

# Aromatic Annulation Strategy for Naphthalenes Fused at 1,2- and 3,4-Positions with Two Heterocycles

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## Supplementary Data

### General experimental procedures

Melting points are uncorrected. <sup>1</sup>H-NMR were recorded on a Bruker DPX 300 at 300MHz with CDCl<sub>3</sub> as solvent. Chemical shifts are reported in δ (ppm). *J* values are given in Hz. Infrared spectra were taken with a Shimadzu IR 440 spectrometer in KBr pellets. Mass spectra were recorded with a VG ZAB-HS spectrometer. Elemental analyses were done on a Perkin-Elmer-240C analyzer. UV spectra were recorded with a Shimadzu UV-2401PC UV-VIS recording spectrophotometer. Fluorescence spectra were done on a SLM Aminco Bowman Series 2 Luminescence Spectrometer. Benzene (A R grade) was dried with sodium and distilled before use.

### Preparative photolysis for *N*-methyl-3, 4-dichloromaleimide (1) and 3, 4-dichlorocoumarin (2)

The light source was a medium-pressure mercury lamp (500W) in a glass cooling water jacket. The solution of *N*-methyl-3, 4-dichloromaleimide (**1**) or 3, 4-dichlorocoumarin (**2**) (0.05M) and an excess amount of phenylbenzofuran and phenylbenzothiophene (**3a-3d**) in benzene was purged with dry nitrogen for 30 min and then irradiated with light of λ > 300nm under continuous nitrogen purging. The reaction course was monitored by TLC. After the photolysis, the solvent was removed *in vacuo* and the residue was separated by flash chromatography on a silica gel column with petroleum ether (b. p. 60-90°C)-ethyl acetate as eluents.

**Photolysis of 1 with 2-phenylbenzofuran (3a):** A solution of **1** (540 mg, 3.0 mmol) and **3a** (756 mg, 3.9 mmol) in benzene (60 ml) was photolyzed for 21h to reach a complete conversion of **1**. Work-up as described above gave **4a** (73.4%) and **5a** (24.2%).

Heating **4a** (200mg) on silica gel (2g) at 100°C for 12h, a 98% conversion of **4a** was reached. Separation of the reaction mixture by flash chromatography on a silica gel gave **5a** (1.2%) and **6a** (93.9%). Irradiating a solution of **6a** (50mg) in acetone (20ml) with light of λ > 400nm under continuous nitrogen purging for 10h resulted in a complete conversion of **6a** and gave **5a** (96.1%).

If the mixture of **5a** and **6a** obtained by heating **4a** on silica was not separated but was dissolved in acetone and irradiated with light of λ > 400nm under continuous nitrogen purging for 10h to reach a complete conversion of **4a**, this gave **5a** (92.2%). The total yield of **5a** in photoreaction and in subsequent conversion of **4a** was 94.2%

#### **4a**

MP 160-161°C

<sup>1</sup>H-NMR (300 MHz; CDCl<sub>3</sub>) δ 2.83(3H, s, NCH<sub>3</sub>), 4.97(1H, s), 7.05(1H, d, *J* = 8.2), 7.11(1H, dd, *J*<sub>1</sub>=7.5, *J*<sub>2</sub>=0.8), 7.35(1H, d, *J*=7.9), 7.38-7.46(6H, m)

IR (KBr) ν 3066, 2963, 1795, 1726, 1594, 1479, 1457, 1427, 1370, 1279, 1237, 1007, 971, 779, 742, 699cm<sup>-1</sup>

MS *m/z* (%) 337(M<sup>+</sup>-HCl, 35.31), 302(M<sup>+</sup>-2Cl, 9.08), 245(33.84), 217(20.82), 194(Base), 189(27.67), 165(16.12), 151(4.06), 44(5.41)

EA Found: C, 60.81; H, 3.34; N, 3.56. C<sub>19</sub>H<sub>13</sub>Cl<sub>2</sub>N<sub>1</sub>O<sub>3</sub> requires C, 60.98; H, 3.50; N 3.74.

#### **5a**

MP 244-245°C

<sup>1</sup>H-NMR (300 MHz; CDCl<sub>3</sub>) δ 3.31(3H, s, NCH<sub>3</sub>), 7.53(1H, t, *J*=7.5), 7.62(1H, td, *J*<sub>1</sub>=7.8, *J*<sub>2</sub>=1.2), 7.73-8.82(3H, m), 8.51(1H, m), 8.80(1H, d, *J*=8.1), 9.09(1H, m)

IR (KBr)  $\nu$  3071, 2924, 1756, 1699, 1600, 1572, 1458, 1373, 1350, 1286, 1191, 1102, 990, 753 $\text{cm}^{-1}$   
MS  $m/z$  (%) 301(M, base), 257(13.09), 216(37.48), 187(16.16), 151(4.16), 108(5.13)  
EA Found: C, 75.84; H, 3.41; N, 4.68.  $\text{C}_{19}\text{H}_{11}\text{N}_1\text{O}_3$  requires C, 75.74; H, 3.68; N, 4.65.  
UV (nm) (log  $\epsilon$ ): 397(3.85), 292(4.50)

#### **6a**

MP 164-165 $^{\circ}\text{C}$

$^1\text{H-NMR}$  (300 MHz;  $\text{CDCl}_3$ )  $\delta$  3.20(3H, s,  $\text{NCH}_3$ ), 7.32(1H, d,  $J=7.3$ ), 7.39(1H, d,  $J=7.7$ ), 7.44-7.48(3H, m), 7.53(1H, d,  $J=7.7$ ), 7.61(1H, d,  $J=7.8$ ), 7.68-7.70(2H, m)

IR (KBr)  $\nu$  1781, 1720, 1650, 1442, 1386, 1224, 1141, 988, 830, 775, 688 $\text{cm}^{-1}$

MS  $m/z$  (%) 339( $\text{M}^++2$ , 26.46), 337( $\text{M}^+$ , 77.23), 302( $\text{M}^+-\text{Cl}$ , 14.71), 301( $\text{M}^+-\text{HCl}$ , 11.26), 245(Base), 217(53.64), 189(47.52), 163 (7.62), 95(14.87), 77(7.55)

EA Found: C, 67.57; H, 3.41; N, 4.08.  $\text{C}_{19}\text{H}_{12}\text{N}_1\text{O}_3\text{Cl}$  requires C, 67.66; H, 3.56; N, 4.15.

**Photolysis of 1 with 3-phenylbenzofuran (3b):** A solution of **1** (540 mg, 3 mmol) and **3b** (756 g, 3.9 mmol) in benzene (60 ml) was photolyzed for 48h to reach a 63.3% conversion of **1**. Work-up as described above gave **4b** (86.8%) and **5b** (5.84%).

Heating **4b** (200mg) on silica gel (2g) at 90 $^{\circ}\text{C}$  for 12h, a complete conversion was reached and this gave the only product **6b** (95.9%). Irradiating a solution of **6b** (100mg) in acetone (20ml) for 18h with light of  $\lambda > 400\text{nm}$  under continuous nitrogen purging resulted in a complete conversion of **6b** and gave **5b** (98.9%).

If **6b** obtained by heating **4b** on silica was not separated but was dissolved in acetone and irradiated with light of  $\lambda > 400\text{nm}$  under continuous nitrogen purging for 18h to reach a complete conversion of **4b**, this gave **5b** (97.2%). The total yield of **5b** in photoreaction and in subsequent conversion of **4b** was 90.0%.

#### **4b**

MP 193-194 $^{\circ}\text{C}$

$^1\text{H-NMR}$  (300 MHz;  $\text{CDCl}_3$ )  $\delta$  3.01(3H, s,  $\text{NCH}_3$ ), 5.58(1H, s), 7.00(1H, d,  $J=7.5$ ), 7.04(1H, d,  $J=7.3$ ), 7.28-7.34(3H, m), 7.37-7.39(1H, m), 7.42-7.46(3H, m)

IR (KBr)  $\nu$  3063, 2983, 3948, 2363, 1790, 1723, 1592, 1465, 1426, 1369, 1284, 1227, 1117, 1078, 1016, 939, 832, 790, 753, 698 $\text{cm}^{-1}$

MS  $m/z$ (%) 337( $\text{M}^+-\text{HCl}$ , 92.87), 302 (57.69), 257 (4.08), 252 (8.21), 245 (Base), 223 (4.51), 217 (60.61), 194 (73.99), 189 (58.78), 187 (24.59), 165 (34.26), 151 (6.76), 139 (3.97), 126 (6.97), 94 (15.62), 87 (8.04), 82 (2.41), 63 (2.55)

EA Found: C, 60.99; H, 3.56; N, 3.88.  $\text{C}_{19}\text{H}_{13}\text{Cl}_2\text{N}_1\text{O}_3$  requires C, 60.98; H, 3.50; N, 3.74.

#### **5b**

MP 267-268 $^{\circ}\text{C}$

$^1\text{H-NMR}$  (300 MHz;  $\text{CDCl}_3$ )  $\delta$  3.31(3H, s,  $\text{NCH}_3$ ), 7.57(1H, t,  $J=7.5$ ), 7.66(1H, t,  $J=7.6$ ), 7.76(1H, t,  $J=7.6$ ), 7.84-7.90(2H, m), 8.45(1H, d,  $J=7.7$ ), 8.69(1H,  $J=8.3$ ), 9.18(1H, d,  $J=8.3$ )

IR (KBr)  $\nu$  3081, 1767, 1701, 1621, 1573, 1459, 1433, 1375, 1341, 1210, 1066, 993, 864, 757, 691, 611 $\text{cm}^{-1}$

MS  $m/z$ (%) 301( $\text{M}^+$ , Base), 273(2.05), 257(33.25), 244(6.29), 216(83.61), 187(35.03), 161(1.65), 151(3.09), 136(2.78), 122(2.12), 108(9.36), 94(8.99)

EA Found: C, 75.82; H, 3.72; N, 4.57.  $\text{C}_{19}\text{H}_{11}\text{N}_1\text{O}_3$  requires C, 75.74; H 3.68; N, 4.65.

