

Cluster-size dependent internal dynamics and magnetic anisotropy of Ho ions in HoM₂N@C₈₀ and Ho₂MN@C₈₀ families (M = Sc, Lu, Y)

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1. Synthesis and isolation of the $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ and $\text{Ho}_x\text{Lu}_{3-x}\text{N@C}_{80}$ ($x = 1, 2$)

Synthesis

Ho/Lu and Ho/Y MMNCFs were synthesized by the “selective organic solid” (SOS) route as described previously. Briefly, a mixture of Ho_2O_3 , Y_2O_3 or Lu_2O_3 (99.9%, MaTeck GmbH, Germany), guanidine thiocyanate (GT) and graphite powder was used (molar ratio $\text{Ho/M/GT/C}=1/1/2.5/15$, $\text{M} = \text{Lu}$ or Y). After dc-arc discharging, the soot was pre-extracted by acetone and further Soxhlet-extracted by CS_2 for 20h.

Isolation and Characterization

The isolation of Ho-based MMNCFs was performed by two-step HPLC. At the first step a linear combination of two analytical 4.6×250 mm Buckyprep columns (Nacalai Tesque, Japan) was applied on a Hewlett-Packard instrument (series 1100) with toluene as the eluent. Further isolation was performed by a recycling HPLC (Sunchrom, Germany) using a Buckyprep column (10×250 mm; Nacalai Tesque, Japan) and toluene as the eluent. The UV detector set to 320 nm was employed for fullerene detection for all steps. The details of HPLC isolation of Ho-based MMNCFs are described in the Supporting Information. The purity of the isolated products was checked by laser desorption/ionization time-of-flight (LDI-TOF) mass spectrometry (Biflex III, Bruker, Germany), the spectra are shown in supporting information as well. UV-Vis-NIR absorption spectra were measured in toluene solution using Shimadzu 3100 spectrometer.

1.1 Synthesis and isolation of $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ ($x=1, 2$)

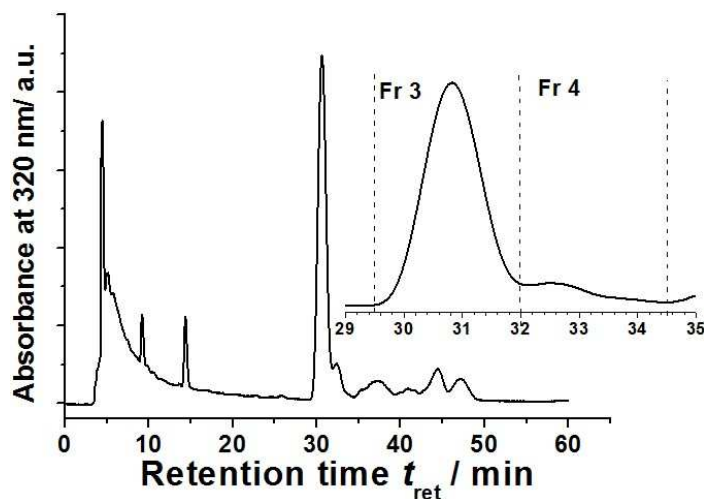


Figure S1. Chromatogram of a raw $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{2n}$ fullerenes extract synthesized by the “selective organic solid” method (linear combination of two 4.6×250 mm Buckyprep columns, flow rate 1.6 ml/min, injection volume 200 μL , toluene as mobile phase, 40 °C). The inset shows the enlarged chromatographic region of 29.5–34.5 min.

