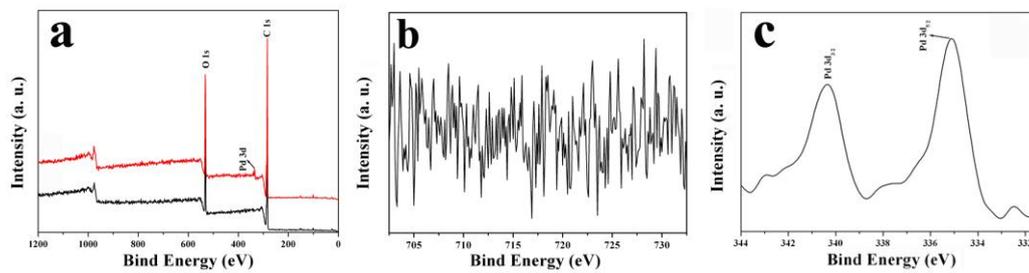


## Supplementary Information

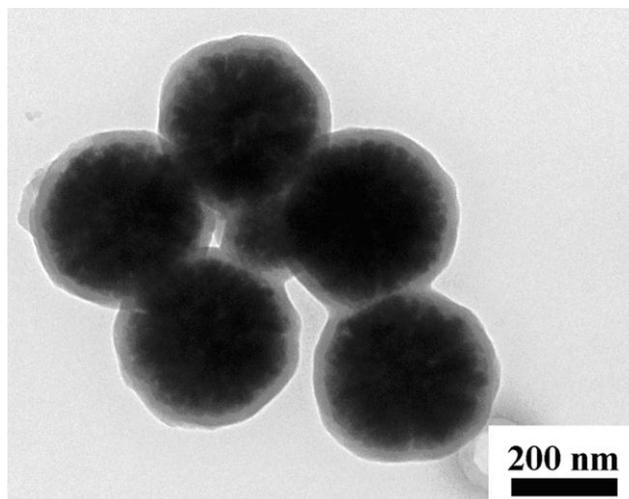
# Preparation of yolk/shell $\text{Fe}_x\text{O}_y/\text{Pd}$ @mesoporous $\text{SiO}_2$ composites with high stability and their application in catalytic reduction of 4-nitrophenol

Tongjie Yao,<sup>ac</sup> Tieyu Cui,<sup>\*a</sup> Xue Fang,<sup>a</sup> Fang Cui<sup>a</sup> and Jie Wu<sup>\*b</sup>

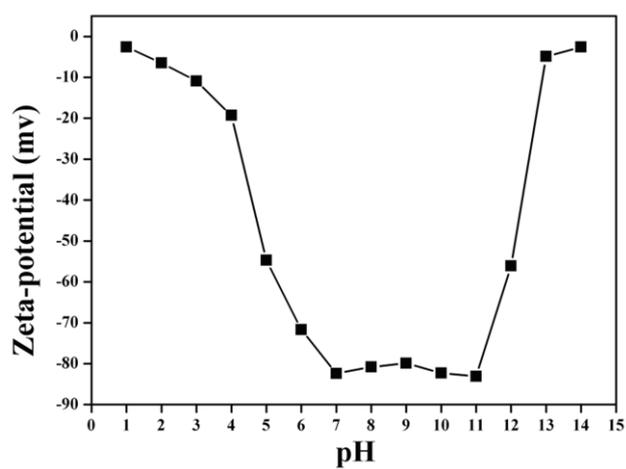
**Fig. S1** (a) Survey XPS spectra of Fe<sub>3</sub>O<sub>4</sub>@C (black line) and Fe<sub>3</sub>O<sub>4</sub>@C/Pd (red line) samples; (b) high-resolution Fe 2p XPS spectrum of Fe<sub>3</sub>O<sub>4</sub>@C composites; (c) high-resolution Pd 3d XPS spectrum of Fe<sub>3</sub>O<sub>4</sub>@C/Pd composites.