UV (nm) (log  $\epsilon$ ): 403(4.06), 374(3.95), 279(4.22)

#### **6b**

MP 148-149 $^{\circ}\text{C}$

$^1\text{H-NMR}$  (300 MHz;  $\text{CDCl}_3$ )  $\delta$  3.12(3H, s,  $\text{NCH}_3$ ), 7.35(1H, t,  $J=7.5$ ), 7.44-7.48(6H, m), 7.64(1H, d,  $J=8.3$ ), 7.73(1H, d,  $J=7.8$ )

IR (KBr)  $\nu$  3040, 2940, 1785, 1720, 1628, 1567, 1437, 1379, 1250, 1163, 1111, 1004, 885, 858, 778, 745, 701 $\text{cm}^{-1}$

MS  $m/z$ (%) 337(M, 25.34), 302(39.51), 287(1.05), 257(4.07), 252(4.78), 245(Base), 223(3.01), 217(45.10), 189(41.41), 163(4.75), 151(4.22), 126(4.39), 94(11.29), 84(10.21), 63(1.29), 44(21.99)

EA Found: C, 67.48; H, 3.62; N, 4.11.  $\text{C}_{19}\text{H}_{12}\text{ClNO}_3$  requires C, 67.56; H, 3.58; N, 4.15.

**Photolysis of 1 with 2-phenylbenzothiophene (3c):** A solution of **1** (540 mg, 3 mmol) and **3c** (819 g, 3.9 mmol) in benzene (60 ml) was photolyzed for 27h to reach a 79.3% conversion of **1**. Work-up as described above gave **4c** (84.0%) and **5c** (7.36%).

Heating **4c** (100mg) on silica gel (1g) to 100 $^{\circ}\text{C}$  for 16h, and then dissolving the products in benzene (20 ml) and irradiating the solution with light of  $\lambda > 400\text{nm}$  under continuous nitrogen purging for 6h to reach a complete conversion of **4c**, this gave **5c** (93.9%). The total yield of **5c** in photoreaction and in subsequent conversion of **4c** was 86.5%.

#### **4c**

MP 150-151 $^{\circ}\text{C}$

$^1\text{H-NMR}$  (300 MHz;  $\text{CDCl}_3$ )  $\delta$  2.82(3H, s), 5.08(1H, s), 7.20-7.29(4H, m), 7.31-7.42(5H, m)

IR (KBr)  $\nu$  3059, 1786, 1724, 1580, 1468, 1446, 1430, 1373, 1286, 1257, 1194, 1176, 1157, 1123, 1087, 1070, 1014, 940, 895, 737, 696 $\text{cm}^{-1}$

MS  $m/z$ (%) 389( $M^+$ , 0.47), 353(40.28), 318(43.69), 261(56.63), 233(45.22), 210(base), 165(25.44), 116(20.75), 87(13.51), 44(2.53)

EA Found: C, 58.54; H, 3.52; N, 3.46.  $C_{19}H_{11}N_1O_2S_1Cl_2$  requires C, 58.46; H, 3.33; N, 3.59.

#### 5c

MP 244-245°C

$^1H$ -NMR (300 MHz;  $CDCl_3$ )  $\delta$  3.29(3H, s,  $NCH_3$ ), 7.57-7.61(2H, m), 7.70-7.74(2H, m), 7.93(1H, dd,  $J_1=7.1$ ,  $J_2=1.0$ ), 8.14(1H, dd,  $J_1=7.1$ ,  $J_2=1.2$ ), 9.13(1H, dd,  $J_1=8.1$ ,  $J_2=1.5$ ), 9.70(1H, dd,  $J_1=8.1$ ,  $J_2=1.3$ )

IR (KBr)  $\nu$  3058, 1758, 1702, 1545, 1450, 1441, 1381, 1327, 1258, 1243, 1075, 1003, 957, 752, 728,  $650cm^{-1}$

MS  $m/z$ (%) 317( $M^+$ , base), 273(13.64), 232(40.36), 187(11.34), 158(3.23), 116(13.04), 57(3.37), 44(5.67)

EA Found: C, 71.94; H, 3.50; N, 4.20.  $C_{19}H_{11}N_1O_2S_1$  requires C, 71.92; H, 3.47; N, 4.42.

UV (nm) (log  $\epsilon$ ): 405(3.74), 321(4.18), 302(4.42), 278(4.51)

**Photolysis of 1 with 3-phenylbenzothiophene (3d):** A solution of **1** (540 mg, 3 mmol) and **3d** (819 g, 3.9 mmol) in benzene (60 ml) was photolyzed for 44h to reach a 94.8% conversion of **1**. Work-up as described above gave **4d** (74.8%) and **5d** (8.33%).

Heating **4d** (100mg) absorbed on silica gel (1g) at 100°C for 6h, and then dissolving the products in benzene (20 ml) and irradiating the solution with light of  $\lambda > 400nm$  under continuous nitrogen purging for 8h to reach a complete conversion of **4d**, this gave **5d** (85.5%). The total yield of **5d** in photoreaction and in subsequent conversion of **4d** was 72.3%.

#### 4d

MP 215-216°C

$^1H$ -NMR (300 MHz;  $CDCl_3$ )  $\delta$  3.05(3H, s), 4.89(1H, s), 7.09-7.14(1H, m), 7.24-7.27(2H, m), 7.29-7.39(4H, m), 7.55-7.58(2H, m)

IR (KBr)  $\nu$  3059, 2973, 1792, 1718, 1463, 1448, 1423, 1368, 1277, 1126, 1012, 941, 749, 727, 700,  $631cm^{-1}$

MS  $m/z$ (%) 354( $M^+-Cl$ , 43.25), 318(27.74), 261(39.58), 210(base), 165(36.24), 116(16.83), 87(20.35), 45(10.39)

EA Found: C, 58.43; H, 3.24; N, 3.43.  $C_{19}H_{11}N_1O_2S_1Cl_2$  requires C, 58.46; H, 3.33; N, 3.59.

#### 5d

MP 251-252°C

$^1H$ -NMR (300 MHz;  $CDCl_3$ )  $\delta$  3.30(3H, s), 7.62(1H, td,  $J_1=7.4$ ,  $J_2=1.5$ ), 7.68(1H, td,  $J_1=7.6$ ,  $J_2=1.4$ ), 7.77(1H, td,  $J_1=7.6$ ,  $J_2=1.0$ ), 7.87(1H, ddd,  $J_1=8.5$ ,  $J_2=7.0$ ,  $J_3=1.5$ ), 8.10(1H, dd,  $J_1=8.1$ ,  $J_2=1.5$ ), 8.89(1H, dd,  $J_1=8.0$ ,  $J_2=1.3$ ), 9.08(1H, d,  $J=8.6$ ), 9.17(1H, dd,  $J_1=8.7$ ,  $J_2=1.1$ )

IR (KBr)  $\nu$  3059, 2939, 1764, 1695, 1551, 1450, 1421, 1386, 1371, 1361, 1327, 1256, 1185, 1059, 991, 780, 755, 732,  $663cm^{-1}$

MS  $m/z$ (%) 317( $M^+$ , base), 273(5.54), 232(44.04), 187(13.05), 116(10.07), 43(18.14)

EA Found: C, 71.87; H, 3.36; N, 4.58.  $C_{19}H_{11}N_1O_2S_1$  requires C, 71.92; H, 3.47; N, 4.42.

UV (nm) (log  $\epsilon$ ): 401(4.00), 279(4.40)

**Photolysis of 2 with 2-phenylbenzofuran (3a):** A solution of **2** (645 mg, 3 mmol) and **3a** (756 mg, 3.9 mmol) in benzene (60 ml) was photolyzed for 48h to reach a 42.6% conversion of **2**. After evaporation of the solvent, the reaction mixture was separated by flash column chromatography on silica gel with petroleum ether (b.p. 60-90°C) ethyl acetate as eluents to give **7a**(58.4%).

#### 7a

MP 189-190°C

$^1H$ -NMR (300 MHz;  $CDCl_3$ )  $\delta$  7.36(1H, m), 7.44-7.57(4H, m), 7.66-7.70(2H, m), 7.80(1H, t,  $J=8.0$ ), 8.33(1H, d,  $J=7.9$ ), 8.52(1H, d,  $J=7.2$ ), 8.81(1H, d,  $J=8.5$ ), 9.18(1H, d,  $J=7.9$ )

IR (KBr)  $\nu$  3059, 2922, 1736, 1608, 1487, 1368, 1347, 1225, 1060, 976,  $740cm^{-1}$

MS  $m/z$ (%) 336( $M^+$ , Base), 308(26.65), 279(19.32), 250(15.84), 154(5.61), 125(6.22)

EA Found: C, 82.35; H, 3.41.  $C_{23}H_{12}O_3$  requires C, 82.13; H, 3.60.

UV (nm) (log  $\epsilon$ ): 403(4.36), 384(4.32), 342(4.21), 327(4.19), 291(4.59)

**Photolysis of 2 with 3-phenylbenzofuran (3b):** A solution of **2** (645 mg, 3 mmol) and **3b** (756 mg, 3.9 mmol) in benzene (60 ml) was photolyzed for 48h to reach a 35.7% conversion of **2**. Work-up as described above gave **7b** (70.8%).

