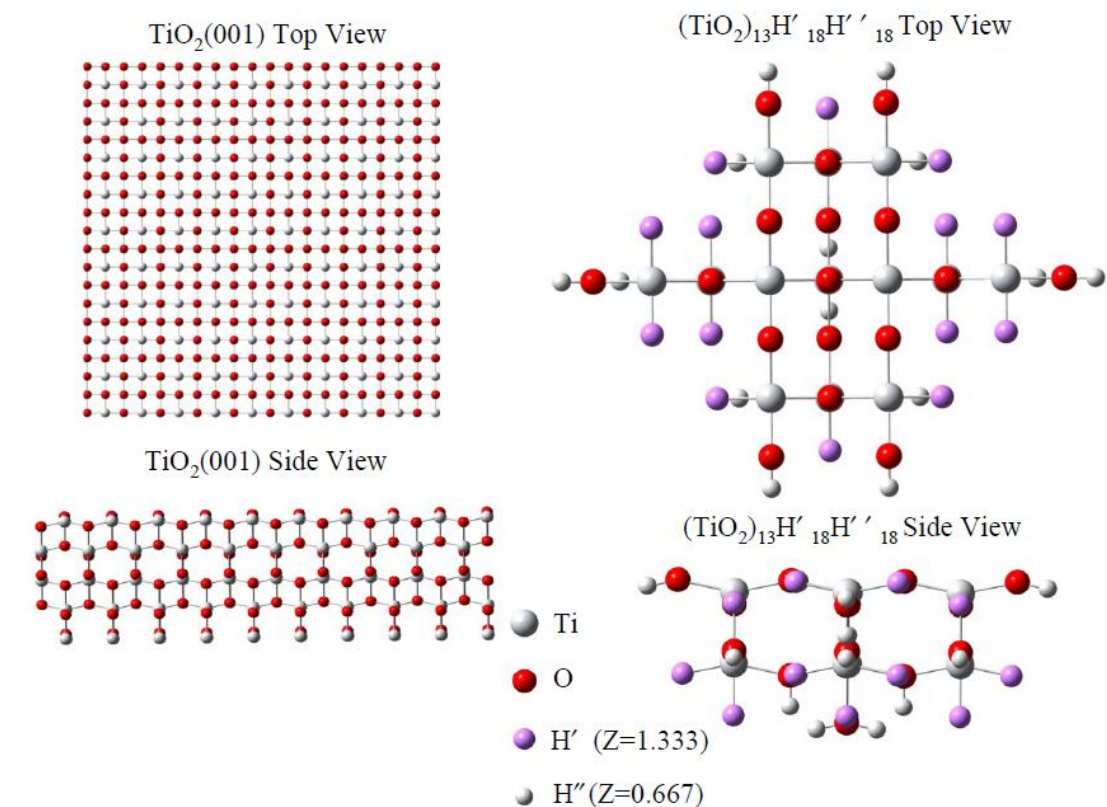


Figure S4. (TiO₂)₁₃H'₁₈H''₁₈ cluster model mimicking anatase(001) surface.

