

Cite this: DOI: 10.1039/c0xx00000x

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Supporting Information

Incorporation of Cl in sequentially deposited lead halide perovskite films for highly efficient mesoporous solar cells.

Sabba Dharani^{a,b}, Herlina Arianita Dewi^b, Rajiv Ramanujam Prabhakar^b, Tom Baikie^b, Chen Shi^c, Du Yonghua^d, Nripan Mathews^{a,b,e}, Pablo P. Boix^{b*}, Subodh G. Mhaisalkar^{a,b*}

⁵ Received (in XXX, XXX) Xth XXXXXXXXXX 20XX, Accepted Xth XXXXXXXXXX 20XX
DOI: 10.1039/b000000x

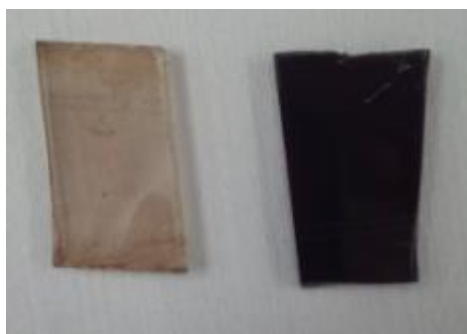


Fig.S1 On the left is the sample with PbCl₂ in MAI while the one on the right is that of PbI₂ + PbCl₂ in MAI.

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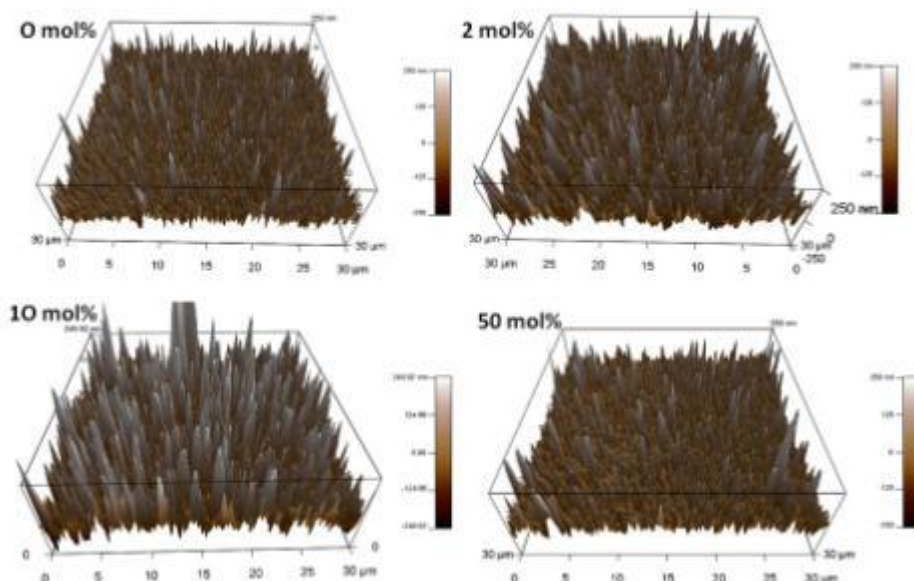


Fig.S2 AFM images illustrating the effect of Cl addition to CH₃NH₃PbI₃ on surface roughness of the films.

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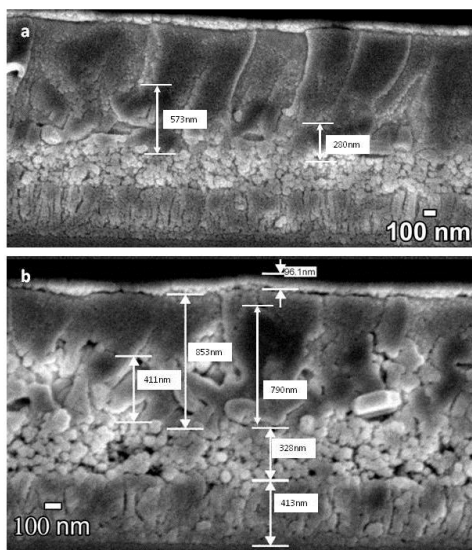
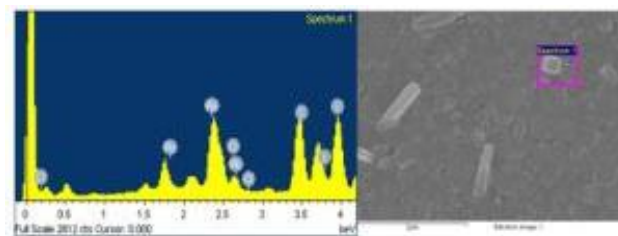
Table. S1 Effect of addition of Cl on lattice parameters of perovskite crystal structure

