

**Theoretical (DFT, GIAO-NMR, NICS) Study of Carbocations (M+H)<sup>+</sup>, Dications (M<sup>2+</sup>) and Dianions (M<sup>2-</sup>) from Dihydro-dicyclopenta[*ef,kl*]heptalene (Dihydro-azupyrene), Dihydro-dicyclohepta[*ed,gh*]pentalene, and Related Bridged [14]annulenes**

Takao Okazaki<sup>a,b</sup> and Kenneth K. Laali<sup>\*b</sup>

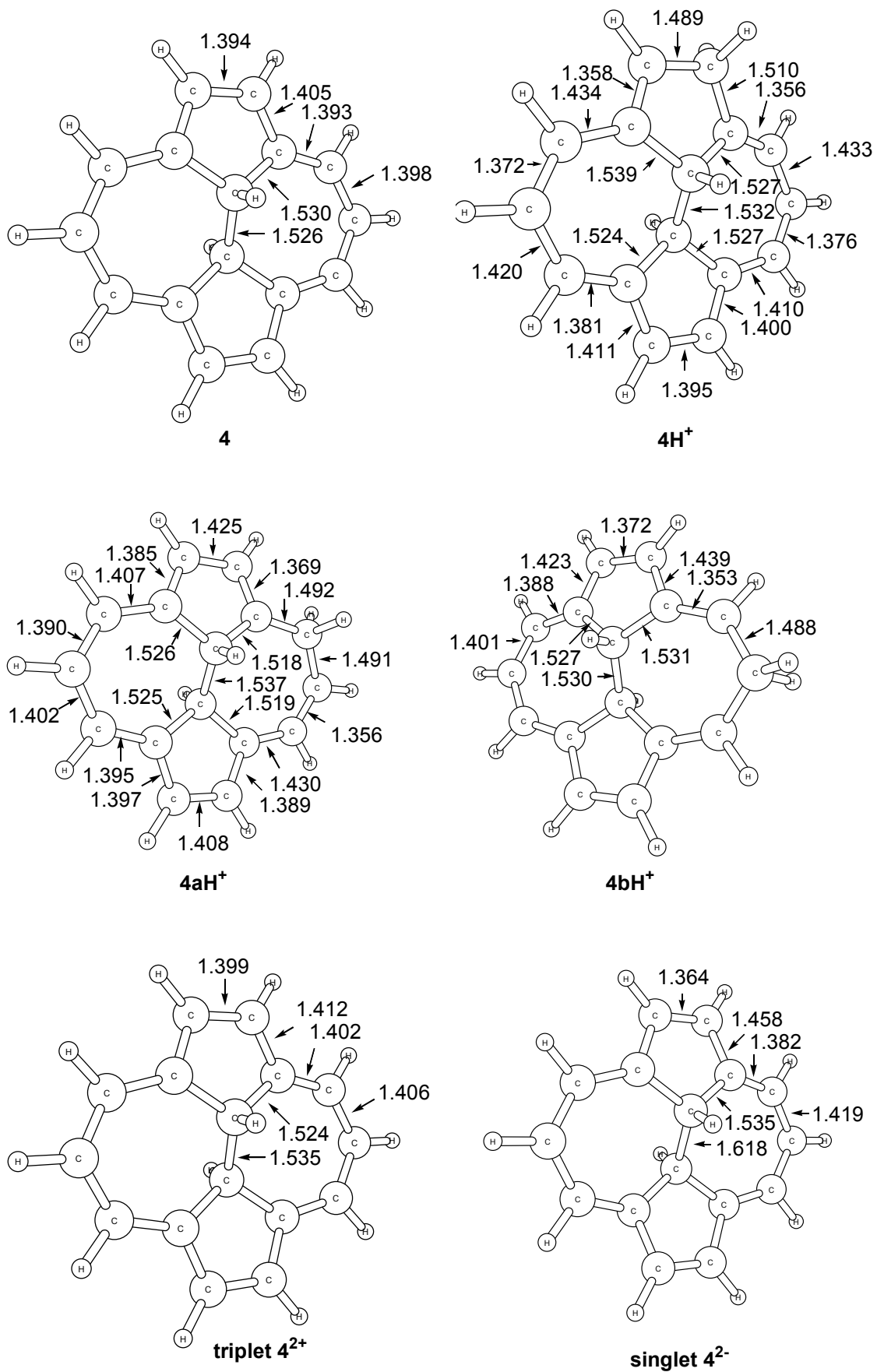
<sup>a)</sup> Department of Energy and Hydrocarbon Chemistry, Kyoto University, Kyoto, Japan

<sup>b)</sup> Department of Chemistry, Kent State University, Kent, OH 44242, USA

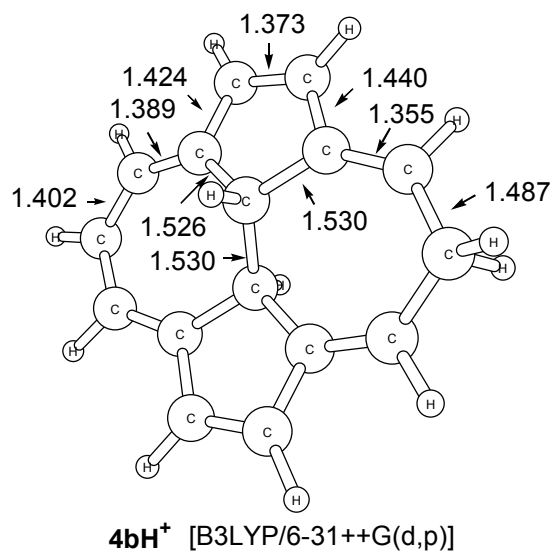
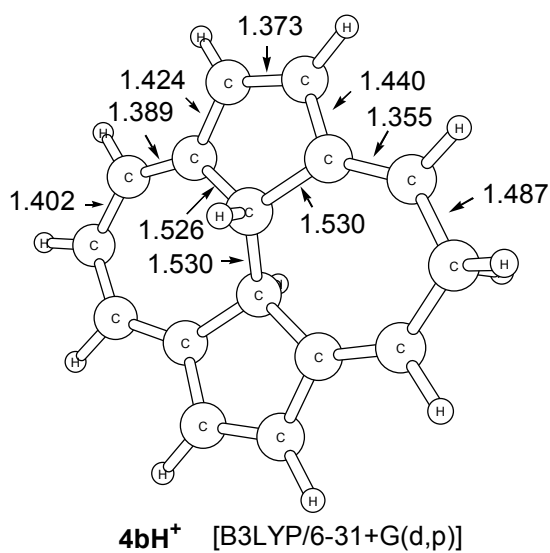
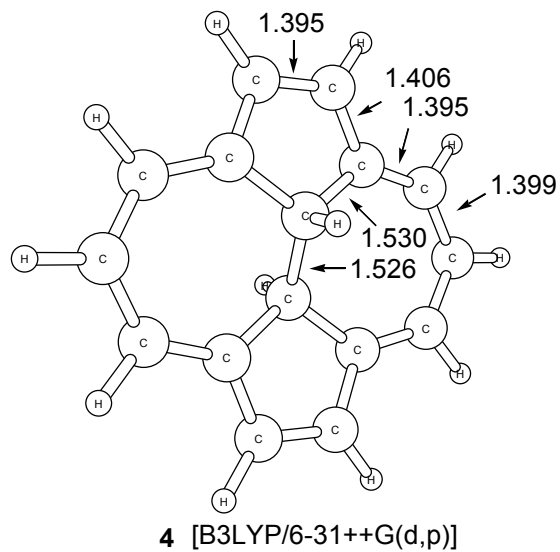
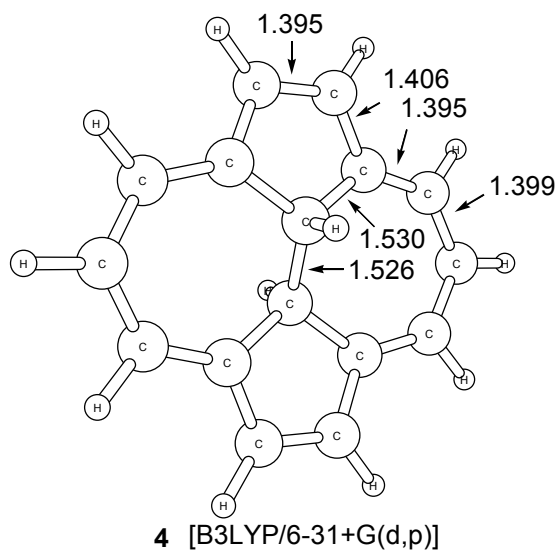
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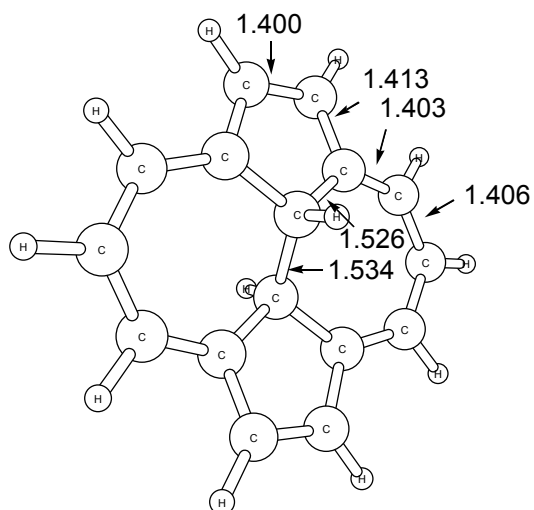
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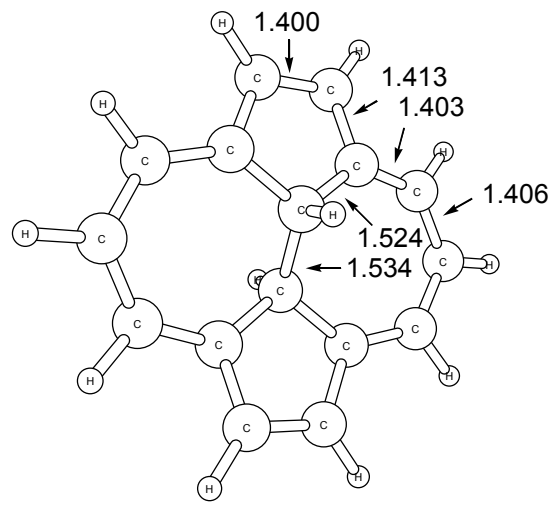
**Fig S1.** Optimized geometries for **4**, **4H<sup>+</sup>**, **4aH<sup>+</sup>**, **4bH<sup>+</sup>**, triplet **4<sup>2+</sup>**, and singlet **4<sup>2-</sup>** at B3LYP/6-31G(d) level (bond length, Å).



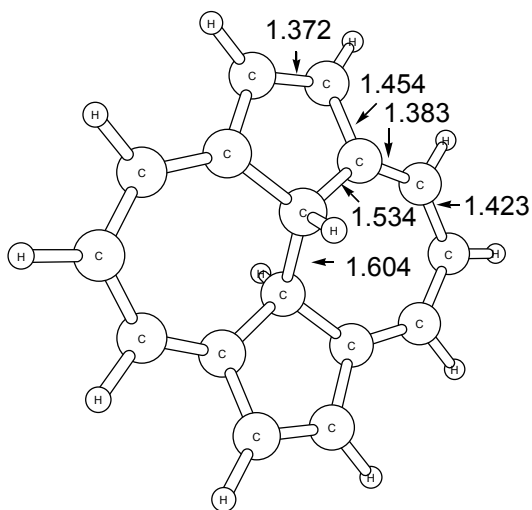
**Fig S1a.** Optimized geometries for **4**, **4bH<sup>+</sup>**, triplet **4<sup>2+</sup>**, and singlet **4<sup>2-</sup>** at B3LYP/6-31+G(d,p) and B3LYP/6-31++G(d,p) levels (bond length, Å).



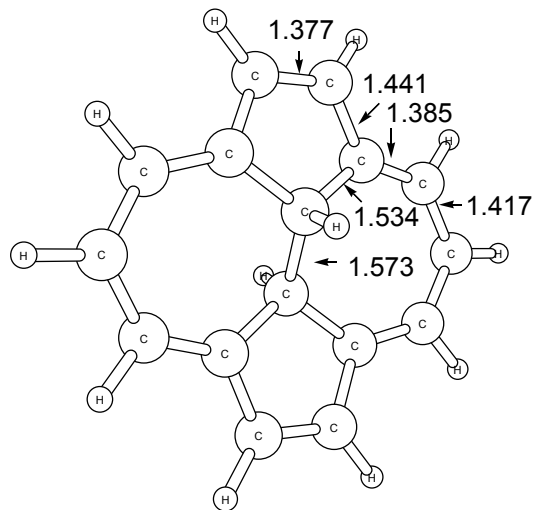
**triplet  $4^{2+}$  [B3LYP/6-31+G(d,p)]**



**triplet  $4^{2+}$  [B3LYP/6-31++G(d,p)]**

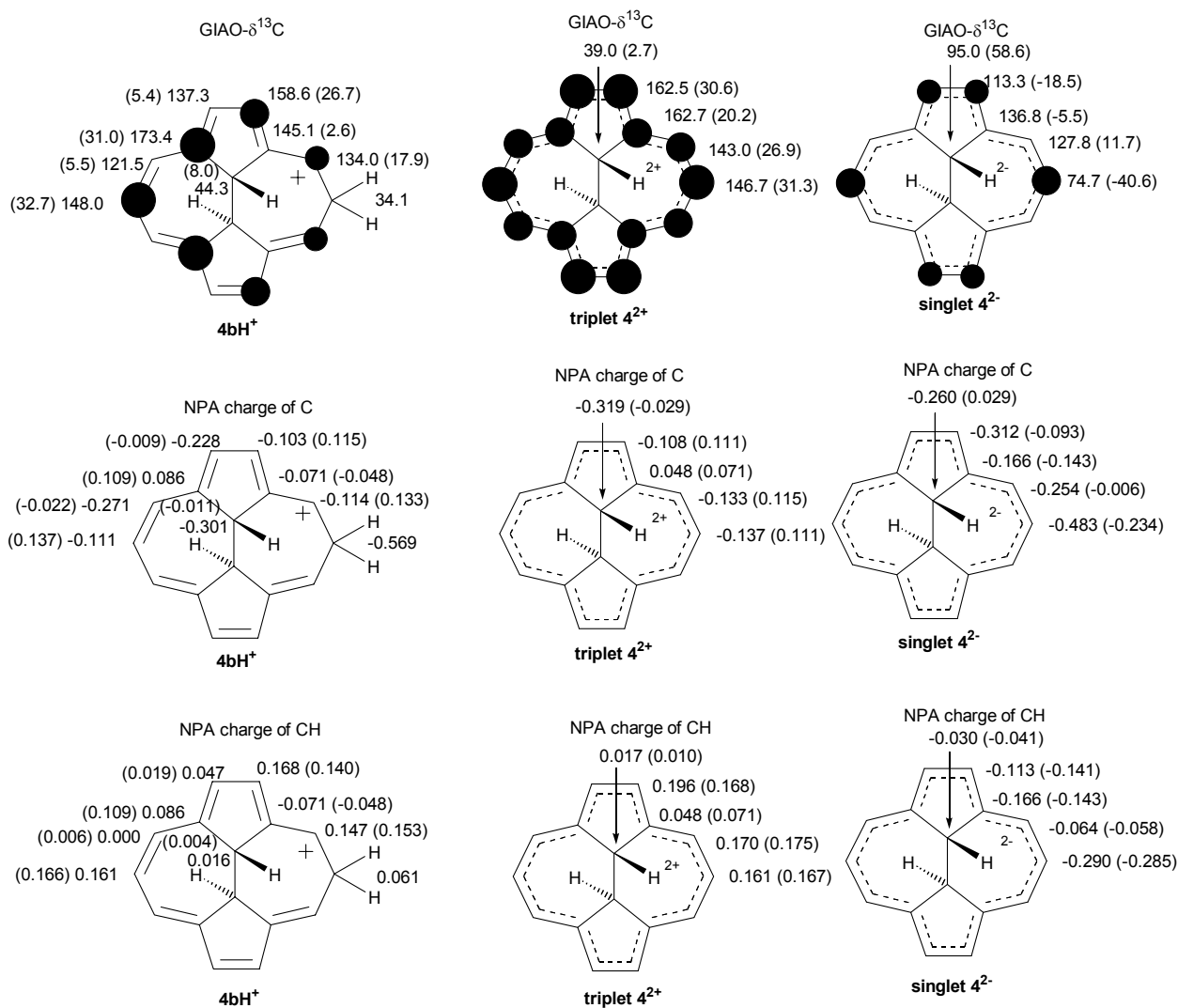


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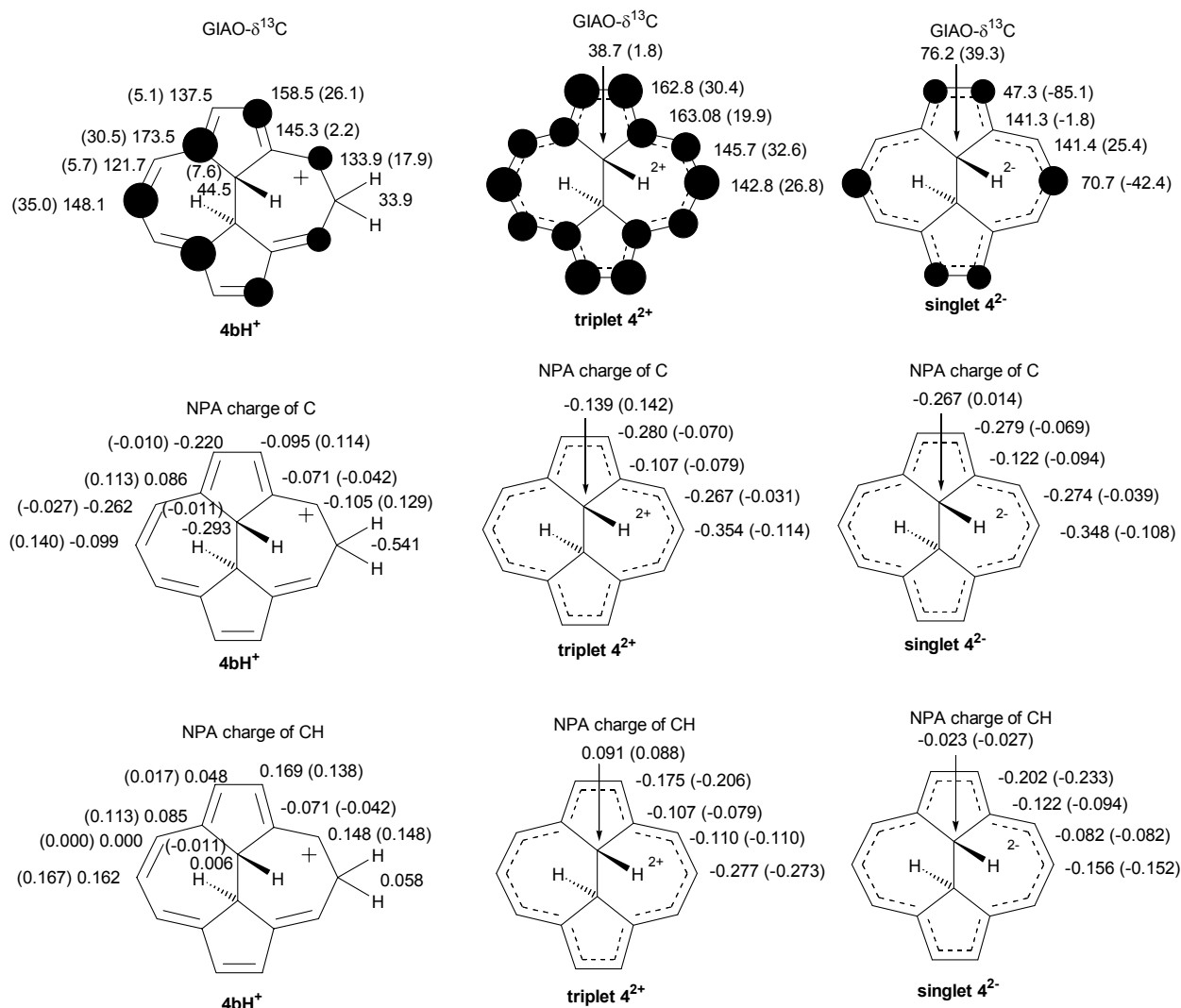


**singlet  $4^{2-}$  [B3LYP/6-31++G(d,p)]**

**Fig S1a (contined).**

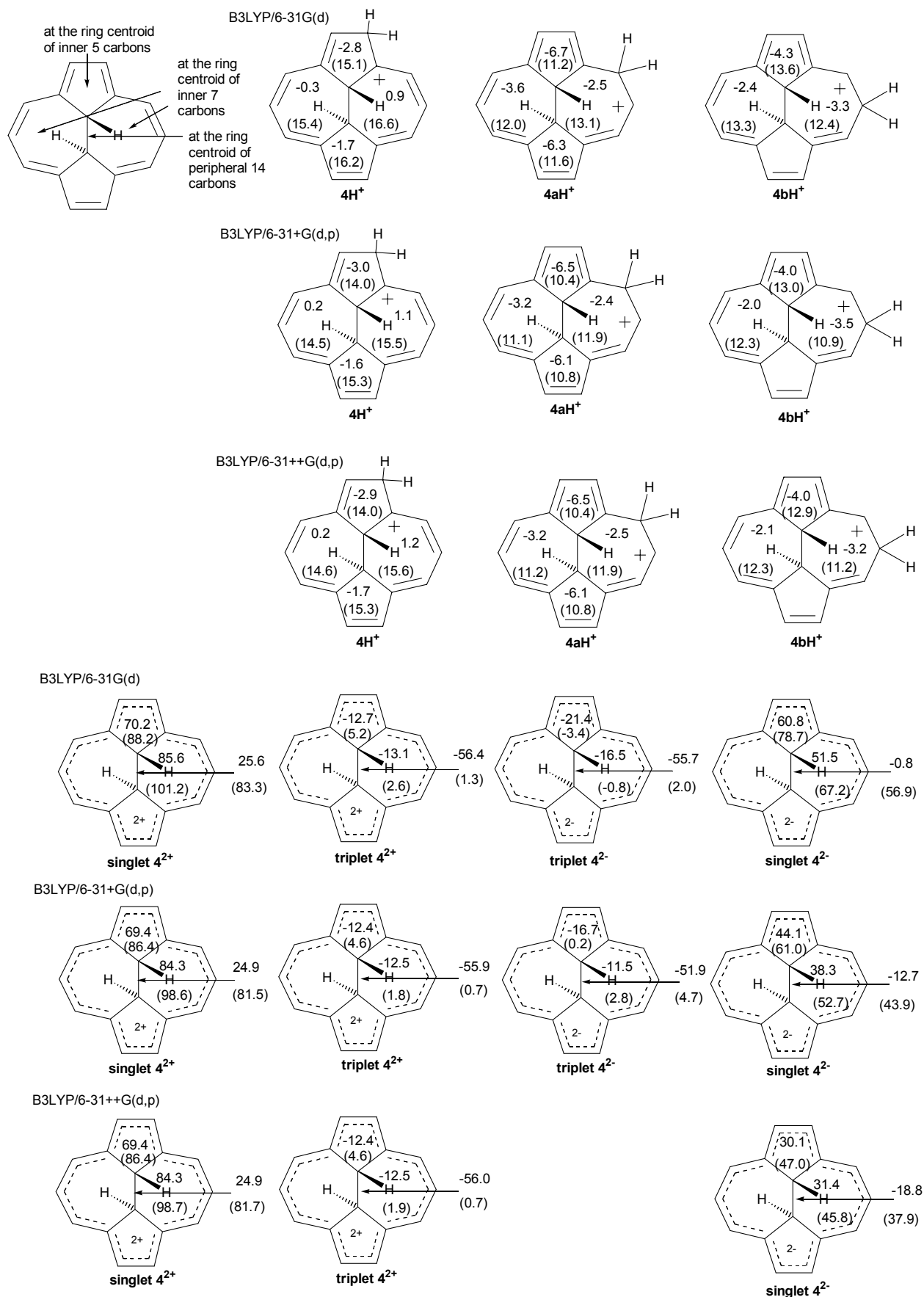


**Fig S1b.** Computed  $^{13}\text{C}$  NMR chemical shifts, NPA-derived carbon charges, and NPA-derived overall charges over CH units for **4bH<sup>+</sup>**, triplet **4<sup>2+</sup>**, and singlet **4<sup>2-</sup>** ( $\Delta\delta^{13}\text{C}$ 's and  $\Delta\text{charges}$  relative to **4** in parentheses) at B3LYP/6-31+G(d,p) level. [Dark circles are roughly proportional to the magnitude of  $\Delta\delta^{13}\text{C}$ s (positive/downfield for the carbocations/dication and negative/upfield for the dianion) ; threshold was set to 10 ppm].

