

## Supporting information

### **Facile synthesis of highly luminescent nitrogen-doped graphene quantum dots for the detection of 2,4,6-trinitrophenol in aqueous solution**

*Liping Lin<sup>a</sup>, Xinhong Song<sup>a</sup>, Mingcong Rong<sup>a</sup>, Tingting Zhao<sup>c</sup>, Yiru Wang<sup>a</sup>, Xi Chen<sup>a,b\*</sup>*

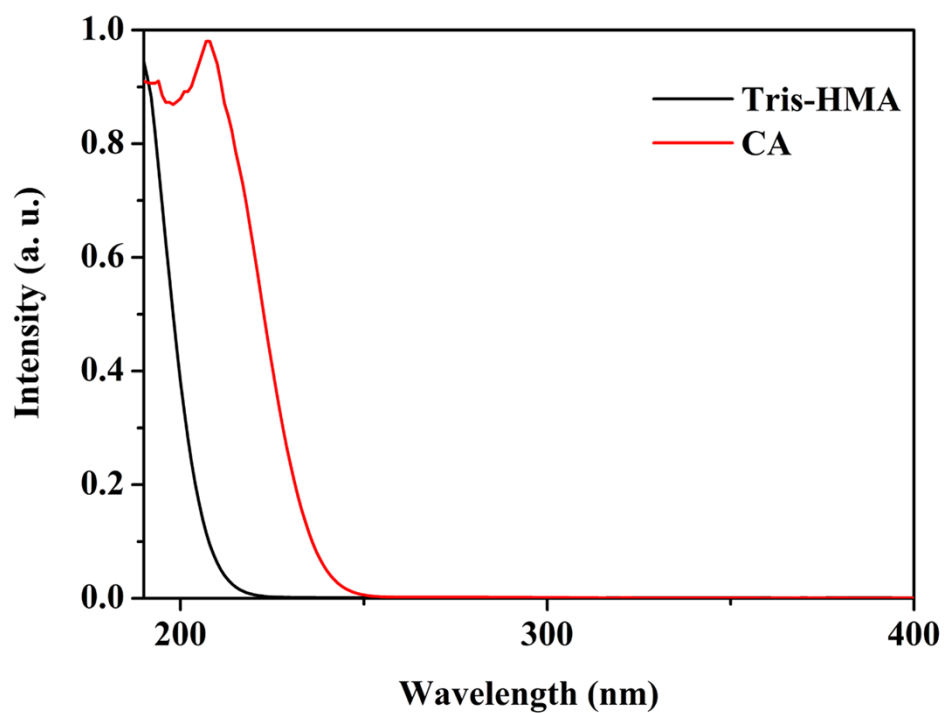
<sup>a</sup>Department of Chemistry and the MOE Key Laboratory of Spectrochemical Analysis & Instrumentation, College of Chemistry and Chemical Engineering, Xiamen University, Xiamen 361005, China

<sup>b</sup>State Key Laboratory of Marine Environmental Science, Xiamen University, Xiamen, 361005, China

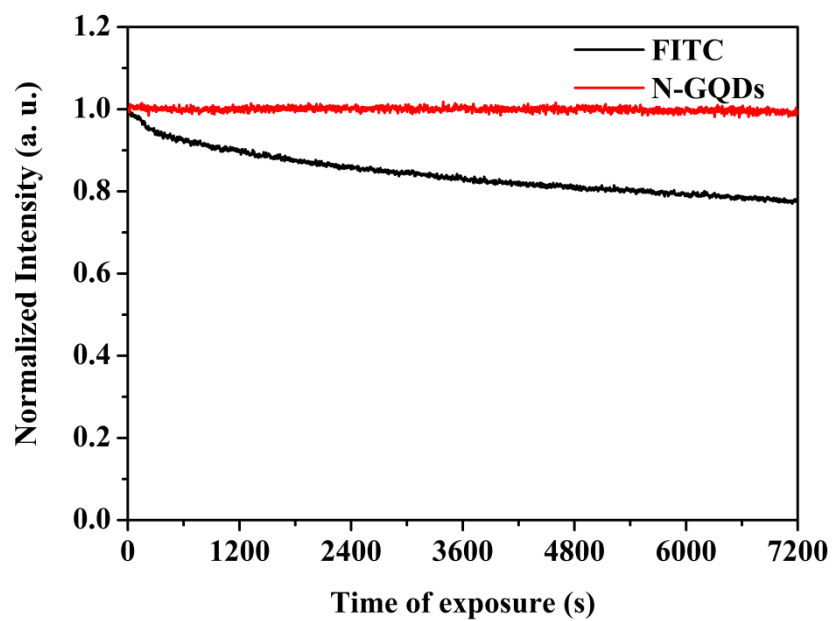
<sup>c</sup>Xiamen Huaxia Vocational College, Xiamen, 361005, China

