## Supplementary Information

One-pot Hydrothermal Synthesis of Peony-like Ag/Ag<sub>0.68</sub>V<sub>2</sub>O<sub>5</sub> Hybrid as

High-Performance Anode and Cathode Materials for Rechargeable

Lithium Batteries

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Figure S1. EDS of the  $Ag/Ag_{0.68}V_2O_5$  hybrid.



Figure S2 The first three consecutive CVs of the composite at a scan rate of 0.1 mV s<sup>-1</sup> between 1.5 and 3.5 V (a);  $0.01 \sim 3.00$  V (b).

Method	Composition	Current density (mA/g)		Capacity	(mAh/g)	Capacity retention (cycle	
				initial-last		number)	
		Cathode	Anode	Cathode	Anode	Cathode	Anode
Based on β-AgVO <sub>3</sub> [9]	β-AgVO <sub>3</sub> /PANI	30		211-131		62%(20)	
Substrate-assisted [10]	$\beta$ -AgVO <sub>3</sub> cluster	100		220-100		45%(50)	
Hydrothermal [S1]	$\beta$ -Ag <sub>0.33</sub> V <sub>2</sub> O <sub>5</sub> nanowire	20		240-160		67% (8)	
Solid approach [17]	Ag/AgVO3 nanorod	50		242-111		46%(30)	
This work	$Ag/Ag_{0.68}V_2O_5$ flower	100	400	321-150	499-524	47%(65)	105%(1000)

**Table S1.** Electrochemical properties of SVOs prepared by different methods.

[S1] Hu, W.; Zhang, X.; Cheng, Y.; Wu, C.; Cao, F.; Wang, L., ChemSusChem 2011, 4, 1091.



Figure S3 SEM images of the  $Ag/Ag_{0.68}V_2O_5$  hybrid after a few cycles as a cathode material.



Figure S4 TEM images of the  $Ag/Ag_{0.68}V_2O_5$  hybrid with different reaction time a) 1 h; b) 6 h; c) 12 h; and d) 24 h.



Figure S5 Electrochemical impedance spectrum of the samples with different reaction time.