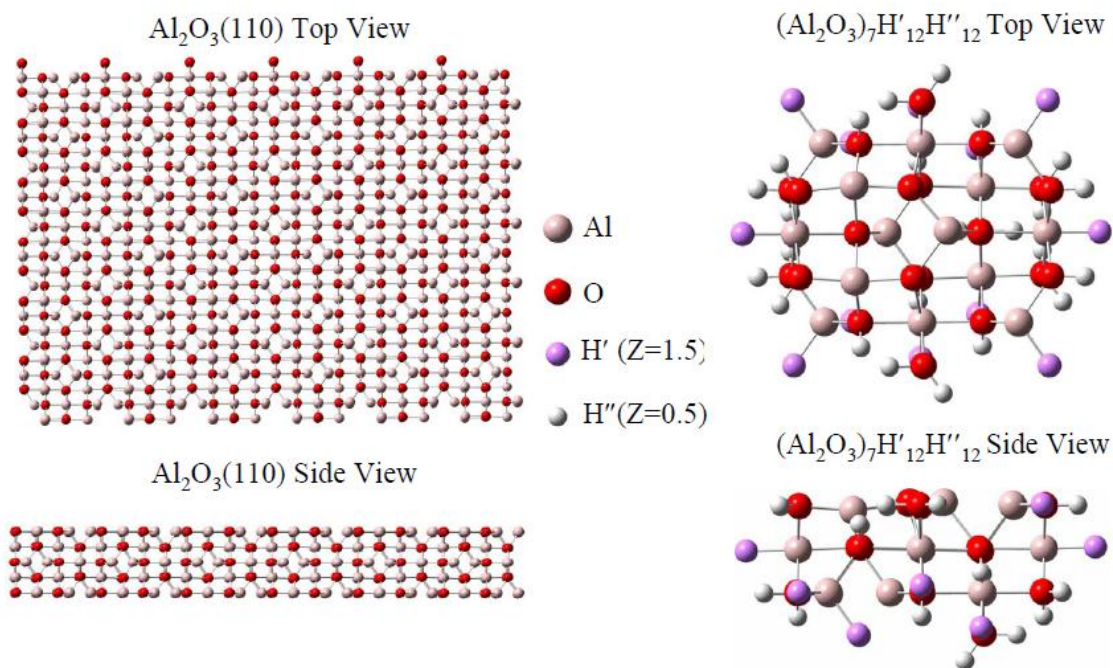


Figure S5. (Al₂O₃)₇H'₁₂H''₁₂ cluster model mimicking γ -Al₂O₃(110) surface.

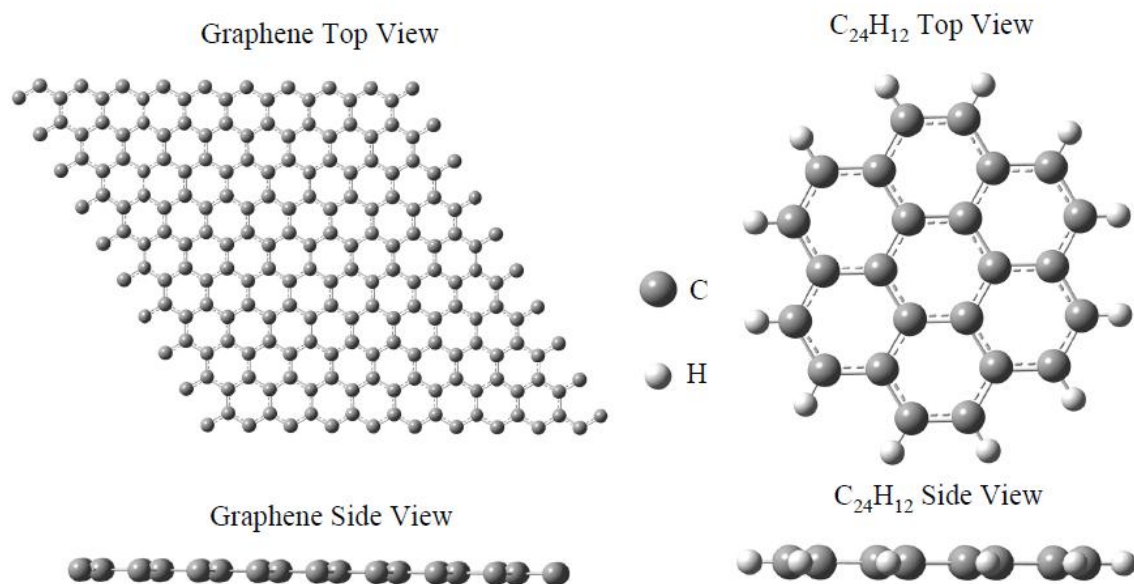


Figure S6. $C_{24}H_{12}$ cluster model mimicking graphene.