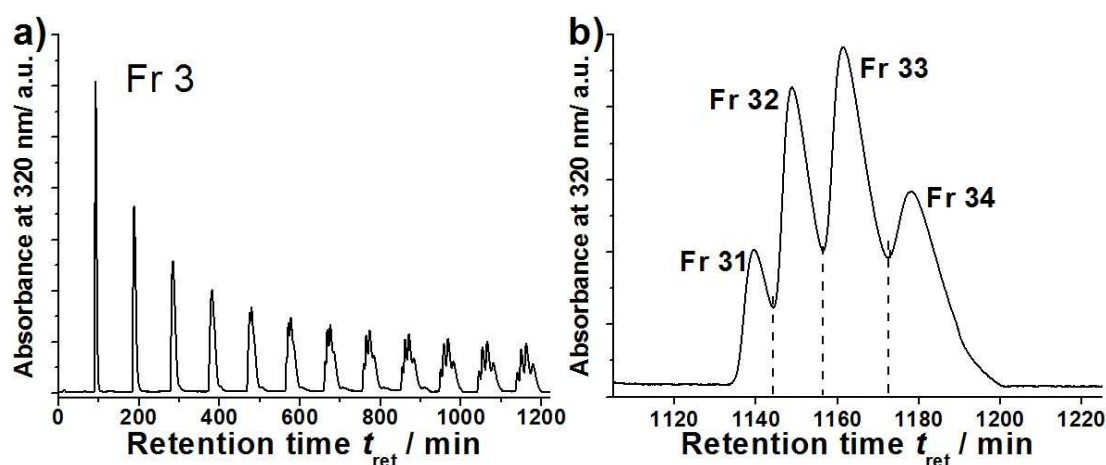


Figure S2. The HPLC isolation of fraction Fr 3. (10×250 mm Buckyprep column; flow rate 1.5 ml/min; injection volume 5 ml; toluene as eluent; 20 °C).

The synthesis of $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ ($x = 1, 2$) was achieved by “selective organic solid” route. The process of dc-arc discharging and solution extraction was the same in the production of $\text{Ho}_x\text{Sc}_{3-x}\text{N@C}_{80}$ (I; $x = 1, 2$). A mixture of Ho_2O_3 and Y_2O_3 (99.9%, MaTeck GmbH, Germany), guanidine thiocyanate (GT) and graphite powder was used (molar ratio $\text{Ho}/\text{Y}/\text{GT}/\text{C} = 1:1:2.5:15$). The chromatogram of the extracted $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{2n}$ fullerenes is shown in Figure S1. The $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ (I, $x = 1, 2$) were isolated by multistep HPLC (see Figure S2-5). Firstly, the analytical HPLC was employed to collect $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ (I) (Fraction 3) and $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ (II) (Fraction 4) respectively. Different to $\text{Ho}_x\text{Sc}_{3-x}\text{N@C}_{80}$ (I, $x = 1, 2$), the retention time of $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ (I) in Buckyprep column (4.6×250 mm) are identical. Secondly, the Fr 3 was subjected to isolation by recycling HPLC on a Buckyprep column (10×250 mm), see Figure S2. After 12 cycles, four sub-fractions could be obtained which marked as Fr 31 to Fr 34. The relative yield of $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ ($x = 0-3$) could be estimated from the integrated areas of the corresponding chromatographic peaks which agrees well with mass spectrum result of Fr 3 (Figure S 2b).

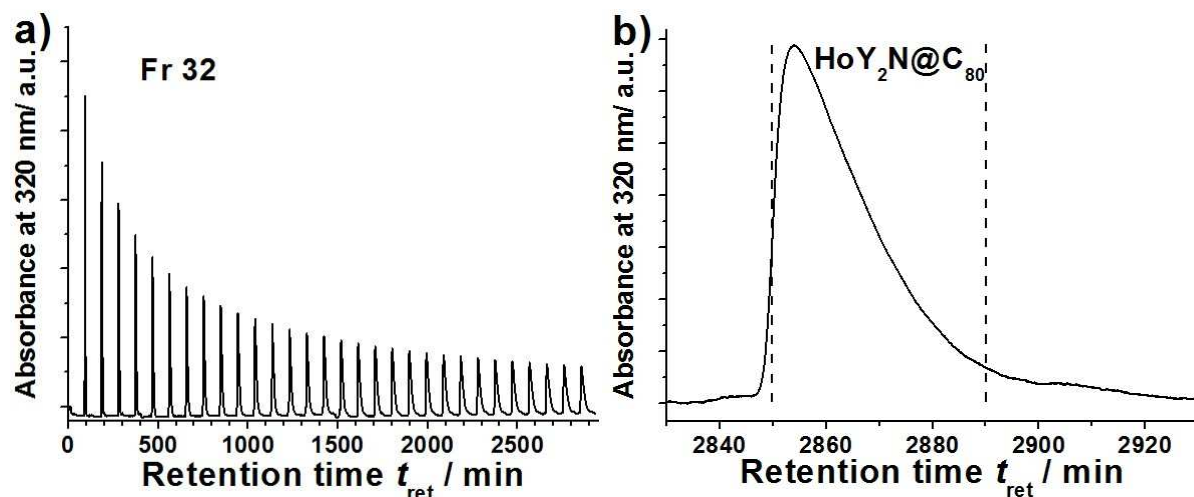


Figure S3. The isolation of fraction Fr 32. (10×250 mm Buckyprep column; flow rate 1.5 ml/min; injection volume 5 ml; toluene as eluent; 20 °C).