---

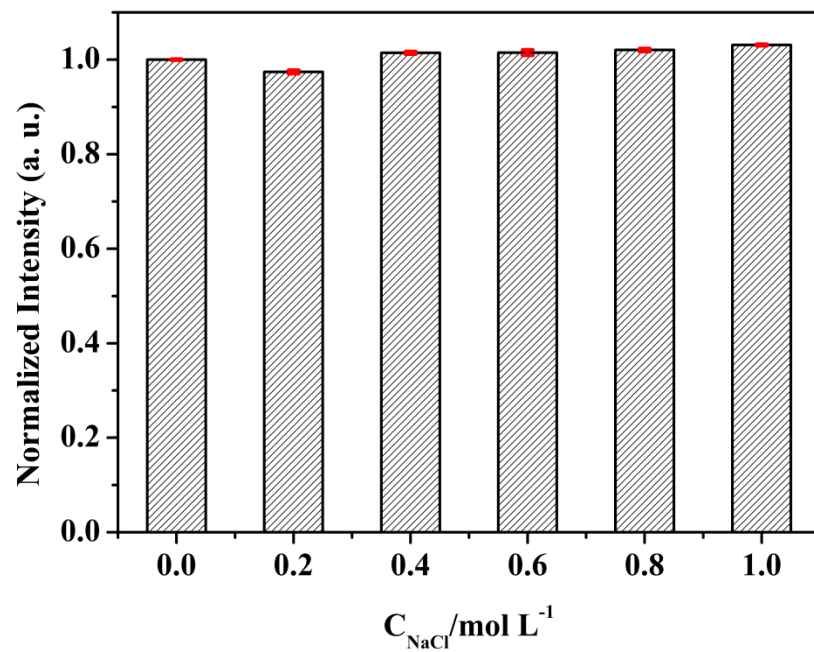
\* Corresponding author. Tel: +86 592 2184530; Fax: +86 592 2184530; E-mail address: [xichen@xmu.edu.cn](mailto:xichen@xmu.edu.cn)



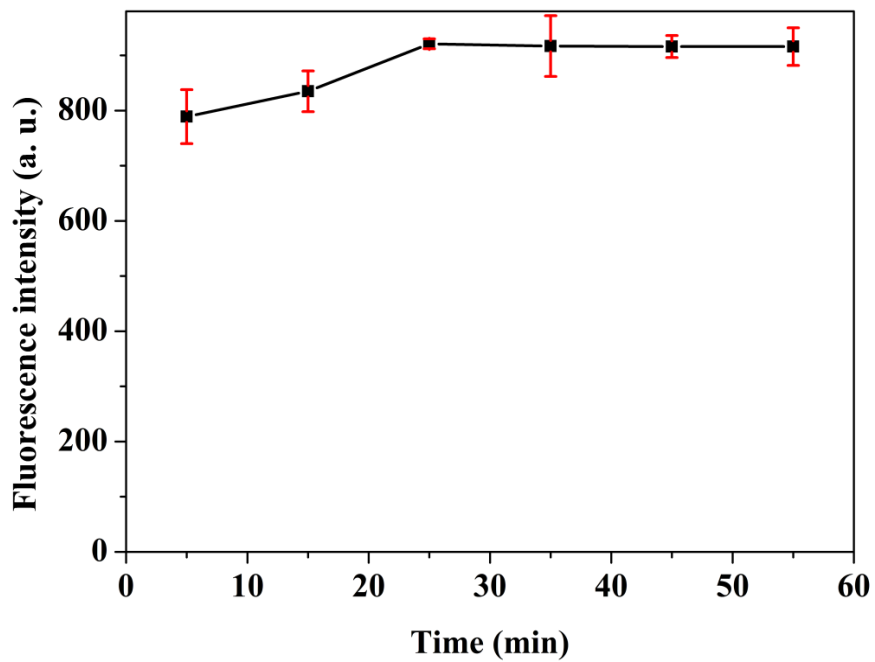
**Fig. S1** UV-Vis absorption spectra of CA and Tris-HMA.