#### 7b

MP 204-5°C

$^1H$ -NMR (300 MHz;  $CDCl_3$ )  $\delta$  7.45(1H, td,  $J_1=4.1$ ,  $J_2=1.9$ ), 7.51-7.62(3H, m), 7.70(1H, t,  $J=7.8$ ), 7.90(2H, t,  $J=7.8$ ), 8.39(1H, d,  $J=7.5$ ), 8.47(1H, d,  $J=8.1$ ), 8.73(1H, d,  $J=8.3$ ), 8.93(1H, d,  $J=8.6$ )

IR (KBr)  $\nu$  3080, 1745, 1612, 1550, 1464, 1353, 1271, 1213, 1178, 1110, 1056, 1026, 759,  $687cm^{-1}$

MS  $m/z$ (%) 337( $M^++1$ , 12.31), 336( $M^+$ , Base), 308(60.63), 279(19.27), 250(11.08), 224(2.82), 168(2.52), 154(4.23), 140(4.39), 125(7.88), 112(3.63)

EA Found: C, 81.96; H, 3.55.  $C_{23}H_{12}O_3$  requires C, 82.13; H, 3.60.

UV (nm) (log  $\epsilon$ ): 410(4.20), 391(4.20), 347(4.13), 331(3.96), 280(4.36)

**Photolysis of 2 with 3-phenylbenzothiophene (3d):** A solution of **2** (645 mg, 3 mmol) and **3d** (819 mg, 3.9 mmol) in benzene (60 ml) was photolyzed for 72h to reach a 36.1% conversion of **2**. Work-up as described above gave **7d** (66.6%).

**7d**

MP 219-221 °C

<sup>1</sup>H-NMR (300 MHz; CDCl<sub>3</sub>) δ 7.45-7.49(1H, m), 7.57-7.67(4H, m), 7.75(1H, td, *J*=7.2), 7.93(1H, t, *J*=7.2), 8.14(1H, d, *J*=7.5), 8.50(1H, d, *J*=8.0), 8.85(1H, d, *J*=8.0), 8.98(1H, d, *J*=8.5), 9.18(1H, d, *J*=8.5)

IR (KBr) ν 3052, 1712, 1607, 1531, 1454, 1359, 1252, 1190, 1127, 1112, 1007, 764, 753, 727, 660 cm<sup>-1</sup>

MS *m/z*(%) 352(base), 324(13.95), 295(13.05), 44(15.22)

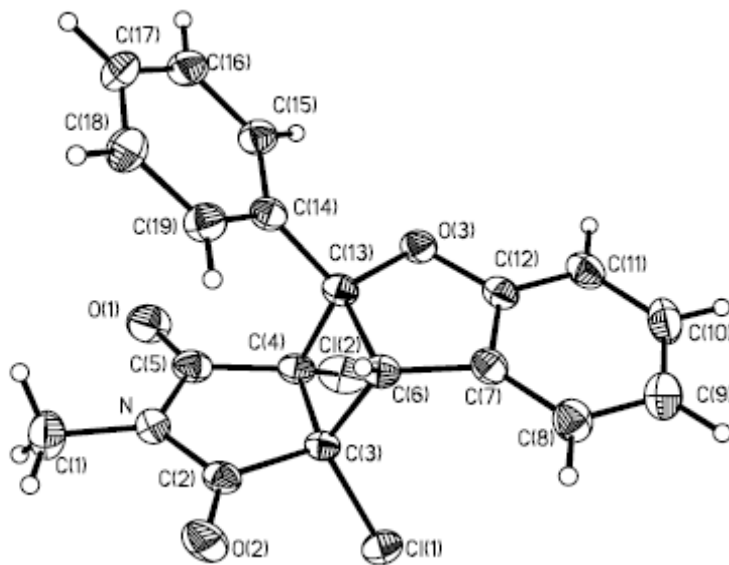
EA Found: C, 78.38; H, 3.49. C<sub>23</sub>H<sub>12</sub>O<sub>2</sub>S<sub>1</sub> requires C, 78.40; H, 3.41.

UV (nm) (log ε): 418(3.96), 398(3.90), 366(4.19), 352(4.06), 323(4.01), 310(3.96), 280(4.52)

**X-ray Crystallographic Study:**

Crystal data for **4a** (Fig. 1): C<sub>19</sub>H<sub>13</sub>Cl<sub>2</sub>NO<sub>3</sub>, M = 374.20, colorless blocks, Enraf-Nonius CAD4 diffractometer, Mo-*K*<sub>α</sub> radiation (λ = 0.71073 Å), 0.40 × 0.30 × 0.28 mm, T = 293(2) K. Monoclinic, space group *P*2<sub>1</sub>/*c*, a = 12.895(3) Å, b = 9.910(2) Å, c = 13.175(3) Å, α = 90.00°, β = 90.54(3)°, γ = 90.00°, V = 1683.6(6) Å<sup>3</sup>, Z = 4, D<sub>c</sub> = 1.476 g cm<sup>-3</sup>, μ = 0.404 mm<sup>-1</sup>, F(000) = 768.00. The structure was solved by direct method (SHELXL) and refined on F<sup>2</sup> by full-matrix least-squares method. A total of 2971 independent reflections [R (int) = 0.0426] were used in the refinement, which converged with R = 0.0536 and wR = 0.1931 (GOF = 1.004).

Crystal data for **5b** (Fig. 2): C<sub>19</sub>H<sub>11</sub>NO<sub>3</sub>, M = 301.29, yellow needles, Enraf-Nonius CAD4 diffractometer, Mo-*K*<sub>α</sub> radiation (λ = 0.71073 Å), 0.42 × 0.30 × 0.24 mm, T = 293(2) K. Monoclinic, space group *P*2<sub>1</sub>/*c*, a = 9.5260(19) Å, b = 18.319(4) Å, c = 7.9420(16) Å, α = 90.00°, β = 96.25(3)°, γ = 90.00°, V = 1377.7(5) Å<sup>3</sup>, Z = 4, D<sub>c</sub> = 1.453 g cm<sup>-3</sup>, μ = 0.099 mm<sup>-1</sup>, F(000) = 624.00. The structure was solved by direct method (SHELXL) and refined on F<sup>2</sup> by full-matrix least-squares method. A total of 2439 independent reflections [R (int) = 0.0433] were used in the refinement, which converged with R = 0.0551 and wR = 0.1188 (GOF = 1.000).



**Fig. 1** ORTEP drawing of **4a**

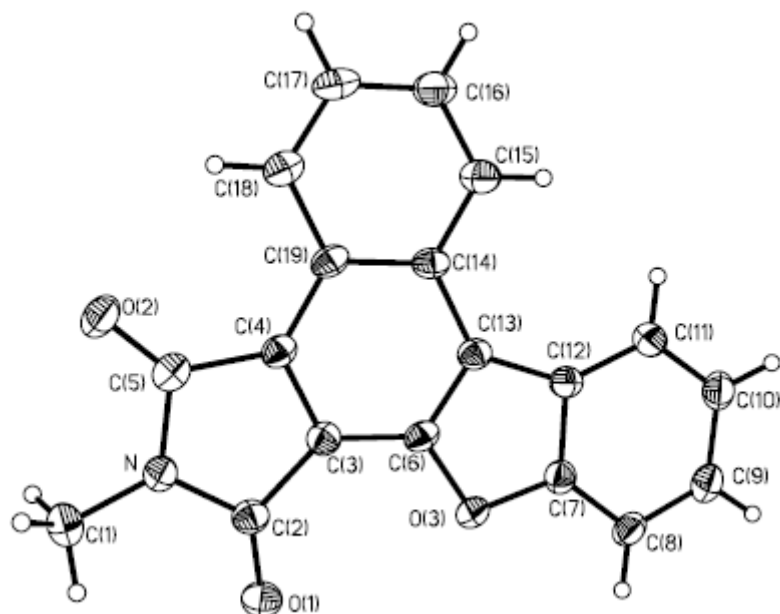


Fig. 2 ORTEP drawing of **5b**

**Table** Luminescence Quantum Yield

Compd <sup>a</sup>	$\lambda_{\text{ex}}$ <sup>b</sup> (nm)	$\lambda_{\text{em}}$ <sup>c</sup> (nm)	$\Phi_f$ <sup>d</sup>
<b>5a</b>	342	450	0.41
<b>5b</b>	372	453	0.63
<b>5c</b>	405	463	0.33
<b>5d</b>	402	461	0.38
<b>7a</b>	403	441	0.2
<b>7b</b>	347	449	0.73
<b>7d</b>	406	440	0.42

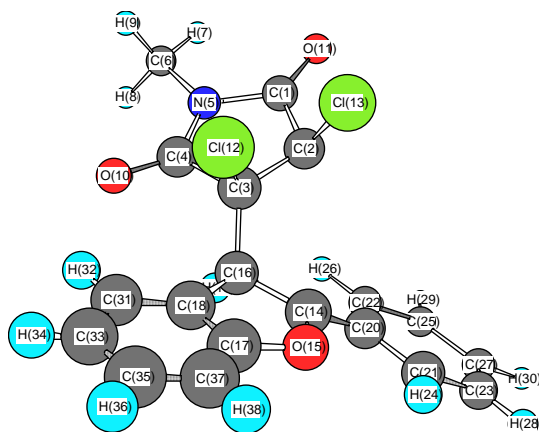
<sup>a</sup> Correct emission spectra were obtained in dilute benzene solution. <sup>b</sup> Only the absorption maxima appropriate to measure the quantum yield are shown. <sup>c</sup> Emission maximum wavelength excited at the absorption maximum. <sup>d</sup> Quantum efficiencies of **5a** and **7b** using quinine sulfate in 0.1N H<sub>2</sub>SO<sub>4</sub> as a standard at the corresponding absorption maximum. The others were measured using **5a** in benzene as a standard at the corresponding absorption maximum.