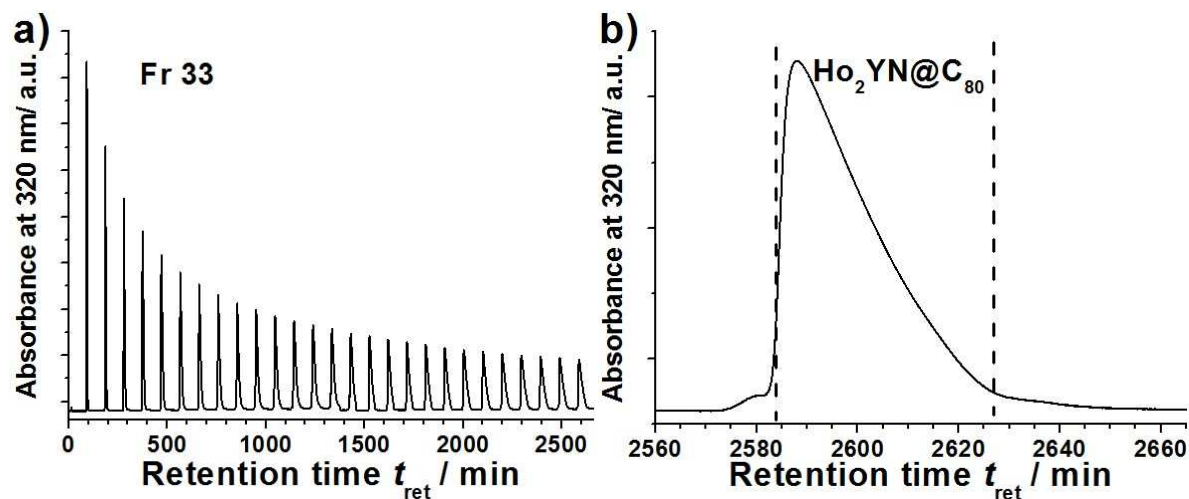


Figure S4. The isolation of fraction Fr 33. (10×250 mm Buckyprep column; flow rate 1.5 ml/min; injection volume 5 ml; toluene as eluent; 20 °C).

Isolation of $\text{HoY}_2\text{N@C}_{80}$ (I) was accomplished by removing the small amount of $\text{Y}_3\text{N@C}_{80}$ (I) and $\text{Ho}_2\text{YN@C}_{80}$ (I) from fraction 32 after 30 cycles. Similarly, the pure $\text{Ho}_2\text{YN@C}_{80}$ (I) could be obtained by removing the minor structures ($\text{HoY}_2\text{N@C}_{80}$ (I) and $\text{Ho}_3\text{N@C}_{80}$ (I)) in fraction 33 through 27 cycles. The purity of $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ (I, $x = 1, 2$) were confirmed by LDI-TOF mass spectroscopy (Figure S5).

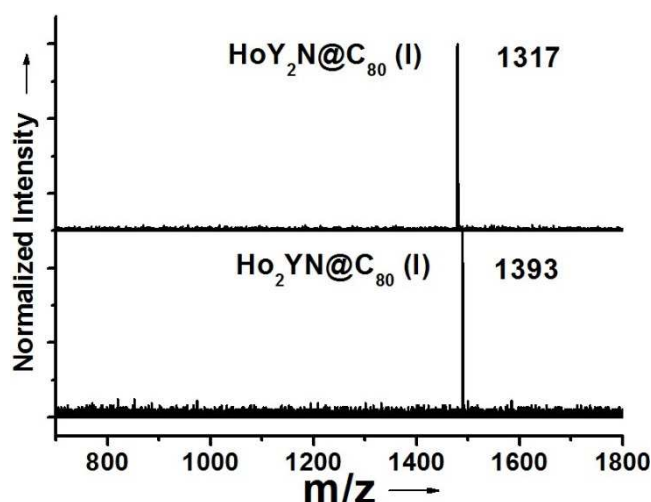


Figure S5. The isolated samples of $\text{Ho}_x\text{Y}_{3-x}\text{N}@\text{C}_{80}$ (I; $x=1, 2$) were identified by laser-desorption/ionization time-of-flight (LDI-TOF) mass spectrum analysis, which confirmed their high purity.

