

Protein-directed synthesis of pH-responsive red fluorescent copper nanoclusters and applications in cellular imaging and catalysts

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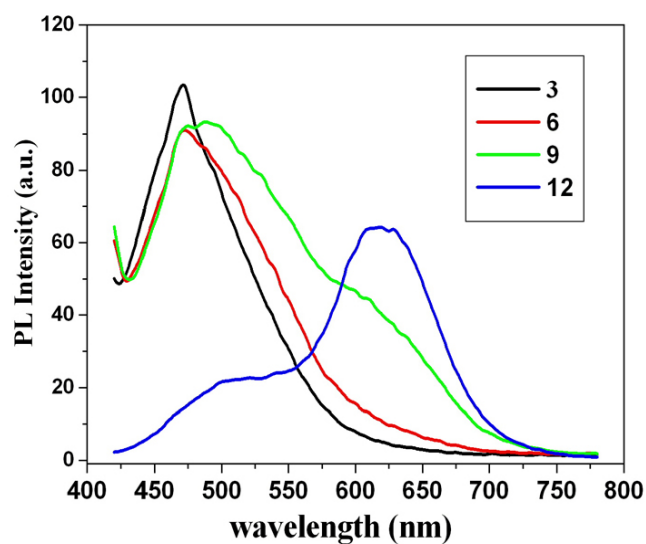


Fig. S1 The fluorescence spectra of resultant BSA-Cu NCs under different pH of reaction solution

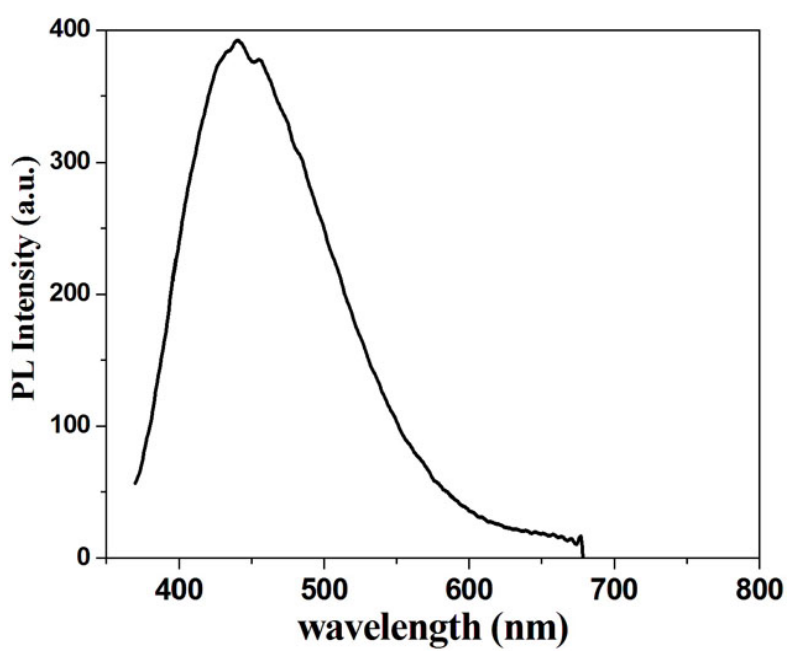


Fig S2 The PL spectrum of BSA without copper source

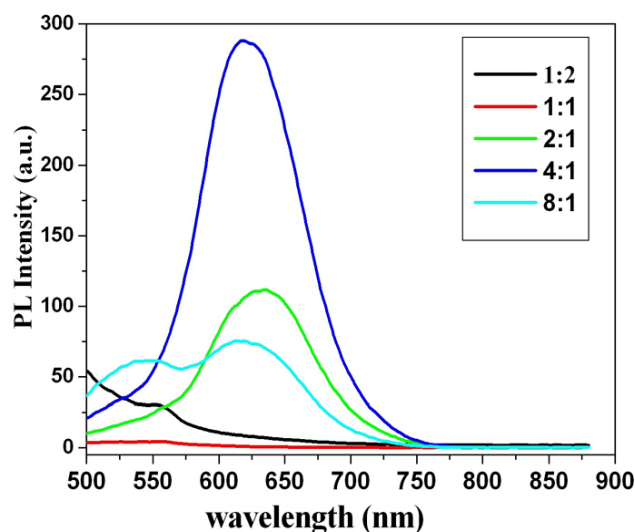


Fig. S3 The fluorescence spectra of resultant BSA-Cu NCs under different the molar ratio of BSA and Cu ions