#### DFT calculation results:

Optimized structures of triplet diradical intermediates **A(1, 3a)**, **A(1, 3b)**, **E(2, 3a)** and **E(2, 3b)**

Provided are the Cartesian coordinates for **A(1, 3a)**, **A(1, 3b)**, **E(2, 3a)** and **E(2, 3b)** along with the  $S^2$  values. The structural optimizations were performed by standard procedures within Gaussian 03 package.

<b>A(1, 3a)</b>
-----------------



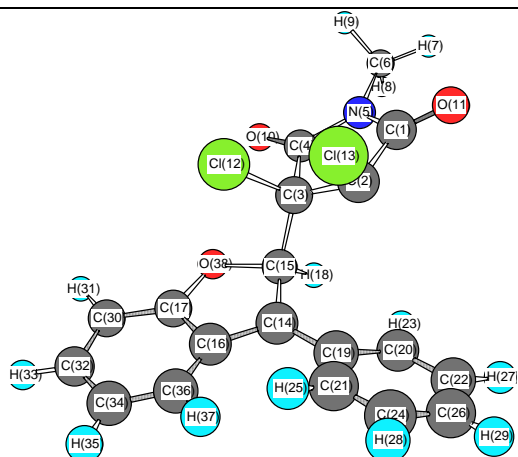
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E(UB+HF-LYP) =  
-1932.27632326

S\*\*2 = 2.0328

S\*\*2 before  
annihilation 2.0328, after  
2.0006

Center Number	Atomic Number	Coordinates (Angstroms)			Total atomic spin densities:
		X	Y	Z	
1	6	1.502419	2.682500	0.265718	1 C -0.101489
2	6	1.009398	1.384016	0.713377	2 C 0.752716
3	6	-0.331964	1.097725	0.156376	3 C -0.045619
4	6	-0.553486	2.312712	-0.782348	4 C 0.000520
5	7	0.521442	3.173703	-0.635875	5 N 0.054859
6	6	0.642415	4.449540	-1.336838	6 C -0.005172
7	1	1.663123	4.808251	-1.200739	7 H -0.000116
8	1	0.425767	4.307659	-2.397943	8 H 0.003946
9	1	-0.060822	5.179959	-0.926217	9 H 0.004746
10	8	-1.497286	2.473336	-1.564003	10 O -0.001496
11	8	2.550145	3.274299	0.560714	11 O 0.152371
12	17	-1.568828	1.378529	1.639116	12 Cl 0.112893
13	17	1.822191	0.513903	1.999944	13 Cl 0.093494
14	6	0.093430	-1.494322	-0.090109	14 C 0.682369
15	8	0.827614	-2.406269	0.473456	15 O 0.066222
16	6	-0.633514	-0.259947	-0.590037	16 C -0.043056
17	6	-2.121124	-1.950731	0.155382	17 C -0.009602
18	6	-2.085841	-0.722156	-0.508824	18 C 0.025332
19	1	-0.346220	-0.055413	-1.632692	19 H 0.027120
20	6	1.408388	-1.955066	-0.353847	20 C -0.225399
21	6	1.835884	-3.234088	0.108739	21 C 0.243823
22	6	2.336385	-1.171241	-1.098417	22 C 0.241698
23	6	3.122251	-3.689471	-0.158397	23 C -0.137704
24	1	1.139174	-3.844159	0.670184	24 H -0.009777
25	6	3.618627	-1.643124	-1.360512	25 C -0.136369
26	1	2.042289	-0.196047	-1.473844	26 H -0.009820
27	6	4.024828	-2.902421	-0.892323	27 C 0.253438
28	1	3.428636	-4.665553	0.204656	28 H 0.004449
29	1	4.307612	-1.029759	-1.932559	29 H 0.004487
30	1	5.027084	-3.263929	-1.096217	30 H -0.011084
31	6	-3.277746	-0.147875	-0.948026	31 C -0.004579
32	1	-3.271423	0.809807	-1.453919	32 H 0.000166
33	6	-4.479498	-0.829147	-0.697938	33 C 0.011984
34	1	-5.416395	-0.397942	-1.033652	34 H -0.000474
35	6	-4.485862	-2.054337	-0.013948	35 C -0.004878
36	1	-5.426843	-2.561173	0.172850	36 H 0.000136
37	6	-3.291413	-2.642230	0.428290	37 C 0.010211
38	1	-3.272657	-3.591916	0.947968	38 H -0.000346

A(1, 3b)



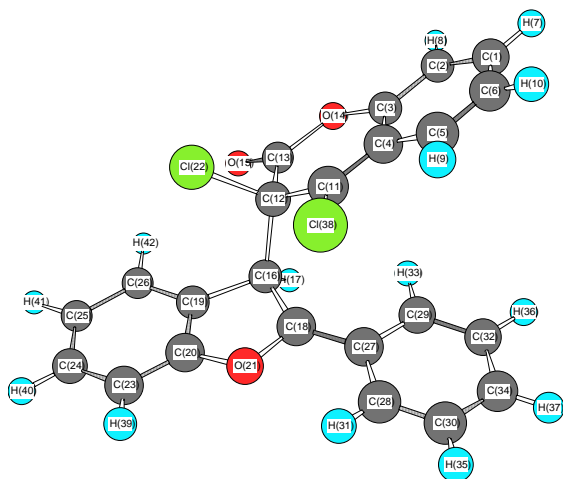
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S\*\*2 = 2.0353

S\*\*2 before  
annihilation 2.0353, after  
2.0008

Center Number	Atomic Number	Coordinates (Angstroms)			Total atomic spin densities:
		X	Y	Z	
1	6	-3.269743	0.371772	0.175277	1 C -0.103470
2	6	-1.859839	0.455706	0.542806	2 C 0.772671
3	6	-1.118100	-0.767154	0.143215	3 C -0.049598
4	6	-2.243582	-1.618022	-0.525289	4 C -0.000835
5	7	-3.421384	-0.875728	-0.471517	5 N 0.061464
6	6	-4.696326	-1.360834	-0.993206	6 C -0.006069
7	1	-5.444603	-0.590678	-0.804220	7 H -0.000068
8	1	-4.616726	-1.552640	-2.066534	8 H 0.005683
9	1	-4.978602	-2.289764	-0.490724	9 H 0.004218
10	8	-2.139961	-2.730709	-1.038659	10 O 0.003050
11	8	-4.180825	1.191037	0.367242	11 O 0.144277
12	17	-0.600903	-1.697662	1.742041	12 Cl 0.087730
13	17	-1.306466	1.723931	1.621257	13 Cl 0.093286
14	6	1.286737	0.238938	-0.384141	14 C 0.653062
15	6	0.119247	-0.640913	-0.810524	15 C -0.036165
16	6	2.391456	-0.637978	-0.130580	16 C -0.200607
17	6	1.993977	-1.955841	-0.481793	17 C 0.179416
18	1	-0.289119	-0.360671	-1.790364	18 H 0.027696
19	6	1.274691	1.685008	-0.459658	19 C -0.161444
20	6	0.415797	2.367737	-1.361551	20 C 0.158049
21	6	2.132306	2.476836	0.348060	21 C 0.158163
22	6	0.430247	3.757544	-1.462394	22 C -0.088357
23	1	-0.250177	1.803885	-2.006761	23 H -0.006570
24	6	2.137874	3.867123	0.246436	24 C -0.088162
25	1	2.754795	1.997306	1.093041	25 H -0.006952
26	6	1.292596	4.518483	-0.661733	26 C 0.161019
27	1	-0.230369	4.249773	-2.169044	27 H 0.003321
28	1	2.795592	4.446856	0.886443	28 H 0.003090
29	1	1.299763	5.600513	-0.739023	29 H -0.007026
30	6	2.824296	-3.059172	-0.411572	30 C -0.103612
31	1	2.464880	-4.040576	-0.693641	31 H 0.003132
32	6	4.139605	-2.841078	0.030618	32 C 0.222918
33	1	4.822269	-3.680505	0.107210	33 H -0.009942
34	6	4.584659	-1.547058	0.361230	34 C -0.105828
35	1	5.610948	-1.402429	0.681762	35 H 0.003161
36	6	3.732692	-0.444244	0.277188	36 C 0.210280
37	1	4.108325	0.543839	0.507743	37 H -0.007476
38	8	0.685343	-1.990641	-0.959292	38 O 0.026495