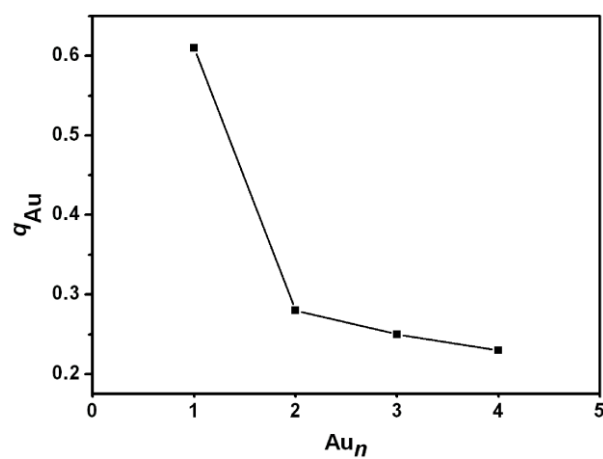


Figure S7. The average charge state of gold atom in Au_n ($n = 2 - 4$) clusters on $(CeO_2)_{14}H'_{36}H''_{12}$ cluster.

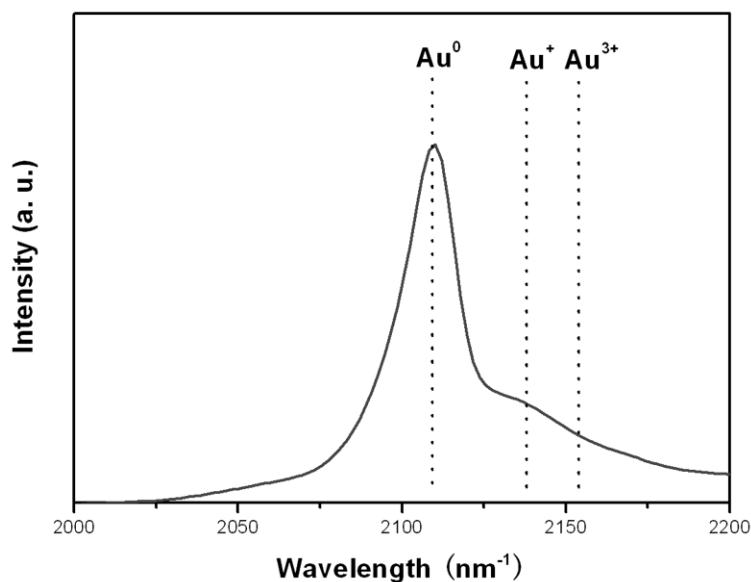


Figure S8. FTIR spectra of CO adsorbed on gold supported on CeO₂ nanocube at a CO pressure of 30 Torr and at room temperature.

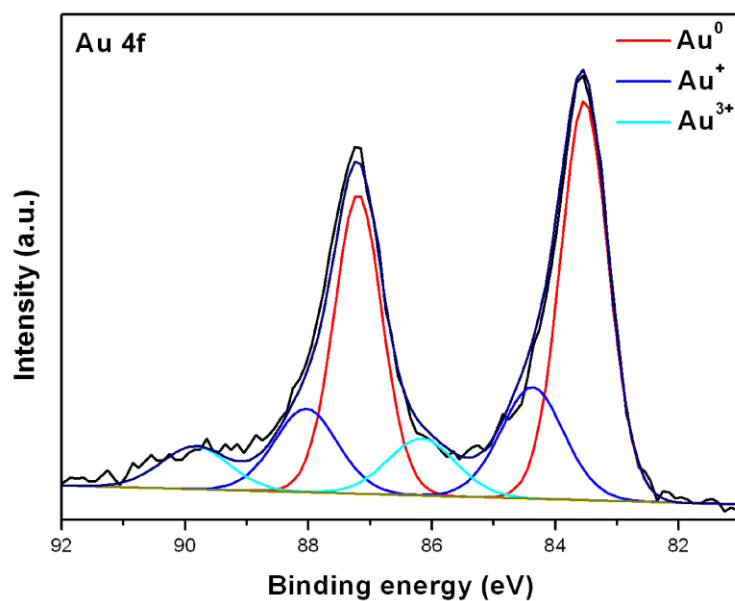


Figure S9. X-ray photoelectron spectroscopic scan survey in the region of Au 4d of gold supported on CeO₂ nanocube.