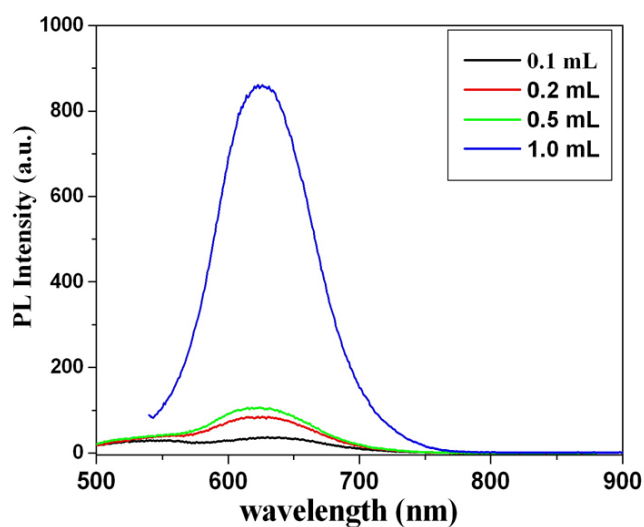


Fig. S4 The fluorescence spectra of resultant BSA-Cu NCs under different the amount of $\text{N}_2\text{H}_4 \cdot 2\text{H}_2\text{O}$

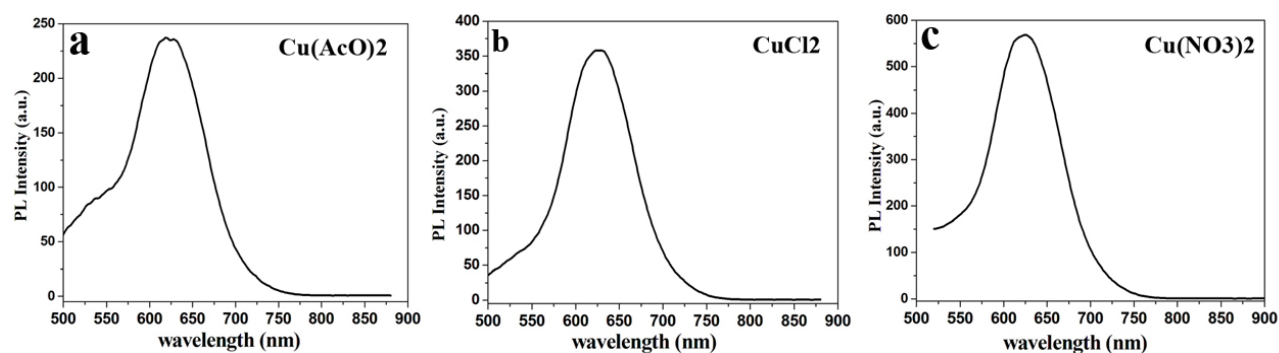


Fig. S5 The fluorescence spectra of resultant BSA-Cu NCs under different the Cu source including $\text{Cu}(\text{AcO})_2$, CuCl_2 , and $\text{Cu}(\text{NO}_3)_2$

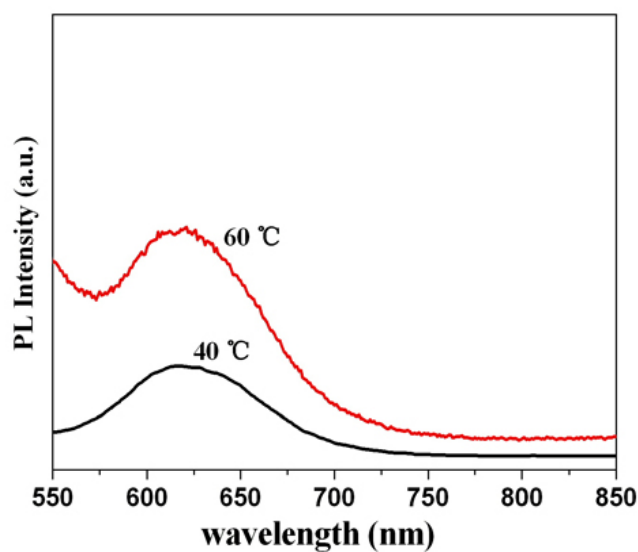


Fig. S6 The fluorescence spectra of resultant BSA-Cu NCs under different the reaction temperature