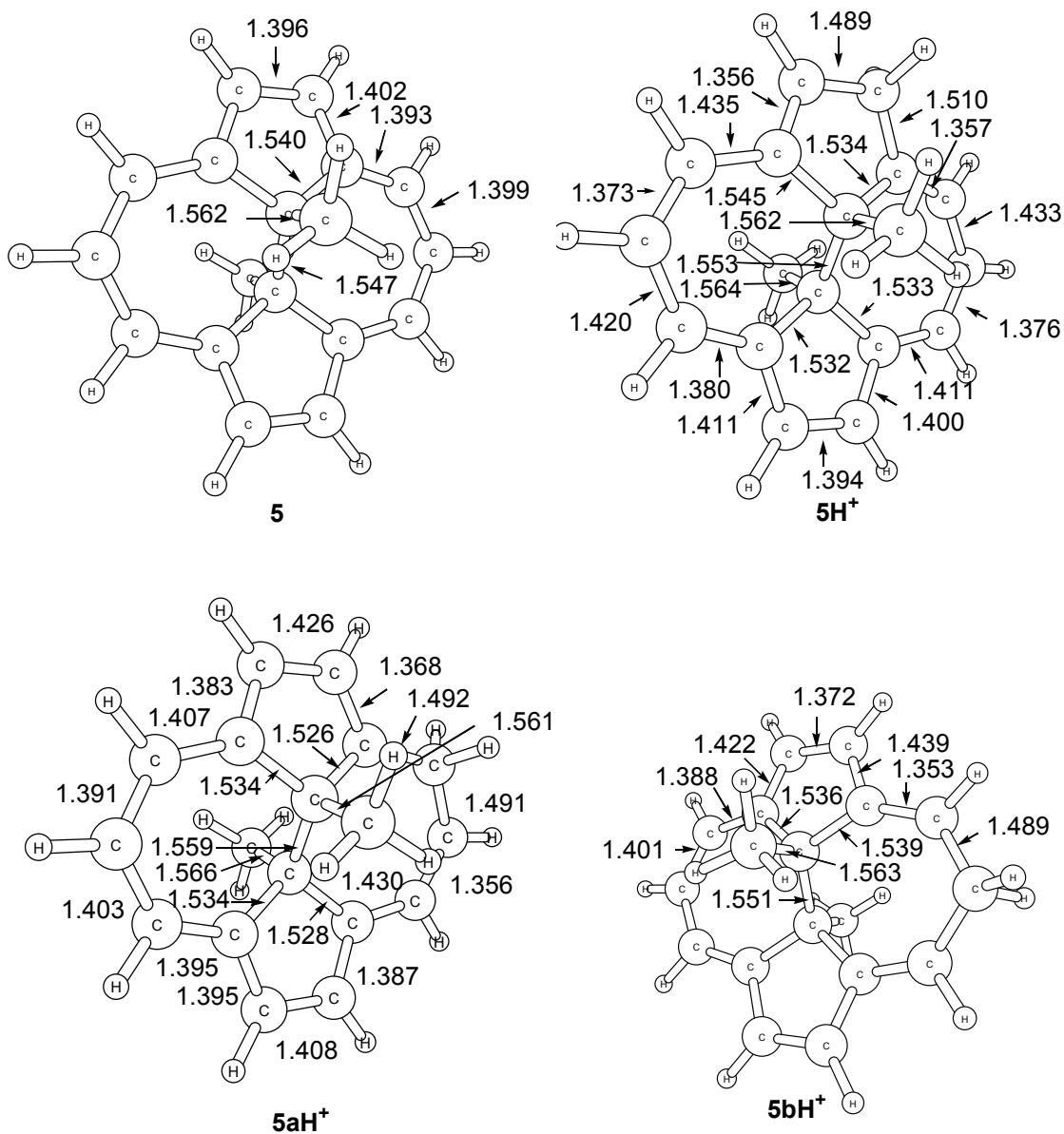


**Fig S1c.** Computed  $^{13}\text{C}$  NMR chemical shifts, NPA-derived carbon charges, and NPA-derived overall charges over CH units for **4bH<sup>+</sup>**, triplet **4<sup>2+</sup>**, and singlet **4<sup>2-</sup>** ( $\Delta\delta^{13}\text{C}$ 's and  $\Delta\text{charges}$  relative to **4** in parentheses) at B3LYP/6-31++G(d,p) level. [Dark circles are roughly proportional to the magnitude of  $\Delta\delta^{13}\text{C}$ s (positive/downfield for the carbocations/dication and negative/upfield for the dianion); threshold was set to 10 ppm].

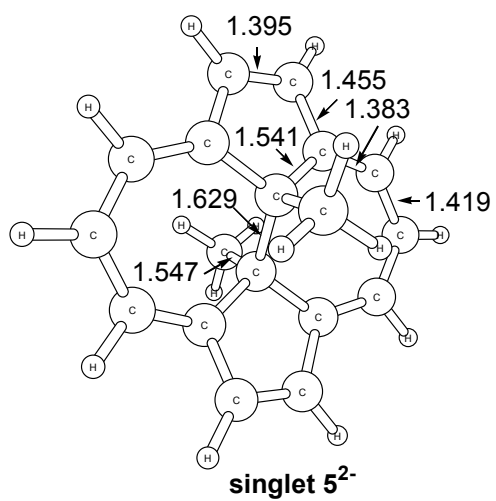
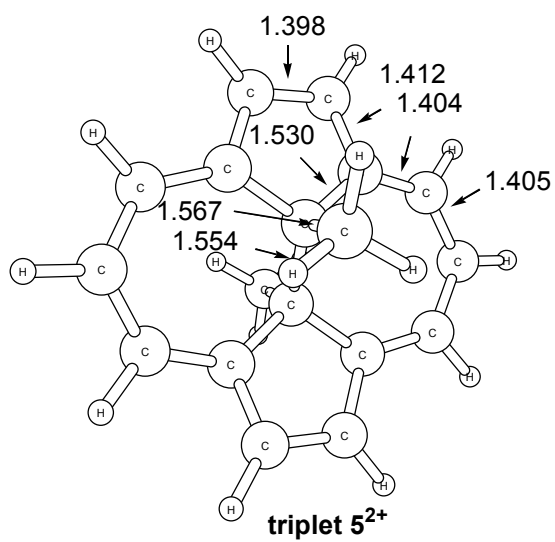
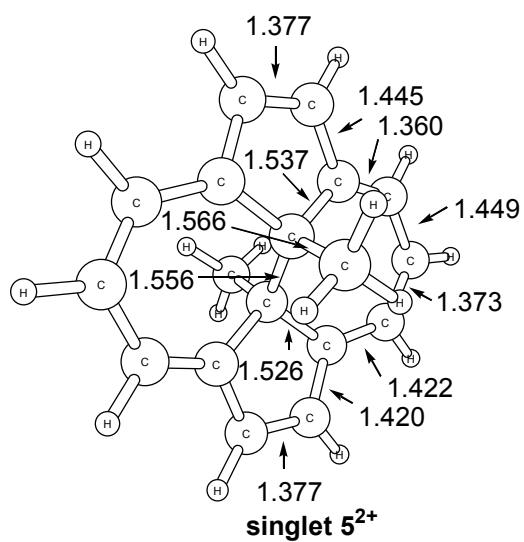




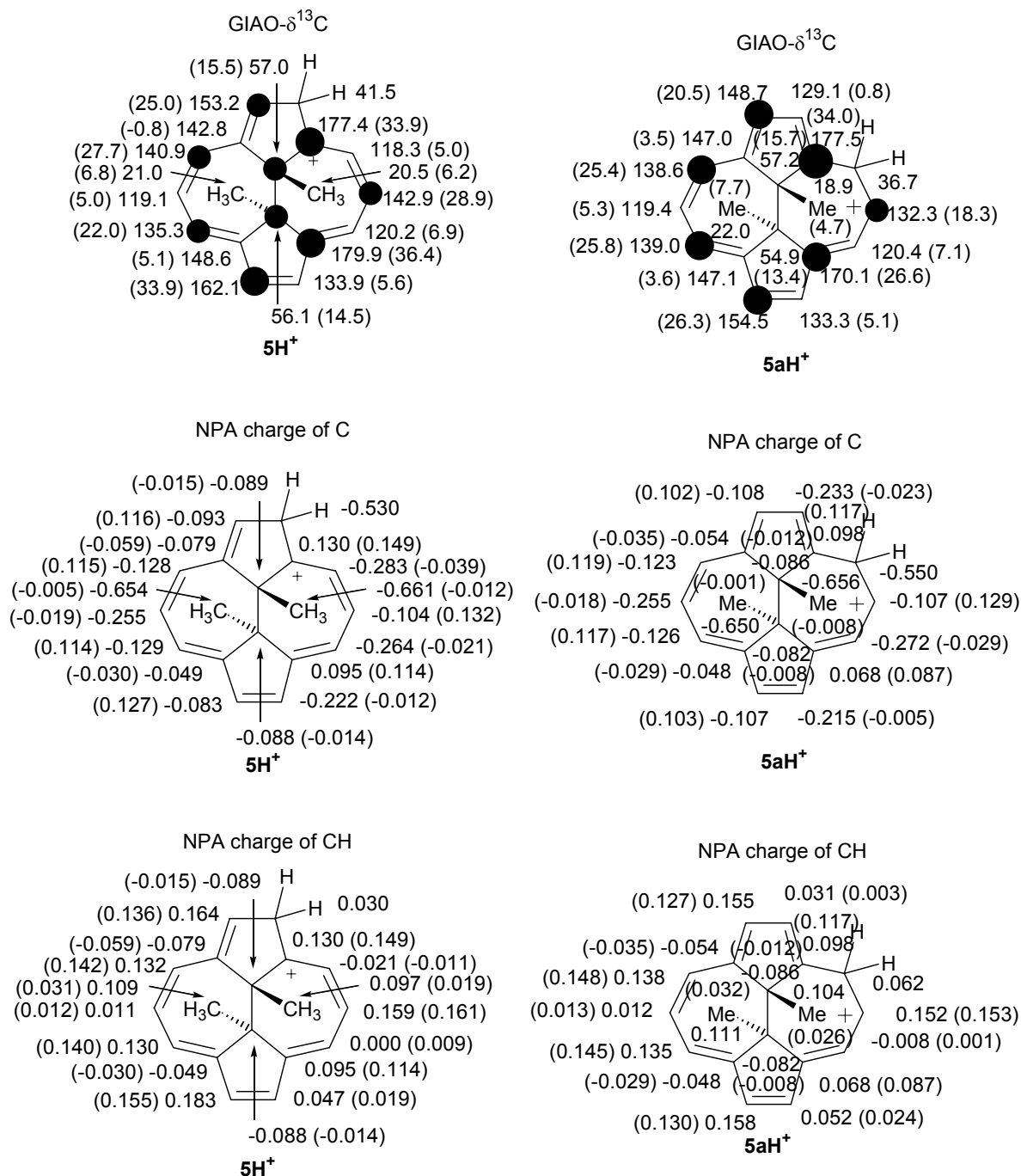
**Fig S2.** NICS values for  $4H^+$ ,  $4aH^+$ ,  $4bH^+$ ,  $4^{2+}$ , and  $4^{2-}$  at B3LYP/6-31G(d), 6-31+G(d,p), or 6-31++G(d,p) level ( $\Delta$ NICS values relative to those of **4** in parentheses).



**Fig S3.** B3LYP/6-31G(d) optimized geometries for **5**, **5H<sup>+</sup>**, **5aH<sup>+</sup>**, **5bH<sup>+</sup>**, **5<sup>2+</sup>**, and **5<sup>2-</sup>** (bond length, Å).



**Fig S3 (continued).**



**Fig S4.** Computed  $^{13}\text{C}$  NMR chemical shifts, NPA-derived carbon charges, and NPA-derived overall charges over CH units for  $5\text{H}^+$ ,  $5\text{aH}^+$ ,  $5\text{bH}^+$ ,  $5^{2+}$ , and  $5^{2-}$  at B3LYP/6-31G(d) level ( $\Delta\delta^{13}\text{C}$ 's and  $\Delta$ charges relative to **5** in parentheses). [Dark circles are roughly proportional to the magnitude of  $\Delta\delta^{13}\text{C}$ s (positive/downfield for the carbocations/dication and negative/upfield for the dianion); threshold was set to 10 ppm].

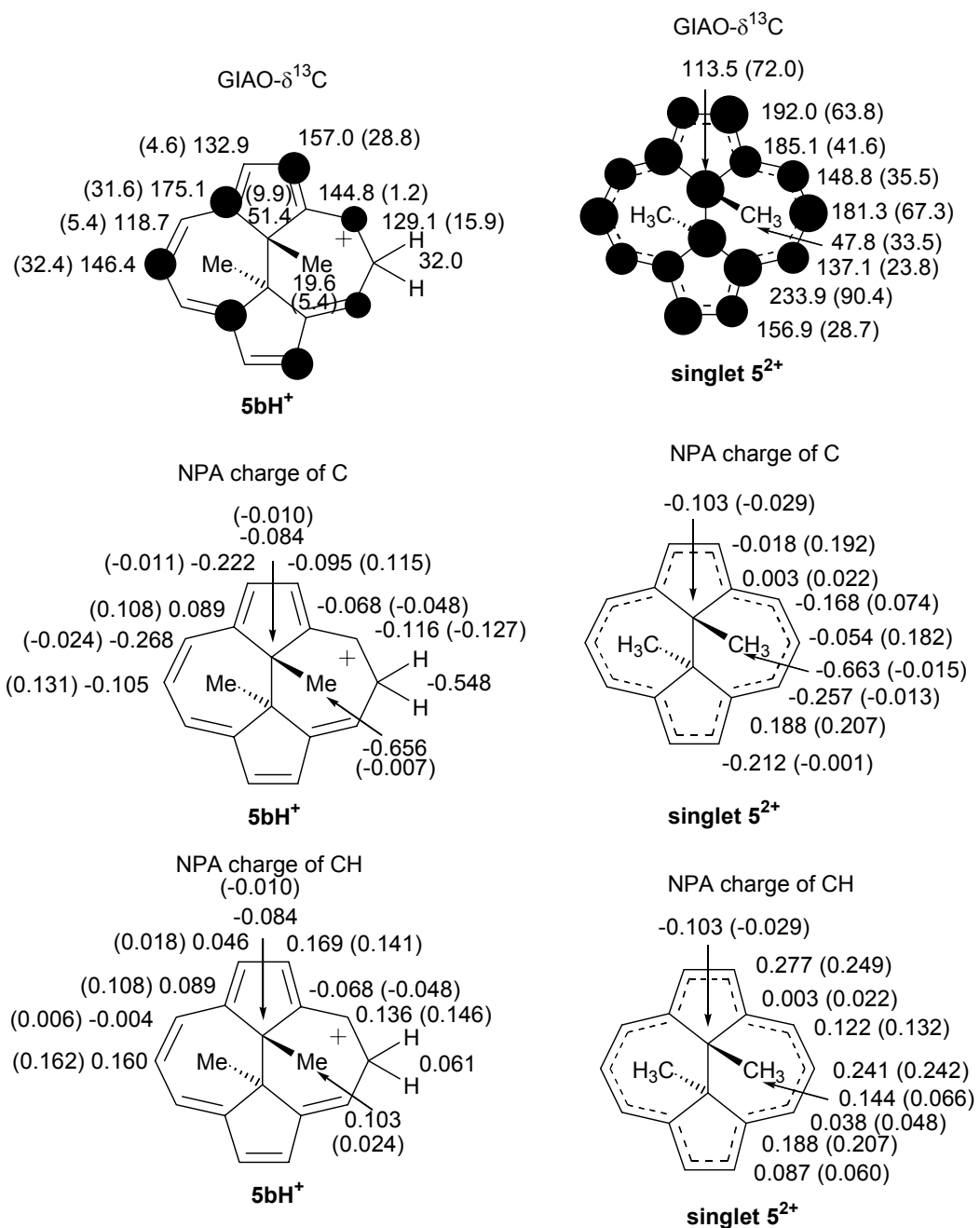


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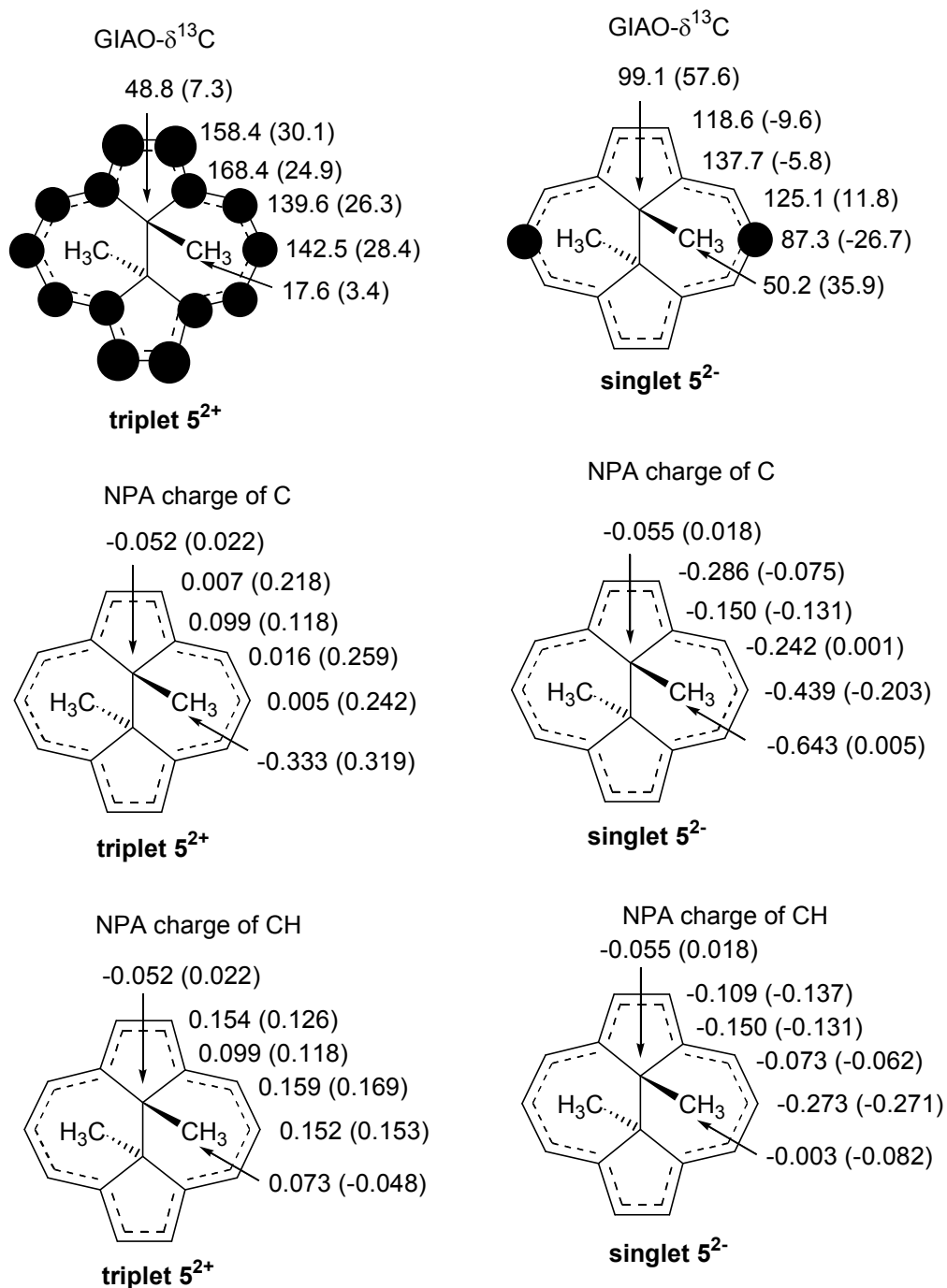
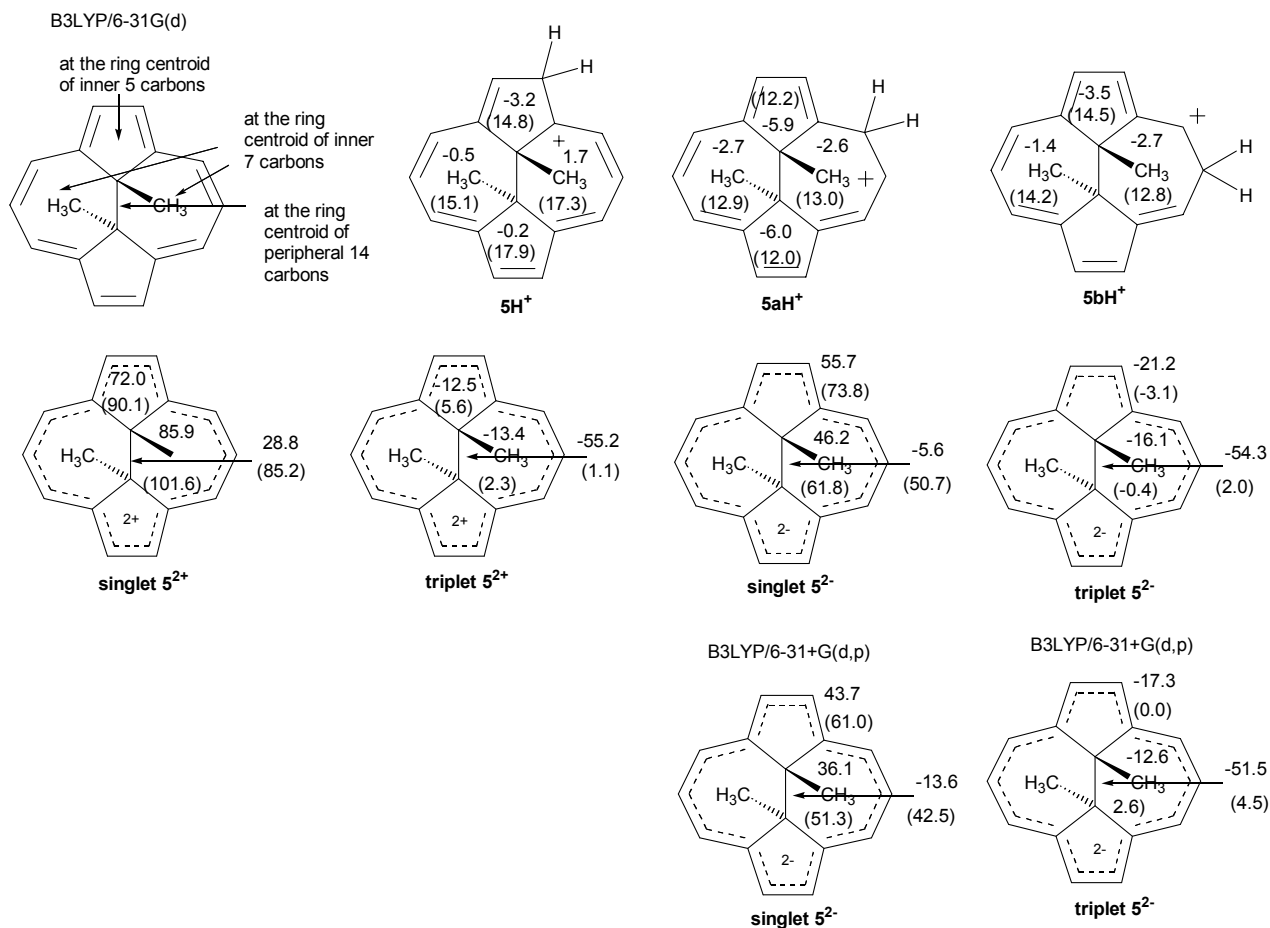
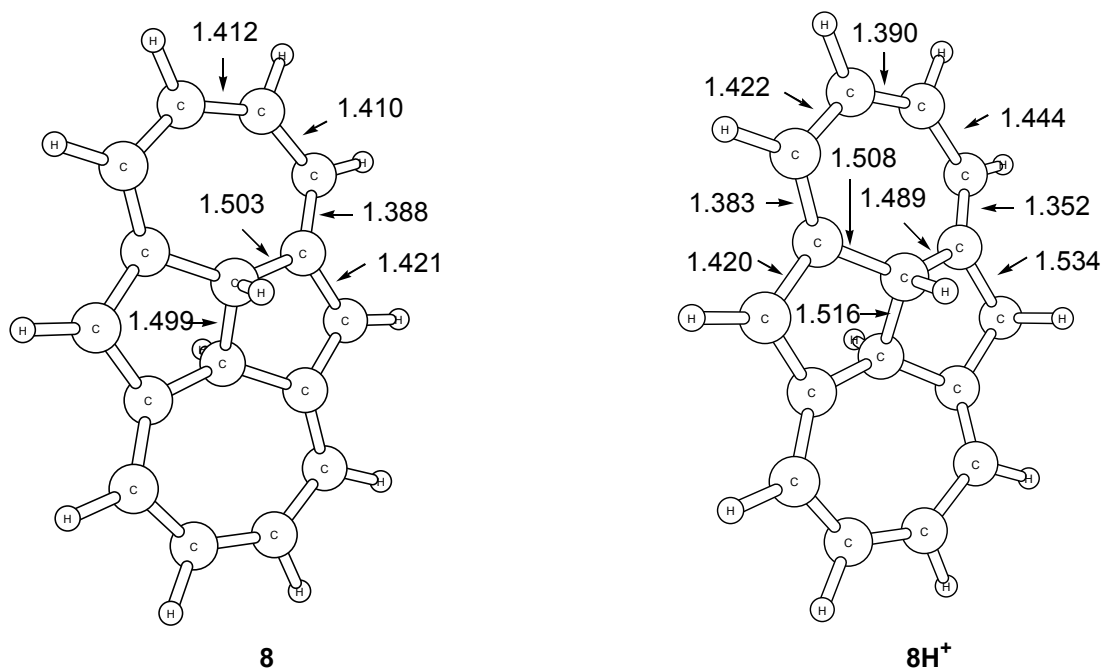


Fig S4 (continued).



