

**Supplementary Information**

**High-Areal-Capacity Lithium Storage of Kirkendall Effect-driven  
Hollow Hierarchical NiS<sub>x</sub> Nanoarchitecture**

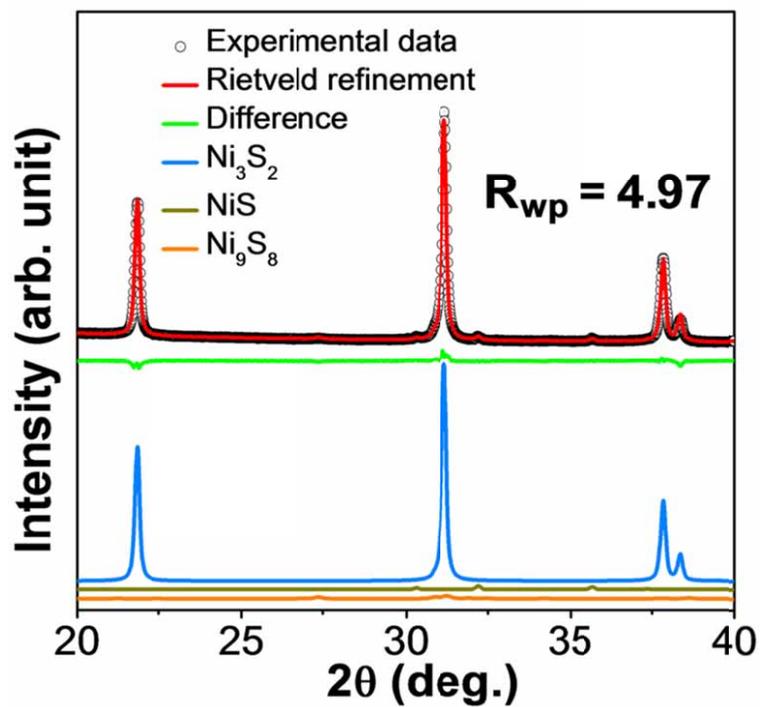
Chan Woo Lee,<sup>ab</sup> Seung-Deok Seo,<sup>c</sup> Hoon Kee Park,<sup>a</sup> Sangbaek Park,<sup>a</sup> Hee Jo Song,<sup>a</sup> Dong

Wan Kim\*<sup>c</sup> and Kug Sun Hong\*<sup>ab</sup>

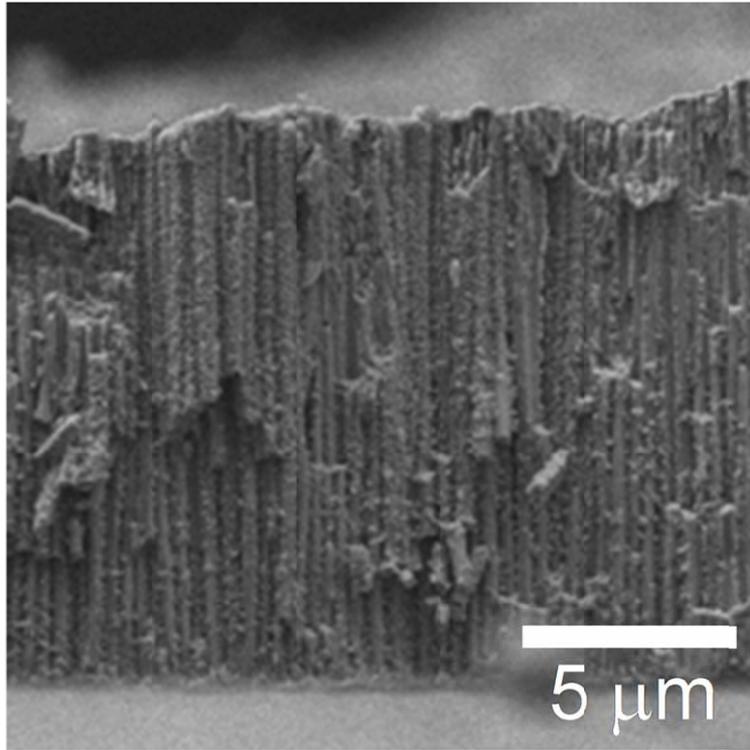
<sup>a</sup> Department of Materials Science and Engineering, Seoul National University, Seoul, 151-744, Korea

<sup>b</sup> WCU Hybrid Materials Program, Department of Materials Science and Engineering, Seoul National University, Seoul, 151-744, Korea

