Supporting Information

Eco-friendly synthesis of size-controllable amine-functionalized

graphene quantum dots with antimycoplasma property

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Content:

solution.

Figure S1. Sephadex G-25 gel column chromatography for the separation of GQDs (left) and photograph of F2 dispersed in PBS, saline, DMEM and FBS at concentration of 0.5 mg/mL. Figure S2. UV-Vis absorption spectra (left) and fluorescent spectra (right) of F1 in aqueous

Figure S3.XPS spectra of GO and GQDs (a), and curve fitting for N 1s binding energy of GQDs (b), curve fitting for C 1s of GQDs (c) and GO (d).

Table S1 Percentage of chemical elements in GO and GQDs

Table S2 Fitting of C1s binding energy for GO and GQDs.

Figure S4 (a) FT–IR spectra and (b) Raman spectra of GO and GQDs.

Figure S5 ¹H NMR spectra of GO (a) and GQDs (b) in DMSO-d₆ with proposed structural

fragment in inset.



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Figure S3.XPS spectra of GO and GQDs (a), and curve fitting for N 1s binding energy of GQDs (b), curve fitting for C 1s of GQDs (c) and GO (d).

	Table S1	Percentage	of	chemical	elements	in	GO	and	GÇ)Ds
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Sample	C (at. %)	O (at. %)	N (at. %)
GO	70.9	29.1	0
GQDs	56.6	34.5	8.9

Table S2 Fitting of C1s binding energy for GO and GQDs.

Bond (C1s)	Binding energy (eV)	GO	GQDs
C-C	284.8±0.1	56%	44%
C-O, C-N	286.5±0.3	35%	37%
C=O, O=C-R (R=NH ₂ , OH)	288.2-290	9%	19%



Figure S4 (a) FT-IR spectra and (b) Raman spectra of GO and GQDs.



Figure S5 1 H NMR spectra of GO (a) and GQDs (b) in DMSO-d₆ with proposed structural fragment in inset.