1.2 Synthesis and isolation of $\text{Ho}_x\text{Lu}_{3-x}\text{N}@\text{C}_{80}$ (I, $x=1, 2$)

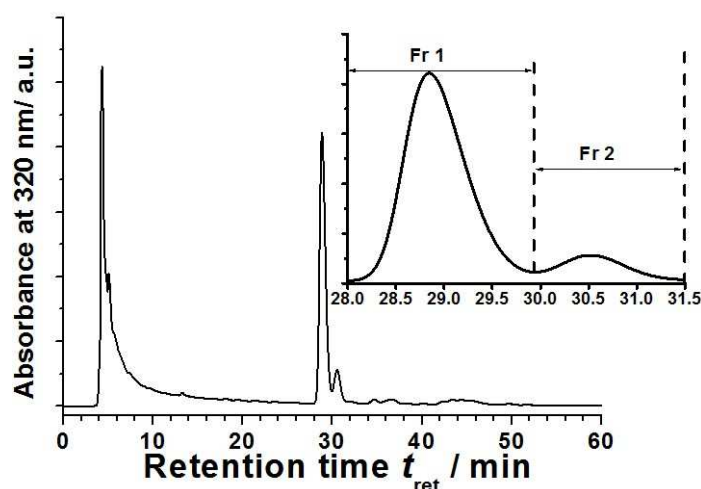


Figure S6. Chromatogram of a raw $\text{Ho}_x\text{Lu}_{3-x}\text{N}@\text{C}_{2n}$ fullerenes extract synthesized by the “selective organic solid” method (linear combination of two 4.6×250 mm Buckyprep columns, flow rate 1.6 ml/min, injection volume 200 μL , toluene as mobile phase, 40 $^\circ\text{C}$). The inset shows the enlarged chromatographic region of 28.0–31.5 min.

The synthesis of $\text{Ho}_x\text{Lu}_{3-x}\text{N}@\text{C}_{80}$ ($x=1, 2$) was achieved by “selective organic solid” route as). A mixture of Ho_2O_3 and Lu_2O_3 (99.9%, MaTeck GmbH, Germany), guanidine thiocyanate (GT) and graphite powder was used (molar ratio $\text{Ho/Lu/GT/C}=1:1:2.5:15$).

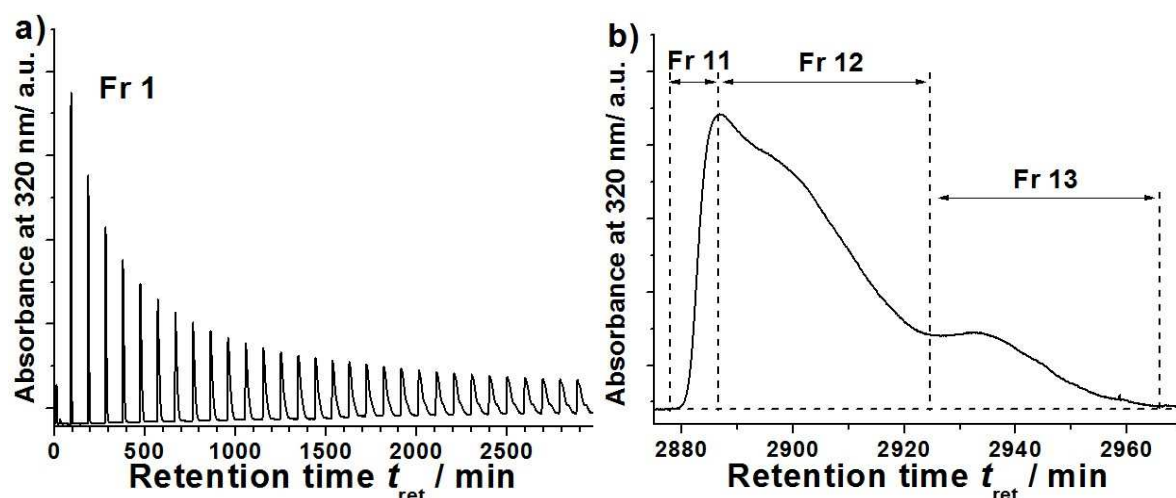


Figure S7. The isolation of fraction Fr 1. (10×250 mm Buckyprep column; flow rate 1.5 ml/min; injection volume 5 ml; toluene as eluent; 20 °C).