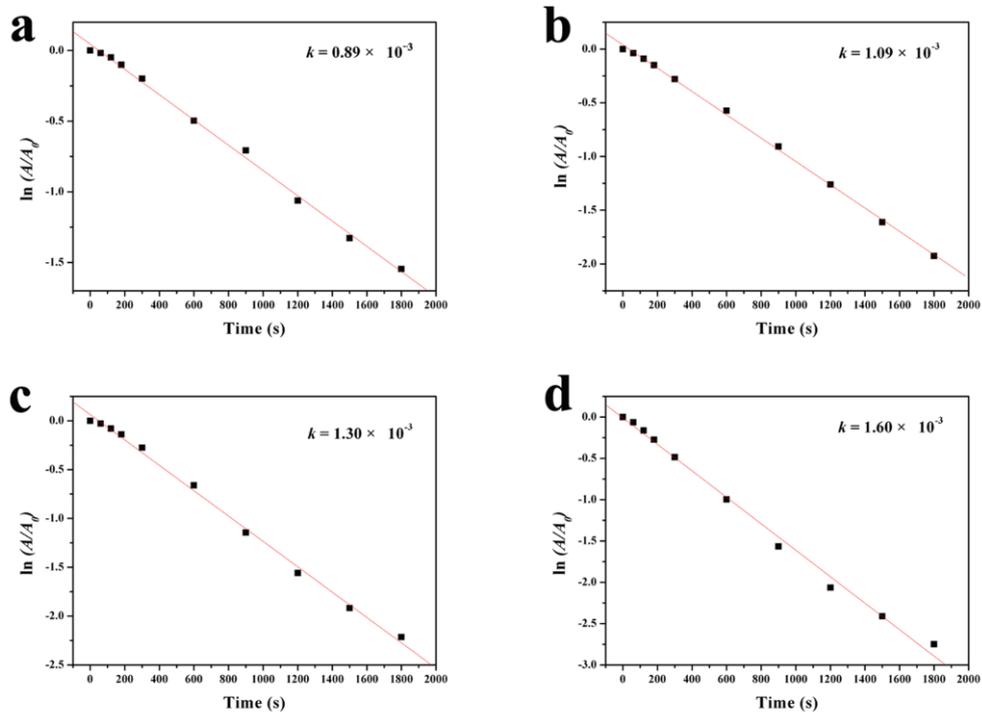
**Fig. S2** TEM image of  $\text{Fe}_3\text{O}_4@\text{C}$  composites prepared with glucose dosage of 5.0 g.  
The thickness of carbon shell is around 25 nm.



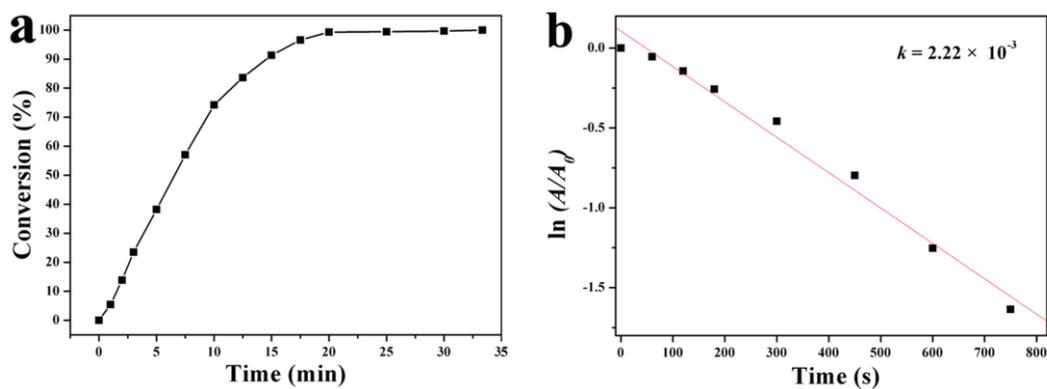
**Fig. S3** Zeta-potentials of the  $\text{Fe}_3\text{O}_4@\text{C}$  composites in aqueous solution at different pH values. When pH value was adjusted above 12, plenty of hydroxyl ions were introduced into the solution, which resulted in the increase of ion concentration in solution and compression of the electric double layer; and hence, the absolute value of zeta-potential gradually reduced.