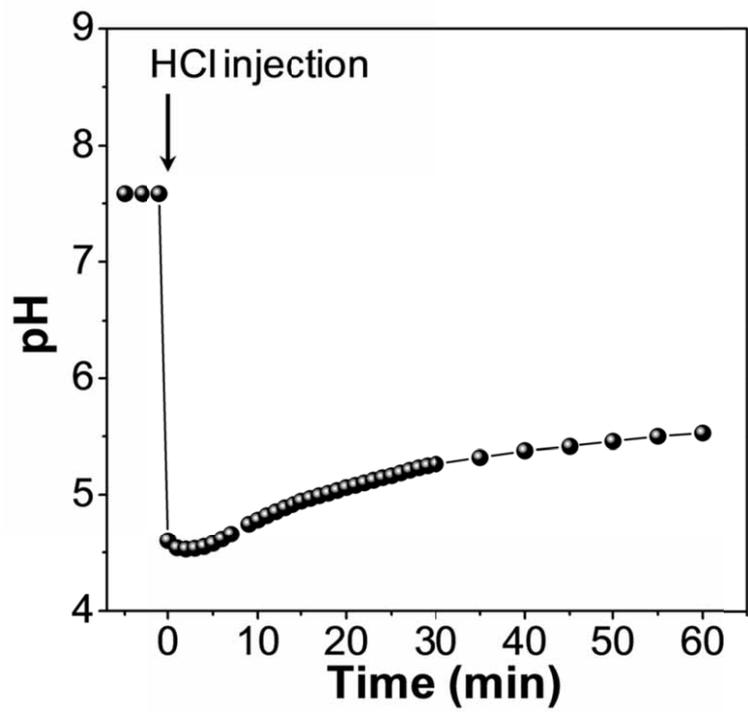
<sup>c</sup> School of Civil, Environmental and Architectural Engineering, Korea University, Seoul 136-713, Korea



**Figure S1.** Rietveld refinement results of the annealed 3-D NiS<sub>x</sub> electrodes. The Ni<sub>3</sub>S<sub>2</sub>, Ni<sub>9</sub>S<sub>8</sub>, and NiS phases occupy 93.39, 5.74 and 0.87 wt.% of the total phase, respectively.



**Figure S2.** A cross-view FESEM image of the 3-D NiS<sub>x</sub> electrodes.



**Figure S3.** The plot of pH value versus reaction time.

**Table S1.** A summary of literature results of 3-D lithium ion microbattery electrodes.

| <b>Material</b>   | <b>Areal capacity/<br/>mA h cm<sup>-2</sup></b> | <b>Rate/<br/>mA cm<sup>-2</sup></b> | <b>References</b> |
|---|---|-------------------------------------|-------------------|
| <b>Hollow NiS<sub>x</sub> nanoarchitecture</b>              | 1.5   | 0.5                                 | This work         |
| Porous Ni <sub>3</sub> S <sub>2</sub> film grown on Ni foam | 1.53  | 0.4                                 | 1                 |
| CuS-coated $\mu$ -channel silicon substrate                 | 1.015   | 0.12                                | 2                 |
| TiO <sub>x</sub> S <sub>y</sub> nanotubes                   | 0.225   | 0.07                                | 3                 |
| Polyaniline-coated Au nanorods                              | 0.032   | 0.04                                | 4                 |
| V <sub>2</sub> O <sub>5</sub> -coated MWCNT sponge          | 0.818   | 1.1                                 | 5                 |
| LiFePO <sub>4</sub> -coated porous carbon foam              | 1.72  | 0.1                                 | 6                 |
| LiCoO <sub>2</sub> -deposited Al nanorods                   | 0.11  | C/5                                 | 7                 |

## References

1. N. Feng, D. Hu, P. Wang, X. Sun, X. Li and D. He, *Phys. Chem. Chem. Phys.*, 2013, **15**, 9924–9930.
2. H. Mazor, D. Golodnitsky, L. Burstein and E. Peled, *Electrochem. Solid-State Lett.*, 2009, **12**, A232–A235.
3. N. A. Kyeremateng, N. Plylahan, A. C. S. dos Santos, L. V. Taveira, L. F. P. Dick and T. Djenizian, *Chem. Commun.*, 2013, **49**, 4205–4207.
4. S. R. Gowda, A. L. M. Reddy, X. Zhan, H. R. Jafry and P. M. Ajayan, *Nano Lett.* 2012, **12**, 1198–1202.
5. X. Chen, H. Zhu, Y.-C. Chen, Y. Shang, A. Cao, L. Hu and G. W. Rubloff, *ACS Nano*, 2012, **6**, 7948–7955.
6. H. D. Asfaw, M. R. Roberts, C.-W. Tai, R. Younesi, M. Valvo, L. Nyholm and K. Edstrom, *Nanoscale*, 2014, **6**, 8804–8813.
7. M. M. Shajumon, E. Perre, B. Daffos, P.-L. Taberna, J.-M. Tarascon and P. Simon, *Adv. Mater.*, 2010, **22**, 4978–4981.