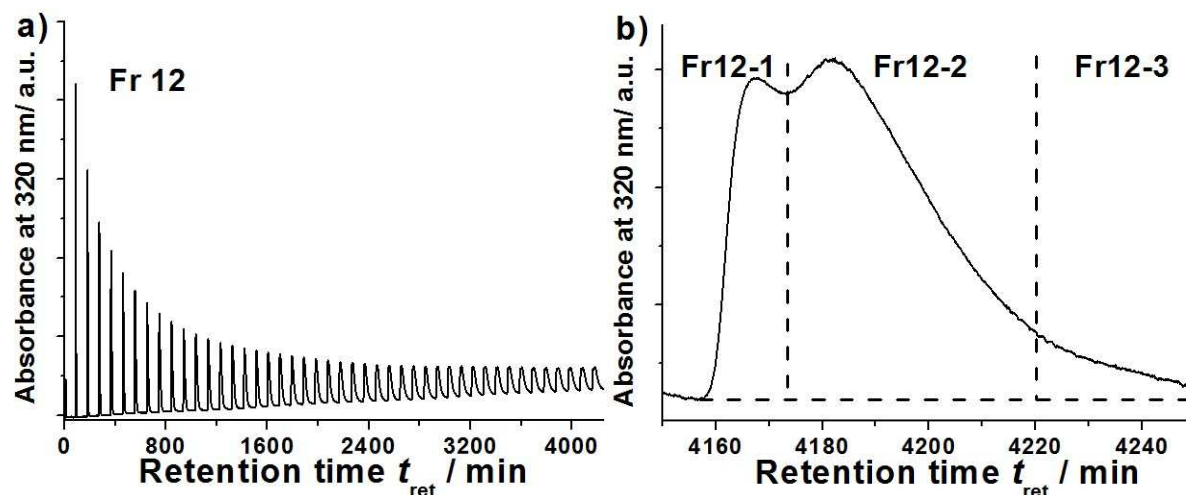


Figure S8. The isolation of fraction Fr 12. (10×250 mm Buckyprep column; flow rate 1.5 ml/min; injection volume 5 ml; toluene as eluent; 20 °C).

The mixture of $\text{Ho}_x\text{Lu}_{3-x}\text{N}@\text{C}_{2n}$ was subjected to isolation by analytical HPLC in the first step (Figure S6). $\text{Ho}_x\text{Lu}_{3-x}\text{N}@\text{C}_{80}$ (I) and $\text{Ho}_x\text{Lu}_{3-x}\text{N}@\text{C}_{80}$ (II) were separated into Fr 1 (28.0-29.9 min) and Fr 2 (29.9-31.5 min) respectively according to the difference of their cage symmetry. In the second step, three sub-fractions could be obtained after recycling fraction 1 over 30 times which then named as Fr 11, Fr 12 and Fr 13. Checking by mass spectrum, the dominant structure in Fr 12 is $\text{Ho}_x\text{Lu}_{3-x}\text{N}@\text{C}_{80}$ (I) (Figure S7). In the third step, the Fr 12 was subjected to recycling HPLC again for removing minor structures ($\text{HoLu}_2\text{N}@\text{C}_{80}$ (I) and $\text{Ho}_3\text{N}@\text{C}_{80}$ (I)). As shown in Figure S8, Fr 122 was collected after 44 cycles. In the fourth step, after running another 44 cycles, the isolation of $\text{Ho}_2\text{LuN}@\text{C}_{80}$ (I) was successfully achieved and its purity was confirmed by LDI-TOF mass spectroscopy (Figure S11).