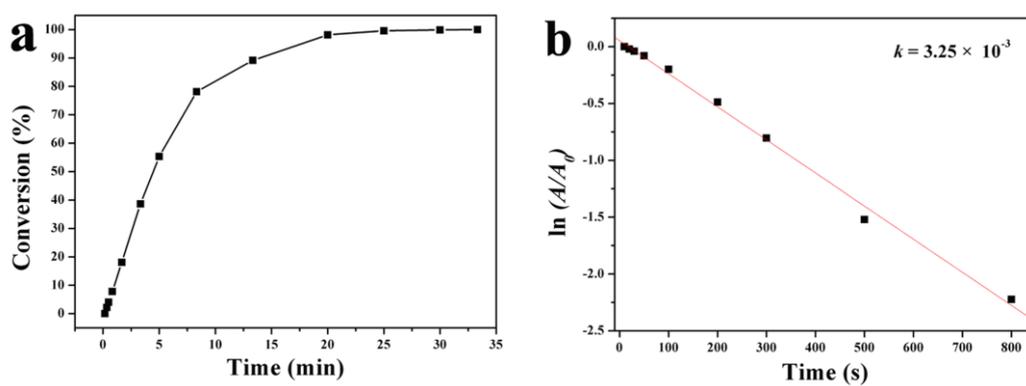
**Fig. S4** The rate constant  $k$  estimated by the slopes of straight lines of  $\ln(A_t/A_0)$  vs. reduction time using  $\text{Fe}_x\text{O}_y/\text{Pd}@m\text{SiO}_2$  catalysts prepared with different Pd nanoparticles loading: (a) 0.43 wt%; (b) 1.1 wt%; (c) 1.4 wt%; (d) 1.9 wt%.



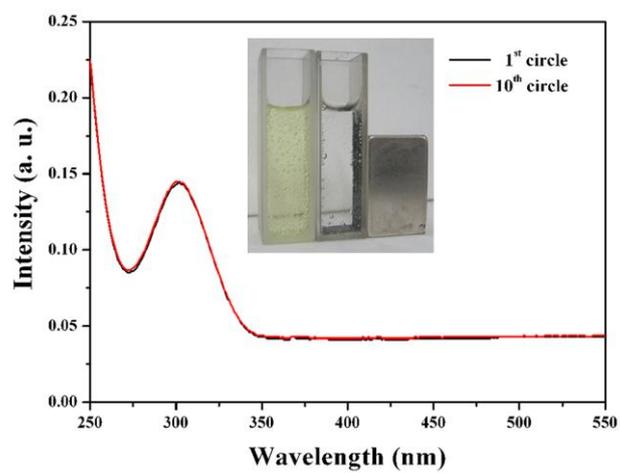
**Fig. S5** (a) Conversion of 4-NP using obtained Pd@mSiO<sub>2</sub> composites as catalysts (0.1 mg) for various time; (b) The rate constant  $k$  estimated by the slopes of straight lines of  $\ln(A_t/A_0)$  vs. reduction time.



**Fig. S6** (a) Conversion of 4-NP using  $\text{Fe}_3\text{O}_4@\text{C}/\text{Pd}$  composites with the same Pd nanoparticle loading as corresponding  $\text{Fe}_x\text{O}_y/\text{Pd}@m\text{SiO}_2$  catalysts for various time; (b) The rate constant  $k$  estimated by the slopes of straight lines of  $\ln(A_t/A_0)$  vs. reduction time.



**Fig. S7** UV-Vis spectra of 4-NP reduction with  $\text{Fe}_x\text{O}_y/\text{Pd@mSiO}_2$  catalysts in different cycles. Inset showed the digital image of reaction solution after reaction for 10 times



**Fig. S8** (a) SEM image of  $\text{Fe}_x\text{O}_y/\text{Pd}@m\text{SiO}_2$  composites after used in catalytic tests for 10 times. The inset showed the magnified image. The yolk/shell structure could be confirmed from the ruptured  $m\text{SiO}_2$  shell which was indicated by arrow. (b) TEM image of obtained  $\text{Pd}@m\text{SiO}_2$  composites which were prepared by dissolution of magnetic core in corresponding  $\text{Fe}_x\text{O}_y/\text{Pd}@m\text{SiO}_2$  samples (*Fig. S8a*). The inset showed the magnified image. It could be seen that both the size and dispersity of Pd nanoparticles were well preserved.