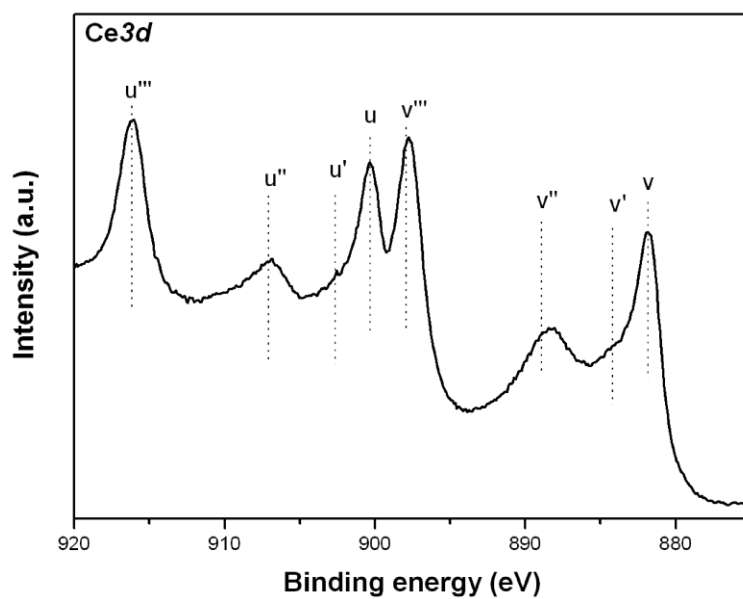


Figure S10. X-ray photoelectron spectroscopic scan survey in the region of Ce 3d of gold supported on CeO₂ nanocube.

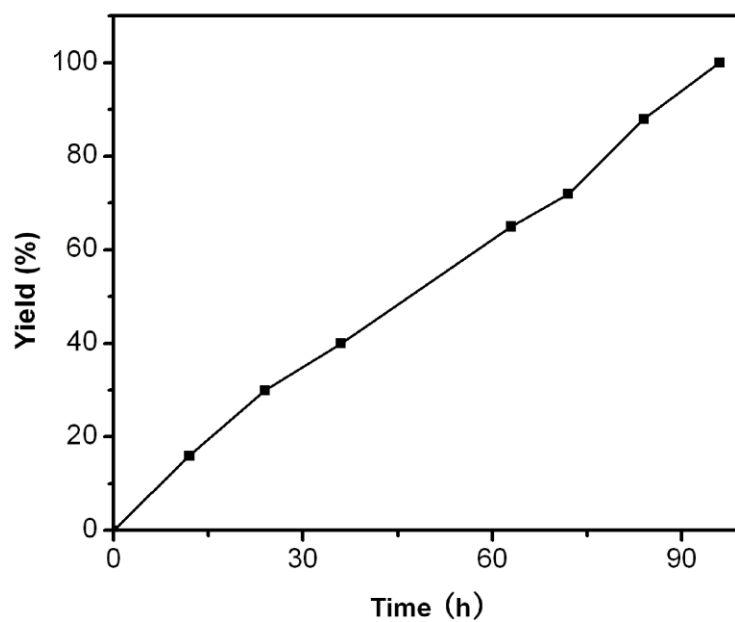


Figure S11. The reaction yield as a function of time for Au/CeO₂ nanocube catalysts.