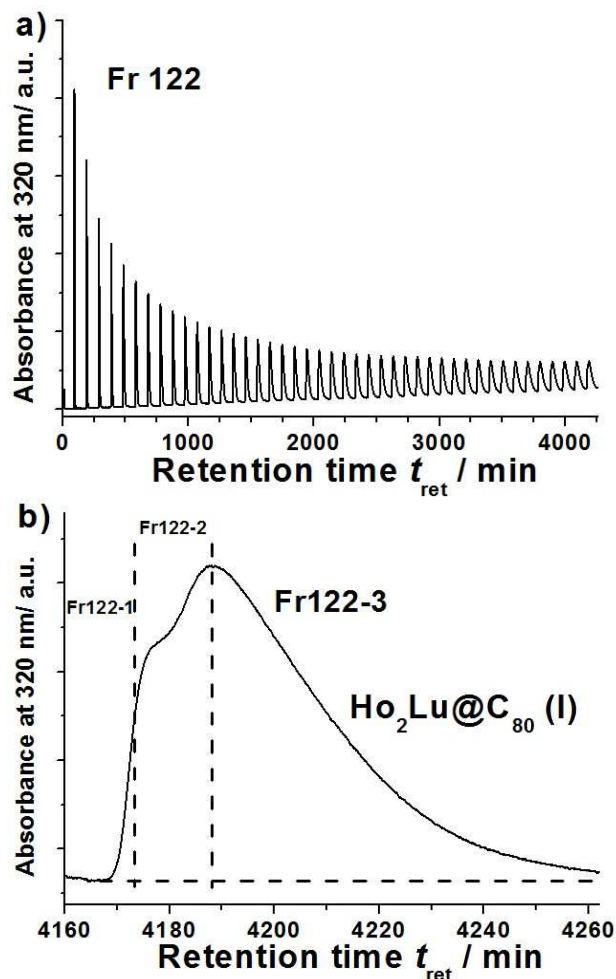


Figure S9. The isolation of fraction Fr 122. (10×250 mm Buckyprep column; flow rate 1.5 ml/min; injection volume 5 ml; toluene as eluent; 20 °C).

Similar to $\text{Ho}_2\text{LuN}@C_{80}$ (I), the isolation of $\text{HoLu}_2\text{N}@C_{80}$ (I) is extremely time-consuming due to the retention time of $\text{Lu}_3\text{N}@C_{80}$ (I) and $\text{HoLu}_2\text{N}@C_{80}$ (I) is almost identical. Only by running on recycling HPLC over 73 cycles, small amount of $\text{HoLu}_2\text{N}@C_{80}$ (I) could be obtained, see Figure S10. The purity of $\text{HoLu}_2\text{N}@C_{80}$ (I) was confirmed by LDI-TOF mass spectroscopy (Figure S11).

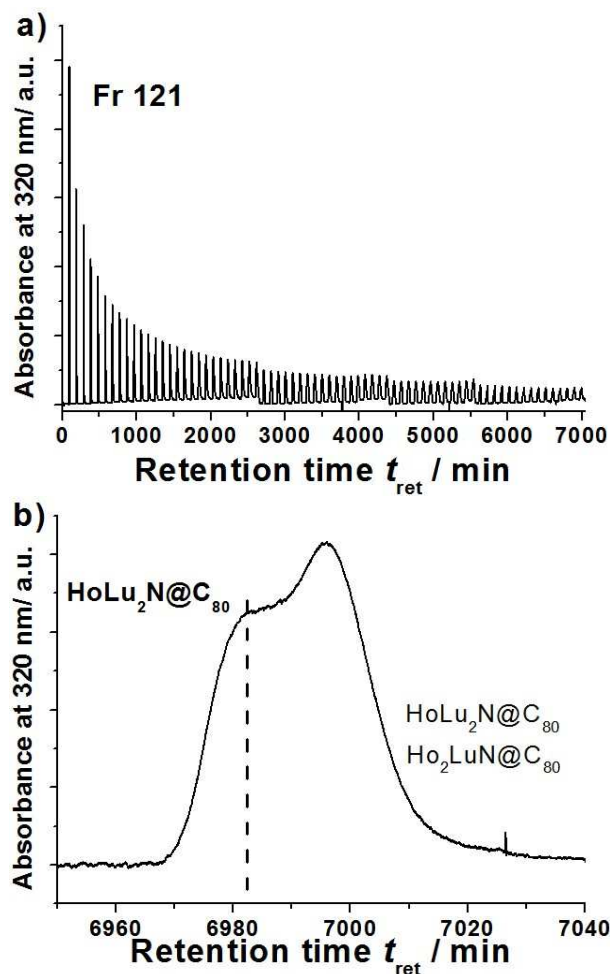


Figure S10. The isolation of fraction Fr 121. (10×250 mm Buckyprep column; flow rate 1.5 ml/min; injection volume 5 ml; toluene as eluent; 20 °C).

