Electronic Supplementary Information:

Aldolase Peptide Dendrimers from Combinatorial Libraries

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HPLC-integration for sequence determination of single peptide dendrimers on resin beads.

HPLC-intergration data for amino acid analysis of single beads of dendrimers after total hydrolysis with 6M HCl solution at 110°C for 22 h and derivitazation with phenyl isothiocyanate (PITC). The sequence of dendrimers is deduced from the HPLC-peak integration of each amino acid PITC-derivative relative to the reference integration of this derivative. The Dap (2,3-diaminopropanoic acid) branching unit co-elutes with phenylalanine (relative integration =91.0). The value is substracted before calculating integration for this amino acid.

HPLC-trace for analysis of the bead of catalytic dendrimer L2D1 identified as catalytic.



RT	RT (STD)	PW(50%)	Area	Height	Amount	Peak Name
min	min	min	mAU*min	mAU	lomg	
2.01	2.00	0.067	0.04	0.55	3.0	Asp
2.21	2.20	0.072	0.08	1.07	6.3	Glu
3.41	3.40	0.098	0.12	0.96	6.7	Ser
3.65	3.64	0.097	0.19	1.64	10.9	Gly
5.22	5.22	0.102	3.89	29.49	181.3	Pro
7.11	7.09	0.102	0.83	7.25	51.1	Tyr
8.06	8.10	0.187	0.10	0.52	3.8	Val
10.11	10.10	0.089	0.13	1.47	10.1	lle
10.34	10.33	0.099	0.73	6.92	50.7	Leu
10.70	10.69	0.152	4.61	27.26	540.7	NH3
11.21	11.23	0.115	2.52	16.45	124.7	Phe
12.37	12.37	0.105	5.05	40.34	165.5	Lvs
Total:			18.292	133.923	1154.80	

HPLC-trace for analysis of the bead of catalytic dendrimer L2K4 identified as catalytic.



RT	RT (STD)	PW(50%)	Area	Height Amoun		Peak Name
min	min	min	mAU*min	mAU	pmol	
2.03	2.03	0.065	0.18	2.35	6.5	Asp
2.23	2.23	0.074	0.30	3.73	8.7	Glu
3.42	3.41	0.084	0.25	2.70	8.9	Ser
3.66	3.66	0.090	1.30	13.00	50.7	Gly
3.84	3.83	0.092	1.51	14.28	98.6	His
4.26	4.26	0.081	0.09	1.10	3.9	Arg
4.66	4.66	0.090	1.32	13.31	93.7	Thr
4.84	4.84	n.a.	0.14	1.57	7.1	Ala
5.26	5.26	0.098	1.97	18.92	134.8	Pro
7.16	7.15	0.098	1.74	16.25	55.0	Tyr
8.16	8.16	0.099	0.90	6.81	31.8	Val
8.55	8.55	0.099	0.45	4.03	10.1	Met
10.17	10.17	0.096	0.10	0.98	7.0	lle
10.63	10.63	0.138	6.23	38.84	728.7	NH3
11.27	11.31	0.119	4.08	28.06	203.8	Phe
12.46	12.46	0.104	4.32	36.59	150.2	Lys
Total:			24.880	202.502	1599.40	



RT	RT (STD)	PW(50%)	Area	Height	Amount	Peak Name
min	min	min	mAU*min	mAU	pmol	
2.00	2.00	0.071	0.11	1.49	8.0	Asp
2.20	2.20	0.076	0.32	3.76	22.1	Glu
3.40	3.40	0.089	0.91	9.05	62.6	Ser
3.64	3.64	0.084	0.16	1.78	11.8	Gly
4.22	4.22	0.091	1.17	10.93	75.3	Arg
4.65	4.63	0.111	0.06	0.54	4.3	Thr
4.82	4.81	0.101	0.20	1.76	12.5	Ala
5.22	5.22	0.104	2.57	18.30	112.5	Pro
7.10	7.09	0.105	1.06	8.80	62.1	Tyr
8.11	8.10	0.108	1.02	7.79	56.4	Val
8.50	8.49	0.103	0.14	1.22	9.6	Met
10.11	10.10	0.085	0.06	0.74	5.1	lle
10.34	10.33	0.102	0.70	6.26	45.9	Leu
10.70	10.69	0.154	5.12	29.72	589.3	NH3
11.21	11.23	0.120	3.03	18.98	143.9	Phe
12.38	12.37	0.107	4.11	32.25	132.4	Lys
Total:			20.723	153.370	1353.65	