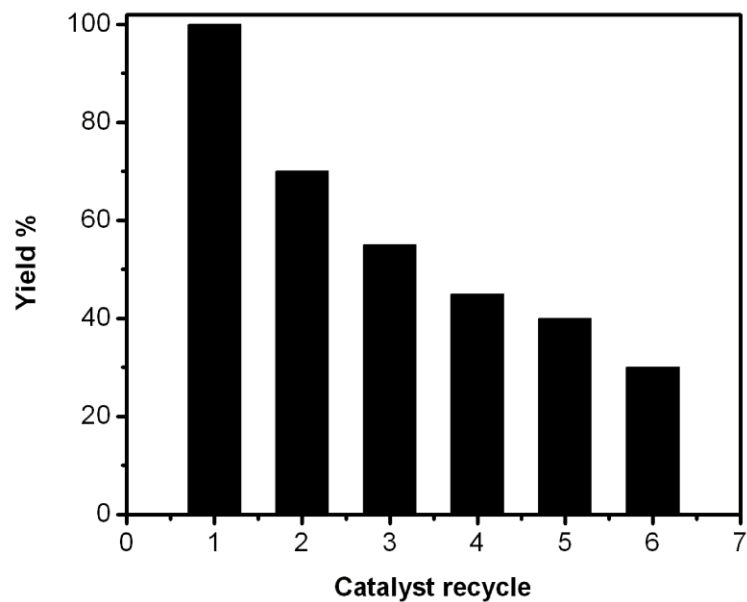


Figure S12. Au/CeO₂ nanocube catalysts for cycloisomerizations of enynes after different uses.

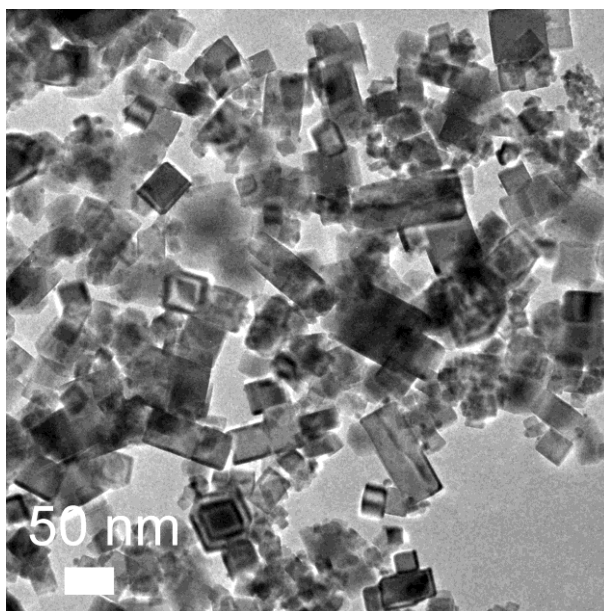


Figure S13. Transmission electron microscope (TEM) images of Au/CeO₂ nanocubes after the reaction. No leaching or aggregation is observed.

