

**An efficient one-step condensation and activation strategy to
synthesize porous carbons with optimal micropore sizes for highly
selective CO₂ adsorption**

Jiacheng Wang* and Qian Liu*

State Key Laboratory of High Performance Ceramics and Superfine Microstructure, Shanghai
Institute of Ceramics, Chinese Academy of Sciences, Shanghai 200050, P. R. China. Tel:
0086-21-52412612; Email: jiacheng.wang@mail.sic.ac.cn; qianliu@sunm.shenc.ac.cn

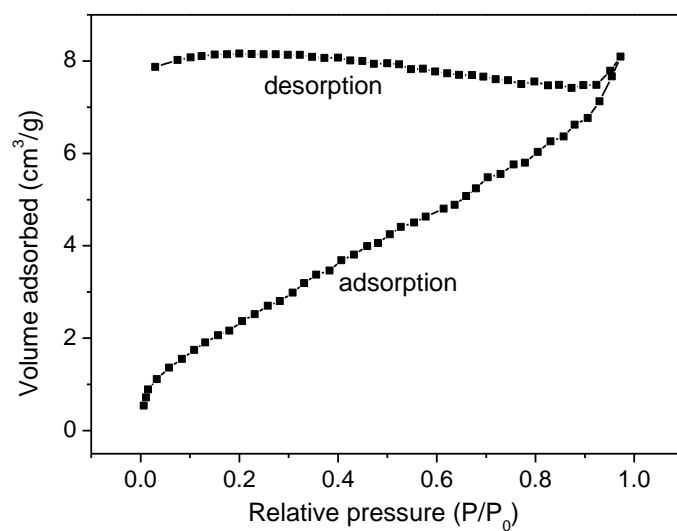


Fig. S1 Nitrogen adsorption-desorption isotherms of the resulting carbon prepared by direct carbonization of the ground *m*-phenylenediamine and terephthalaldehyde mixture. The specific surface area determined by the BET method is $\sim 7 \text{ m}^2/\text{g}$.

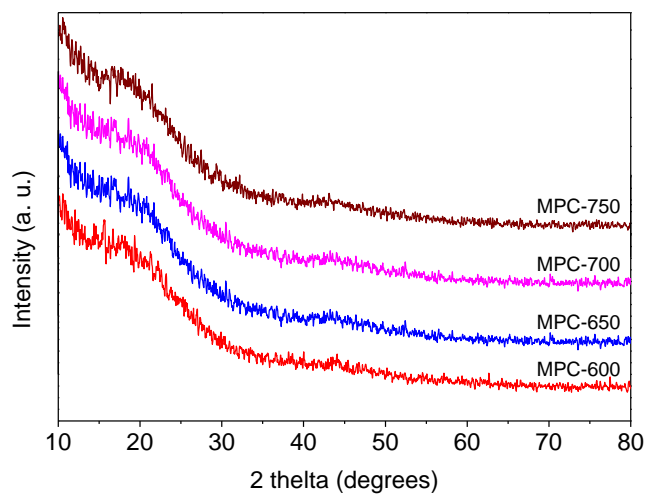


Fig. S2 Powder XRD patterns of various MPCs prepared at different temperatures.

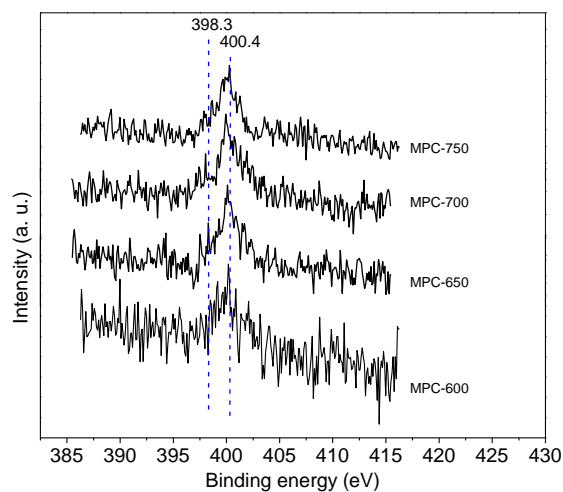


Fig. S3 N1s XPS spectra of various MPCs prepared at different temperatures.

Table S1. Elemental analysis of various MPCs determined by XPS analysis.

Sample	N [wt%]	C [wt%]
MPC-600	0.66	93.3
MPC-650	0.81	91.1
MPC-700	0.78	92.7
MPC-750	0.33	96.3

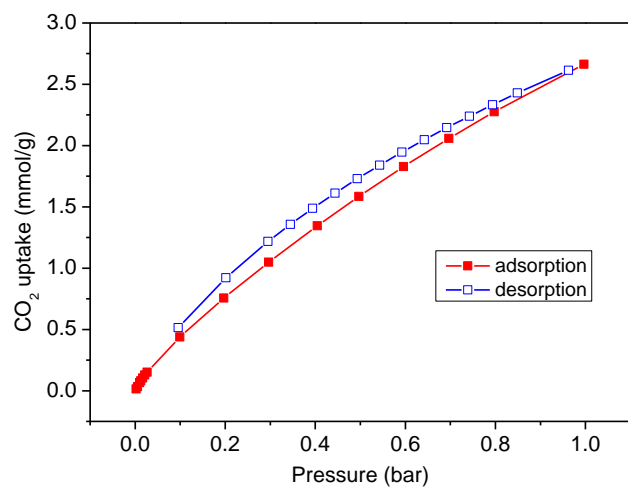


Fig. S4 CO₂ sorption isotherms of MPC-600 at 25 °C.

