Electronic Supplementary Information

 ${\bf Table}$. Optimization of conditions for the reaction of dipyrromethane ${\bf 30}$ with aldehyde ${\bf 11}^a$

Entry	TFA	Time of acid	DPM 30	DDQ (mmol)/	X-fold	Yield of corrole
	(mM)	catalyzed	(mM)	30 (mmol)	$dilution^b$	35 (%) ^c
		step				
1	3	5 h	33	1	-	18
2	3	5 h	33	1	27	22
3	3	5 h	33	1.3	-	24
4	3	5 h	33	1.3	27	28
5 ^d	3	5 h	33	1.3	27	27
6^e	3	5 h	33	1.3	27	27
7	1.3	5 h	133	1.3	27	28
8	4.3	3 h	133	1.3	-	26
9	13	3 h	133	1.3	-	26
10	13	20 min	133	1.3	-	29
11	13	10 min	133	1.3	-	29
12	1.3	5 h	133	1.3	-	23
13	13	5 h	133 ^f	1.3	-	19
14	13	10 min	133	1.3	g	26
15	1.3	5 h	133	1.3	13	29
16	1.3	5 h	133	1.3	27^{h}	26
17	1.3	5 h	133	1.3	g	31
18^i	1.3	16 h	133	1.4	27	31
19	18	1 min	40	1.5	-	8
20	18	3 min	40	1.5	-	16
21	18	12 min	40^{j}	1.5	-	16
22	18 ^k	3.5 min	40	1.3	27	24

^a All reactions were performed under the following constant conditions: For 1^{st} step: CH₂Cl₂, MesDPM:aldehyde **11** = 2:1, RT. For 2^{nd} step: RT). ^b Where X indicates how many times the reaction mixture was diluted before addition of DDQ. ^c Isolated

yields. ^d THF was used in the 2nd step. ^e Toluene was used in the 2nd step. ^f MeCN was used in 2 nd step ^g The reaction mixture dissolved in CH₂Cl₂ and the solution of DDQ were added simultaneously to CH₂Cl₂ during 10 min. ^h MeCN with NH₄Cl was used in 2nd step. ⁱ Aldehyde 11 was added during 3 h. ^j Toluene was used as a solvent in both steps. ^k Aldehyde 11 was added during 3 min.