**Fig S5.** NICS values for  $5H^+$ ,  $5aH^+$ ,  $5bH^+$ ,  $5^{2+}$ , and  $5^{2-}$  at B3LYP/6-31G(d) or 6-31+G(d,p) level ( $\Delta$ NICS values relative to those of **5** in parentheses).



**Fig S6.** B3LYP/6-31G(d) optimized geometries for **8**,  **$8H^+$** , triplet  **$8^{2+}$** , and singlet  **$8^{2-}$**  (bond length, Å).

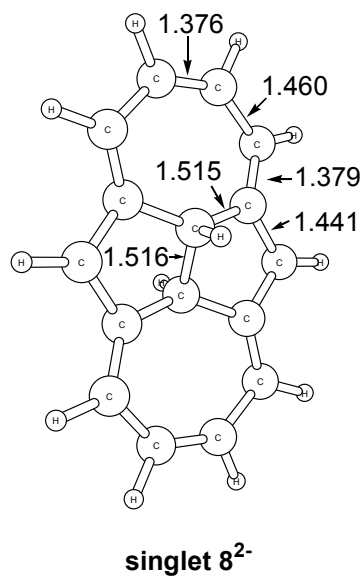
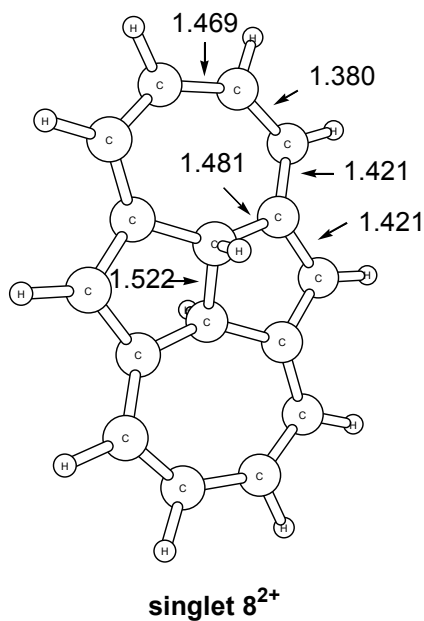


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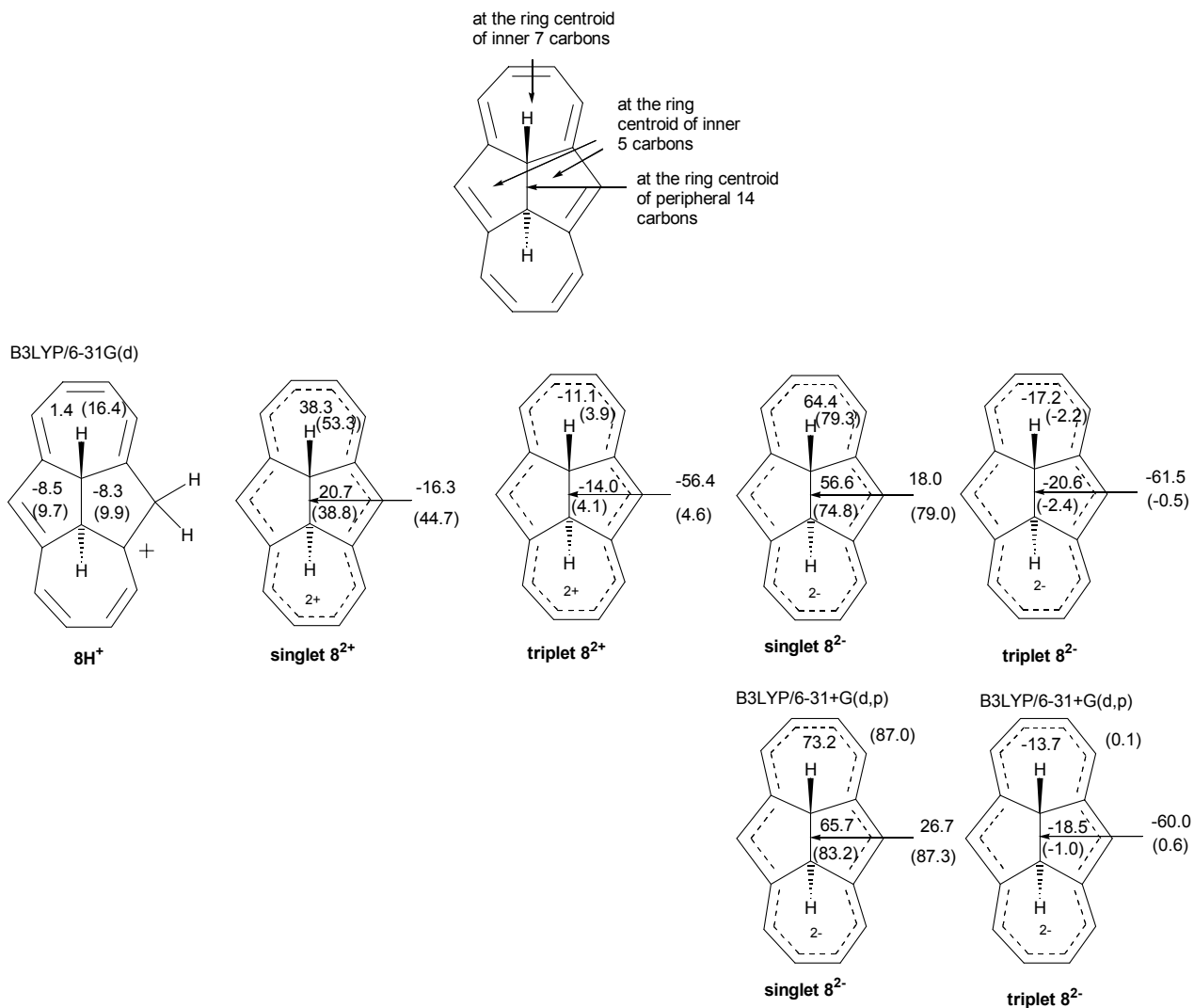
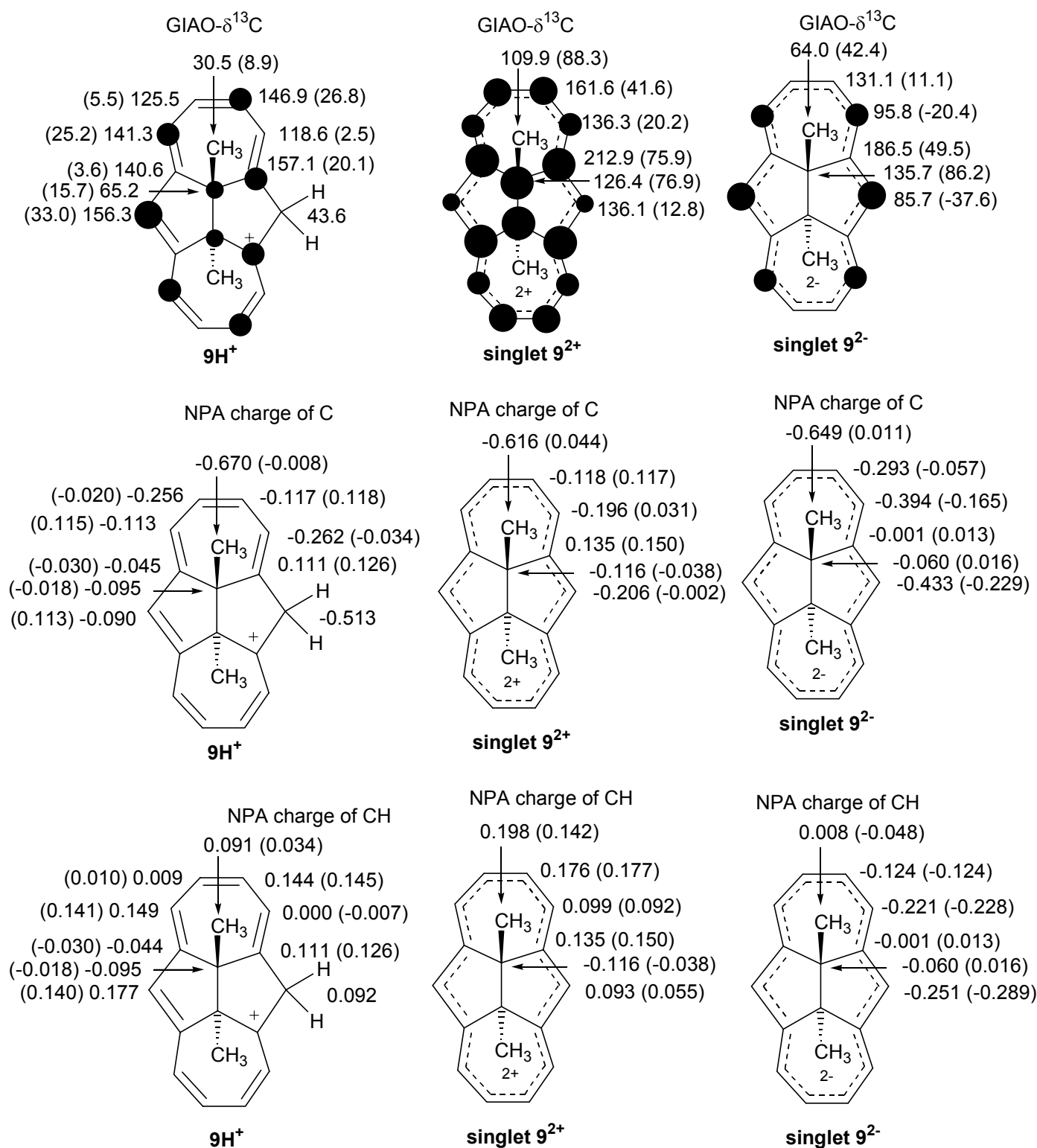
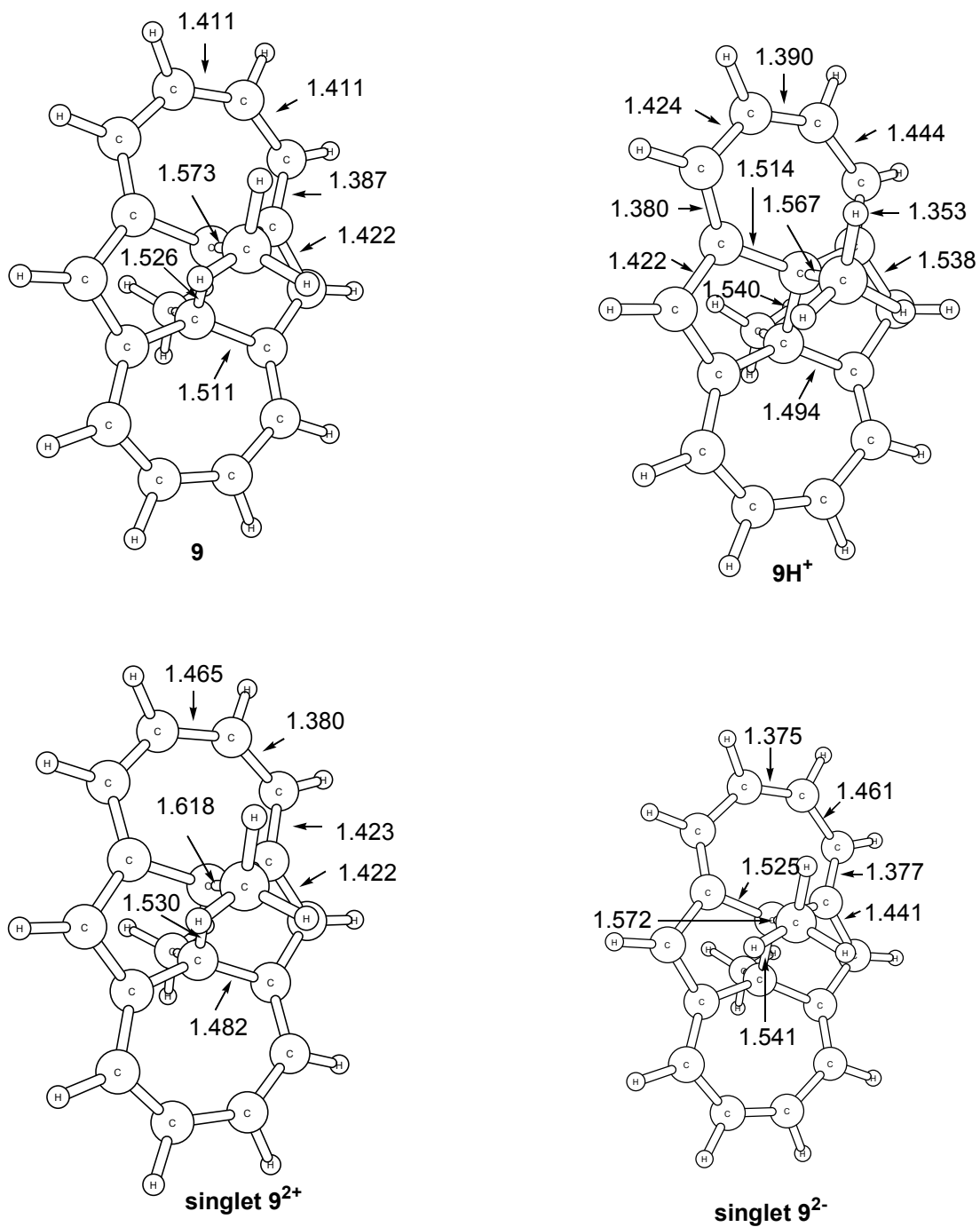


Fig S7. NICS values for  $8H^+$ ,  $8^{2+}$ , and  $8^{2-}$  at B3LYP/6-31G(d) or 6-31+G(d,p) level ( $\Delta$ NICS values relative to those of **8** in parentheses).

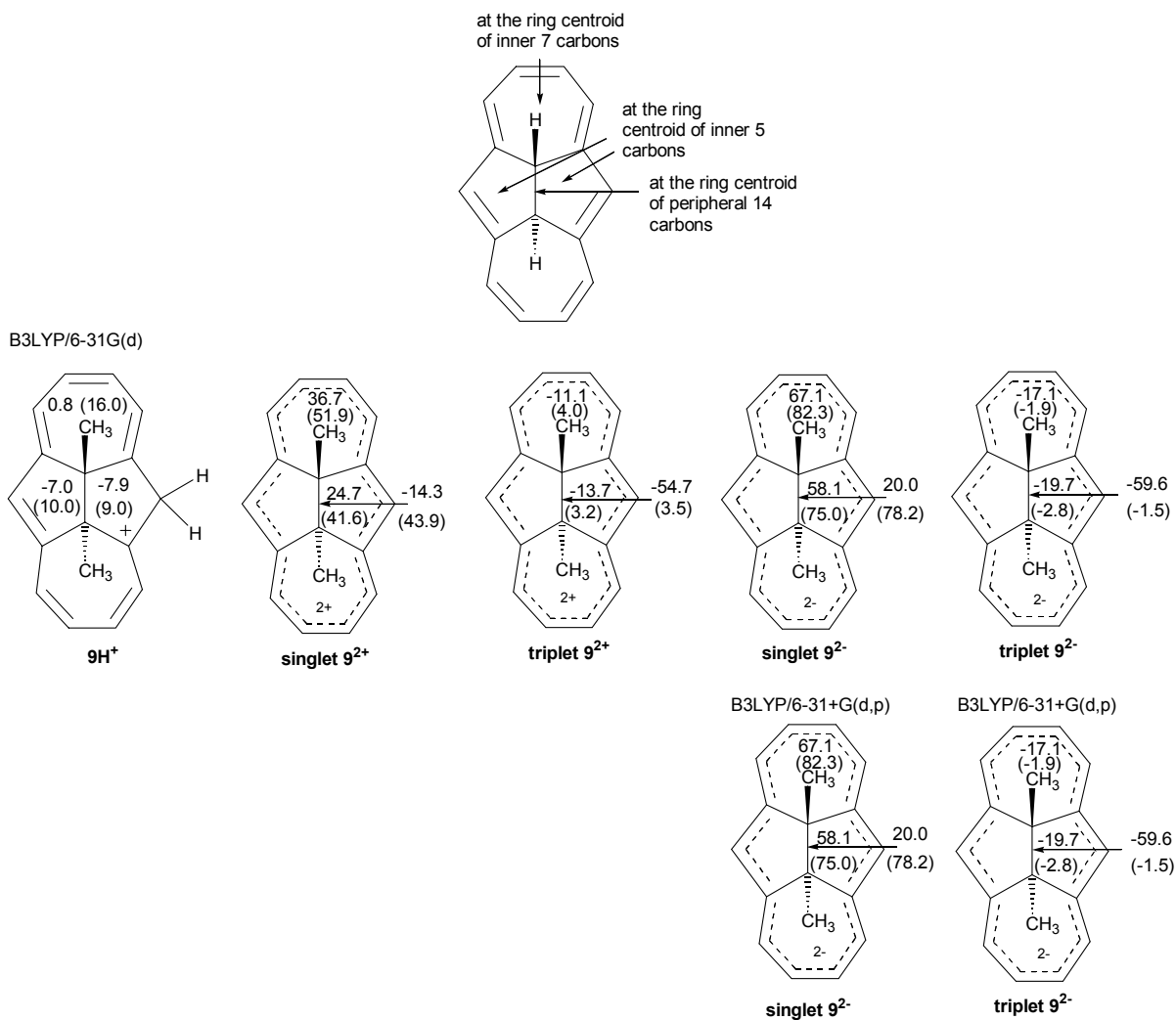




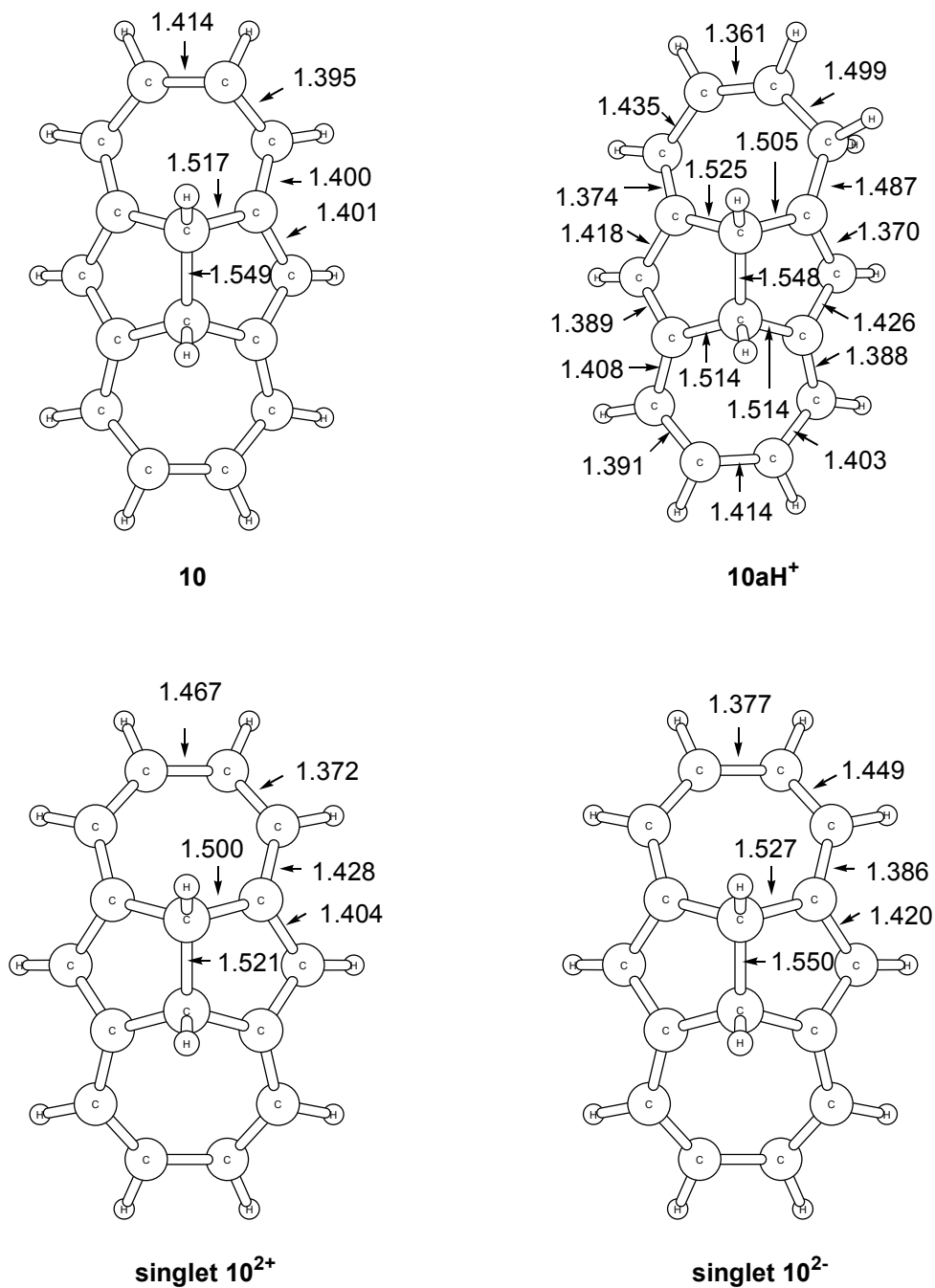
**Fig S8.** Computed  $^{13}\text{C}$  NMR chemical shifts, NPA-derived carbon charges, and NPA-derived overall charges over CH units for  $9\text{H}^+$ ,  $9^{2+}$ , and  $9^{2-}$  at B3LYP/6-31G(d) level ( $\Delta\delta^{13}\text{C}$ 's and  $\Delta$ charges relative to **9** in parentheses). [Dark circles are roughly proportional to the magnitude of  $\Delta\delta^{13}\text{C}$ s (positive/downfield for the carbocations/dication and negative/upfield for the dianion); threshold was set to 10 ppm].