Kinetic measurements with 6. The kinetic measurements were carried out by using a Cytofluor II plate reader from Perseptive Biosystems (λ_{ex} = 460 nm, λ_{em} = 530 nm) at 25 °C. 200 µL assays were followed in individual wells of round-bottom polypropylene 96-well-plates (Costar). Kinetic experiments were followed for 6 hrs. The dendrimers were stored at -20°C in 1 mM stock solution in acetonitrile/water (1:1, v/v). Dendrimer stock solutions were freshly diluted to 80.0 µM solution in 20 mM aq. bicine buffer pH 8.5. The 20 mM bicine buffer, pH 8.5 was prepared using MilliQ deionized water. Initial reaction rates were calculated from the steepest part observed during the first 5000 sec of each curve. In a typical experiment, 140 µL of ag. bicine pH 8.5 (20 mM) were first added in a well, then 50 µL of a dendrimer solution (80.0 µM in aq. bicine pH 8.5, concentration in the well: 20 µM), and last 10 µL of substrate solution (1.0 mM in aq. bicine buffer pH=8.5/acetonitrile (10:1,: v/v), concentration in the well: 50 µM). Fluorescence data were converted to product concentration by means of a calibration curve with pure product. The initial reaction rates observed under these conditions is the apparent rate V_{app}. V_{uncat} is the initial rate observed under the same conditions without dendrimer. Michaelis-Menten parameters were obtained from the linear double reciprocal plot of $1/V_{net}$ ($V_{net} = V_{app}-V_{uncat}$) vs. 1/[S] measured similarly with 20 µM dendrimer (Vapp) or no dendrimer (Vuncat) and 20, 50, 100, 200, 500 µM substrate. The catalytic rate constant k_{cat} for the hydrolysis is given by $k_{cat} = V_{max}/[D]$, where [D] indicates the concentration of dendrimers. The second order rate constants k_2 were calculated from linear regression of the experimentally measured pseudo first order rate constants k' as a function of proline or lysine concentration.



Catalyst		$k_{\text{cat}} (s^{-1})$	$K_{\rm M}(\mu{ m M})$	$k_{\rm cat}/k_{\rm uncat}$	$k_{\rm cat}/K_{\rm M}~({\rm s}^{-1}~\mu{\rm M}^{-1})$	$(k_{\rm cat}/K_{\rm M})/k_2$
PK.PK.YL.IG	L2D1	1.2×10^{-4}	230	56	5.2×10^{-7}	5'900
EK.SKYA.FV	L2D5	2.0×10^{-4}	460	98	4.3×10^{-7}	4'300 ^[b]
SK.SK.YG.FG	L2D6	1.1×10^{-4}	140	51	7.5×10^{-7}	7'400 ^[b]
PK.ER.βAG.FV	L2D7	2.0×10^{-4}	340	92	3.6×10^{-7}	6'500
PK.TH.TG.FV	L2K4	2.4×10^{-4}	380	120	6.3×10^{-7}	7'200
PK.SR. βAV.YL	L2K7	2.0×10^{-4}	280	96	7.3×10^{-7}	8'300
EK.ED.IG.YA	L2K8	3.0×10^{-4}	820	150	3.7×10^{-7}	3'700 ^[b]
PT.PT	R1G1	9.3×10^{-5}	210	45	4.4×10^{-7}	5'000
PT.PT.PT	R1G2	9.4×10^{-5}	160	45	6.0×10^{-7}	6'800
PT.PT.PT.PT	R1G3	1.3×10^{-4}	180	61	7.1×10^{-7}	8'100
PK.PK	R2G1	6.3×10^{-5}	530	30	1.2×10^{-7}	1'400
PK.PK.PK	R2G2	1.7×10^{-4}	470	82	3.6×10^{-7}	4'100
PK.PK.PK.PK	R2G3	1.9×10^{-4}	300	92	6.4×10^{-7}	7'300
Antibody 38C2		2.6×10^{-3}	80	1800	3.2×10^{-5}	320'000 ^[b]

Kinetic parameters for reaction with coumarin probe 6 with peptide dendrimers from library L2, regular series R1 and R2 and aldolase antibody 38C2.[a]

[a] Conditions: 25–500 μ M ketone **6**, 20 μ M dendrimer, 20 mM aqueous bicine buffer pH 8.5, 25°C. The kinetic constants given are derived from the linear double-reciprocal plots of $1/V_{net}$ versus 1/[S]. $V_{net} = V_{app} - V_{uncat}$ with V_{app} being the apparent hydrolysis rate in the presence of dendrimer and V_{uncat} the hydrolysis rate in buffer alone ($k_{uncat} = 2.1 \times 10^{-6} \text{ s}^{-1}$). $k_2 = 8.8 \times 10^{-11} \mu \text{M}^{-1} \text{ s}^{-1}$ is the catalytic rate constant for β -elimination catalyzed by L-proline under the same conditions. [b] $k_2 = 1.01 \times 10^{-10} \mu \text{M}^{-1} \text{ s}^{-1}$ is the catalytic rate constant for β -elimination catalyzed by L-lysine under the same conditions.

6-(4-Nitro-phenyl)-hexane-2,4-dione (5)





Succinic acid mono-(2-{ethyl-[4-(4-nitro-phenylazo)-phenyl]-amino}-ethyl) ester (8)





0 8

C₁₂H₁₀O₃ Mol. Wt.: 202.206 C, 71.28; H, 4.98; O, 23.74





C₁₂H₁₂O₅ Mol. Wt.: 236.22068 C, 61.01; H, 5.12; O, 33.87





C₁₈H₂₆O₅Si Mol. Wt.: 350.48154 C, 61.68; H, 7.48; O, 22.82; Si, 8.01





C₁₈H₂₄O₅Si Mol. Wt.: 348.46566 C, 62.04; H, 6.94; O, 22.96; Si, 8.06





C₁₂H₁₀O₅ Mol. Wt.: 234.2048 C, 61.54; H, 4.30; O, 34.16























Dendrimer R1G1



Dendrimer R1G2





Dendrimer R2G1





Dendrimer R2G2



Dendrimer R2G3