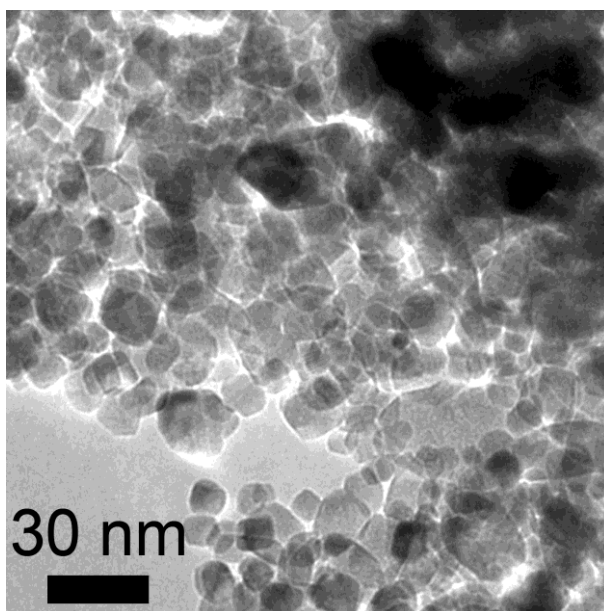
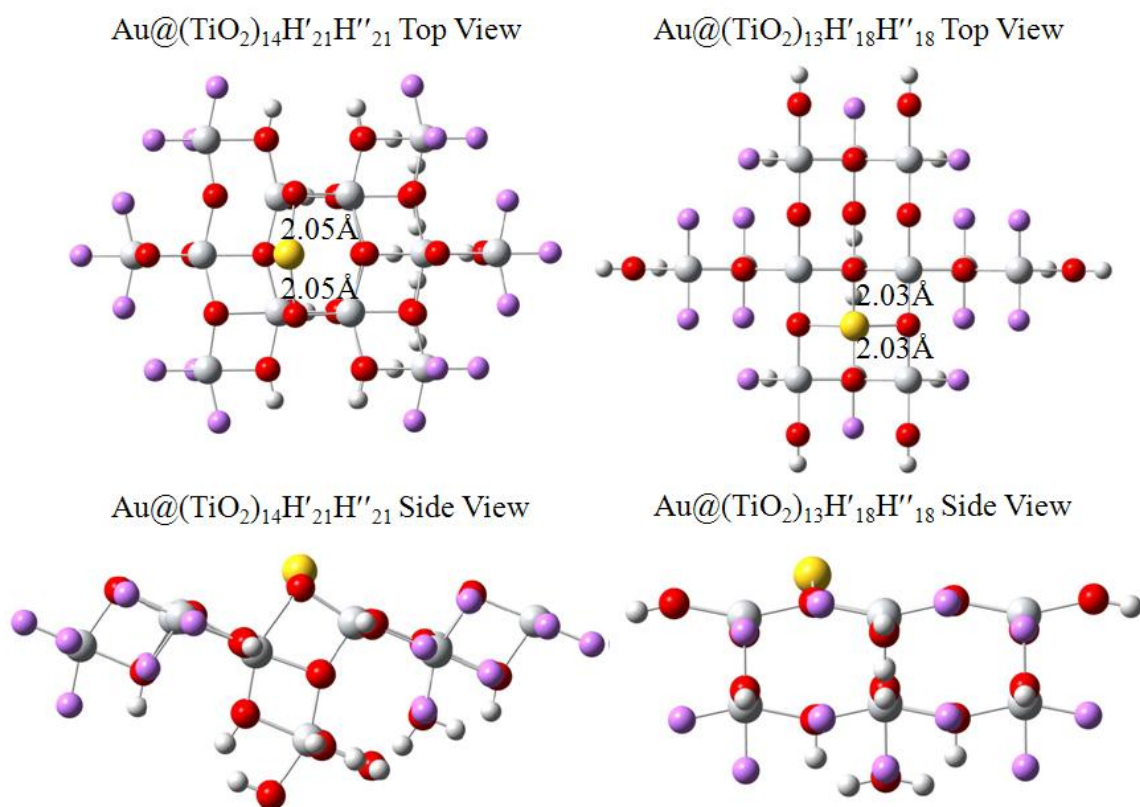


Figure S14. Transmission electron microscope (TEM) images of Au/TiO₂ nanoparticles.