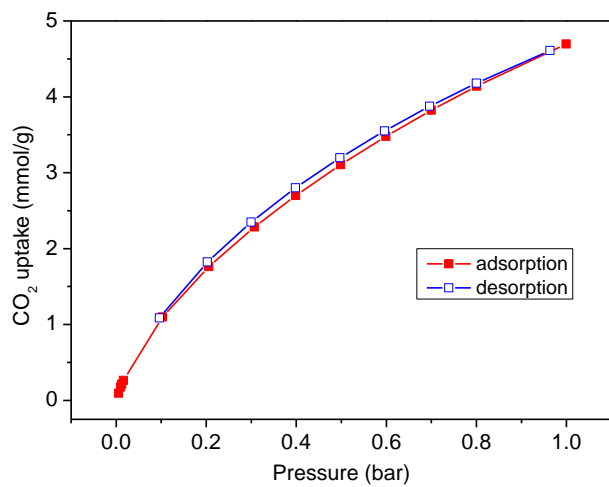


Fig. S5 CO₂ sorption isotherms of MPC-600 at 0 °C.

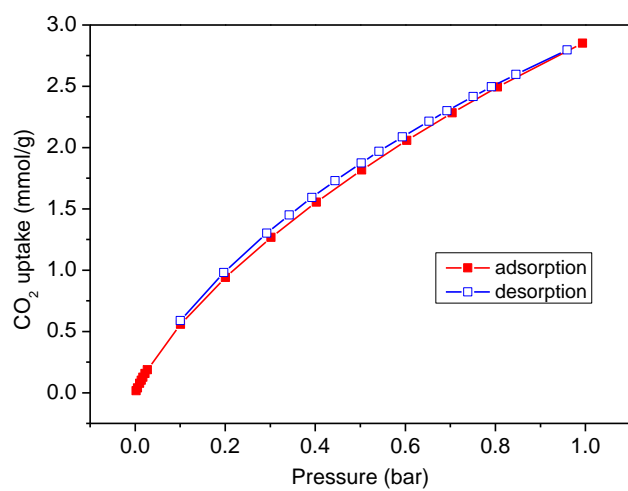


Fig. S6 CO₂ sorption isotherms of MPC-650 at 25 °C.

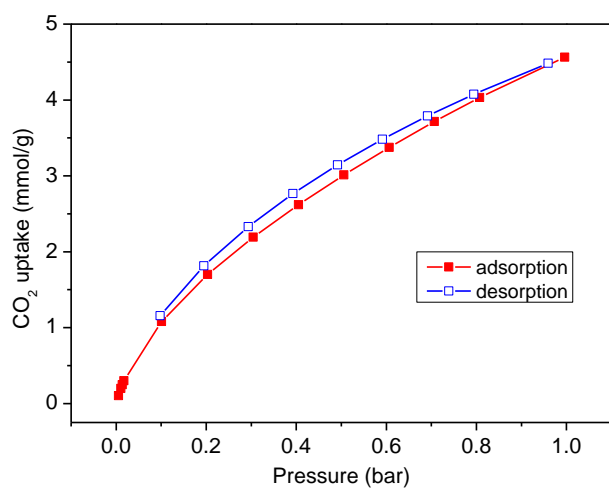


Fig. S7 CO₂ sorption isotherms of MPC-650 at 0 °C.

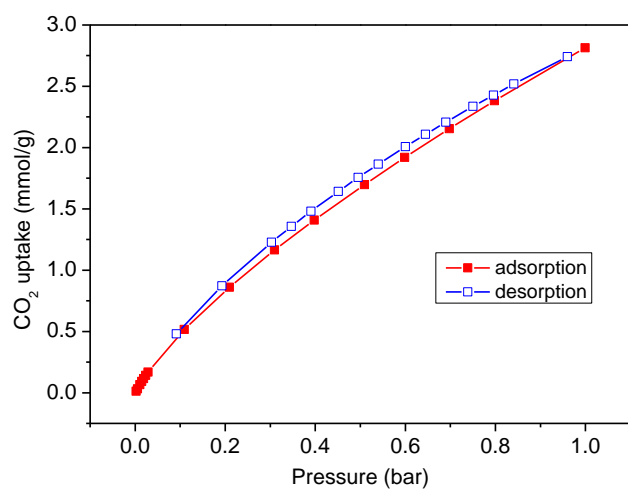


Fig. S8 CO₂ sorption isotherms of MPC-700 at 25 °C.