**E(2, 3a)**



SCF Done:  
E(UB+HF-LYP) =  
-2020.35494883

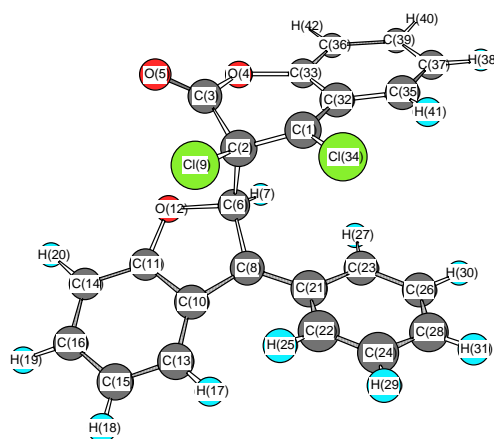
S\*\*2 = 2.0431

S\*\*2 before  
annihilation 2.0431, after  
2.0012

Center Number	Atomic Number	Coordinates (Angstroms)			Total atomic spin densities:
		X	Y	Z	
1	6	-4.982791	-1.559010	-0.178238	1 C 0.243765
2	6	-3.916323	-2.180125	-0.826572	2 C -0.120745
3	6	-2.617568	-1.843723	-0.494549	3 C 0.200590
4	6	-2.328007	-0.876995	0.496436	4 C -0.194162
5	6	-3.432443	-0.255921	1.144641	5 C 0.219995
6	6	-4.730940	-0.592586	0.810545	6 C -0.119289
7	1	-5.996234	-1.823292	-0.438387	7 H -0.011116
8	1	-4.067596	-2.928423	-1.587059	8 H 0.004030
9	1	-3.231396	0.487037	1.897751	9 H -0.009452
10	1	-5.553766	-0.107828	1.313281	10 H 0.003981
11	6	-0.980525	-0.580131	0.777230	11 C 0.645176
12	6	0.167145	-1.193665	0.062440	12 C -0.011477
13	6	-0.242487	-2.253268	-0.980650	13 C 0.005355
14	8	-1.594819	-2.503659	-1.182201	14 O 0.004595
15	8	0.564139	-2.828893	-1.685814	15 O 0.021227
16	6	1.108943	-0.128075	-0.689642	16 C -0.033583
17	1	0.798826	-0.156628	-1.736674	17 H 0.024548
18	6	1.010556	1.297619	-0.177000	18 C 0.647605
19	6	2.614144	-0.377564	-0.546365	19 C 0.027145
20	6	3.181793	0.700112	0.131851	20 C -0.009087
21	8	2.231087	1.699026	0.416386	21 O 0.081872
22	17	1.168871	-2.172438	1.379723	22 Cl 0.110407
23	6	4.526333	0.764706	0.445936	23 C 0.010126
24	6	5.331312	-0.306311	0.037839	24 C -0.003476
25	6	4.785303	-1.391247	-0.654573	25 C 0.012406
26	6	3.415366	-1.437804	-0.946886	26 C -0.003506
27	6	0.043601	2.296496	-0.438234	27 C -0.198812
28	6	0.260144	3.623359	0.026685	28 C 0.221626
29	6	-1.146947	2.027438	-1.166737	29 C 0.218948
30	6	-0.670362	4.618887	-0.225043	30 C -0.119360
31	1	1.165473	3.833358	0.572223	31 H -0.009726
32	6	-2.065303	3.038206	-1.413743	32 C -0.118284
33	1	-1.340829	1.030731	-1.533328	33 H -0.009755
34	6	-1.840024	4.339281	-0.945499	34 C 0.234709
35	1	-0.488093	5.621117	0.136406	35 H 0.004138
36	1	-2.961706	2.815606	-1.974996	36 H 0.004103
37	1	-2.560738	5.120155	-1.140371	37 H -0.010939
38	17	-0.581963	0.522916	2.136500	38 Cl 0.037074
39	1	4.929363	1.614156	0.972985	39 H -0.000354
40	1	6.387858	-0.286612	0.261738	40 H 0.000093
41	1	5.422267	-2.207950	-0.959541	41 H -0.000458
42	1	2.970364	-2.279416	-1.446004	42 H 0.000068



## E(2, 3b)



SCF Done:  
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S\*\*2 = 2.0480

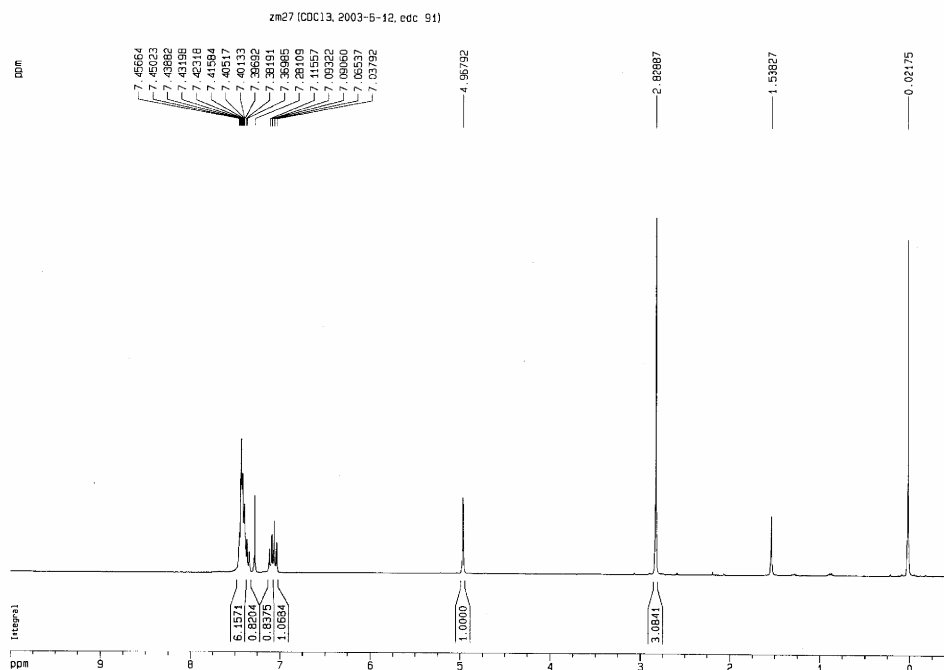
S\*\*2 before  
annihilation 2.0480, after  
2.0015

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1	6	-1.521046	-0.151112	0.944771	1 C 0.691913
2	6	-0.365863	-1.073164	0.724098	2 C -0.024610
3	6	-0.896523	-2.499933	0.485386	3 C 0.008656
4	8	-1.919713	-2.525916	-0.489788	4 O 0.011878
5	8	-0.509398	-3.532509	0.970076	5 O 0.038082
6	6	0.448997	-0.695440	-0.561660	6 C -0.013321
7	1	-0.267504	-0.630454	-1.387518	7 H 0.026607
8	6	1.379392	0.500401	-0.445973	8 C 0.635074
9	17	0.786028	-1.150208	2.181334	9 Cl 0.013254
10	6	2.711813	-0.006107	-0.520502	10 C -0.183338
11	6	2.642998	-1.409762	-0.736573	11 C 0.164875
12	8	1.327791	-1.859718	-0.820189	12 O 0.033645
13	6	3.983099	0.607129	-0.504477	13 C 0.207895
14	6	3.761795	-2.201216	-0.891921	14 C -0.093579
15	6	5.118467	-0.185413	-0.656402	15 C -0.099622
16	6	5.015292	-1.574134	-0.838420	16 C 0.221121
17	1	4.075444	1.678946	-0.396860	17 H -0.008553
18	1	6.097737	0.277591	-0.640548	18 H 0.003181
19	1	5.912632	-2.170263	-0.950359	19 H -0.010336
20	1	3.661434	-3.265522	-1.051751	20 H 0.002986
21	6	0.931208	1.874291	-0.402284	21 C -0.145385
22	6	1.679104	2.867613	0.274967	22 C 0.142691
23	6	-0.262117	2.275727	-1.051624	23 C 0.139823
24	6	1.262671	4.195006	0.282545	24 C -0.075496
25	1	2.556912	2.576212	0.836472	25 H -0.006617
26	6	-0.667469	3.605555	-1.044517	26 C -0.074909
27	1	-0.859503	1.542555	-1.579591	27 H -0.006425
28	6	0.092437	4.575142	-0.381648	28 C 0.144055
29	1	1.844699	4.935170	0.819584	29 H 0.003052
30	1	-1.576568	3.890275	-1.561822	30 H 0.003100
31	1	-0.227401	5.610278	-0.376315	31 H -0.006686
32	6	-2.669350	-0.270907	0.140421	32 C -0.203794
33	6	-2.813489	-1.460319	-0.629886	33 C 0.204974
34	17	-1.365272	1.236311	2.077872	34 Cl 0.036895
35	6	-3.682567	0.719316	0.016796	35 C 0.227627
36	6	-3.873895	-1.629720	-1.504978	36 C -0.121338
37	6	-4.743979	0.538924	-0.853632	37 C -0.121720
38	1	-5.502179	1.307297	-0.941016	38 H 0.004063
39	6	-4.841098	-0.630350	-1.626680	39 C 0.247418
40	1	-5.671000	-0.766500	-2.308424	40 H -0.011327
41	1	-3.589506	1.621101	0.606060	41 H -0.009838
42	1	-3.936740	-2.554443	-2.062050	42 H 0.004026