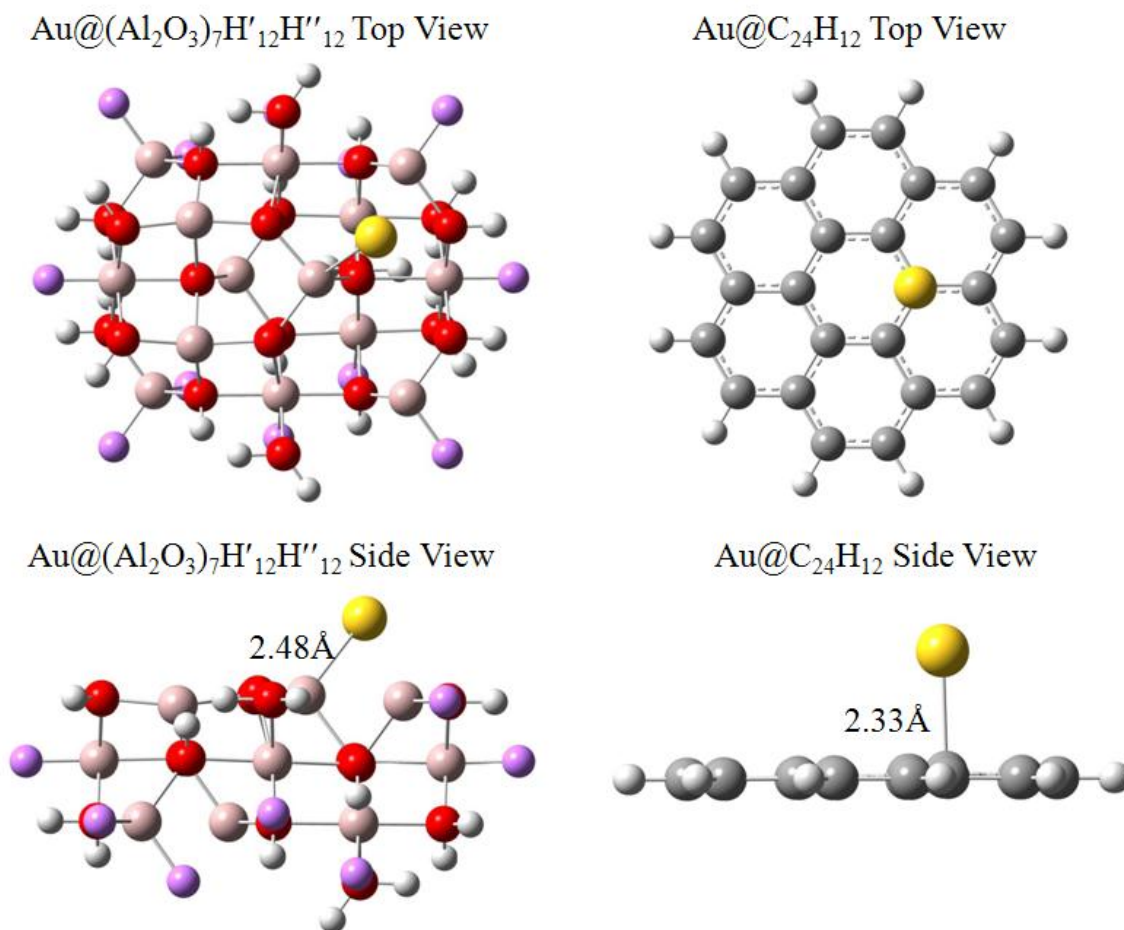


Figure S15. Calculated structures of a Au adsorption at $\text{TiO}_2(101)$, $\text{TiO}_2(001)$, $\gamma\text{-Al}_2\text{O}_3(110)$ and the surface of monolayer GP.

Table S1 The chemisorption energies of carbon monoxide on $(\text{CeO}_2)_{13}\text{H}'_{27}\text{H}''_{27}$ cluster, comparing to the data in the references.

Model	Method	Site	$r(\text{C-Ce})/\text{\AA}$	E_{ads}/eV	Reference
Slab	PW91	atop Ce	2.86	-0.17	Yang, Z. et al. <i>Chem. Phys. Lett.</i> 2004 , 396, 384.
Slab	PW91+U	atop Ce	2.86	-0.26	Nolan, M. et al. <i>Surf. Sci.</i> 2006 , 600, L175.
Slab	PW91+U	atop Ce	2.88	-0.26	Nolan, M. et al. <i>J. Phys. Chem. B</i> 2006 , 110, 16600.
Slab	PBE+U	atop Ce	2.86	-0.17	Huang, M. et al. <i>J. Phys. Chem. C</i> 2008 , 112, 8643.
Slab	PW91+U	atop Ce	2.91	-0.18	Jiang, S. Y. et al. <i>Acta Phys. -Chim. Sin.</i> 2009 , 25, 1629.
Embedded Cluster	LDA	atop Ce	2.856	-0.54	Müller, C. et al. <i>Surf. Sci.</i> 2009 , 603, 2619.
	Exp.			~-0.2	Zaki, M. et al. <i>Spectrochim. Acta</i> 1987 , 43A, 1455.
Saturated Cluster	LDA	atop Ce	2.73	-0.46	This work

Table S2 The vibrations of carbon monoxide on $(\text{CeO}_2)_{13}\text{H}'_{27}\text{H}''_{27}$ cluster, comparing to the data in the references.