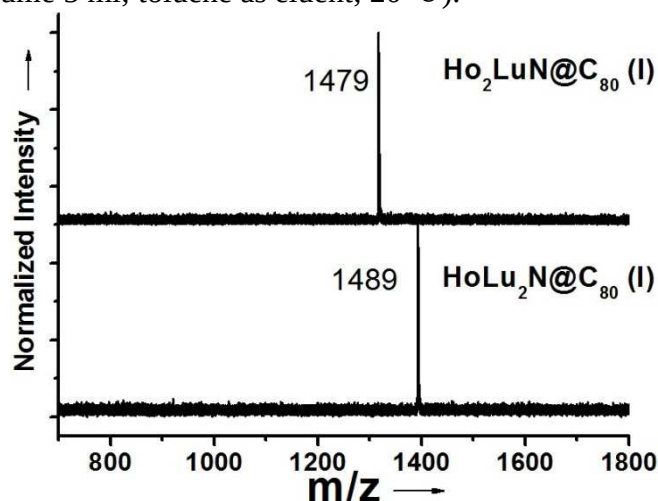


Figure S11. The isolated samples of $\text{Ho}_x\text{Lu}_{3-x}\text{N@C}_{80}$ (I; $x=1, 2$) were identified by LDI-TOF mass spectrum analysis, which confirmed their high purity.

2. Spectroscopic properties of $\text{Ho}_x\text{M}_{3-x}\text{N@C}_{80}$ (I; M= Y and Lu; $x=1, 2$)

2.1 UV-vis-NIR spectra of $\text{Ho}_x\text{M}_{3-x}\text{N@C}_{80}$

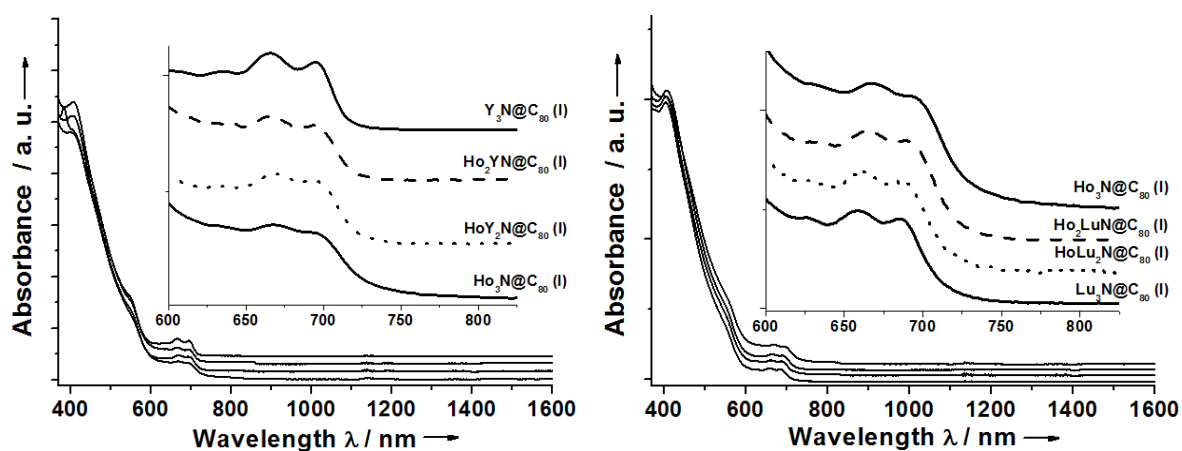


Figure S12. UV-vis-NIR spectra of $\text{Ho}_x\text{Y}_{3-x}\text{N@C}_{80}$ (I; $x=0-3$) (left) and $\text{Ho}_x\text{Lu}_{3-x}\text{N@C}_{80}$ (I; $x=0-3$) (right) in toluene.