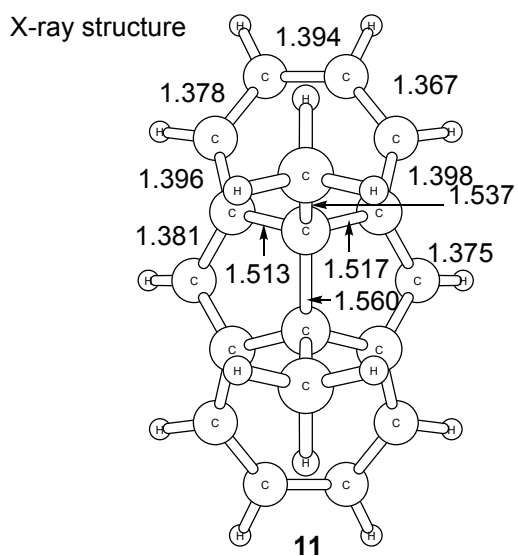
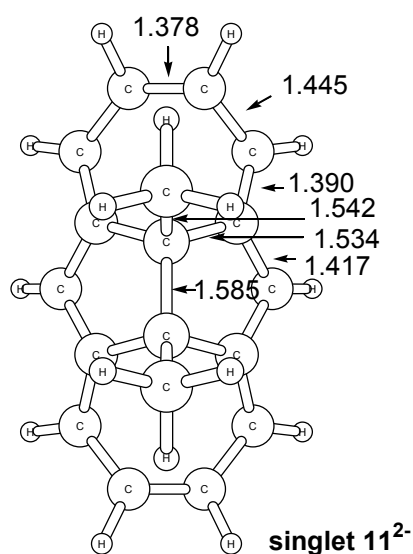
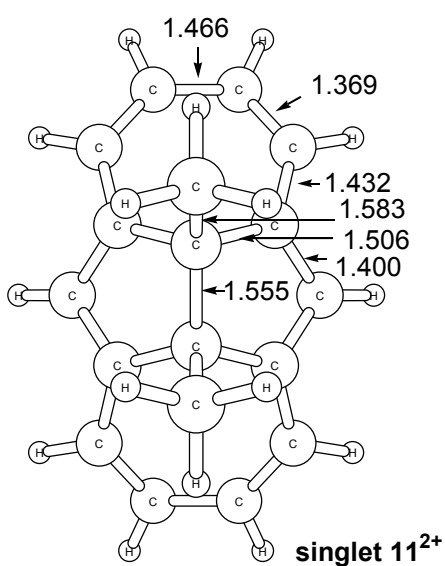
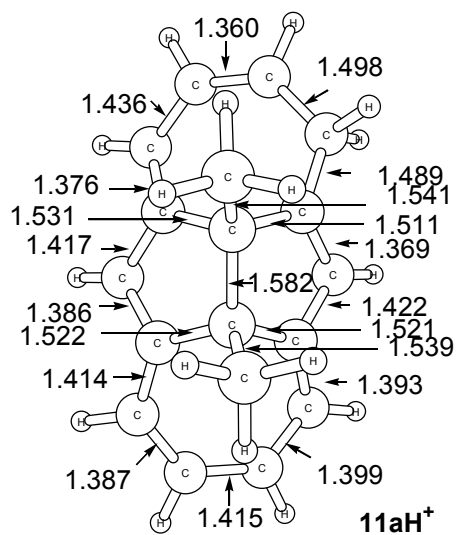
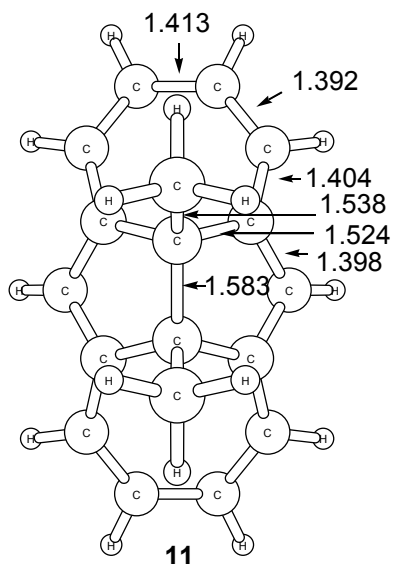
**Fig S9.** B3LYP/6-31G(d) optimized geometries for **9**, **9H<sup>+</sup>**, triplet **9<sup>2+</sup>**, and singlet **9<sup>2-</sup>** (bond length, Å).



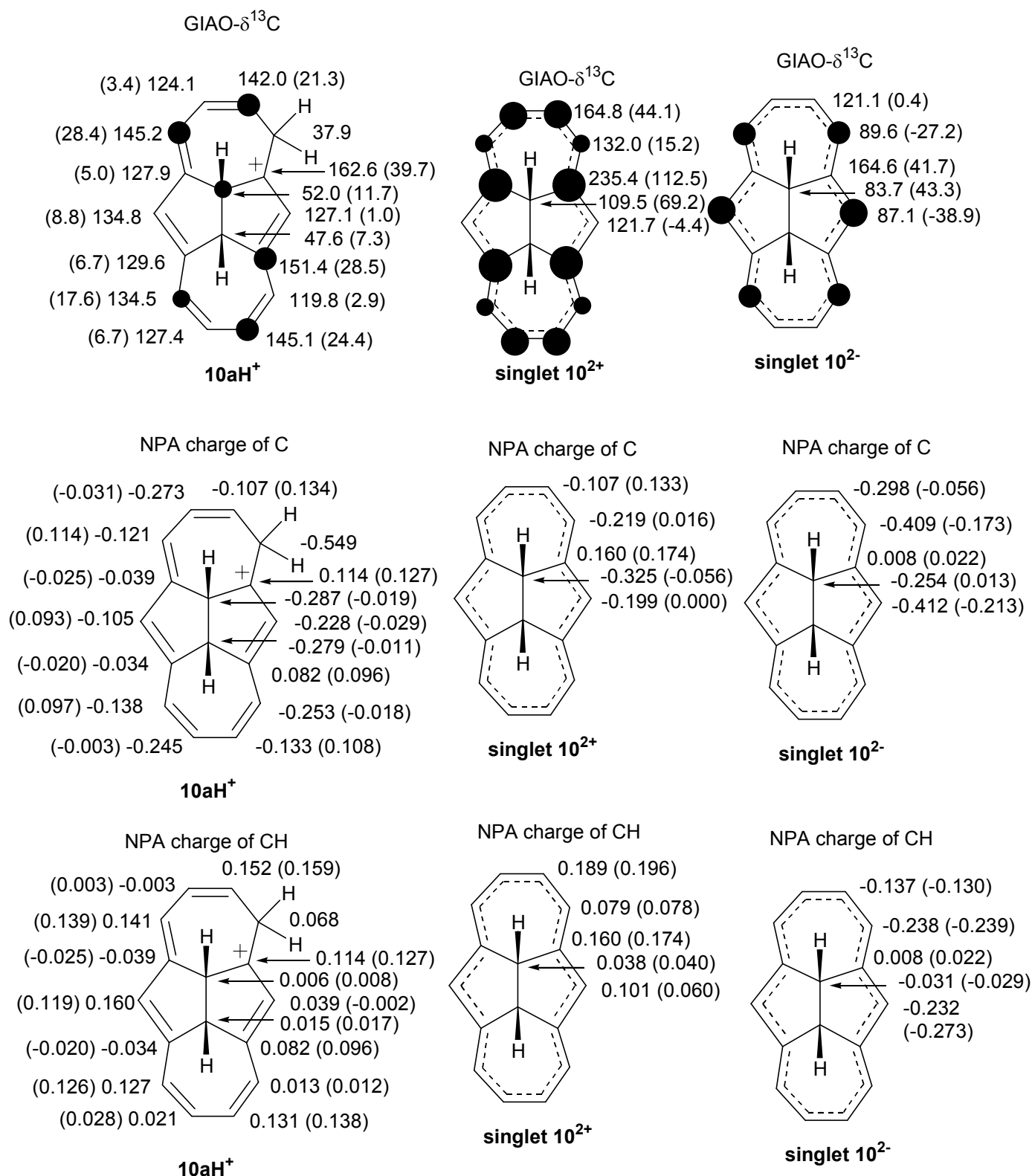
**Fig S10.** NICS values for  $9H^+$ ,  $9^{2+}$ , and  $9^{2-}$  at B3LYP/6-31G(d) or 6-31+G(d,p) level ( $\Delta$ NICS values relative to those of **9** in parentheses).



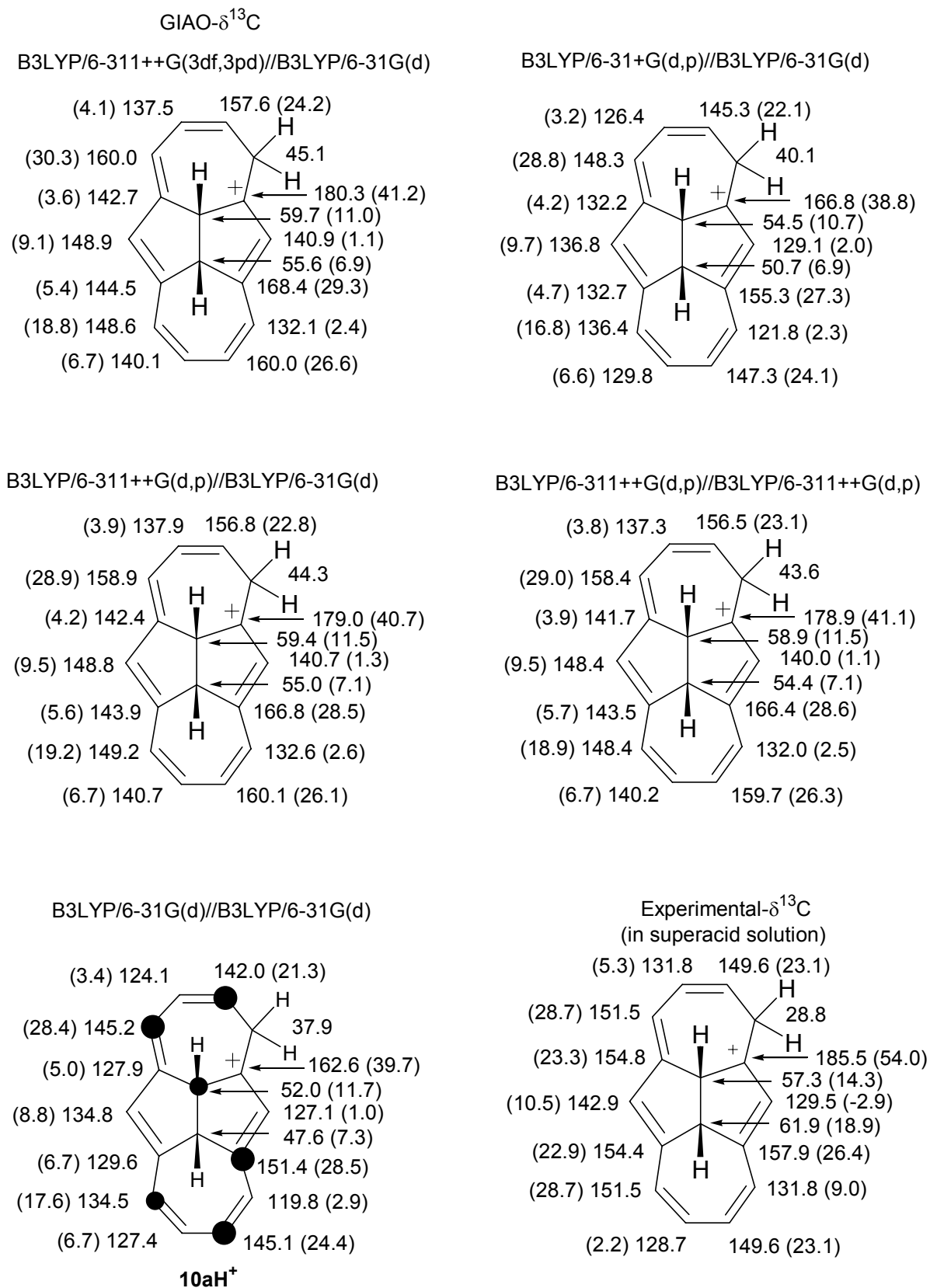
**Fig S11.** B3LYP/6-31G(d) optimized geometries for **10**, **10aH<sup>+</sup>**, singlet **10<sup>2+</sup>**, and singlet **10<sup>2-</sup>** (bond length, Å).



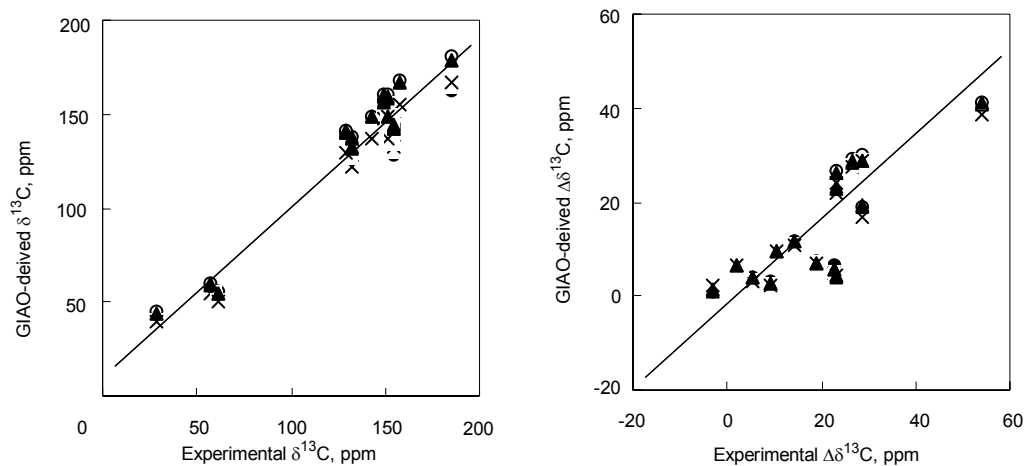
**Fig S12.** B3LYP/6-31G(d) optimized geometries for **11**, **11H<sup>+</sup>**, singlet **11<sup>2+</sup>**, and singlet **11<sup>2-</sup>** and X-ray structure for **11** (bond length, Å).



**Fig S13.** Computed  $^{13}\text{C}$  NMR chemical shifts, NPA-derived carbon charges, and NPA-derived overall charges over CH units for **10aH<sup>+</sup>**, **10<sup>2+</sup>**, and **10<sup>2-</sup>** at B3LYP/6-31G(d) level ( $\Delta\delta^{13}\text{C}$ 's and  $\Delta$ charges relative to **10** in parentheses). [Dark circles are roughly proportional to the magnitude of  $\Delta\delta^{13}\text{C}$ s (positive/downfield for the carbocations/dication and negative/upfield for the dianion); threshold was set to 10 ppm].

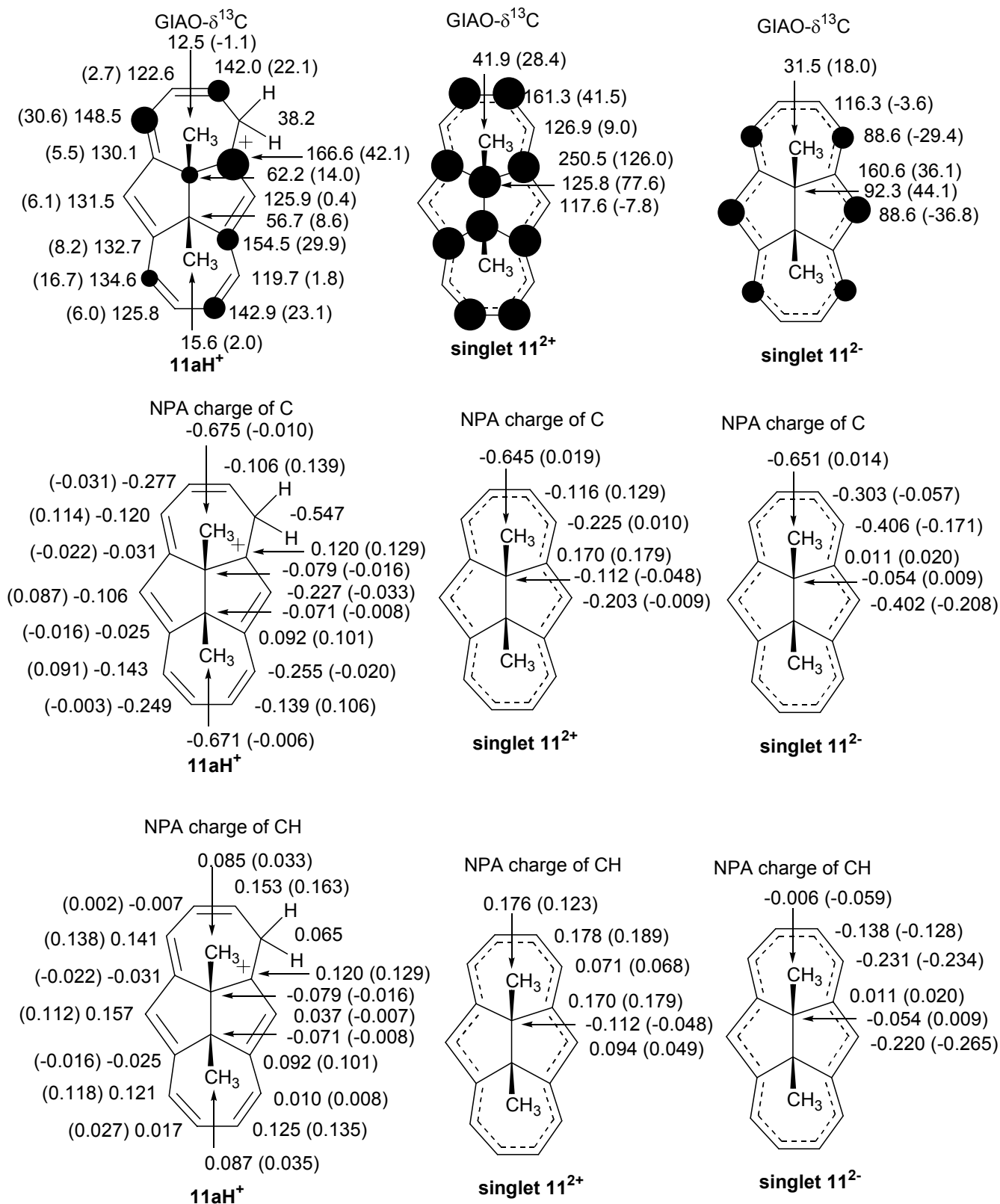


**Fig S13a.** Experimental (Ref 22) and computed  $^{13}\text{C}$  NMR chemical shifts for **10aH<sup>+</sup>**.

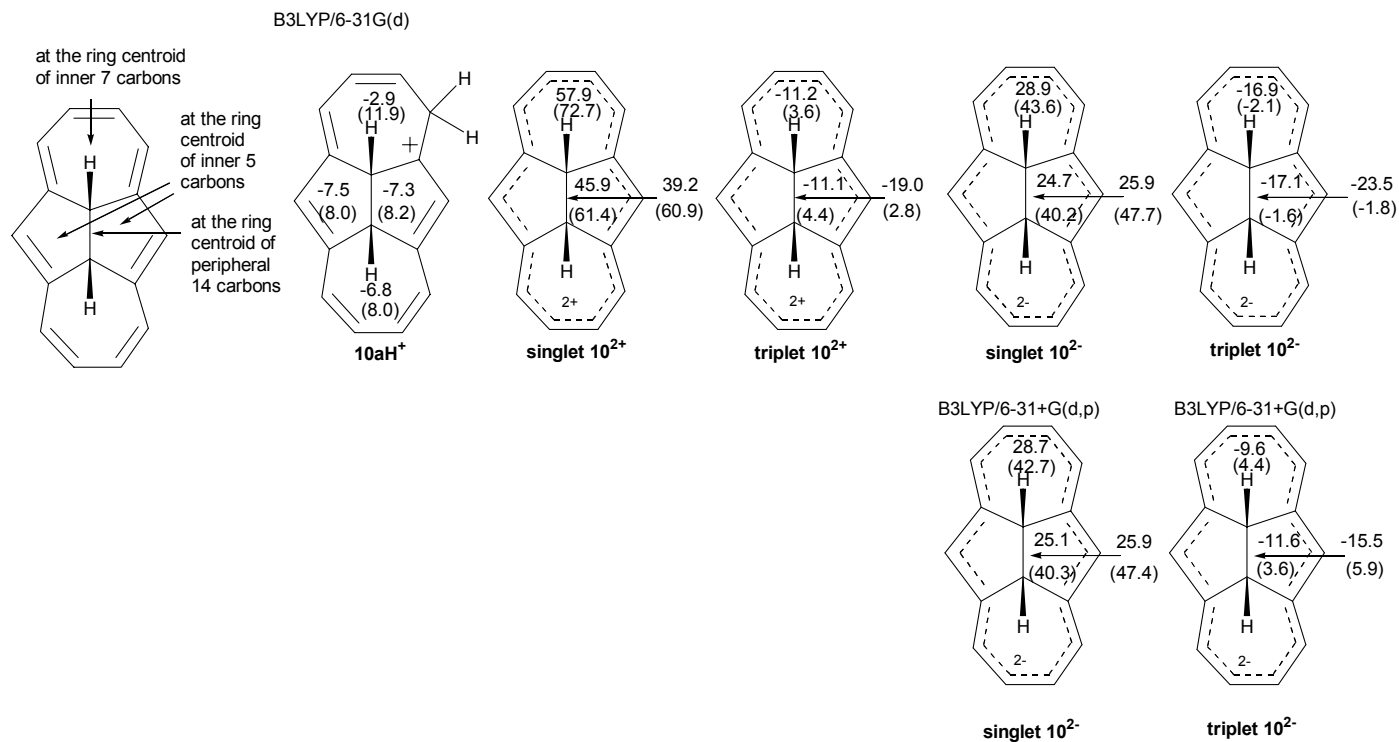


**Figure S13b.** Plots of experimental  $\delta^{13}\text{C}$  vs GIAO-derived  $\delta^{13}\text{C}$  and experimental  $\Delta\delta^{13}\text{C}$  vs GIAO-derived  $\Delta\delta^{13}\text{C}$  for  $10\text{aH}^+$  by (  $\circ$  ) B3LYP/6-31G(d)//B3LYP/6-31G(d), (  $\bullet$  ) B3LYP/6-31+G(d,p)//B3LYP/6-31G(d), (  $\times$  ) B3LYP/6-311++G(d,p)//B3LYP/6-31G(d), (  $\Delta$  ) B3LYP/6-311++G(3df,3pd)//B3LYP/6-31G(d), (  $\blacktriangle$  ) B3LYP/6-311++G(d,p)//B3LYP/6-311++G(d,p) .

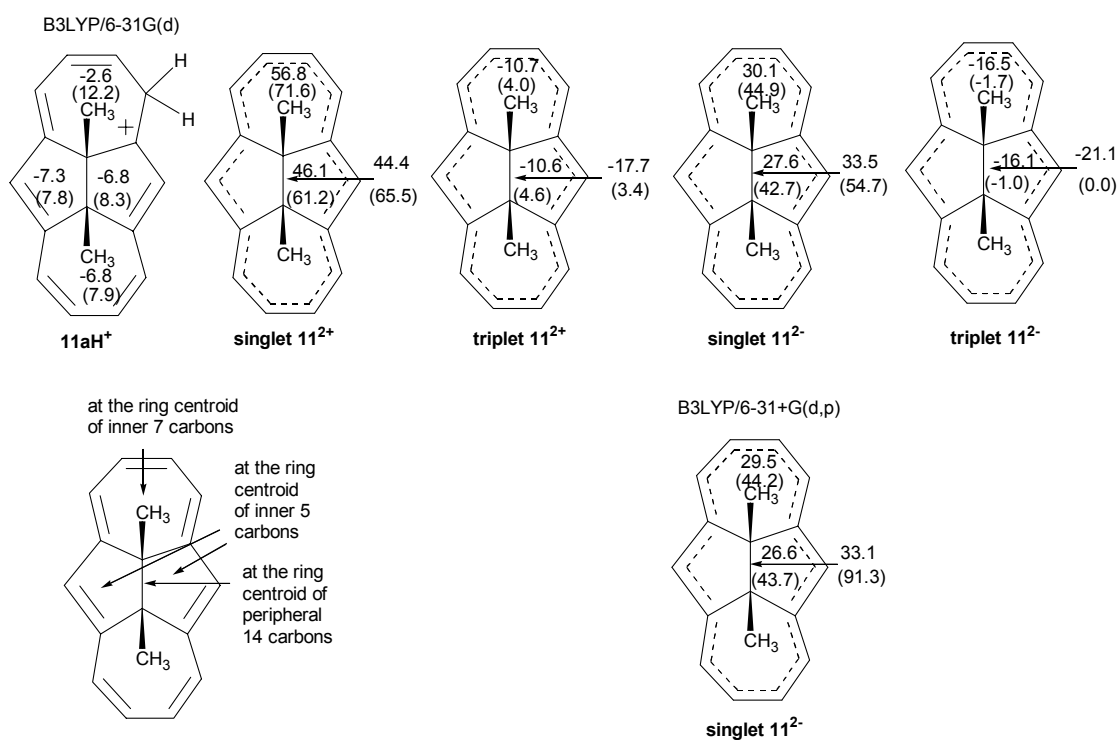




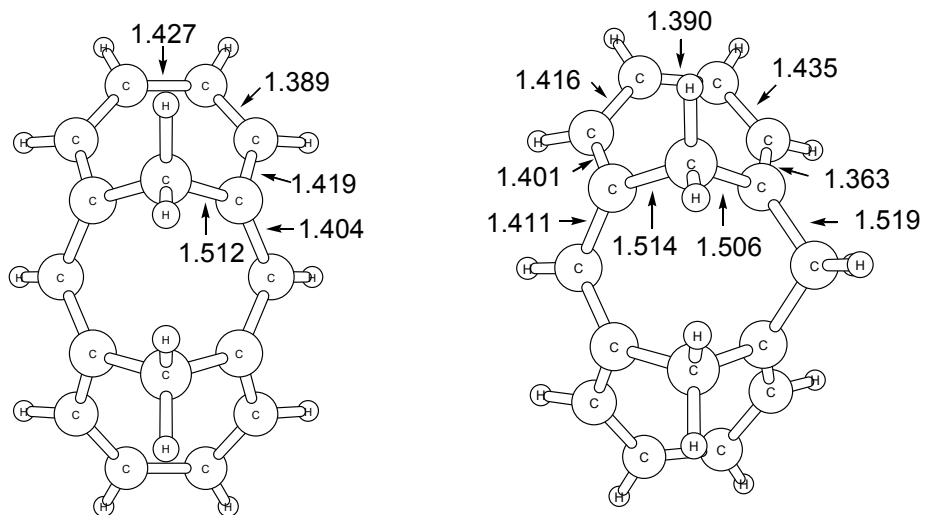
**Fig S14.** Computed  $^{13}\text{C}$  NMR chemical shifts, NPA-derived  $^{13}\text{C}$  carbon charges, and NPA-derived overall charges over CH units for **11aH<sup>+</sup>**, **11<sup>2+</sup>**, and **11<sup>2-</sup>** at B3LYP/6-31G(d) level ( $\Delta\delta^{13}\text{C}$ 's and  $\Delta\text{charges}$  relative to **11** in parentheses). [Dark circles are roughly proportional to the magnitude of  $\Delta\delta^{13}\text{C}$ 's (positive/downfield for the carbocations/dication and negative/upfield for the dianion); threshold was set to 10 ppm].