Model	Method	Site	$\nu(\text{C-O})_{\text{model}}/\text{cm}^{-1}$	$\nu(\text{C-O})_{\text{gas}}/\text{cm}^{-1}$	Reference
	Exp.			2143	Street, S.C. et al. <i>Ann. Rev. Phys. Chem.</i> 1997 , 48, 43.
	Exp.			2148	Reinhardt, P. A. et al. <i>Phys. Rev. B</i> 1996 , 54, 14812.
	Exp.		2165		Bensalem, A. et al. <i>Appl. Catal. A-Gen.</i> 1995 , 121, 81.
	Exp.		2177		Li, C. et al. <i>J. Chem. Soc. Faraday Trans.</i> 1989 , 185, 929.
Slab	PW91	atop Ce	2082	2076	Yang, Z. et al. <i>Chem. Phys. Lett.</i> 2004 , 396, 384.
Slab	PW91+U	atop Ce	2128	2103	Nolan, M. et al. <i>Surf. Sci.</i> 2006 , 600, L175.
Saturated Cluster	LDA	atop Ce	2127	2124	This work

Table S3. The state of charge of gold atom on $(\text{CeO}_2)_{13}\text{H}'_{27}\text{H}''_{27}$ cluster.

Model	Method	Site	q_{Au}	E_{ads}/eV	Reference
Slab	PW91+U	bridge O	+0.33	-1.15	Hernandez, N. C. et al. <i>Phys. Chem. Chem. Phys.</i> 2009 , 11, 5246.
Slab	PW91+U	bridge O	+0.32	-1.17	Zhang, C. J. et al. <i>J. Chem. Phys.</i> 2008 , 129, 194708.
Slab	PBE	atop O	+0.35	-1.26	Liu, Z. P. et al. <i>Phys. Rev. Lett.</i> 2005 , 94, 196102.
Saturated Cluster	LDA	atop O	+0.28	-1.44	This work

Table. S4. Gold loading and XPS Au4f peak analyses for Au on CeO_2 , TiO_2 , Al_2O_3 , C and grapheme.

Entry	Catalyst	Au (wt %)	XPS Au4f peak		
			Au^0	Au^+	Au^{3+}
1	Au/ CeO_2	1.1	64%	23%	13%
2	Au/ TiO_2	1.3	73%	14%	13%
3	Au/ Al_2O_3	1.2	83%	9%	8%
4	Au/C	0.9	85%	10%	5%
5	Au/GP	0.9	100%	—	—

Table S5. The charge state distribution of gold atom on the CeO_2 (100), TiO_2 (101), TiO_2 (001), γ - Al_2O_3 (110) and the surface of monolayer GP.

Support	Model	Site	$r(\text{Au-O/Al/C})/\text{\AA}$	q_{Au}
CeO_2 (100)	$(\text{CeO}_2)_{14}\text{H}'_{36}\text{H}''_{12}$	hollow O	1.95	+0.61
TiO_2 (101)	$(\text{TiO}_2)_{14}\text{H}'_{21}\text{H}''_{21}$	bridge O	2.05	+0.32
TiO_2 (001)	$(\text{TiO}_2)_{13}\text{H}'_{18}\text{H}''_{18}$	bridge O	2.03	+0.31
Al_2O_3 (110)	$(\text{Al}_2\text{O}_3)_7\text{H}'_{12}\text{H}''_{12}$	atop Al	2.48	+0.10
GP	$\text{C}_{24}\text{H}_{12}$	atop C	2.33	+0.03