2.2 FTIR spectra of $\text{Ho}_x\text{M}_{3-x}\text{N@C}_{80}$ (I; $x=0-3$)

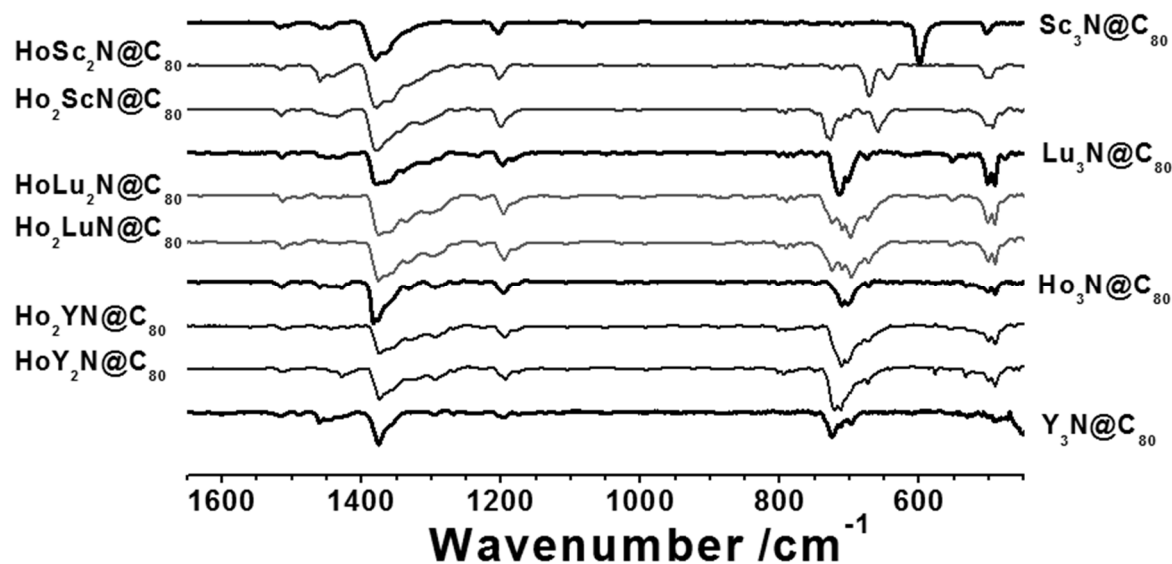
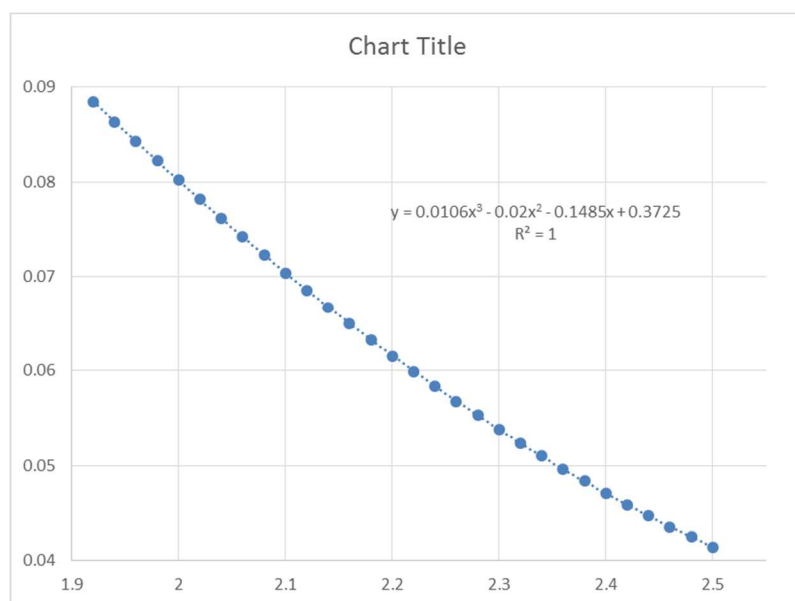


Figure S13. FTIR spectra of $\text{Ho}_x\text{M}_{3-x}\text{N@C}_{80}$ (I; M= Y, Lu; $x=0-3$) compared with $\text{Ho}_x\text{Sc}_{3-x}\text{N@C}_{80}$.

3. Quantum chemical calculations

B3LYP calculations were performed using the Firefly code.¹ The basis set was def2-SVP² for carbon atoms, def2-TZVP² for nitrogen, Stuttgart-Cologne effective core potentials for Sc (ECP10MDF)³ with {3,1,1,1,1,1/2,2,1,1/4,1,1/1,1/1} valence part, ECP28MWB^{4, 5} for Y with {3,1,1,1,1/4,1,1/4,1/1} valence electron part, and 4f-in-core ECP56MWB-II for Ho with {3,1,1,1,1,1/3,1,1,1,1/2,1,1,1,1/1,1,1/1,1} valence part.⁶ For QTAIM calculations, computations were performed using ORCA,⁷ PBE functional, DKH relativistic correction, and DKH-TZVP full electron basis set.⁸

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Correlation between $\Delta\chi$ (arb. units) and Ho-Sc distance. The fitted polynomial was used to compute $\Delta\chi$ at any point of the MD trajectory.