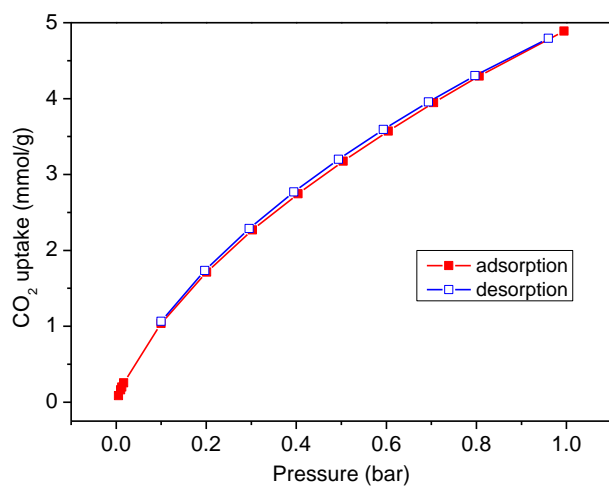


Fig. S9 CO₂ sorption isotherms of MPC-700 at 0 °C.

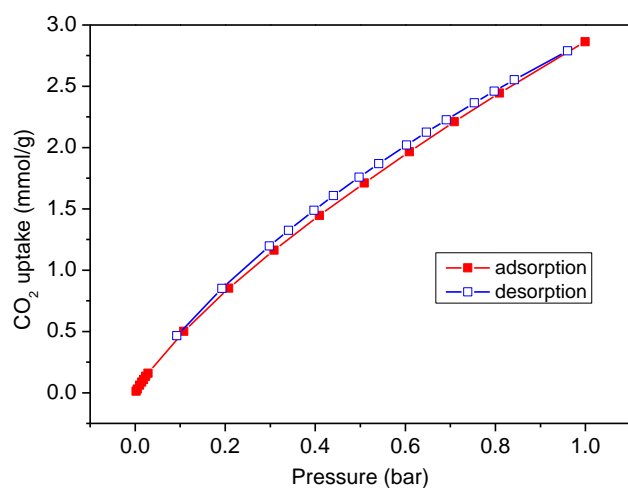


Fig. S10 CO₂ sorption isotherms of MPC-750 at 25 °C.

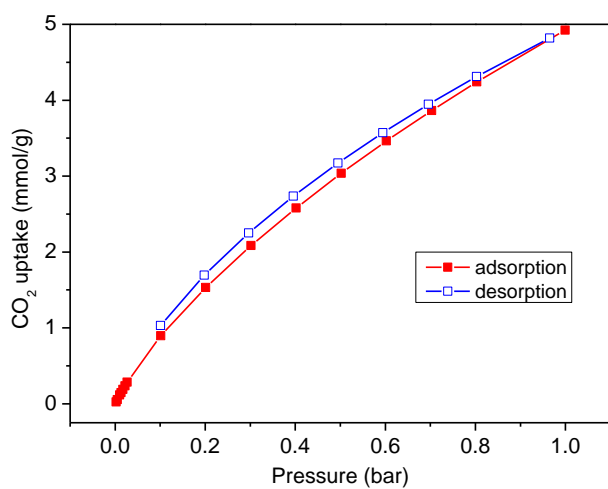


Fig. S11 CO₂ sorption isotherms of MPC-750 at 0 °C.

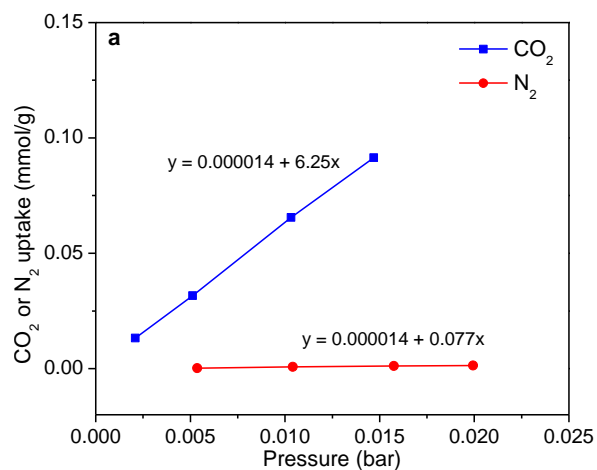


Fig. S12 Initial slopes from CO₂ and N₂ adsorption isotherms at 25 °C for MPC-700.

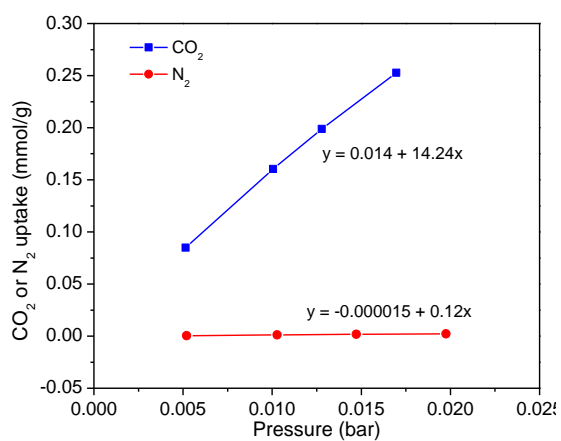


Fig. S13 Initial slopes from CO₂ and N₂ adsorption isotherms at 0 °C for MPC-700.