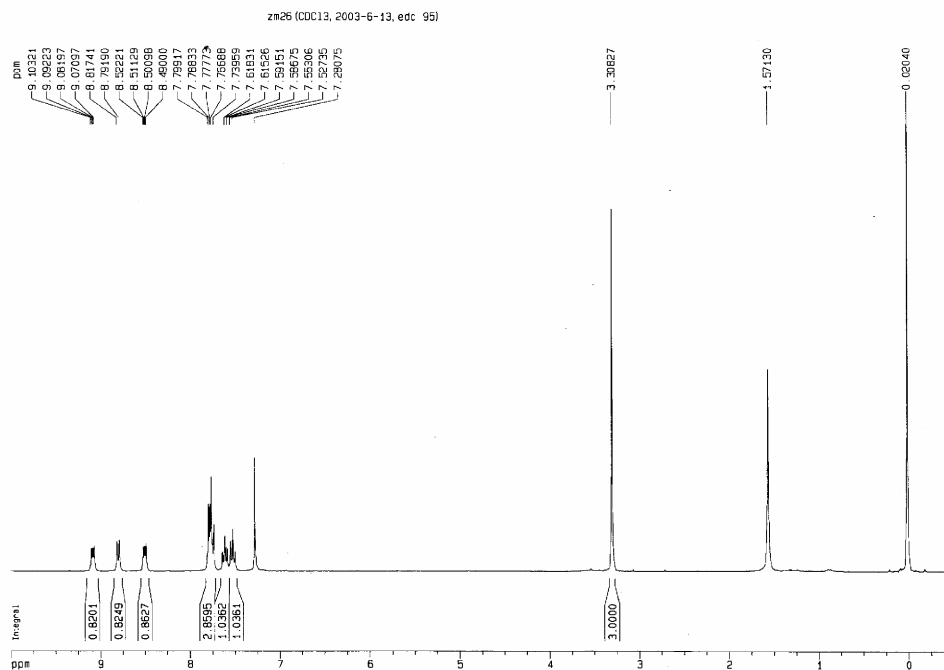
# Title: Aromatic Annulation Strategy for Naphthalenes Fused at 1,2- and 3,4- Positions with Two Heterocycles

Author: Min Zhang, Hui-Ying An, Bao-Guo Zhao and Jian-Hua Xu\*

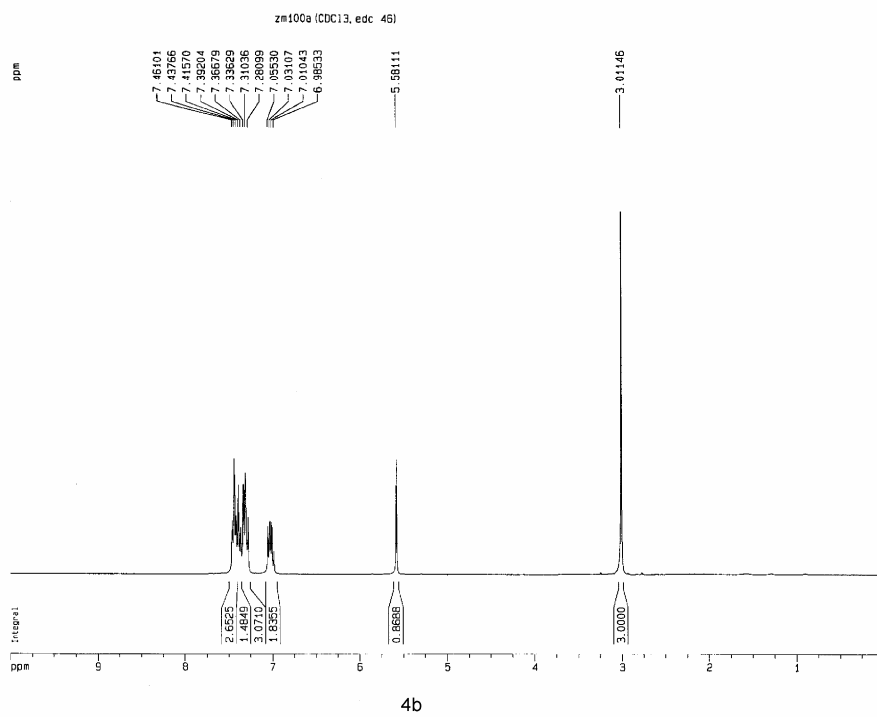
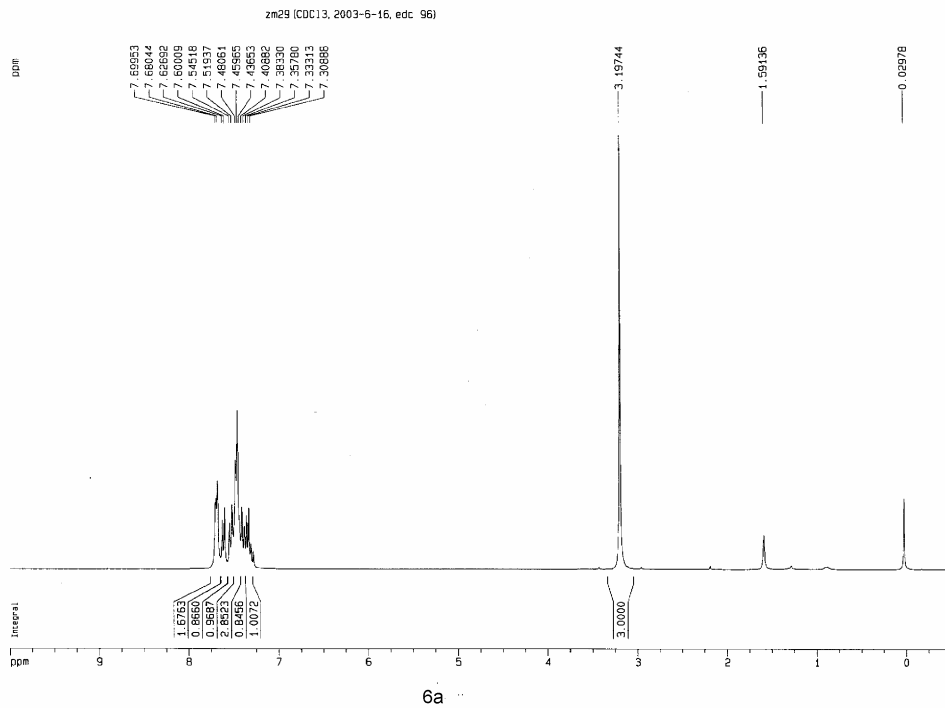
<sup>1</sup>HNMR spectra of all the new compounds (4a, 5a, 6a, 4b, 5b, 6b, 4c, 5c, 4d, 5d, 7a, 7b, 7c):

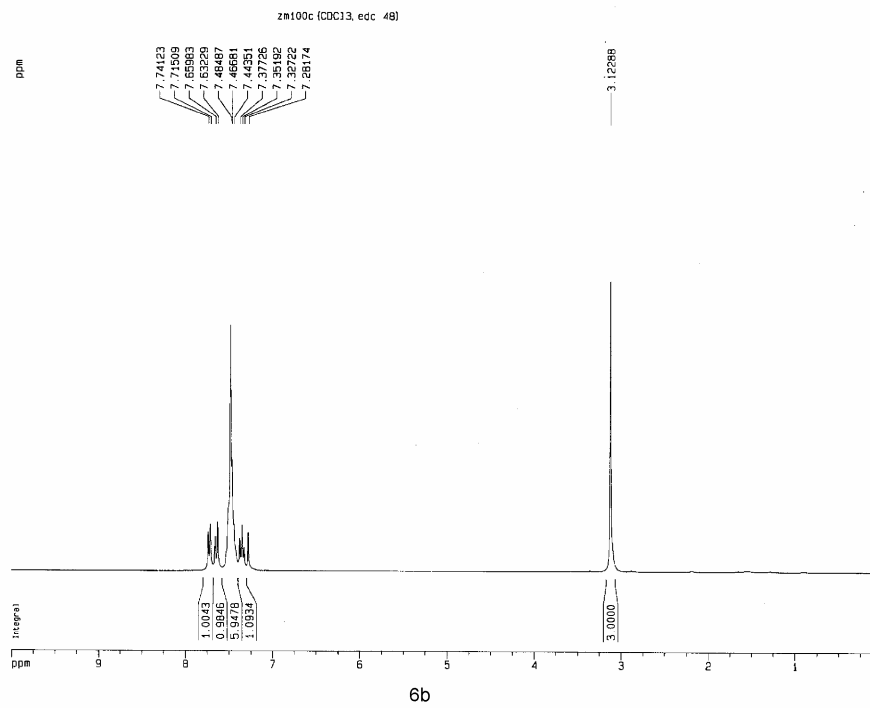
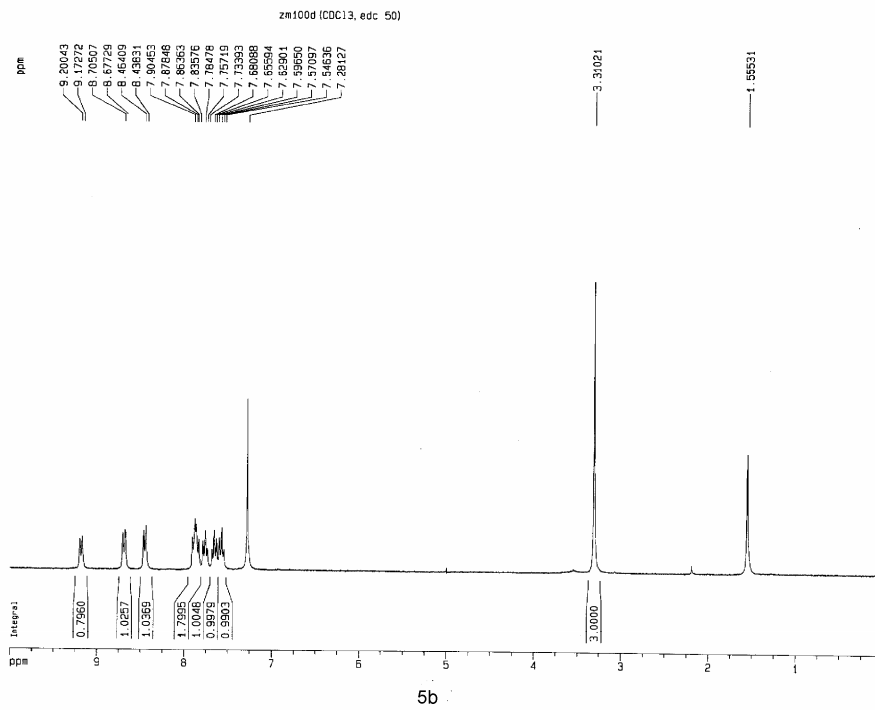