Sample	a (Å)	b (Å)	c (Å)
0 mol % PbCl ₂	8.8811(7)	8.8811(7)	12.6807(19)
2 mol % PbCl ₂	8.8859(6)	8.8859(6)	12.6936(18)
50 mol % PbCl ₂	8.8767(12)	8.8767(12)	12.6910(50)

From the AFM surface roughness characterization it was observed that with increasing concentration of PbCl₂, the surface roughness of the sample increased as indicated by the increasing density of sharp peaks in Fig. S2 and Table S1. However for 50 mol % of PbCl₂, the surface roughness was measured to be similar to the 0 mol % PbCl₂. This could be explained by the vertical growth of all the crystals leading to homogeneous roughness. The increase in the growth of the crystals could be validated from the FESEM cross-section image (Fig. S3).

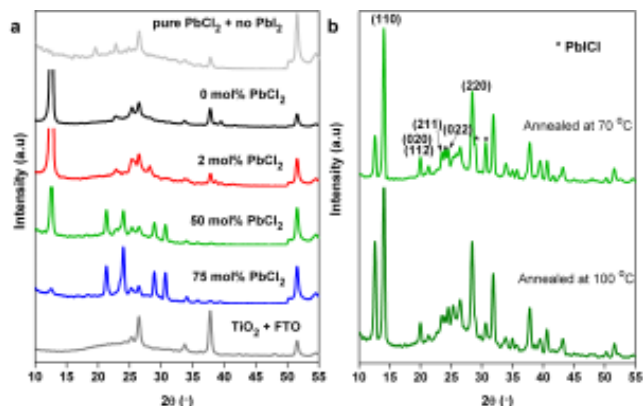
Table. S2 Effect of addition of Cl on surface roughness of perovskite film

Sample	Surface roughness (nm)
0 mol % PbCl ₂	52.74
2 mol % PbCl ₂	70.20
10 mol % PbCl ₂	97.9
50 mol % PbCl ₂	53.34

Fig.S3 Cross-section FESEM image illustrating the height of the perovskite layer for 50 mol % of PbCl₂ addition.Fig.S4 EDX demonstrating the presence of Cl peaks for 50 mol % PbCl₂ addition to methyl ammonium lead iodide.Table. S3 J-V characteristics of 20 mol % of PbCl₂

Sample	J _{sc} (mA/cm ²)	V _{oc} (V)	FF	η (%)
20 mol % PbCl ₂	18.38	0.98	0.46	8.18

From Table. S3, we can observe that for Cl concentration > 10 mol %, the efficiency of the solar cell degrades considerably.

Fig.S5 (a) XRD patterns of the samples after spin-coating PbI₂ or (PbI₂+PbCl₂). (b) Comparison of the XRD patterns of CH₃NH₃PbI_{3-x}Cl_x (x = 0.05) for different annealing conditions.

It was observed from Fig. S5a, that during the first step of inorganic halide deposition itself for Cl concentrations > 2 mol %, additional peaks were observed. Also the annealing temperature had no effect on the diffraction patterns as was observed in Fig. S5b.

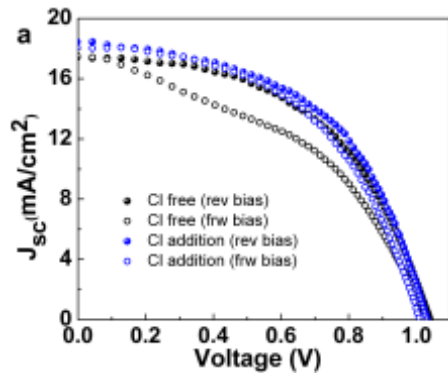


Fig.S6 Effect of Cl on hysteresis for different scan speeds.

10 Table. S3 Hysteresis Index extracted from data in Fig 5(c&d) for different scan rates.

Delay between points	0 mol % PbCl ₂	10 mol% PbCl ₂
0.1 s	0.13	0.02
0.05 s	0.17	0.03