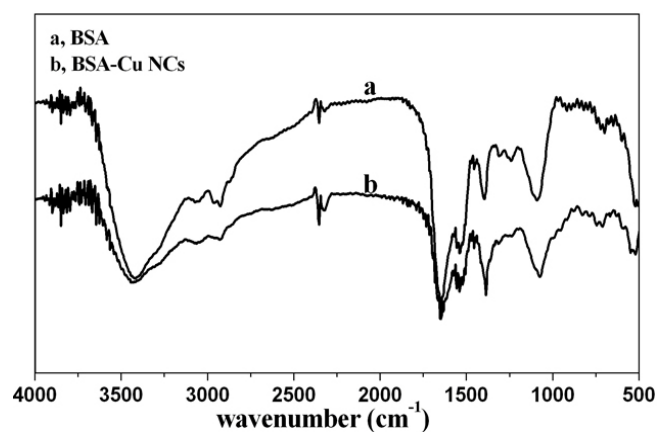


Fig. S7 The FTIR spectra of the free BSA and BSA-Cu NCs bioconjugates.

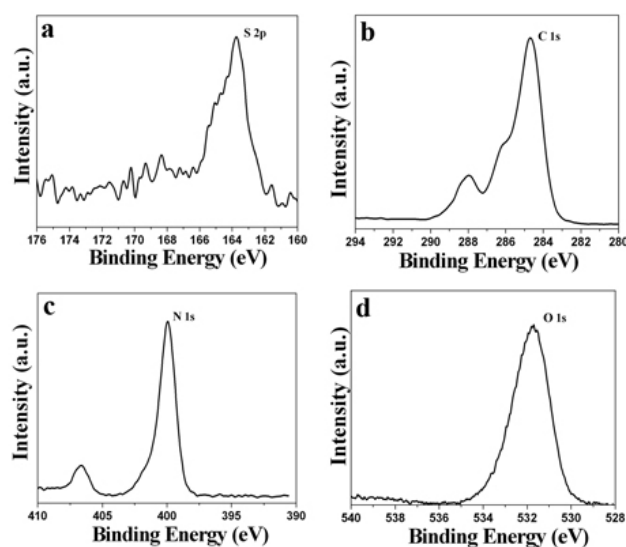


Fig. S8. The XPS spectra of the binding energy of S 2p, C 1s, N 1s, and O 1s

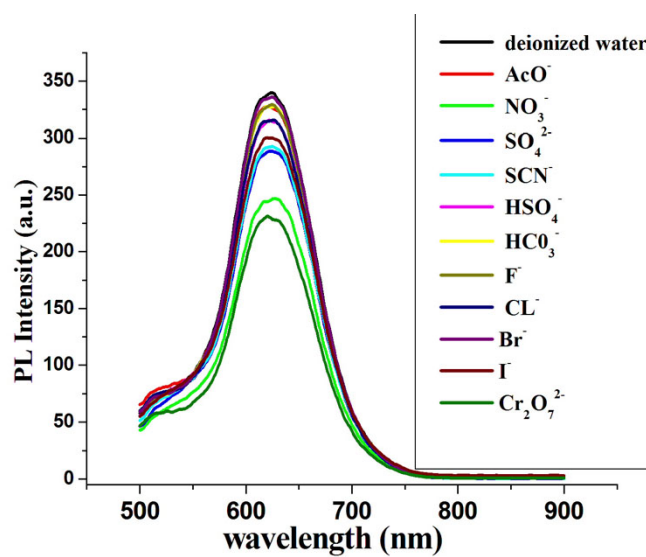


Fig. S9 The fluorescence spectra of BSA-Cu NCs in various common anions in detail

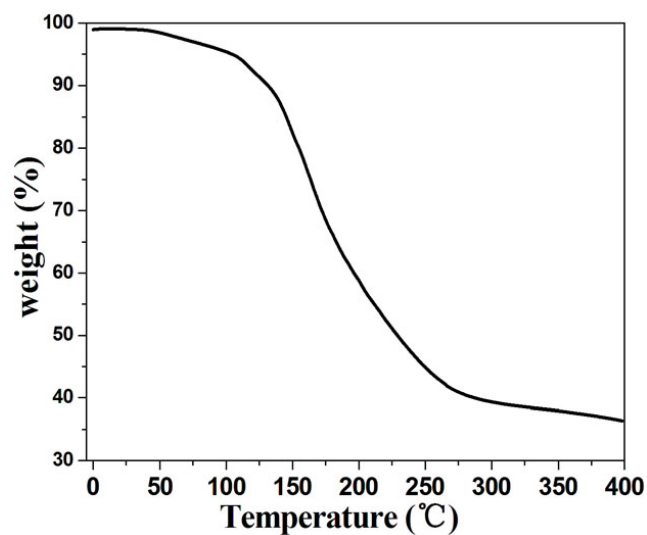


Fig. S10. Thermal gravimetric analysis of as-prepared BSA-Cu NCs