4a

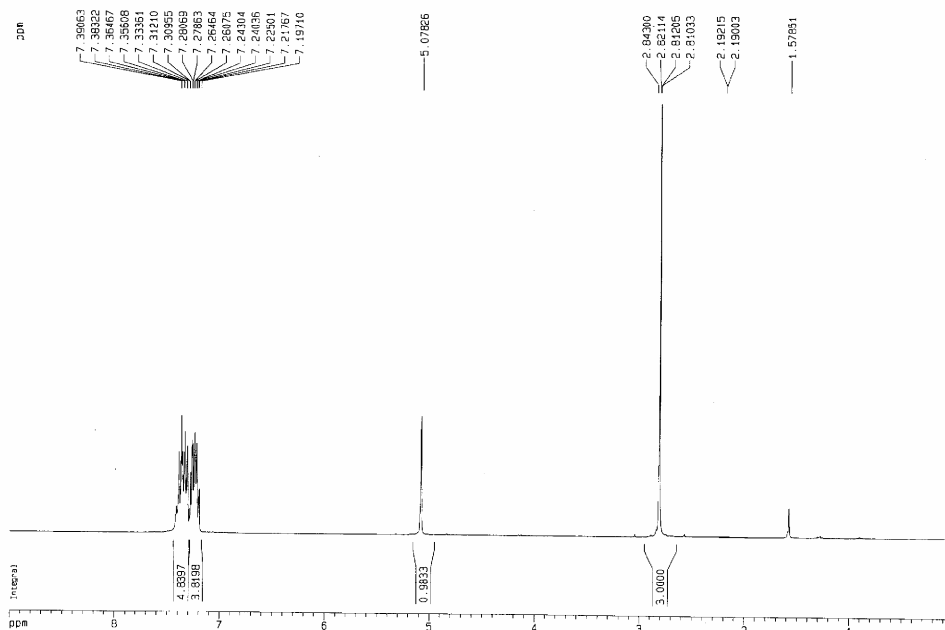


5a



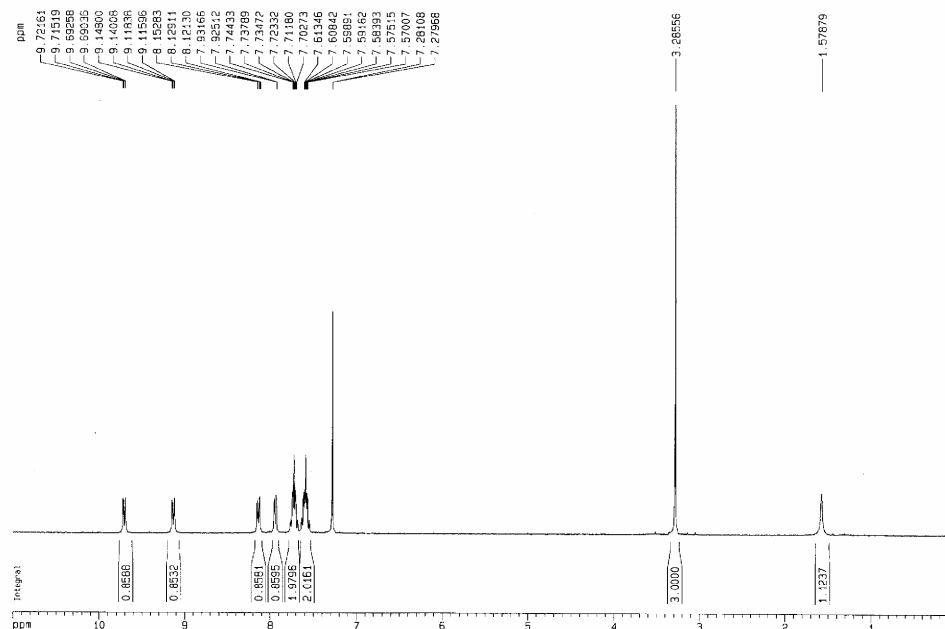


anhy15 (cd 16) 2004.11.25



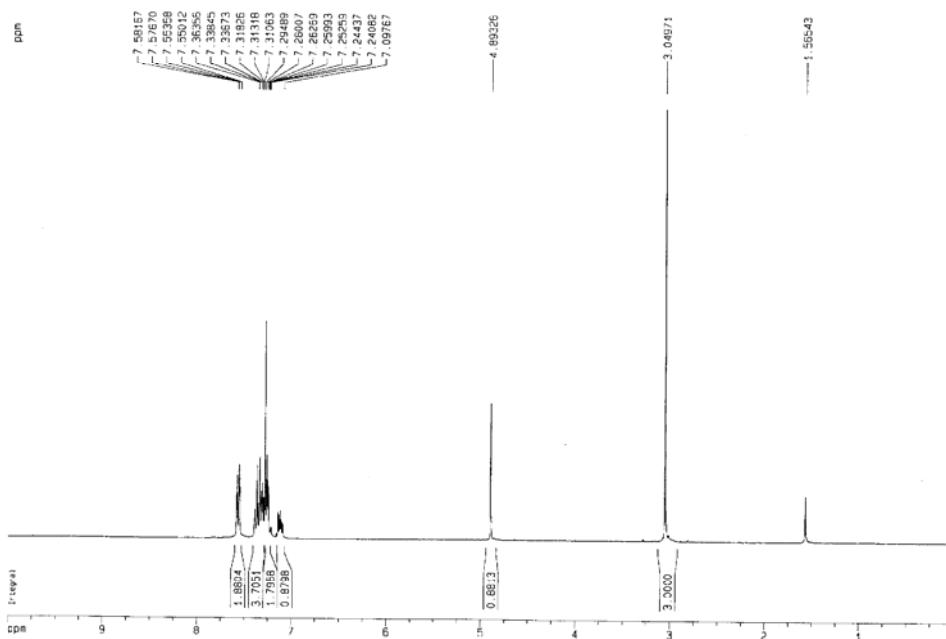
4c

anhy13 (EOC 14) 2004.11.25



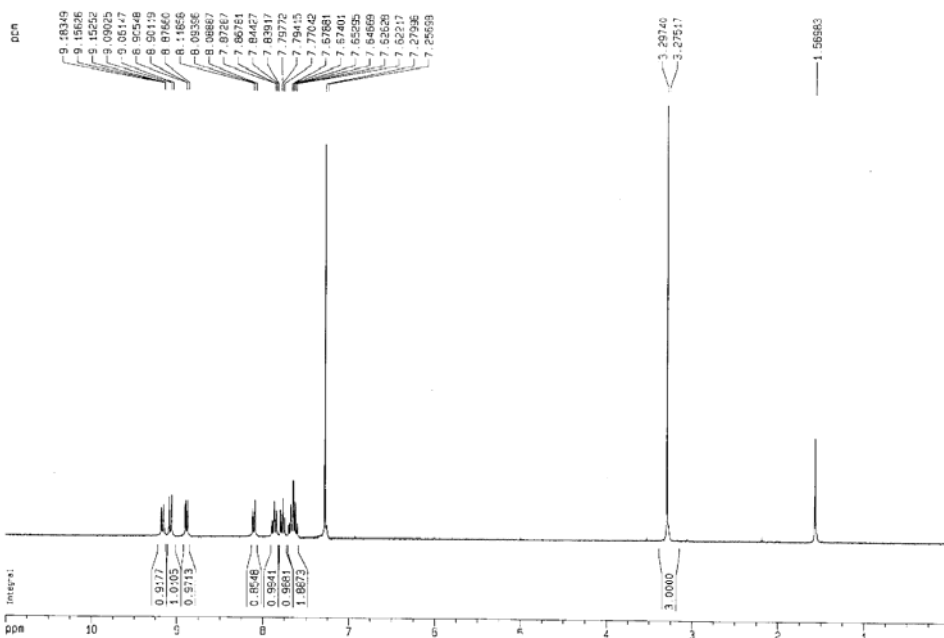
5c

arhy17 2004.12.28 (edc 21)