**Fig. S2** Photostability of the N-GQDs and FITC under continuous irradiation for 2 h.



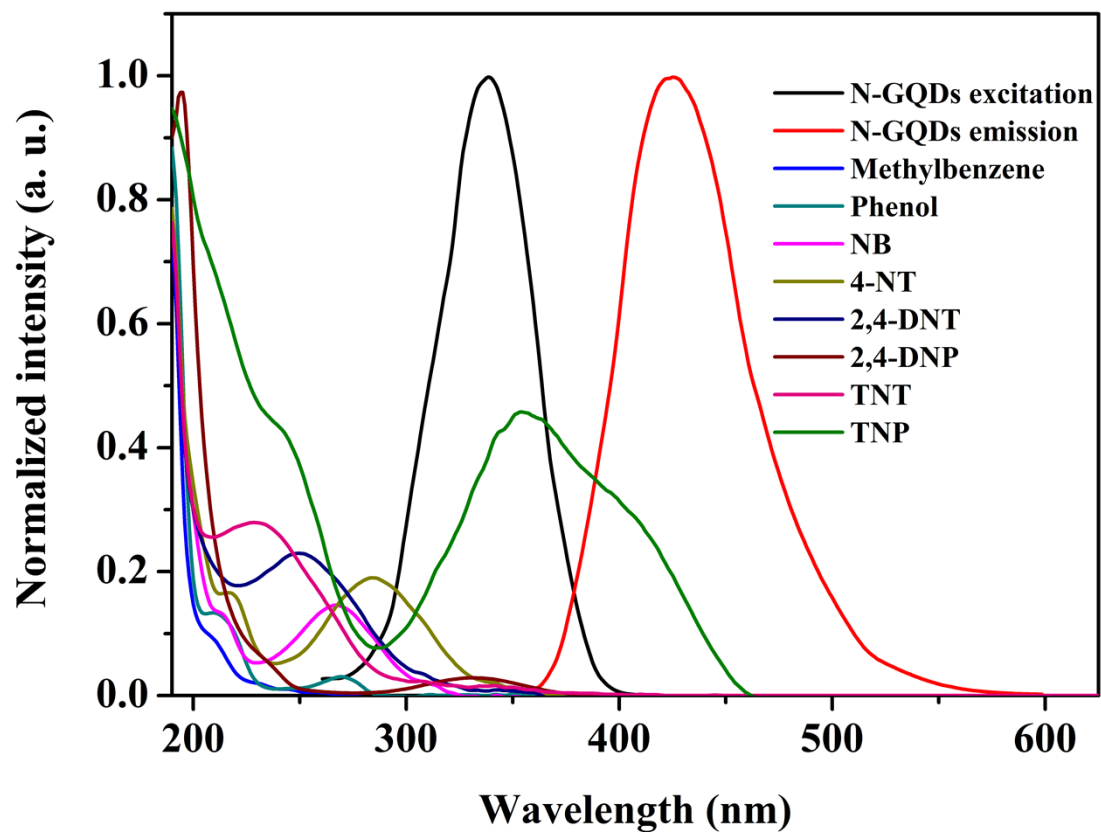
**Fig. S3** The effect of NaCl concentration on N-GQDs fluorescence intensity.



**Fig. S4** Effect of incubation time on the fluorescence quenching of the N-GQDs based fluorescence sensor toward TNP analysis. Conditions: 10  $\mu$ L N-GQDs + 200  $\mu$ L HAC-NaAc (pH 8.0, 0.2 mM) + 10  $\mu$ M TNP.

**Table S1** Fluorescence quenching constant ( $K_{S-V}$ ) and correlation values of different analytes

| Analytes      | $K_{S-V}$           | Correlation coefficient |
|---------------|---------------------|-------------------------|
| TNP           | $2.850 \times 10^4$ | 0.9944                  |
| TNT           | $2.291 \times 10^2$ | 0.9879                  |
| 2,4-DNP       | $7.030 \times 10^2$ | 0.9826                  |
| 2,4-DNT       | $1.688 \times 10^3$ | 0.9910                  |
| 4-NT          | $4.840 \times 10^2$ | 0.9826                  |
| NB            | $1.259 \times 10^3$ | 0.9926                  |
| Phenol        | $1.389 \times 10^3$ | 0.9872                  |
| Methylbenzene | $1.476 \times 10^3$ | 0.9891                  |