**Fig S15.** NICS values for  $10aH^+$ ,  $10^{2+}$ , and  $10^{2-}$  at B3LYP/6-31G(d) or 6-31+G(d,p) level ( $\Delta$ NICS values relative to those of **10** in parentheses).

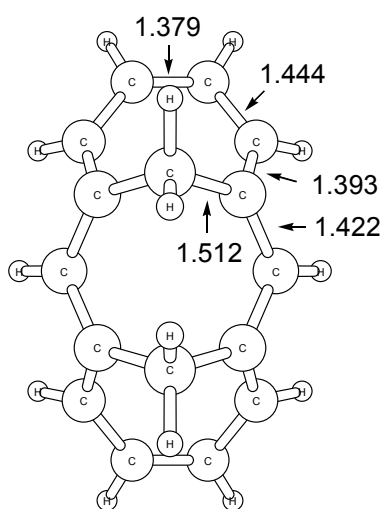


**Fig S16.** NICS values for  $11aH^+$ ,  $11^{2+}$  and  $11^{2-}$  at B3LYP/6-31G(d) or 6-31+G(d,p) level ( $\Delta$ NICS values relative to those of **11** in parentheses).

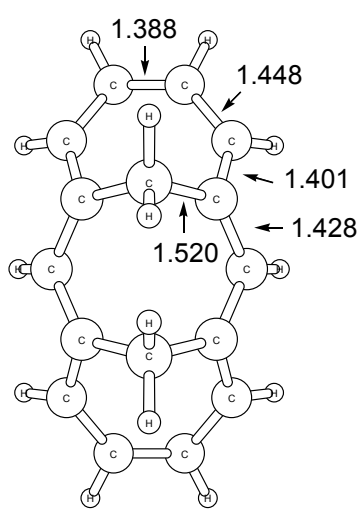


**14**

**14H<sup>+</sup>**

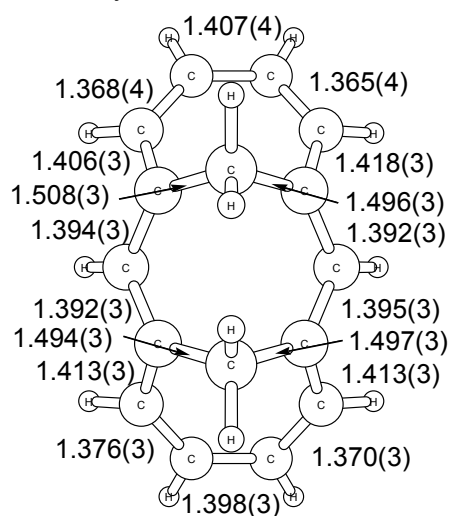


**singlet 14<sup>2+</sup>**



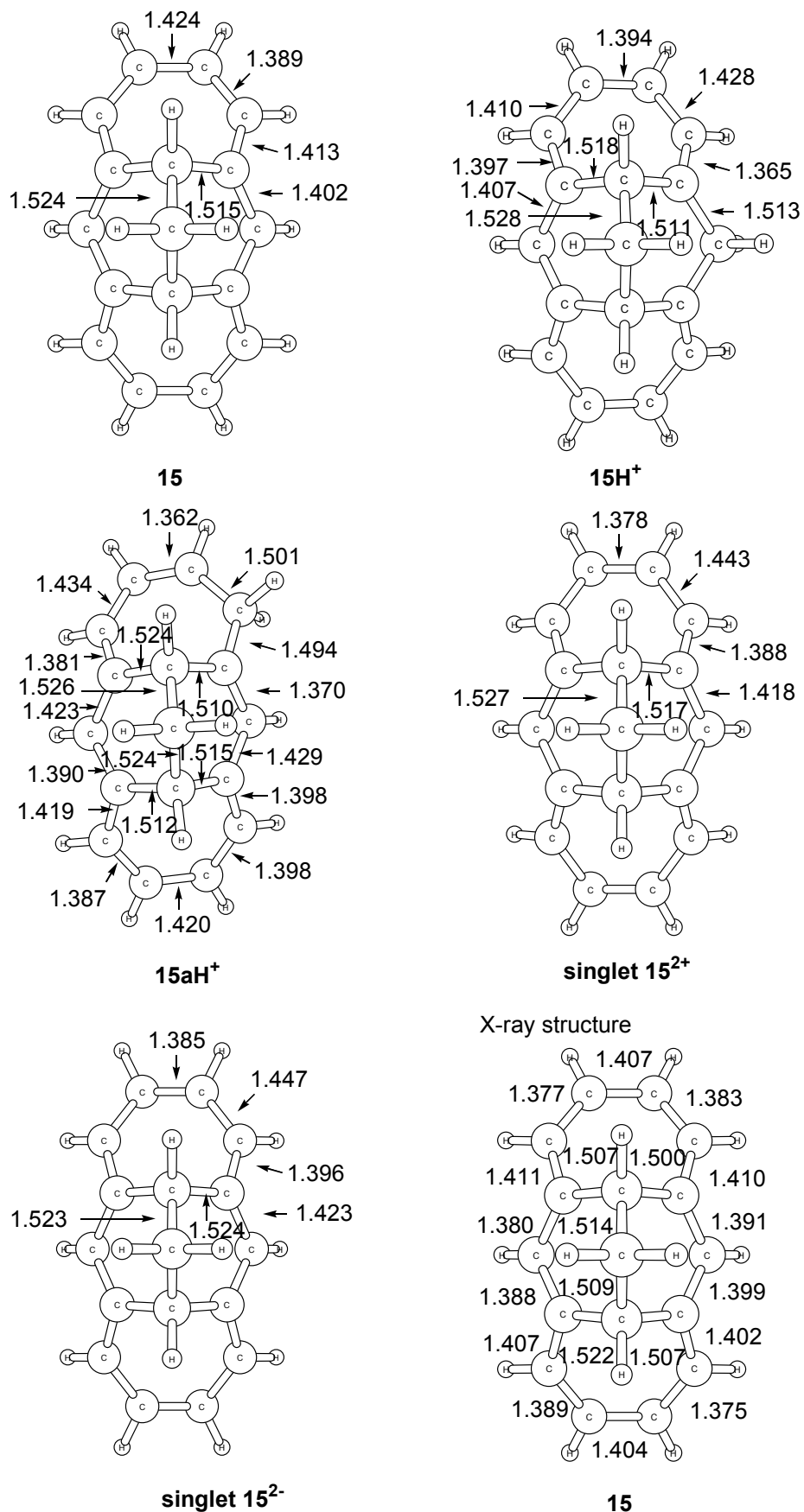
**singlet 14<sup>2-</sup>**

X-ray structure

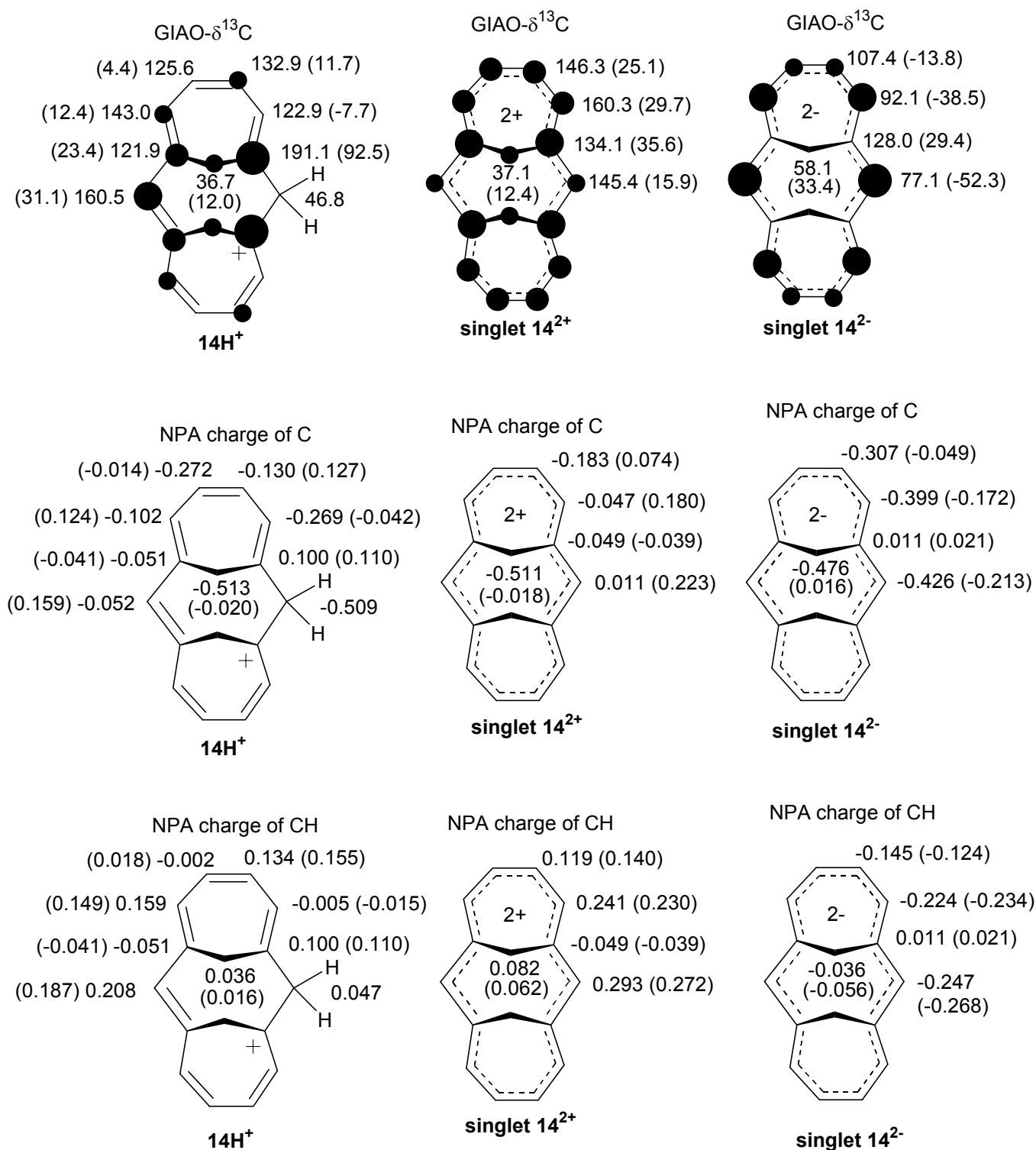


**14**

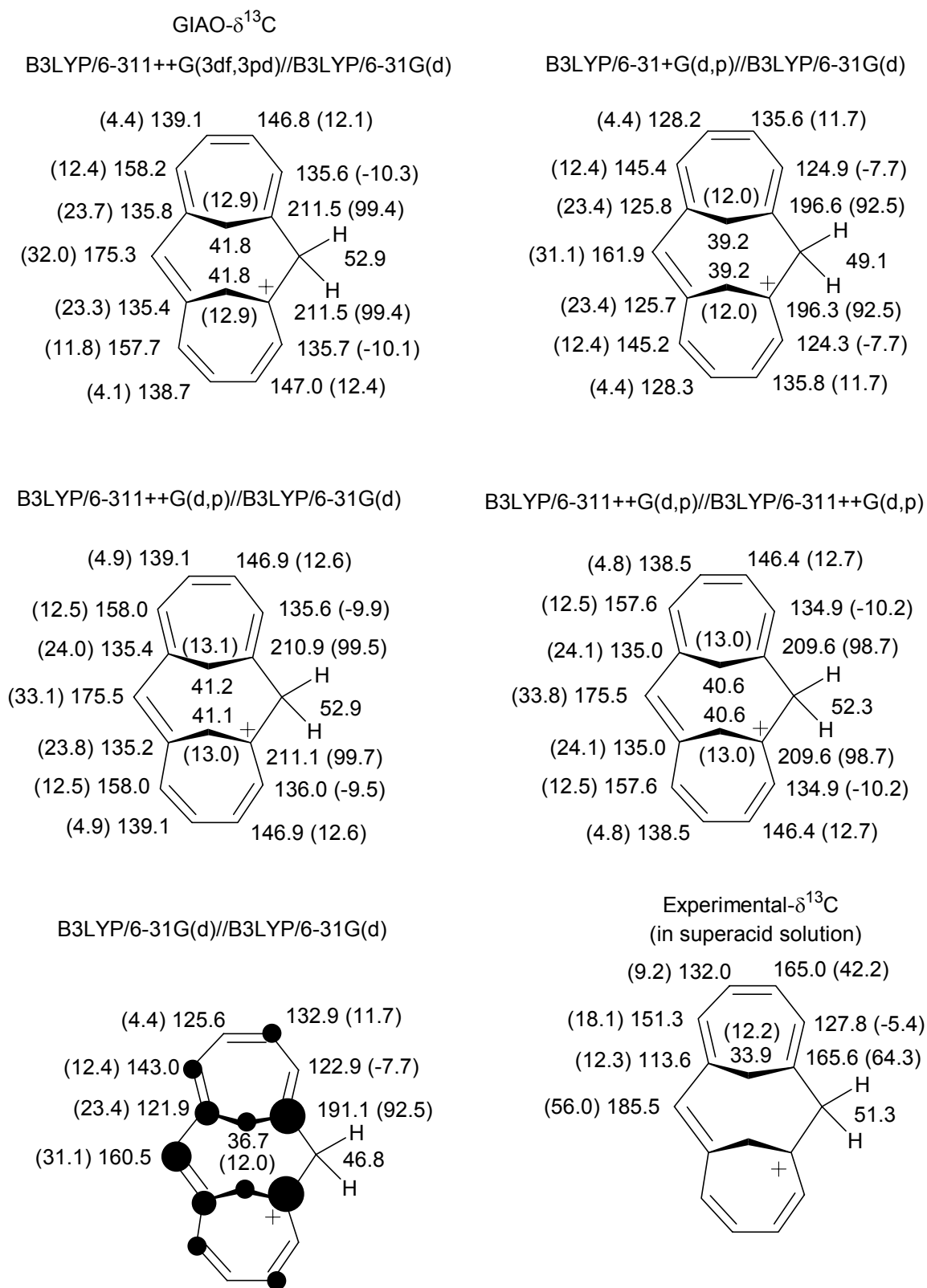
**Fig S17.** B3LYP/6-31G(d) optimized geometries for **14**, **14H<sup>+</sup>**, **singlet 14<sup>2+</sup>** and **singlet 14<sup>2-</sup>** and X-ray structure for **14** (bond length, Å).



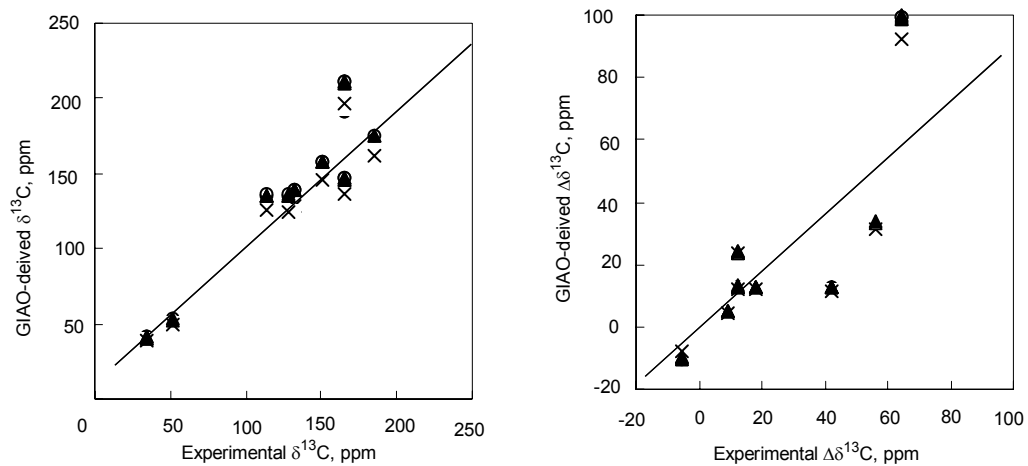
**Fig S18.** B3LYP/6-31G(d) optimized geometries for **15**, **15H<sup>+</sup>**, **15aH<sup>+</sup>**, singlet **15<sup>2+</sup>**, and singlet **15<sup>2-</sup>** and X-ray structure for **15** (bond length, Å).



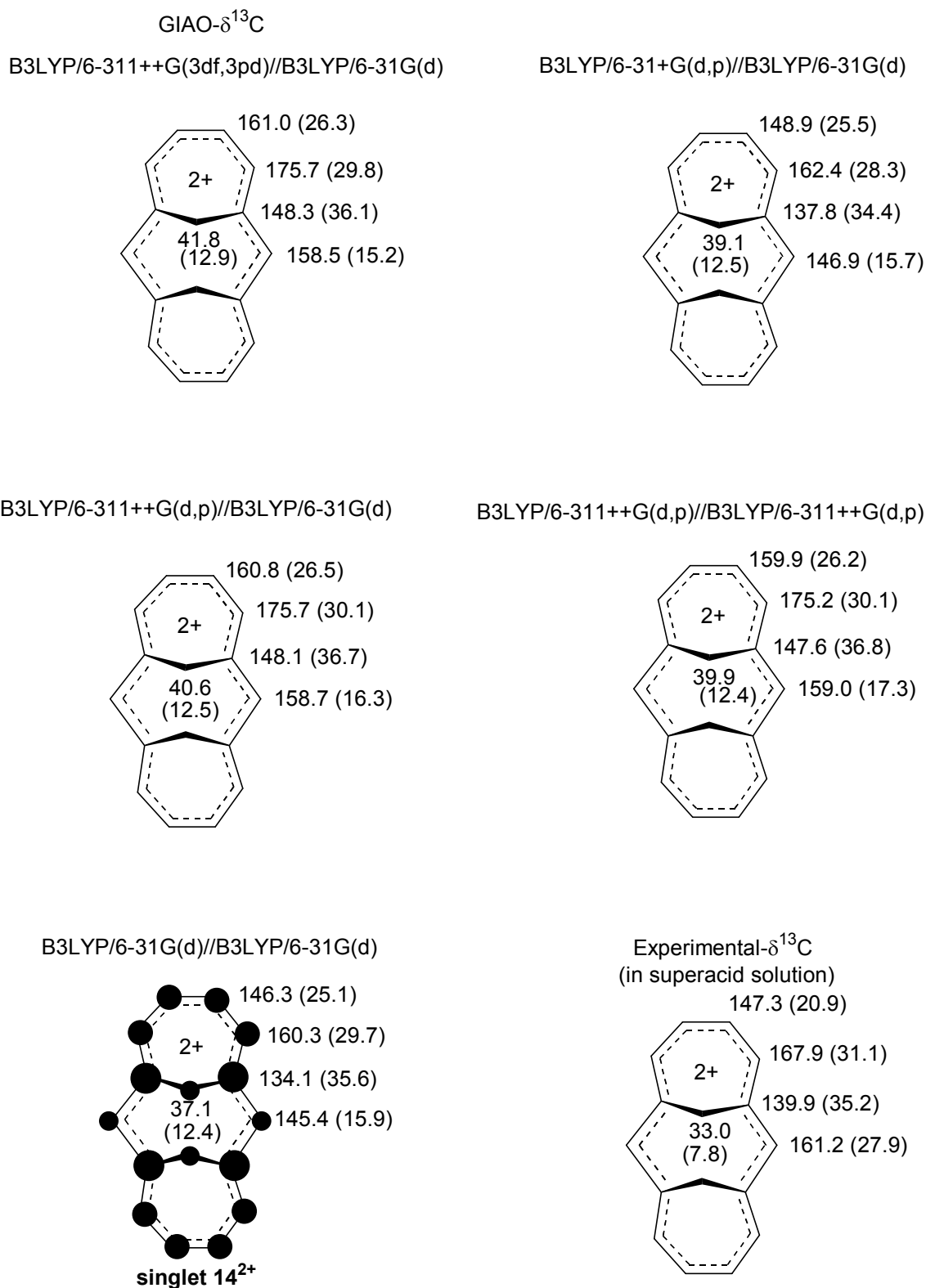
**Fig S19.** Computed  $^{13}\text{C}$  NMR chemical shifts, NPA-derived carbon charges, and NPA-derived overall charges over CH units for  $14\text{H}^+$ ,  $14^{2+}$ , and  $14^{2-}$  at B3LYP/6-31G(d) level ( $\Delta\delta^{13}\text{C}$ 's and  $\Delta$ charges relative to **14** in parentheses). [Dark circles are roughly proportional to the magnitude of  $\Delta\delta^{13}\text{C}$ 's (positive/downfield for the carbocations/dication and negative/upfield for the dianion); threshold was set to 10 ppm].



**Fig S20.** Experimental (Ref 22) and computed  $^{13}\text{C}$  NMR chemical shifts for  $14\text{H}^+$ .

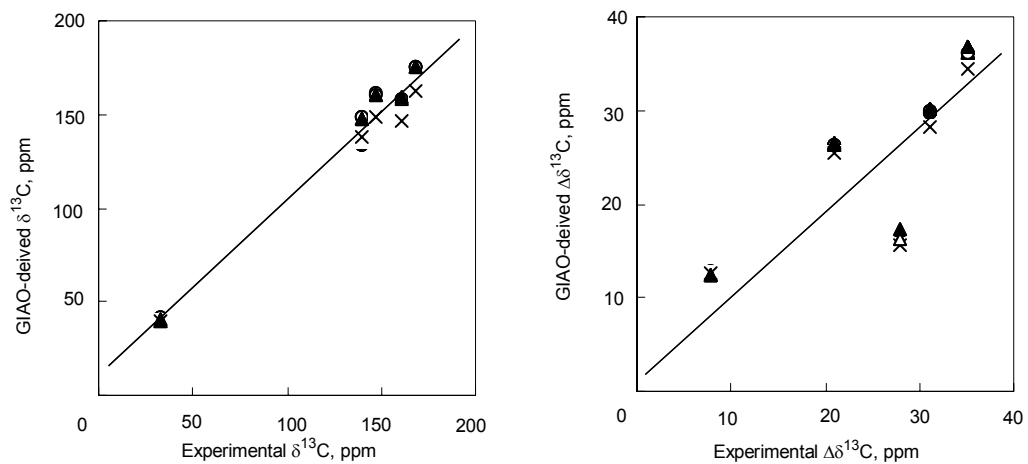


**Figure S20a.** Plots of experimental  $\delta^{13}\text{C}$  vs GIAO-derived  $\delta^{13}\text{C}$  and experimental  $\Delta\delta^{13}\text{C}$  vs GIAO-derived  $\Delta\delta^{13}\text{C}$  for  $14\text{H}^+$  by (  $\circ$  ) B3LYP/6-31G(d)//B3LYP/6-31G(d), (  $\bullet$  ) B3LYP/6-31+G(d,p)//B3LYP/6-31G(d), (  $\times$  ) B3LYP/6-311++G(d,p)//B3LYP/6-31G(d), (  $\Delta$  ) B3LYP/6-311++G(3df,3pd)//B3LYP/6-31G(d), (  $\blacktriangle$  ) B3LYP/6-311++G(d,p)//B3LYP/6-311++G(d,p).