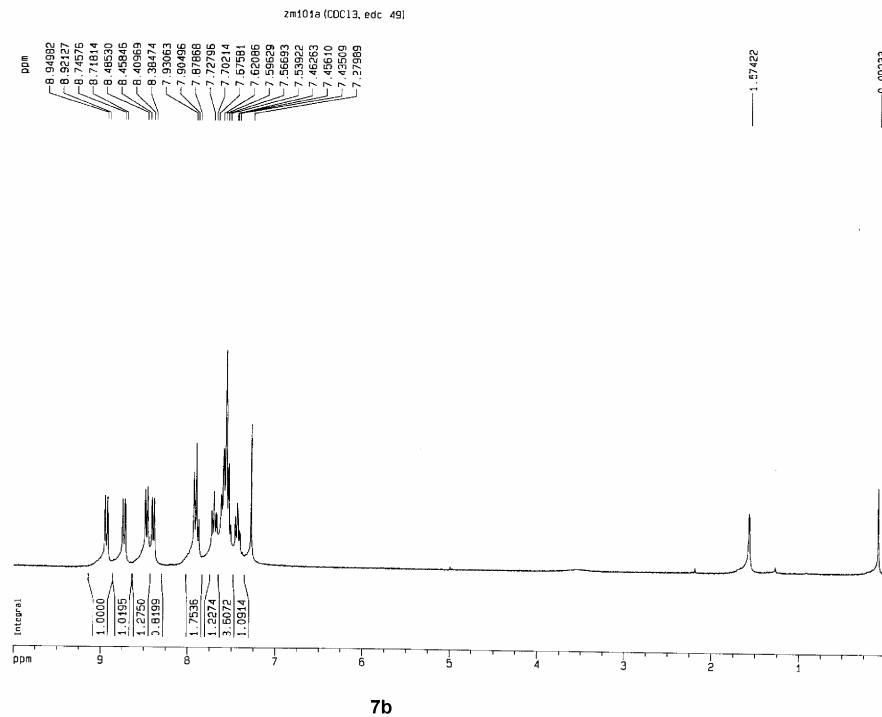
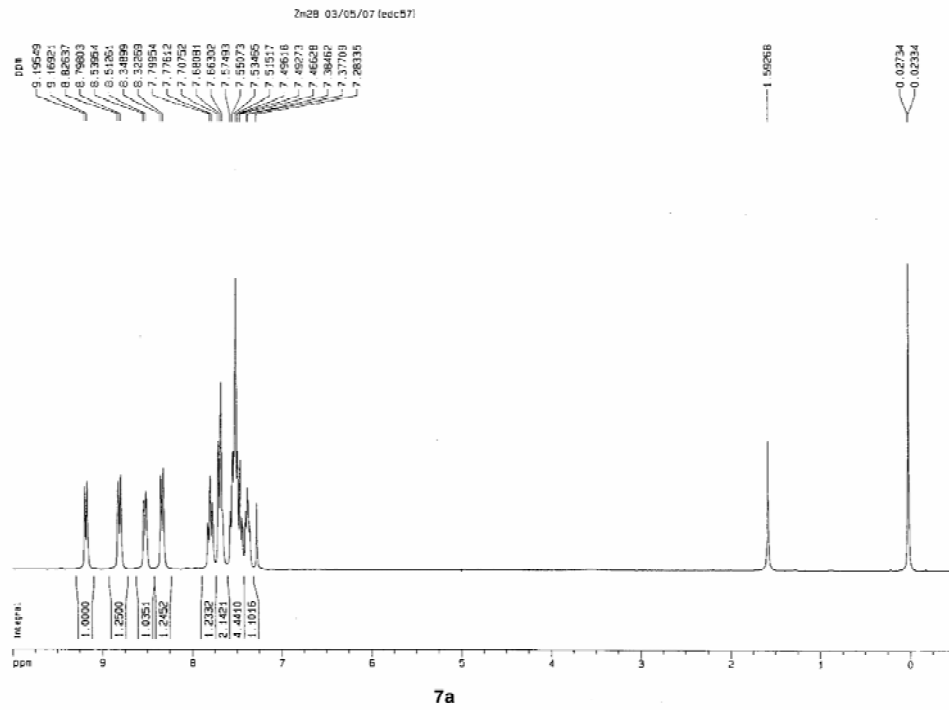


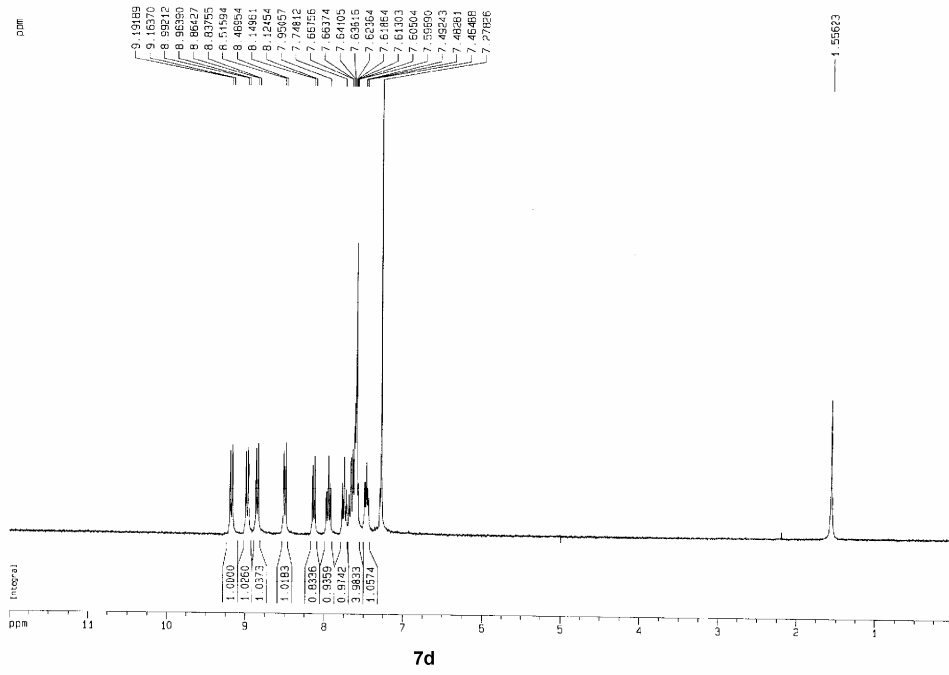
4d

arhy16 2004.12.28 (edc 20)

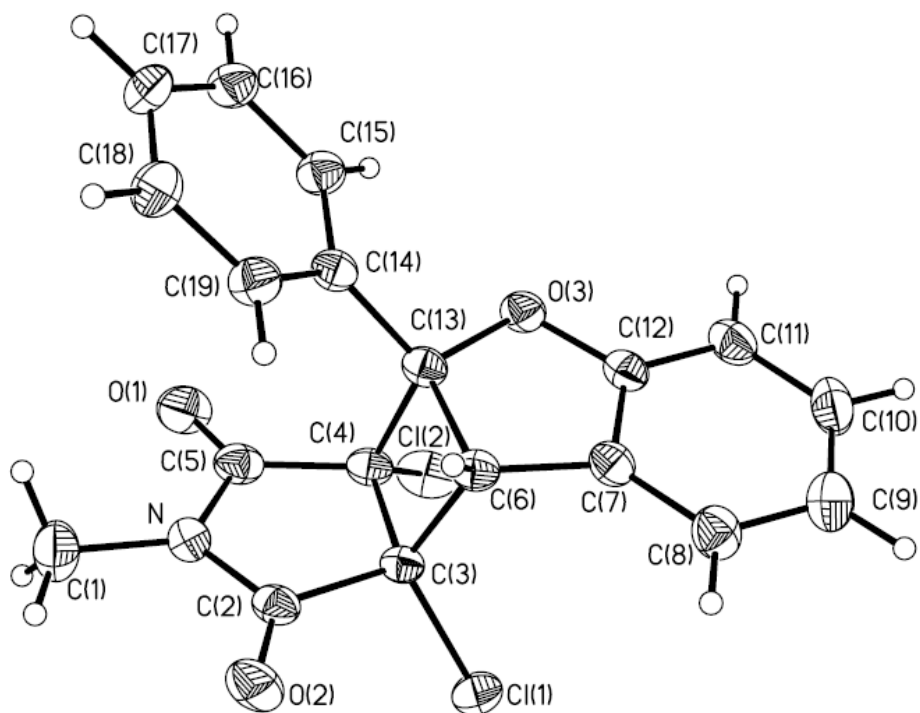


5d

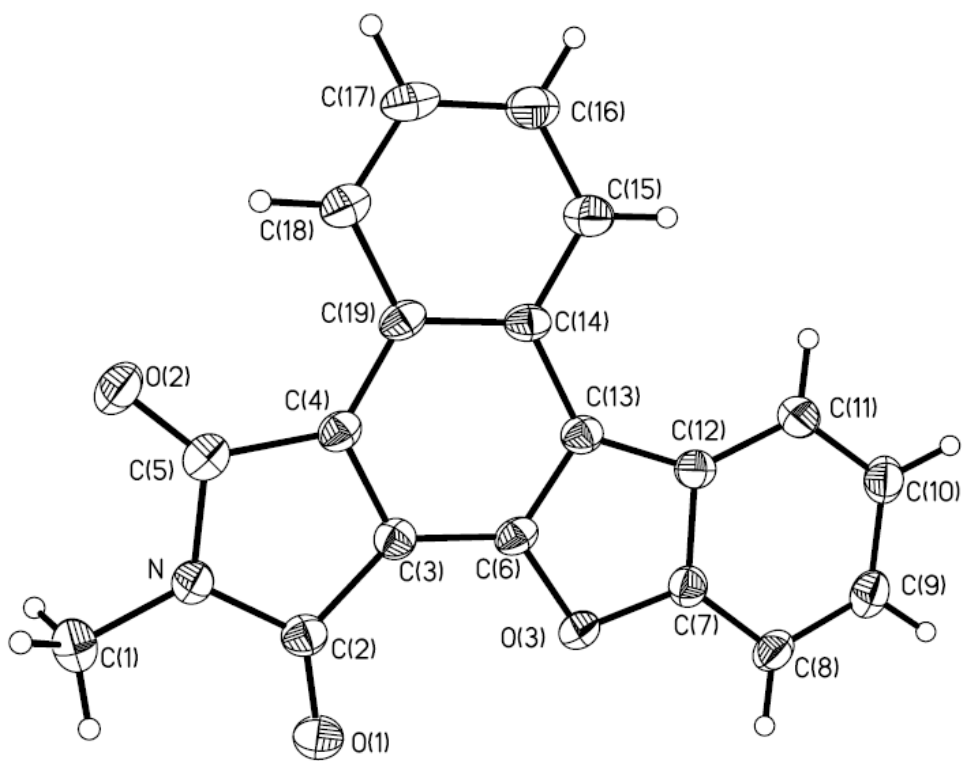








ORTEP drawing of **4a**



ORTEP drawing of **5b**