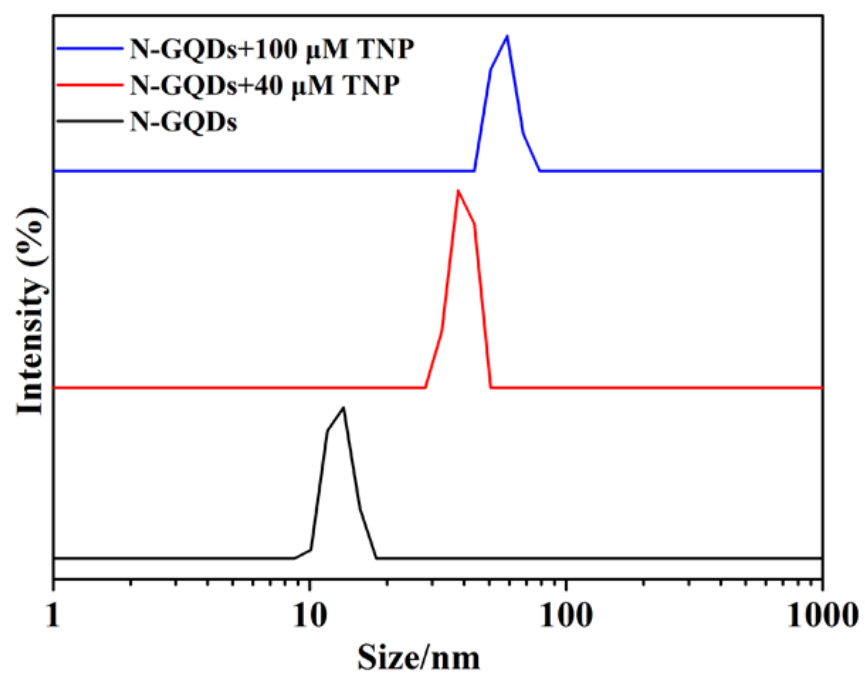


**Fig. S5** Excitation and emission spectra of the N-GQDs and the absorption spectra of different compounds.

**Table S2** The three decay components and their ratios for the fluorescence lifetime of the N-GQDs in different concentrations of TNP analysis

| Sample                   | $\tau_1$ (ns) | %     | $\tau_2$ (ns) | %     | $\tau_3$ (ns) | %    | Lifetime (ns) |
|--------------------------|---------------|-------|---------------|-------|---------------|------|---------------|
| N-GQDs                   | 5.29          | 18.97 | 12.26         | 78.58 | 0.32          | 2.45 | 10.64         |
| N-GQDs + 40 $\mu$ M TNP  | 4.72          | 15.62 | 11.89         | 81.98 | 0.34          | 2.4  | 10.50         |
| N-GQDs + 100 $\mu$ M TNP | 4.73          | 23.55 | 10.84         | 74.3  | 0.39          | 2.15 | 9.18          |





**Fig. S6** Hydrodynamic diameter of the N-GQDs (black line) and the N-GQDs in the presence of 40  $\mu\text{M}$  TNP (red line) and 100  $\mu\text{M}$  TNP (blue line) at pH 8.0 measured using the DLS method.

**Table S3** Comparison of different methods for the determination of TNP.

| Method  | Linear range  | Detection limit                     | References     |
|---|---|-------------------------------------|----------------|
| N-GQDs  | 1-60 $\mu\text{M}$                                      | 0.3 $\mu\text{M}$                   | Present method |
| 2,6-diamino pyridine functionalized graphene oxide system                               | N/A   | $\sim 125$ ppb                      | 1              |
| 8-hydroxyquinoline aluminum (Alq3)-based bluish green fluorescent composite nanospheres | 0.05–7.0 $\mu\text{g/mL}$<br>(0.22-30 $\mu\text{M}$ )   | 32.3 ng/mL<br>(0.14 $\mu\text{M}$ ) | 2              |
| Photoluminescent MoS <sub>2</sub> quantum dots  | 0.099-36.5 $\mu\text{M}$                                | 95 nM                               | 3              |
| Upconversion luminescence NaYF <sub>4</sub> @PSI-NH nanosensor                          | 0.01–4.5 $\mu\text{g/mL}$<br>(0.04-19.6 $\mu\text{M}$ ) | 9.6 ng/mL<br>(0.04 $\mu\text{M}$ )  | 4              |
| Amine-capped carbon dots  | N/A   | 1 $\mu\text{M}$                     | 5              |

## References

- 1 D. Dinda, A. Gupta, B. K. Shaw, S. Sadhu and S. K. Saha, *ACS Appl. Mater. Interfaces* 2014, **6**,10722.
- 2 Y. Ma, H. Li, S. Peng and L. Wang, *Anal. Chem.* 2012, **84**, 8415.
- 3 Y. Wang and Y. Ni, *Anal. Chem.* 2014, **86**, 7463.
- 4 Y. Ma, S. Huang, M. Deng and L. Wang, *ACS Appl. Mater. Interfaces* 2014, **6**, 7790.
- 5 Q. Niu, K. Gao, Z. Lin and W. Wu, *Anal. Methods* 2013, **5**, 6228.