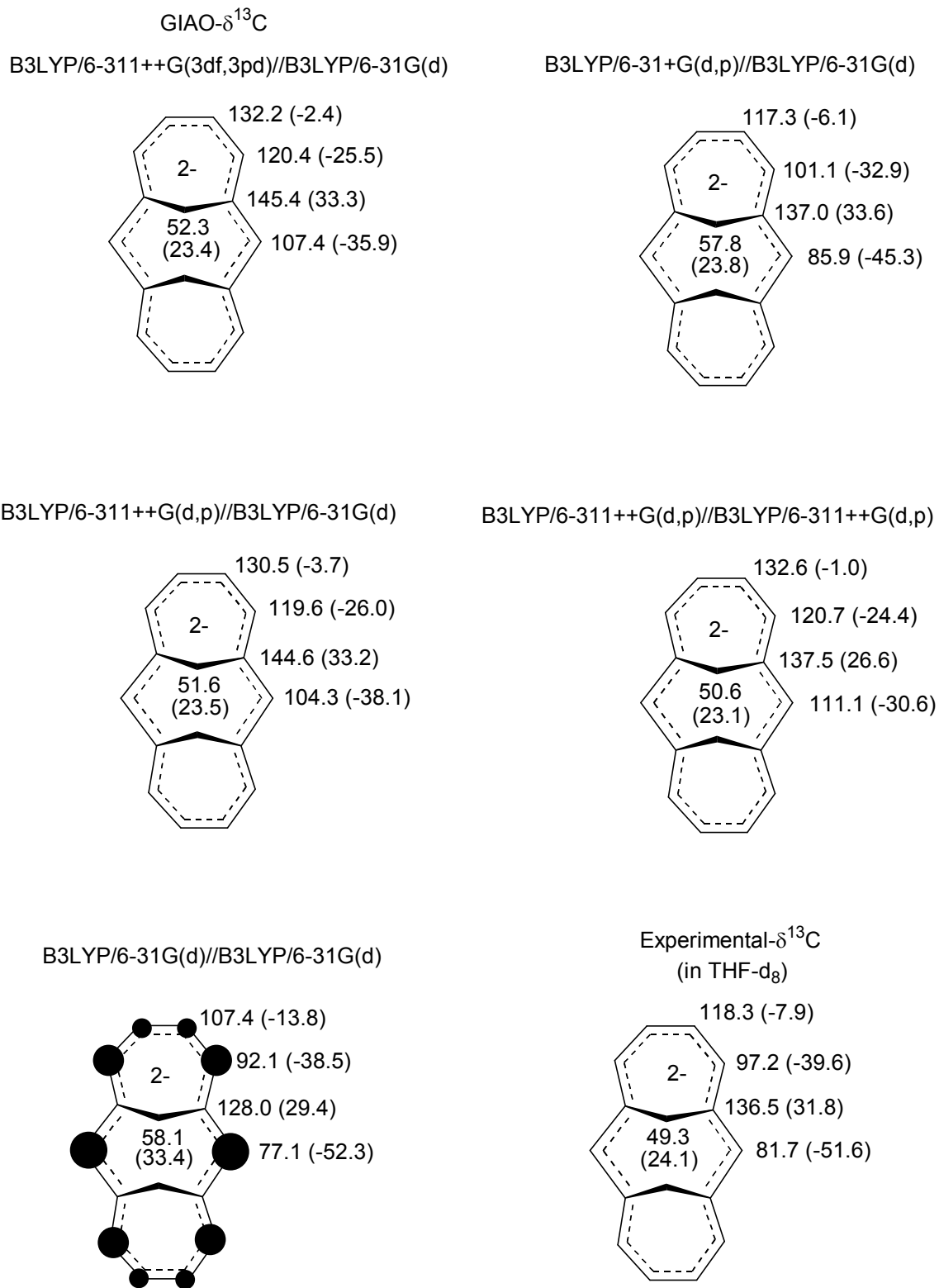


**Fig S21.** Experimental (Ref 22) and computed  $^{13}\text{C}$  NMR chemical shifts for singlet  $14^{2+}$ .

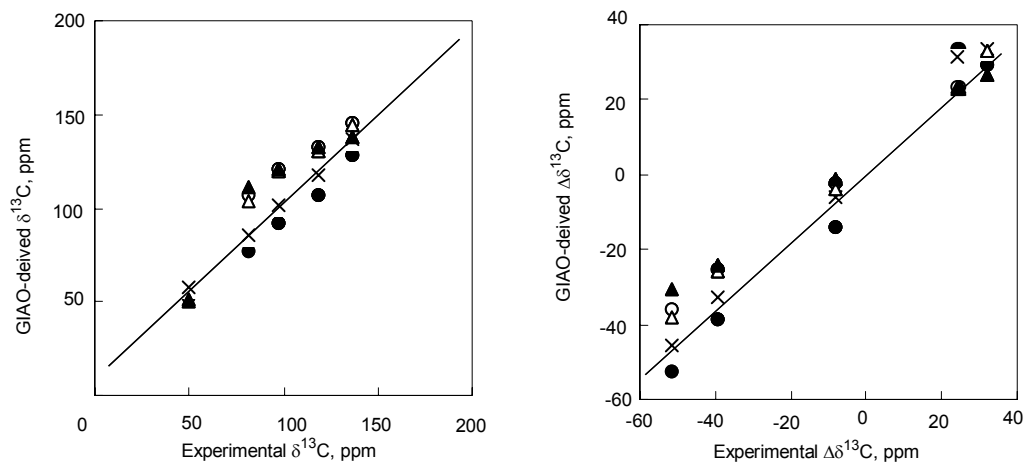




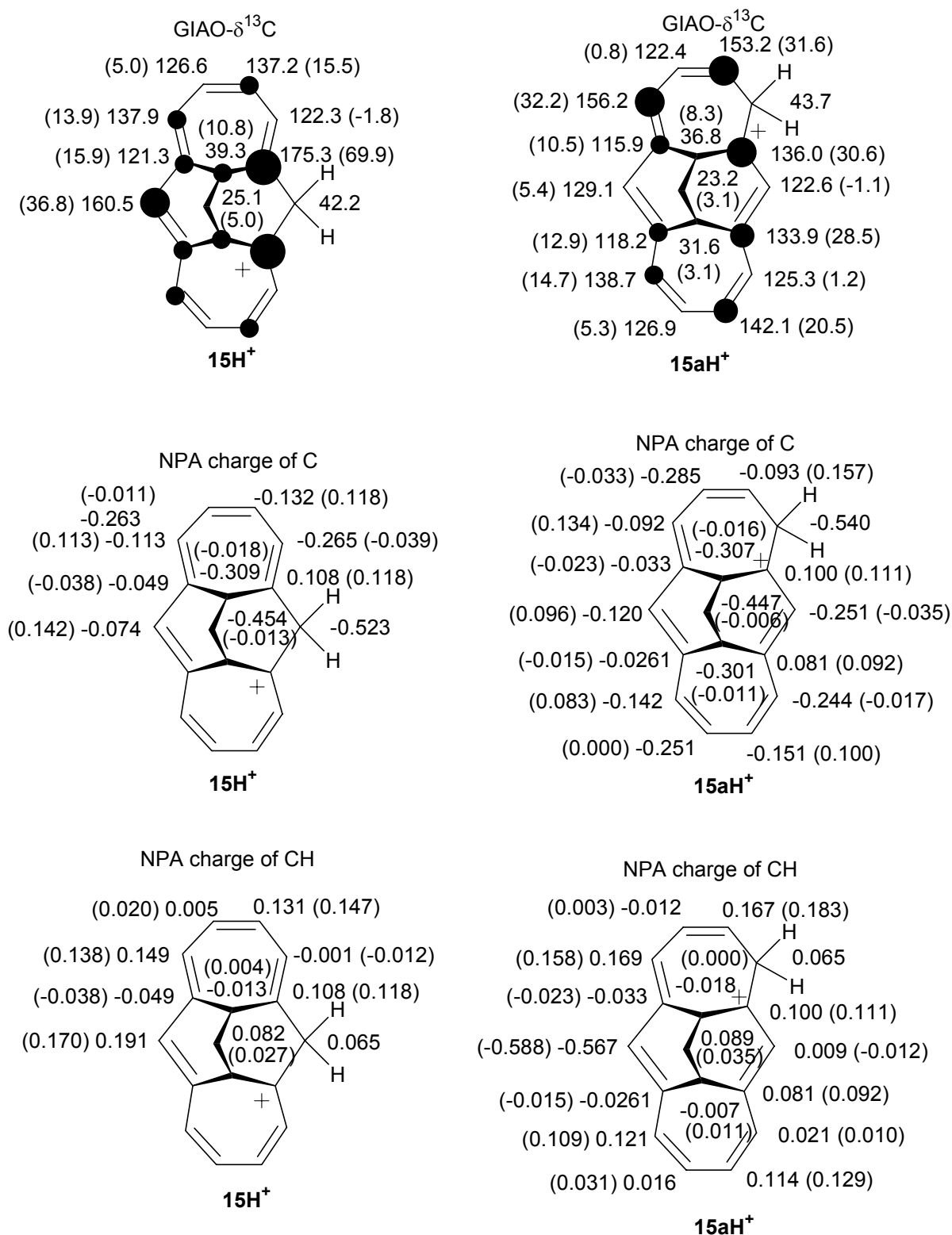
**Figure S21a.** Plots of experimental  $\delta^{13}\text{C}$  vs GIAO-derived  $\delta^{13}\text{C}$  and experimental  $\Delta\delta^{13}\text{C}$  vs GIAO-derived  $\Delta\delta^{13}\text{C}$  for  $14^{2+}$  by (  $\circ$  ) B3LYP/6-31G(d)//B3LYP/6-31G(d), (  $\bullet$  ) B3LYP/6-31+G(d,p)//B3LYP/6-31G(d), (  $\times$  ) B3LYP/6-311++G(d,p)//B3LYP/6-31G(d), (  $\Delta$  ) B3LYP/6-311++G(3df,3pd)//B3LYP/6-31G(d), (  $\blacktriangle$  ) B3LYP/6-311++(d,p)//B3LYP/6-311++G(d,p).



**Fig S21b.** Experimental (Ref 21) and computed  $^{13}\text{C}$  NMR chemical shifts for singlet  $14^{2-}$ .



**Figure S21c.** Plots of experimental  $\delta^{13}\text{C}$  vs GIAO-derived  $\delta^{13}\text{C}$  and experimental  $\Delta\delta^{13}\text{C}$  vs GIAO-derived  $\Delta\delta^{13}\text{C}$  for  $14^{2-}$  by (  $\circ$  ) B3LYP/6-31G(d)//B3LYP/6-31G(d), (  $\bullet$  ) B3LYP/6-31+G(d,p)//B3LYP/6-31G(d), (  $\times$  ) B3LYP/6-311++G(d,p)//B3LYP/6-31G(d), (  $\Delta$  ) B3LYP/6-311++G(3df,3pd)//B3LYP/6-31G(d), (  $\blacktriangle$  ) B3LYP/6-311++G(d,p)//B3LYP/6-311++G(d,p).



**Fig S22.** Computed  $^{13}\text{C}$  NMR chemical shifts, NPA-derived carbon charges, and NPA-derived overall charges over CH units for  $15\text{H}^+$ ,  $15\text{aH}^+$ ,  $15^{2+}$ , and  $15^{2-}$  at B3LYP/6-31G(d) level ( $\Delta\delta^{13}\text{C}$ 's and  $\Delta$ charges relative to **15** in parentheses). [Dark circles are roughly proportional to the magnitude of  $\Delta\delta^{13}\text{C}$ s (positive/downfield for the carbocations/dication and negative/upfield for the dianion); threshold was set to 10 ppm].

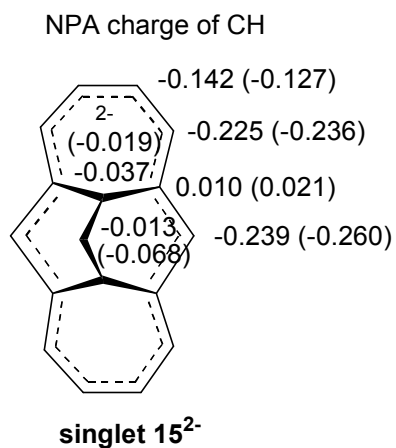
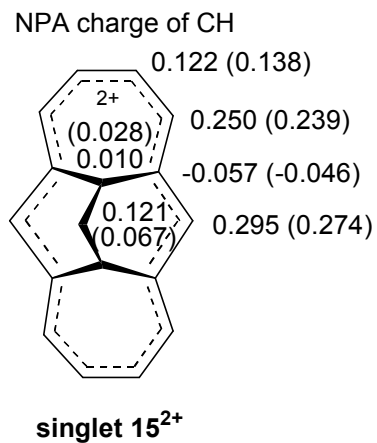
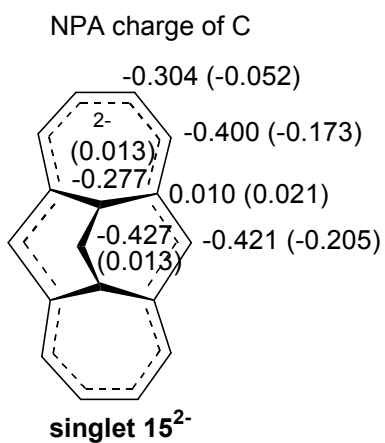
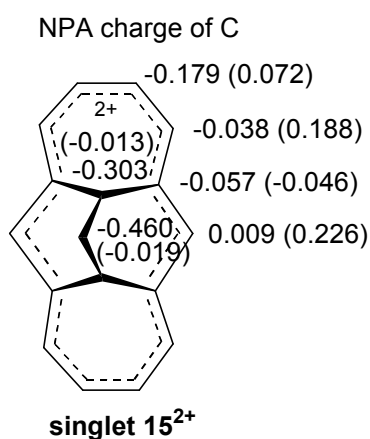
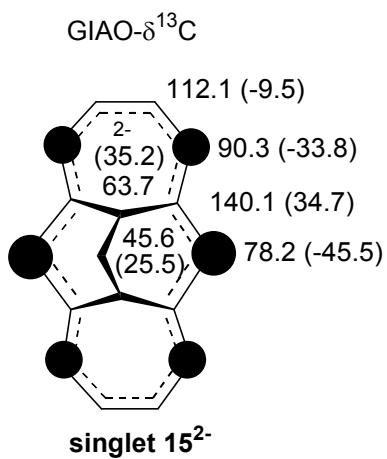
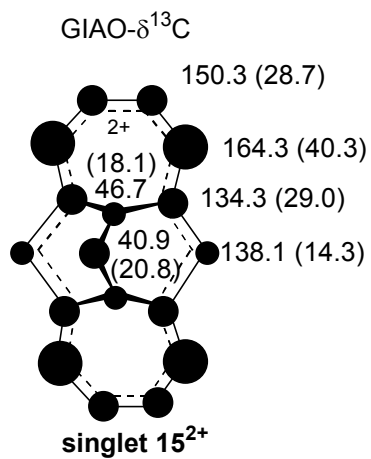
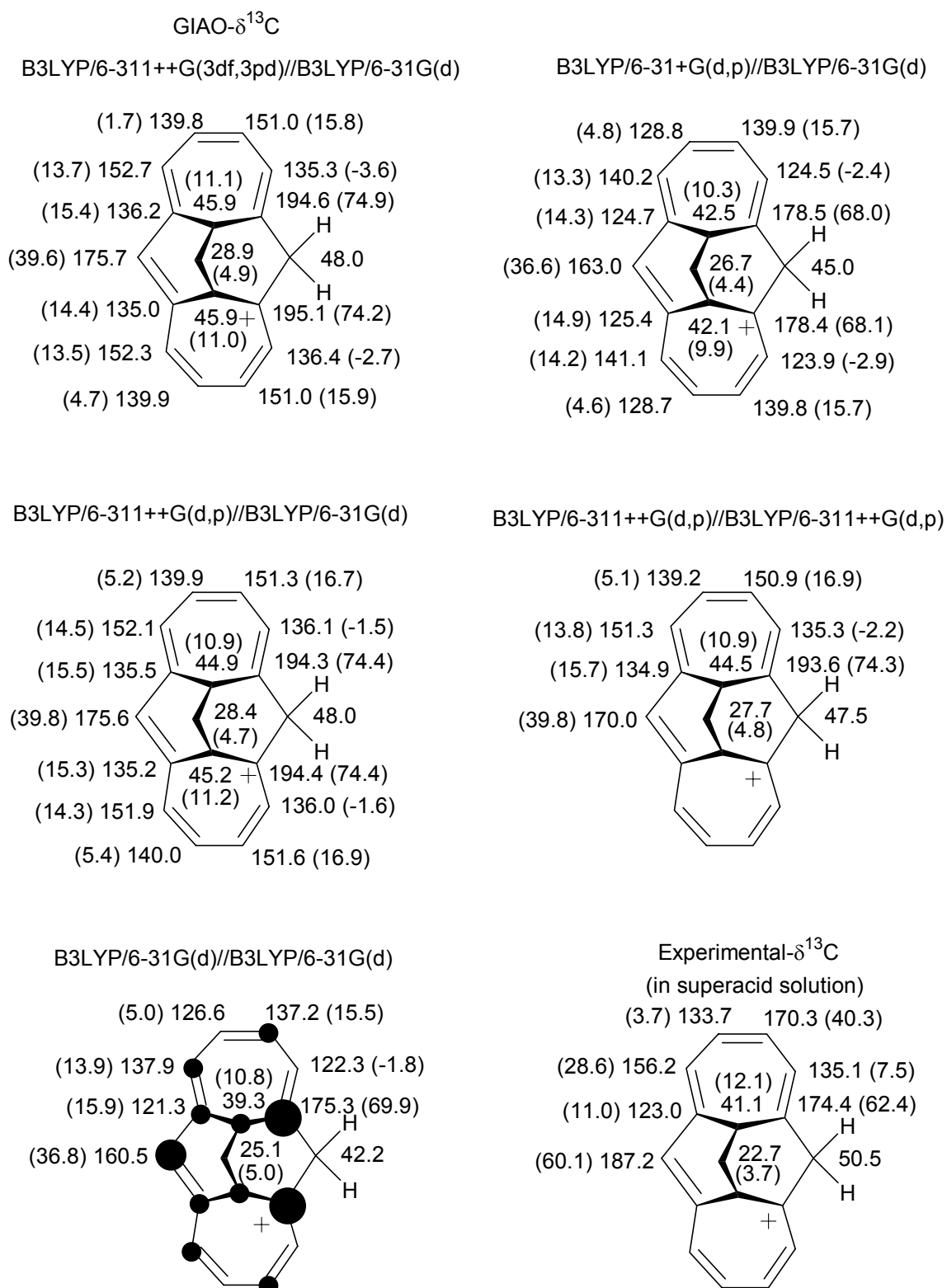
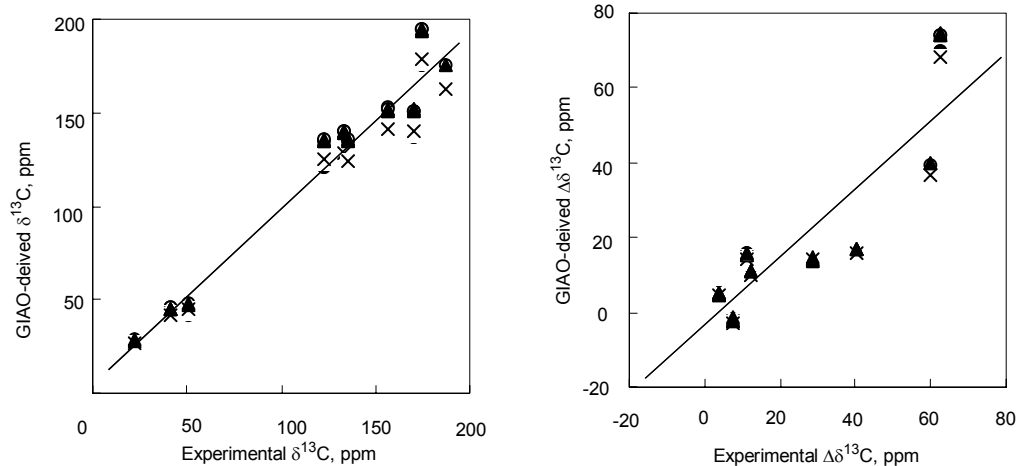


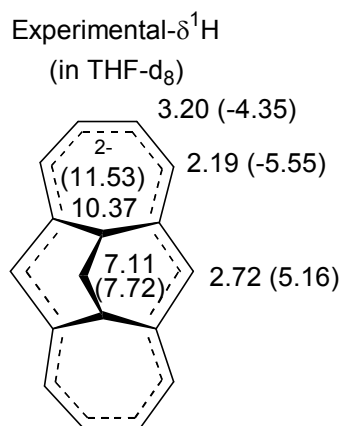
Fig S22 (continued).



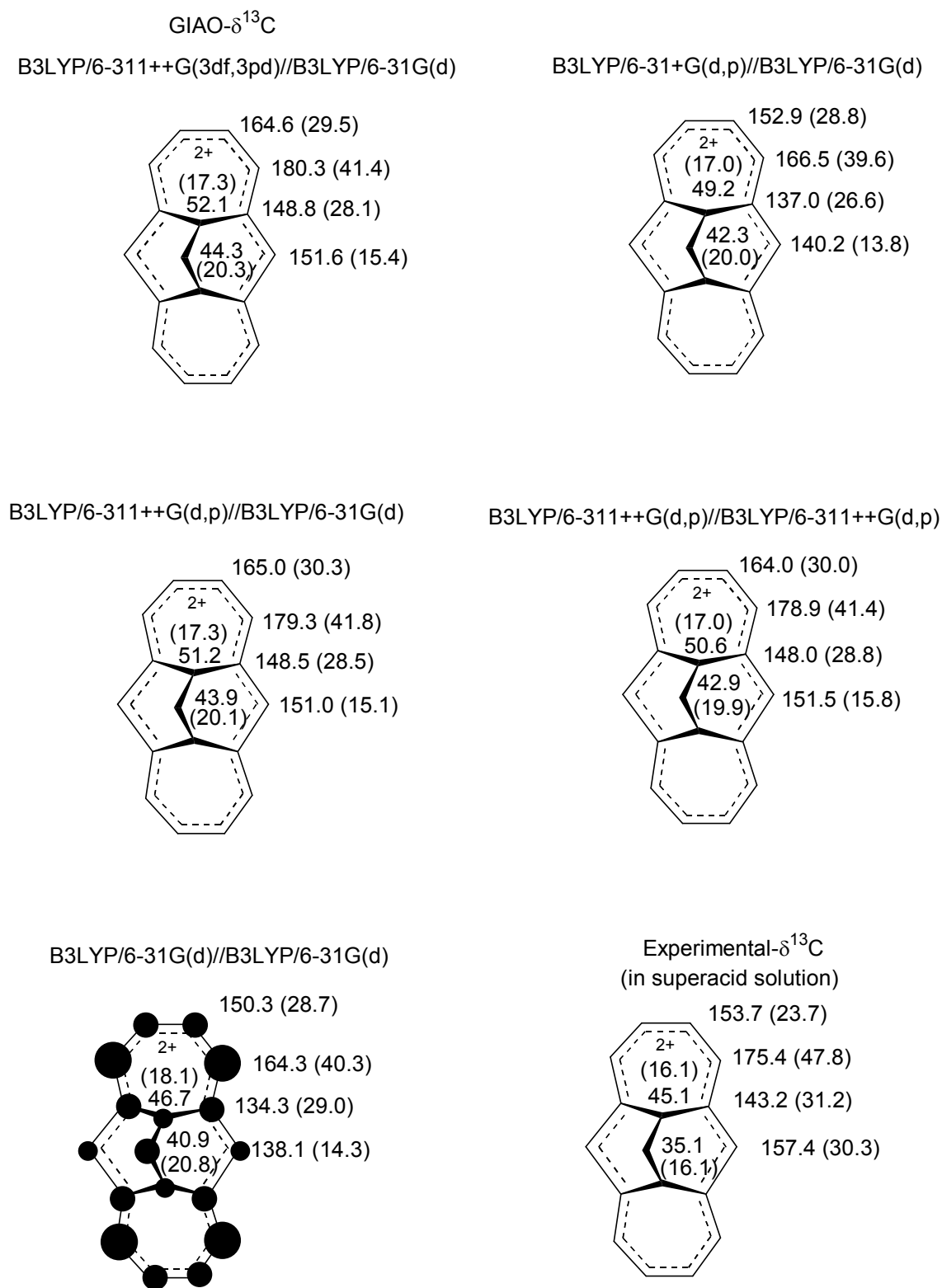
**Fig S22a.** Experimental (Ref 21) and computed  $^{13}\text{C}$  NMR chemical shifts for  $15\text{H}^+$ .



**Figure S22b.** Plots of experimental  $\delta^{13}\text{C}$  vs GIAO-derived  $\delta^{13}\text{C}$  and experimental  $\Delta\delta^{13}\text{C}$  vs GIAO-derived  $\Delta\delta^{13}\text{C}$  for  $15\text{H}^+$  by (  $\circ$  ) B3LYP/6-31G(d)//B3LYP/6-31G(d), (  $\bullet$  ) B3LYP/6-31+G(d,p)//B3LYP/6-31G(d), (  $\times$  ) B3LYP/6-311++G(d,p)//B3LYP/6-31G(d), (  $\Delta$  ) B3LYP/6-311++G(3df,3pd)//B3LYP/6-31G(d), (  $\blacktriangle$  ) B3LYP/6-311++G(d,p)//B3LYP/6-311++G(d,p).

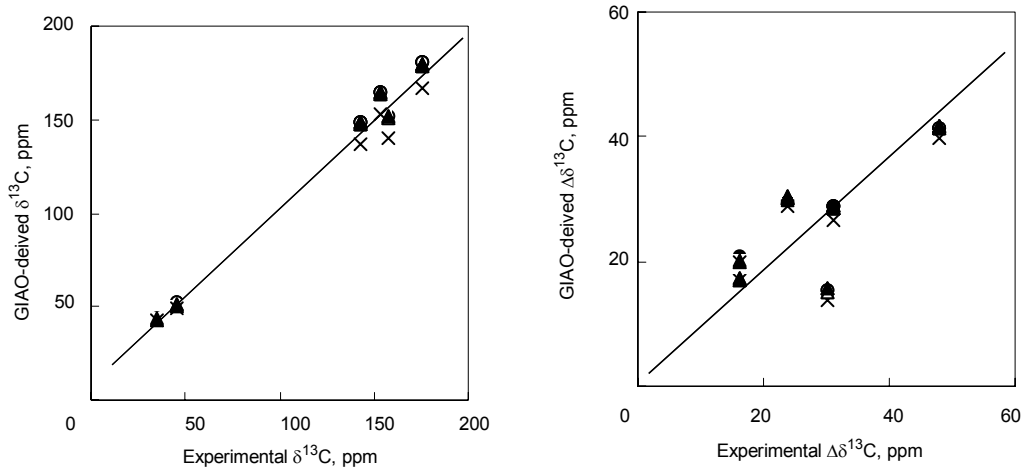


**Fig S22c.** Experimental  $^1\text{H}$  NMR chemical shifts for singlet  $15^{2-}$  (Ref. 21).

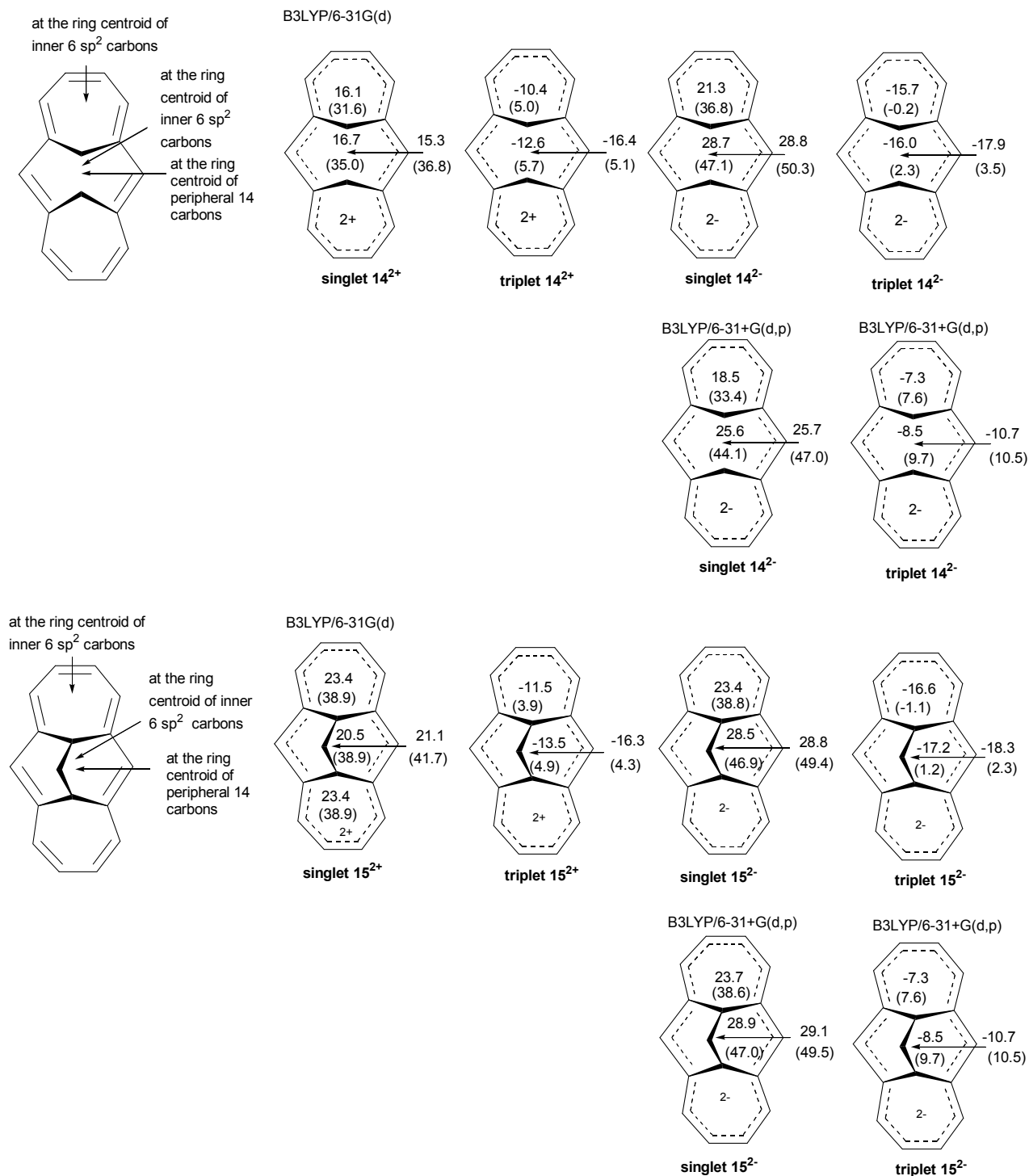


**Fig S22d.** Experimental (Ref 22) and computed  $^{13}\text{C}$  NMR chemical shifts for singlet  $15^{2+}$ .

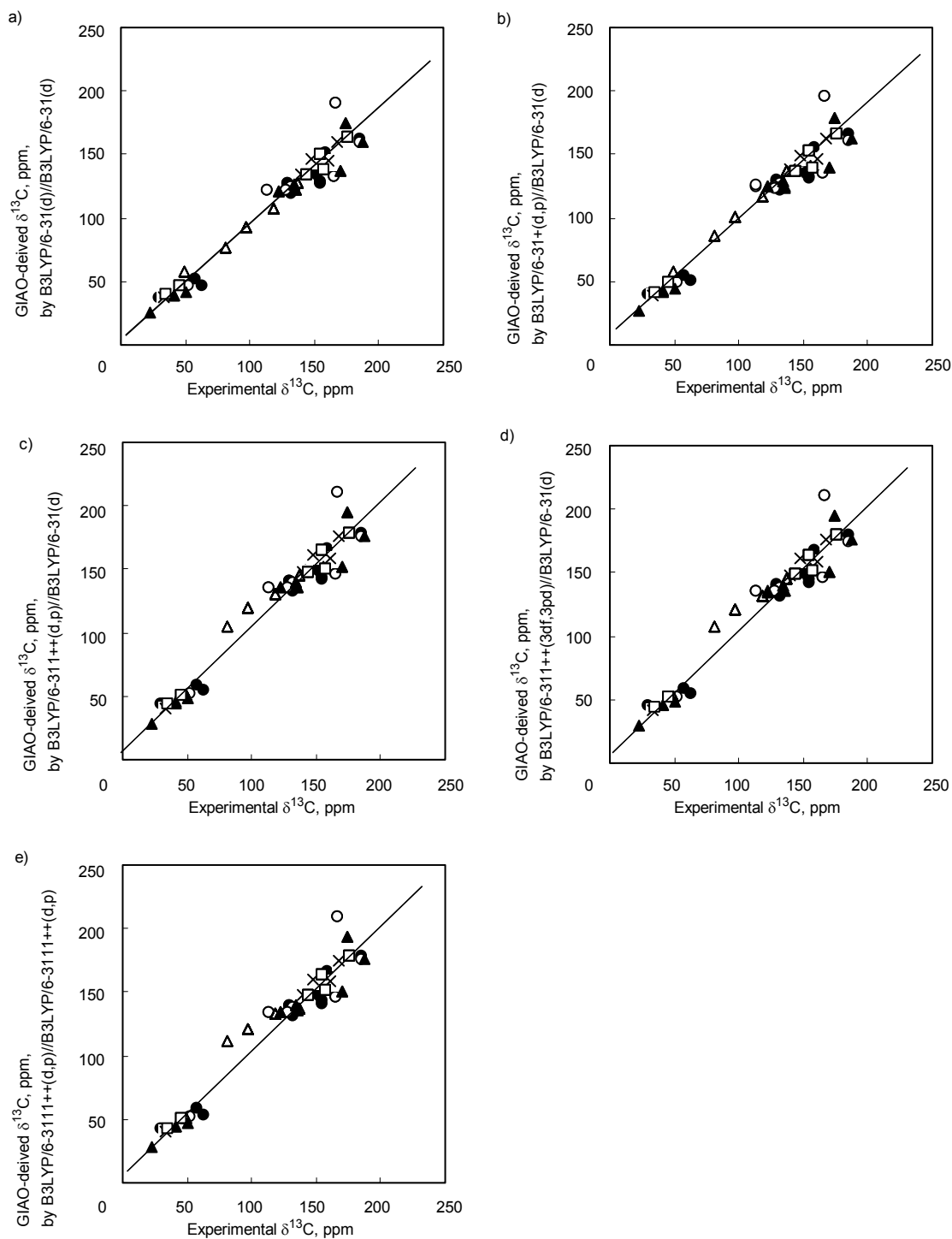




**Figure S22e.** Plots of experimental  $\delta^{13}\text{C}$  vs GIAO-derived  $\delta^{13}\text{C}$  and experimental  $\Delta\delta^{13}\text{C}$  vs GIAO-derived  $\Delta\delta^{13}\text{C}$  for  $15^{2+}$  by (  $\circ$  ) B3LYP/6-31G(d)//B3LYP/6-31G(d), (  $\bullet$  ) B3LYP/6-31+G(d,p)//B3LYP/6-31G(d), (  $\times$  ) B3LYP/6-311++G(d,p)//B3LYP/6-31G(d), (  $\Delta$  ) B3LYP/6-311++G(3df,3pd)//B3LYP/6-31G(d), (  $\blacktriangle$  ) B3LYP/6-311++G(d,p)//B3LYP/6-311++G(d,p).



**Fig S23.** NICS values for  $14^{2+}$ ,  $14^{2-}$ ,  $15^{2+}$ , and  $15^{2-}$  at B3LYP/6-31G(d) or 6-31+G(d,p) level ( $\Delta$ NICS values relative to those of parent hydrocarbons **14** and **15** in parentheses).



**Fig S24.** Plots of experimental  $\delta^{13}\text{C}$  vs GIAO-derived  $\delta^{13}\text{C}$  by (a) B3LYP/6-31G(d)//B3LYP/6-31G(d), (b) B3LYP/6-31+G(d,p)//B3LYP/6-31G(d), (c) B3LYP/6-311++G(d,p)//B3LYP/6-31G(d), (d) B3LYP/6-311++G(3df,3pd)//B3LYP/6-31G(d), (e) B3LYP/6-311++G(d,p)//B3LYP/6-311++G(d,p) ( $\circ$ :  $10\text{aH}^+$ ,  $\bullet$ :  $14\text{H}^+$ ,  $\times$ : singlet  $14^{2+}$ ,  $\Delta$ : singlet  $14^{-2}$ ,  $\blacktriangle$ :  $15\text{H}^+$ ,  $\square$ : singlet  $15^{2+}$ ).

### Addendum to Fig. S24

Computed  $\delta^{13}\text{C}$  values correlate with the experimental data by equations (1-5) (see below) with  $R^2 = 0.94\text{-}0.95$  and a slope of  $0.89\text{-}0.96$ . The slopes at higher basic sets are closer to unity, although correlation coefficients  $R^2$  are similar.

B3LYP/6-31(d)//B3LYP/6-31(d):

$$\delta^{13}\text{C (GIAO)} = 0.89 \times \delta^{13}\text{C (experimental)} + 6.4 \quad R^2 = 0.951 \quad (1)$$

B3LYP/6-31+G(d,p)//B3LYP/6-31G(d):

$$\delta^{13}\text{C (GIAO)} = 0.89 \times \delta^{13}\text{C (experimental)} + 10.2 \quad R^2 = 0.947 \quad (2)$$

B3LYP/6-311++G(d,p)//B3LYP/6-31G(d):

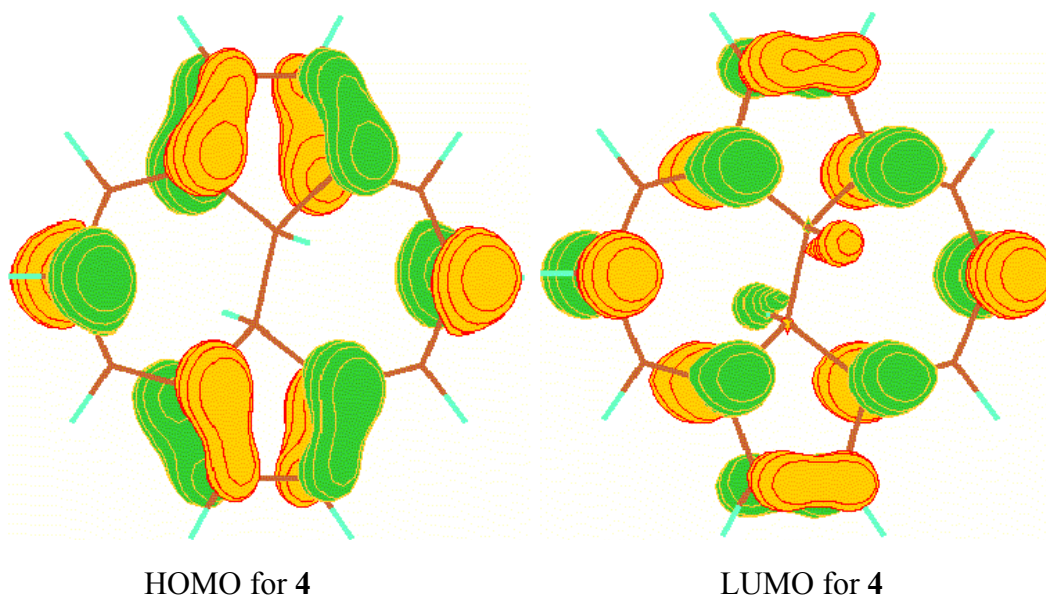
$$\delta^{13}\text{C (GIAO)} = 0.96 \times \delta^{13}\text{C (experimental)} + 11.3 \quad R^2 = 0.942 \quad (3)$$

B3LYP/6-311++G(3df,3pd)//B3LYP/6-31G(d):

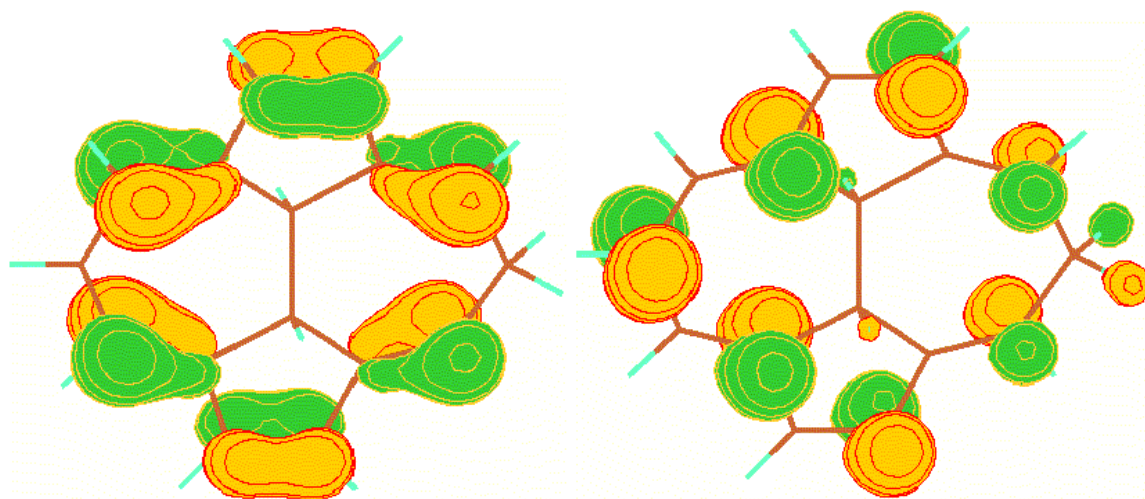
$$\delta^{13}\text{C (GIAO)} = 0.95 \times \delta^{13}\text{C (experimental)} + 12.3 \quad R^2 = 0.940 \quad (4)$$

B3LYP/6-311++G(d,p)//B3LYP/6-311++G(d,p):

$$\delta^{13}\text{C (GIAO)} = 0.96 \times \delta^{13}\text{C (experimental)} + 11.4 \quad R^2 = 0.938 \quad (5)$$

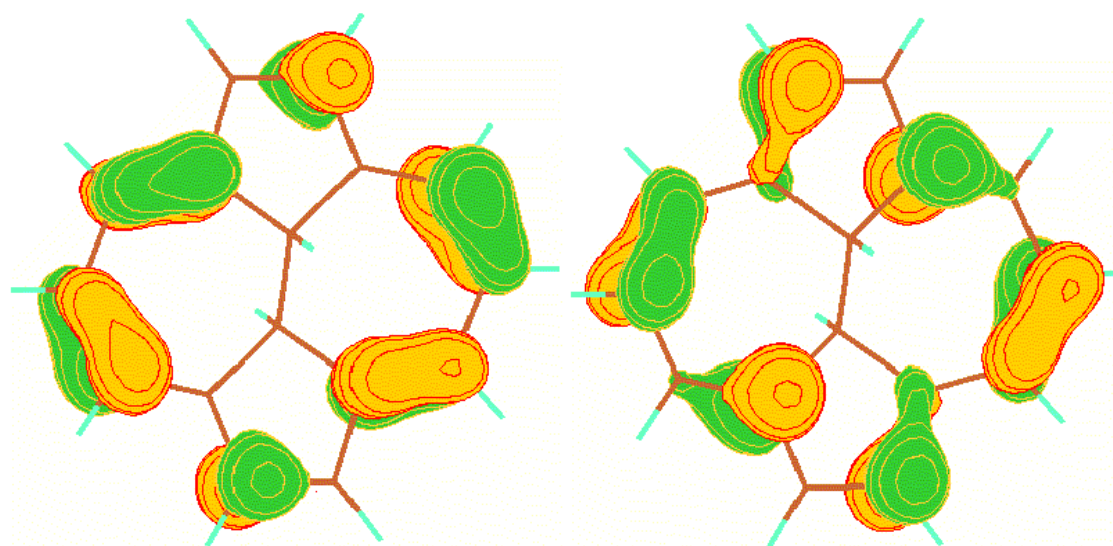


**Fig S25.** Forms of HOMO, LUMO, and SOMO for **4**, **4bH<sup>+</sup>**, triplet **4<sup>2+</sup>**, and singlet **4<sup>2-</sup>** by B3LYP/6-31G(d).



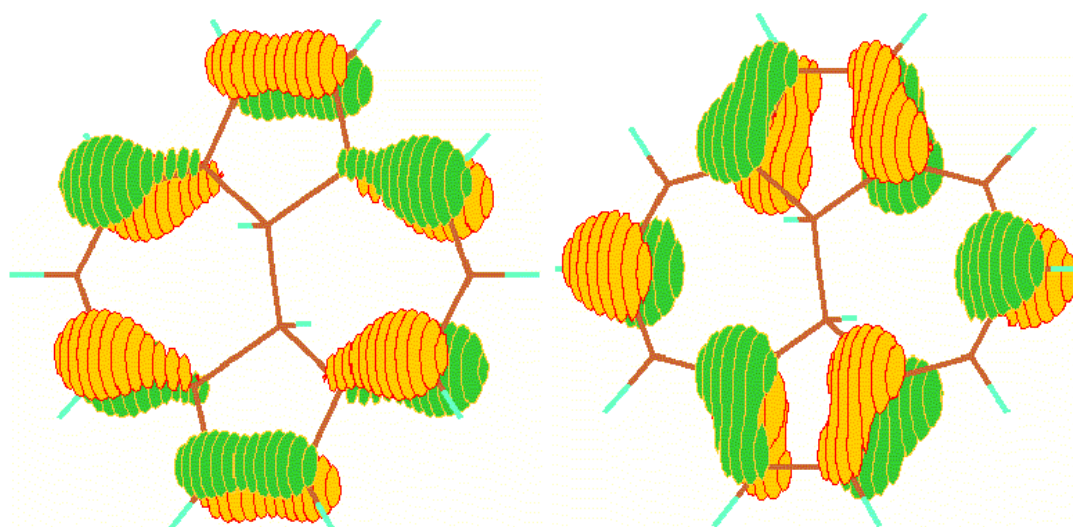
HOMO for 4bH<sup>+</sup>

LUMO for 4bH<sup>+</sup>



HOMO for singlet 4<sup>2+</sup>

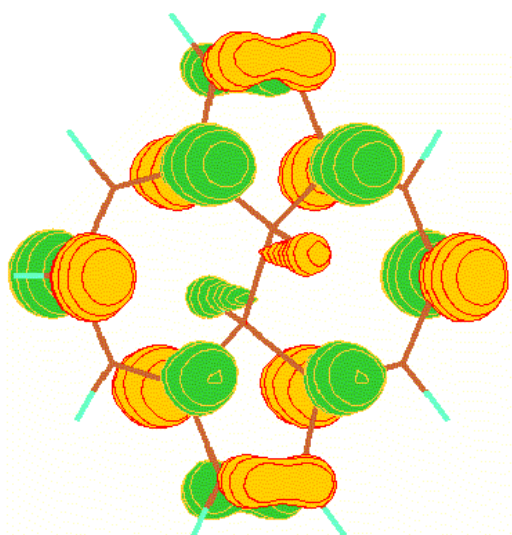
LUMO for singlet 4<sup>2+</sup>



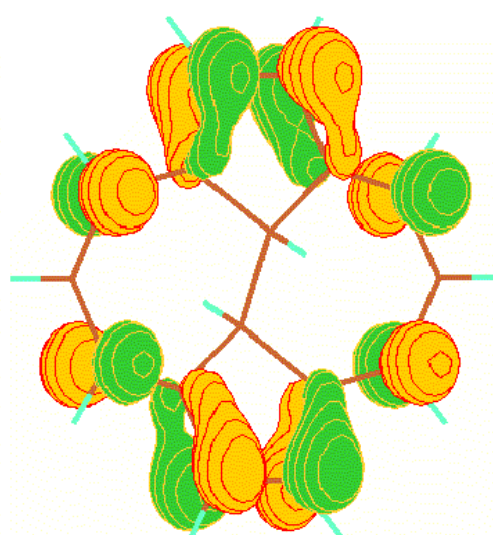
SOMO for triplet 4<sup>2+</sup>

SOMO for triplet 4<sup>2+</sup>

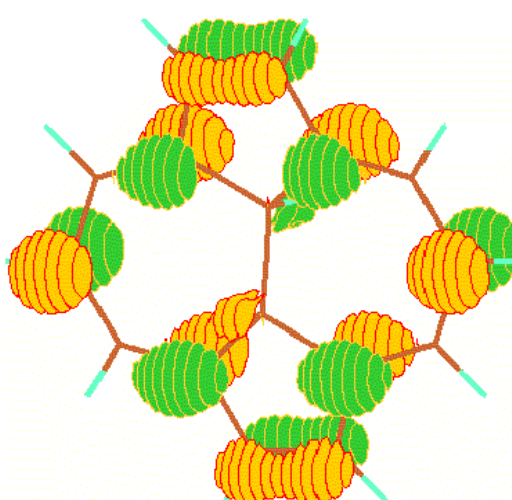
Figure S25 (continued).



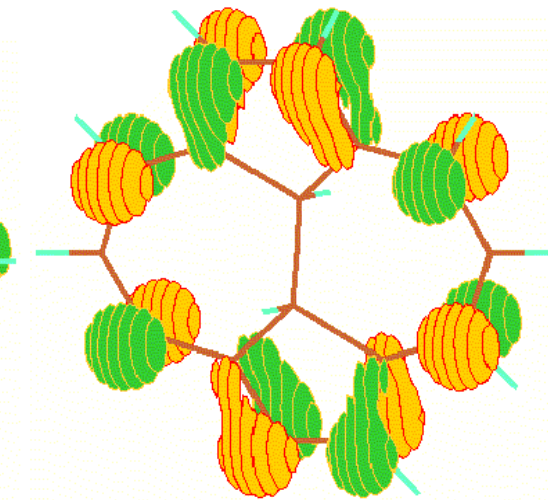
HOMO for singlet  $4^{2-}$



LUMO for singlet  $4^{2-}$



SOMO for triplet  $4^{2-}$



SOMO for triplet  $4^{2-}$

**Figure S25 (continued).**

**Table S1.** Energies (E), Zero Point Energies (ZPE), Gibbs Free Energies (G), and Relative Gibbs Free Energies ( $\Delta G$ ) for B3LYP/6-31G(d) Optimized Structures of Dihydro-derivatives **4-11**.

Compd	Molecular point group	E, hartree	ZPE, hartree	G, hartree	$\Delta G$ , kcal <sup>a</sup>
<b>4</b>	C <sub>1</sub>	-616.8707181	0.228341	-616.678445	(0)
<b>6</b>	C <sub>1</sub>	-616.8465659	0.228190	-616.654624	14.9
<b>5</b>	C <sub>1</sub>	-695.4919074	0.284632	-695.245566	(0)
<b>7</b>	C <sub>1</sub>	-695.4536401	0.284425	-695.207700	23.8
<b>8</b>	C <sub>1</sub>	-616.8234724	0.228569	-616.630880	(0)
<b>10</b>	C <sub>1</sub>	-616.8738954	0.228787	-616.681047	-31.5
<b>9</b>	C <sub>1</sub>	-695.4358718	0.283847	-695.190422	(0)
<b>11</b>	C <sub>1</sub>	-695.5004848	0.284624	-695.254083	-39.9

<sup>a</sup> Relative Gibbs free energies of *syn*-derivatives to those of *anti*-derivatives.

**Table S2.** Energies (E), Zero Point Energies (ZPE), Gibbs Free Energies (G), Relative Gibbs Free Energies ( $\Delta G$ ) for B3LYP/6-31G(d) Optimized Structures for **4-5**, **8-15**, their Monocations, Dications, and Dianions

Compd	Protonation site	Molecular point group	E, hartree <sup>a</sup>	ZPE, hartree <sup>a</sup>	G, hartree <sup>a</sup>	$\Delta G$ , kcal <sup>b</sup>
<b>4</b>		C <sub>1</sub>	-616.8707181	0.228341	-616.678445	(0)
		C <sub>2h</sub>	-616.8706817	0.228333	-616.677761	0.4
<b>4H<sup>+</sup></b>	C(1)	C <sub>1</sub>	-617.2333368	0.240699	-617.029301	-220.2
<b>4aH<sup>+</sup></b>	C(3)	C <sub>1</sub>	-617.234331	0.240751	-617.030422	-220.9
<b>4bH<sup>+</sup></b>	C(4)	C <sub>1</sub>	-617.2384905	0.240730	-617.033830	-223.4
<b>4<sup>2+</sup> (s)<sup>c</sup></b>		C <sub>1</sub>	-616.2230611	0.228640	-616.031314	406.1
		C <sub>i</sub>	-616.2230366	0.228626	-616.031306	406.1
		C <sub>2h</sub> <sup>d</sup>	-616.2214584	0.226870	-616.030650	406.5
<b>4<sup>2+</sup> (t)<sup>c</sup></b>		C <sub>1</sub>	-616.2244602	0.228783	-616.033230	404.9
		C <sub>s</sub>	-616.2244602	0.228777	-616.033237	
		C <sub>2h</sub>	-616.2244166	0.228773	-616.032542	405.3
<b>4<sup>2-</sup> (s)<sup>c</sup></b>		C <sub>1</sub>	-616.7220175	0.218179	-616.540687	86.4
		C <sub>2h</sub>	-616.7219216	0.218157	-616.539974	86.9
<b>4<sup>2-</sup> (t)<sup>c</sup></b>		C <sub>1</sub>	-616.7181169	0.217782	-616.538074	88.1
		C <sub>2h</sub>	-616.7180712	0.217772	-616.537382	88.5
<b>5</b>		C <sub>1</sub>	-695.4919074	0.284632	-695.245566	(0)
		C <sub>2h</sub>	-695.4918974	0.284623	-695.244913	0.4
<b>5H<sup>+</sup></b>	C(1)	C <sub>1</sub>	-695.8578678	0.296845	-695.599860	-222.3
<b>5aH<sup>+</sup></b>	C(3)	C <sub>1</sub>	-695.8578801	0.296912	-695.600027	-222.4
<b>5bH<sup>+</sup></b>	C(4)	C <sub>1</sub>	-695.8615373	0.297000	-695.603381	-224.5
<b>5<sup>2+</sup> (s)<sup>c</sup></b>		C <sub>i</sub>	-694.8542078	0.284409	-694.608866	399.5
		C <sub>2h</sub> <sup>d</sup>	-694.8534797	0.282958	-694.608601	399.7
		C <sub>1</sub>	-694.8542025	0.284498	-694.608694	399.6
<b>5<sup>2+</sup> (t)<sup>c</sup></b>		C <sub>2h</sub>	-694.8545546	0.284792	-694.608847	399.5
		C <sub>1</sub>	-694.8545828	0.284811	-694.609502	399.1
<b>5<sup>2-</sup> (s)<sup>c</sup></b>		C <sub>2h</sub>	-695.3532913	0.273662	-695.118077	80.0
		C <sub>1</sub>	-695.3533188	0.273624	-695.118835	79.5
<b>5<sup>2-</sup> (t)<sup>c</sup></b>		C <sub>2h</sub>	-695.3467395	0.273701	-695.112390	83.6
		C <sub>1</sub>	-695.3467665	0.273722	-695.113043	83.2

<sup>a</sup> 1 hartree = 627.5096 kcal/mol. <sup>b</sup> Relative Gibbs free energies to those of the parent hydrocarbons. <sup>c</sup> (s) and (t) denote triplet and singlet states. <sup>d</sup> Number of imaginary frequencies = 1.



**Table S2 (continued).**

Compd	Protonation site	Molecular point group	E, hartree <sup>a</sup>	ZPE, hartree <sup>a</sup>	G, hartree <sup>a</sup>	$\Delta G$ , kcal <sup>b</sup>
<b>8</b>		C <sub>2h</sub>	-616.8234939	0.228588	-616.630225	0.4
		C <sub>1</sub>	-616.8234724	0.228569	-616.630880	(0)
<b>8H<sup>+</sup></b>	C(5)	C <sub>1</sub>	-617.1897954	0.240435	-616.986002	-222.8
<b>8aH<sup>+</sup></b>	C(1)	C <sub>1</sub>	-617.1836447	0.240668	-616.979726	-218.9
<b>8bH<sup>+</sup></b>	C(2)	C <sub>1</sub>	-617.1886921	0.240909	-616.984366	-221.8
<b>8<sup>2+</sup> (s)<sup>c</sup></b>		C <sub>2h</sub>	-616.1844352	0.229021	-615.991327	401.3
		C <sub>1</sub>	-616.1844054	0.228991	-615.991986	400.9
<b>8<sup>2+</sup> (t)<sup>c</sup></b>		C <sub>2h</sub>	-616.1731913	0.228392	-615.981566	407.5
		C <sub>1</sub>	-616.1731519	0.228345	-615.982242	407.0
<b>8<sup>2-</sup> (s)<sup>c</sup></b>		C <sub>2h</sub>	-616.7091687	0.218790	-616.526239	65.7
		C <sub>1</sub>	-616.7091512	0.218781	-616.526889	65.3
<b>8<sup>2-</sup> (t)<sup>c</sup></b>		C <sub>2h</sub>	-616.6988577	0.218887	-616.516534	71.8
		C <sub>1</sub>	-616.6988251	0.218860	-616.517189	71.3
<b>9</b>		C <sub>2h</sub>	-695.4358090	0.283771	-695.189819	0.4
		C <sub>1</sub>	-695.4358718	0.283847	-695.190422	(0)
<b>9H<sup>+</sup></b>	C(5)	C <sub>1</sub>	-695.8115965	0.296398	-695.554000	-228.1
<b>9aH<sup>+</sup></b>	C(1)	C <sub>1</sub>	-695.7989077	0.295817	-695.542439	-220.9
<b>9bH<sup>+</sup></b>	C(2)	C <sub>1</sub>	-695.8035653	0.295982	-695.546766	-223.6
<b>9<sup>2+</sup> (s)<sup>c</sup></b>		C <sub>2h</sub>	-694.8082124	0.284235	-694.562690	393.9
		C <sub>1</sub>	-694.8081658	0.284146	-694.563478	393.4
<b>9<sup>2+</sup> (t)<sup>c</sup></b>		C <sub>2h</sub>	-694.7976001	0.283432	-694.553371	399.8
		C <sub>1</sub>	-694.7975569	0.283357	-694.554107	399.3
<b>9<sup>2-</sup> (s)<sup>c</sup></b>		C <sub>2h</sub> <sup>b</sup>	-695.3236722	0.273068	-695.088935	63.7
		C <sub>1</sub>	-695.3238175	0.273755	-695.089723	63.2
<b>9<sup>2-</sup> (t)<sup>c</sup></b>		C <sub>2h</sub>	-695.3166255	0.274226	-695.081418	68.4
		C <sub>1</sub>	-695.3166141	0.274213	-695.082080	68.0

**Table S2 (continued).**

Compd	Protonation site	Molecular point group	E, hartree <sup>a</sup>	ZPE, hartree <sup>a</sup>	G, hartree <sup>a</sup>	$\Delta G$ , kcal <sup>b</sup>
<b>10</b>		C <sub>2v</sub>	-616.8738954	0.228787	-616.680393	0.4
		C <sub>1</sub>	-616.8738954	0.228787	-616.681047	(0)
<b>10H<sup>+</sup></b>	C(5)	C <sub>s</sub>	-617.2300612	0.240586	-617.025966	-216.4
		C <sub>1</sub>	-617.2300612	0.240586	-617.025967	-216.4
<b>10aH<sup>+</sup></b>	C(1)	C <sub>1</sub>	-617.2391727	0.240969	-617.034712	-221.9
<b>10bH<sup>+</sup></b>	C(2)	C <sub>1</sub>	-617.235936	0.240777	-617.031813	-220.1
<b>10<sup>2+</sup> (s)<sup>c</sup></b>		C <sub>2v</sub>	-616.2188396	0.228383	-616.026611	410.6
		C <sub>1</sub>	-616.2188396	0.228383	-616.027265	410.3
<b>10<sup>2+</sup> (t)<sup>c</sup></b>		C <sub>2v</sub>	-616.2169067	0.228321	-616.025267	411.5
		C <sub>1</sub>	-616.2169067	0.228321	-616.025921	411.1
<b>10<sup>2-</sup> (s)<sup>c</sup></b>		C <sub>s</sub>	-616.7470000	0.218961	-616.564409	73.2
		C <sub>1</sub>	-616.7469915	0.218991	-616.564379	73.2
<b>10<sup>2-</sup> (t)<sup>c</sup></b>		C <sub>s</sub>	-616.7283051	0.218584	-616.547035	84.1
		C <sub>1</sub>	-616.7282851	0.218561	-616.547050	84.1
<b>11</b>		C <sub>2v</sub>	-695.5004848	0.284624	-695.253430	0.4
		C <sub>1</sub>	-695.5004848	0.284624	-695.254083	(0)
<b>11H<sup>+</sup></b>	C(5)	C <sub>s</sub>	-695.8573173	0.296495	-695.599584	-216.8
		C <sub>1</sub>	-695.8573173	0.296495	-695.599584	-216.8
<b>11aH<sup>+</sup></b>	C(1)	C <sub>1</sub>	-695.8675051	0.297025	-695.609250	-222.9
<b>11bH<sup>+</sup></b>	C(2)	C <sub>1</sub>	-695.8636513	0.296574	-695.606023	-220.8
<b>11<sup>2+</sup> (s)<sup>c</sup></b>		C <sub>2v</sub>	-694.8508447	0.284850	-694.604621	407.5
		C <sub>1</sub>	-694.8508447	0.284850	-694.605275	407.1
<b>11<sup>2+</sup> (t)<sup>c</sup></b>		C <sub>2v</sub>	-694.8487787	0.284259	-694.603589	408.2
		C <sub>1</sub>	-694.8487787	0.284260	-694.604243	407.8
<b>11<sup>2-</sup> (s)<sup>c</sup></b>		C <sub>2v</sub>	-695.3740715	0.274706	-695.137263	73.3
		C <sub>1</sub>	-695.3740715	0.274707	-695.137917	72.9
<b>11<sup>2-</sup> (t)<sup>c</sup></b>		C <sub>2v</sub>	-695.355764	0.274343	-695.120310	83.9
		C <sub>1</sub>	-695.355764	0.274344	-695.120963	83.5

**Table S2 (continued).**

Compd	Protonation site	Molecular point group	E, hartree <sup>a</sup>	ZPE, hartree <sup>a</sup>	G, hartree <sup>a</sup>	$\Delta G$ , kcal <sup>b</sup>
<b>14</b>		C <sub>2v</sub>	-618.0403323	0.252018	-617.823929	0.4
		C <sub>1</sub>	-618.0403323	0.252019	-617.824583	(0)
<b>14H<sup>+</sup></b>	C(5)	C <sub>s</sub>	-618.413039	0.264786	-618.184928	-226.1
		C <sub>1</sub>	-618.413039	0.264786	-618.184928	-226.1
<b>14aH<sup>+</sup></b>	C(1)	C <sub>1</sub>	-618.4120941	0.264625	-618.184096	-225.6
<b>14<sup>2+</sup> (s)<sup>c</sup></b>		C <sub>2v</sub>	-617.3963755	0.252767	-617.179355	404.9
		C <sub>1</sub>	-617.3963755	0.252767	-617.180009	404.5
<b>14<sup>2+</sup> (t)<sup>c</sup></b>		C <sub>2v</sub>	-617.379102	0.251302	-617.164858	414.0
		C <sub>1</sub>	-617.379102	0.251302	-617.165512	413.6
<b>14<sup>2-</sup> (s)<sup>c</sup></b>		C <sub>2v</sub>	-617.904338	0.242016	-617.698569	79.1
		C <sub>1</sub>	-617.904338	0.242016	-617.699222	78.7
<b>14<sup>2-</sup> (t)<sup>c</sup></b>		C <sub>2v</sub>	-617.8882346	0.241606	-617.683725	88.4
		C <sub>1</sub>	-617.8882346	0.241607	-617.684378	88.0
<b>15</b>		C <sub>2v</sub>	-656.1807873	0.258796	-655.957611	0.4
		C <sub>1</sub>	-656.1807873	0.258797	-655.958264	(0)
<b>15H<sup>+</sup></b>	C(5)	C <sub>s</sub>	-656.5443439	0.271000	-656.310041	-220.7
		C <sub>1</sub>	-656.5443439	0.271001	-656.310041	-220.7
<b>15aH<sup>+</sup></b>	C(1)	C <sub>1</sub>	-656.5503302	0.271170	-656.315906	-224.4
<b>15<sup>2+</sup> (s)<sup>c</sup></b>		C <sub>2v</sub>	-655.5347736	0.259031	-655.311582	405.8
		C <sub>1</sub>	-655.5347736	0.259031	-655.312236	405.4
<b>15<sup>2+</sup> (t)<sup>c</sup></b>		C <sub>2v</sub>	-655.5213722	0.258087	-655.300391	412.8
		C <sub>1</sub>	-655.5213722	0.258088	-655.301044	412.4
<b>15<sup>2-</sup> (s)<sup>c</sup></b>		C <sub>2v</sub>	-656.0498283	0.248739	-655.837188	76.0
		C <sub>1</sub>	-656.0498283	0.248740	-655.837842	75.6
<b>15<sup>2-</sup> (t)<sup>c</sup></b>		C <sub>2v</sub>	-656.0312338	0.248407	-655.819851	86.9
		C <sub>1</sub>	-656.0312338	0.248408	-655.820504	86.4

**Table S3.** Energies (E), Zero Point Energies (ZPE), Gibbs Free Energies (G), Relative Gibbs Free Energies ( $\Delta G$ ) for Optimized Structures for **4**, its Monocations, Dications, and Dianions

Compd	Protonation site		E, hartree	ZPE, hartree	G, hartree	$\Delta G$ , kcal <sup>a</sup>
	basic set	Molecular point group				
<b>4</b>						
B3LYP/6-31G(d)	C <sub>1</sub>	-616.8707181	0.228341	-616.678445	(0)	
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.9102648	0.227337	-616.719035	(0)	
B3LYP/6-31++G(d,p)	C <sub>1</sub>	-616.910426	0.227363	-616.719166	(0)	
<b>4H<sup>+</sup></b>	C(1)					
B3LYP/6-31G(d)	C <sub>1</sub>	-617.2333368	0.240699	-617.029301	-220.2	
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-617.2653618	0.239618	-617.062429	-215.5	
B3LYP/6-31++G(d,p)	C <sub>1</sub>	-617.2655807	0.239627	-617.062637	-215.5	
<b>4aH<sup>+</sup></b>	C(3)					
B3LYP/6-31G(d)	C <sub>1</sub>	-617.234331	0.240751	-617.030422	-220.9	
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-617.266431	0.239662	-617.063632	-216.2	
B3LYP/6-31++G(d,p)	C <sub>1</sub>	-617.2667051	0.239676	-617.063890	-216.3	
<b>4bH<sup>+</sup></b>	C(4)					
B3LYP/6-31G(d)	C <sub>1</sub>	-617.2384905	0.240730	-617.033830	-223.4	
B3LYP/6-31+G(d,p)	C <sub>2</sub>	-617.2705073	0.239629	-617.066981	-218.3	
B3LYP/6-31++G(d,p)	C <sub>2</sub>	-617.2707331	0.239645	-617.067190	-218.4	
<b>4<sup>2+</sup> (s)</b>						
B3LYP/6-31G(d)	C <sub>1</sub>	-616.2230611	0.228640	-616.031314	406.1	
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.2489933	0.227759	-616.058145	414.7	
B3LYP/6-31++G(d,p)	C <sub>1</sub>	-616.2492048	0.227777	-616.058331	414.7	
<b>4<sup>2+</sup> (t)</b>						
B3LYP/6-31G(d)	C <sub>1</sub>	-616.2244602	0.228783	-616.033230	404.9	
B3LYP/6-31+G(d,p)	C <sub>s</sub>	-616.2502732	0.227906	-616.059936	413.6	
B3LYP/6-31++G(d,p)	C <sub>s</sub>	-616.2505412	0.227939	-616.060160	413.5	
<b>4<sup>2-</sup> (s)</b>						
B3LYP/6-31G(d)	C <sub>1</sub>	-616.7220175	0.218179	-616.540687	86.4	
B3LYP/6-31+G(d,p)	C <sub>i</sub>	-616.81053	0.218047	-616.629518	56.2	
B3LYP/6-31++G(d,p)	C <sub>i</sub>	-616.8154522	0.220108	-616.632721	54.2	
<b>4<sup>2-</sup> (t)</b>						
B3LYP/6-31G(d)	C <sub>1</sub>	-616.7181169	0.217782	-616.538074	88.1	
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.8027096	0.216199	-616.625230	58.9	
B3LYP/6-31++G(d,p)	C <sub>1</sub>	-616.8257125	0.221942 <sup>b</sup>	-616.640690	49.0	

<sup>a</sup> Relative energies to the neutral. <sup>b</sup> Number of imaginary frequency is 1.

**Table S4.** Energies for Optimized Structures for **5**, **8**, **9**, **10**, **11**, **14**, **15**, and their dianions by B3LYP/6-31+G(d,p)

Compd baisc set	Molecular point group	E, hartree	ZPE, hartree	G, hartree	$\Delta G$ , kcal <sup>a</sup>
<b>5</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.4919074	0.284632	-695.245566	(0)
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.5373455	0.283160	-695.292516	(0)
<b>5<sup>2-</sup>(s)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.3533188	0.273624	-695.118835	79.5
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.4453828	0.273100	-695.211455	50.9
<b>5<sup>2-</sup>(t)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.3467665	0.273722	-695.113043	83.2
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.4356826	0.272290	-695.203664	55.8
<b>8</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-616.8234724	0.228569	-616.630880	(0)
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.8639904	0.227603	-616.672389	(0)
<b>8<sup>2-</sup>(s)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-616.7091512	0.218781	-616.526889	65.3
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.7911433	0.216601	-616.612105	37.8
<b>8<sup>2-</sup>(t)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-616.6988251	0.218860	-616.517189	71.3
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.783247	0.217881	-616.602835	43.6
<b>9</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.4358718	0.283847	-695.190422	(0)
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.4821086	0.282391	-695.238197	(0)
<b>9<sup>2-</sup>(s)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.3238175	0.273755	-695.089723	63.2
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.4115526	0.272225	-695.178851	37.2
<b>9<sup>2-</sup>(t)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.3166141	0.274213	-695.082080	68.0
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.4057966	0.272888	-695.172796	41.0

<sup>b</sup> Relative Gibbs free energies to those of the parent hydrocarbons. <sup>b</sup> Number of imaginary frequencies = 1.

**Table S4 (continued).**

Compd baisc set	Molecular point group	E, hartree	ZPE, hartree	G, hartree	$\Delta G$ , kcal <sup>a</sup>
<b>10</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-616.8738954	0.228787	-616.681047	(0)
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.913608	0.227808	-616.721778	(0)
<b>10<sup>2-</sup>(s)</b>					
B3LYP/6-31G(d)	C <sub>s</sub>	-616.7470000	0.218961	-616.564409	73.2
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.8305483	0.217839	-616.649253	45.5
<b>10<sup>2-</sup>(t)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-616.7282851	0.218561	-616.547050	84.1
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-616.8136409	0.217338	-616.634063	55.0
<b>11</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.5004848	0.284624	-695.254083	(0)
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.5468117	0.283283	-695.301795	(0)
<b>11<sup>2-</sup>(s)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.3740715	0.274707	-695.137917	72.9
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.4634247	0.273319	-695.228806	45.8
<b>11<sup>2-</sup>(t)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-695.355764	0.274344	-695.120963	83.5
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-695.4472602	0.272326 <sup>b</sup>	-695.214100	55.0
<b>14</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-618.0403323	0.252019	-617.824583	(0)
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-618.0831353	0.250841	-617.868622	(0)
<b>14<sup>2-</sup>(s)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-617.904338	0.242016	-617.699222	78.7
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-617.9946934	0.240771	-617.791012	48.7
<b>14<sup>2-</sup>(t)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-617.8882346	0.241607	-617.684378	88.0
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-617.9804216	0.240607	-617.777831	57.0

**Table S4 (continued).**

Compd baisc set	Molecular point group	E, hartree	ZPE, hartree	G, hartree	$\Delta G$ , kcal <sup>a</sup>
<b>15</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-656.1807873	0.258797	-655.958264	(0)
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-656.223446	0.257654	-656.002099	(0)
<b>15<sup>2-</sup>(s)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-656.0498283	0.248740	-655.837842	75.6
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-656.13706	0.247232	-655.926797	47.3
<b>15<sup>2-</sup>(t)</b>					
B3LYP/6-31G(d)	C <sub>1</sub>	-656.0312338	0.248408	-655.820504	86.4
B3LYP/6-31+G(d,p)	C <sub>1</sub>	-656.120833	0.247308	-655.911436	56.9