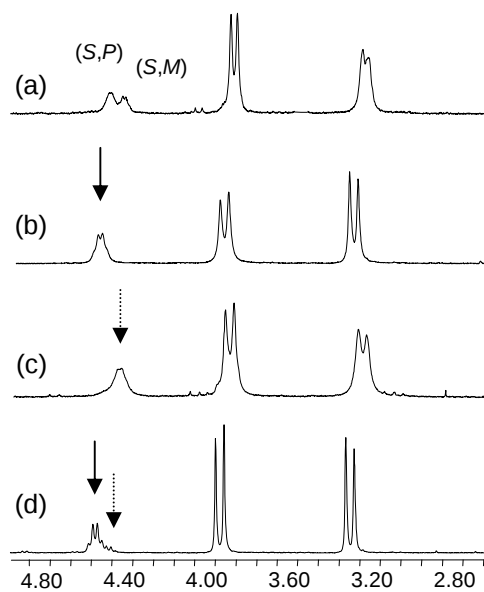
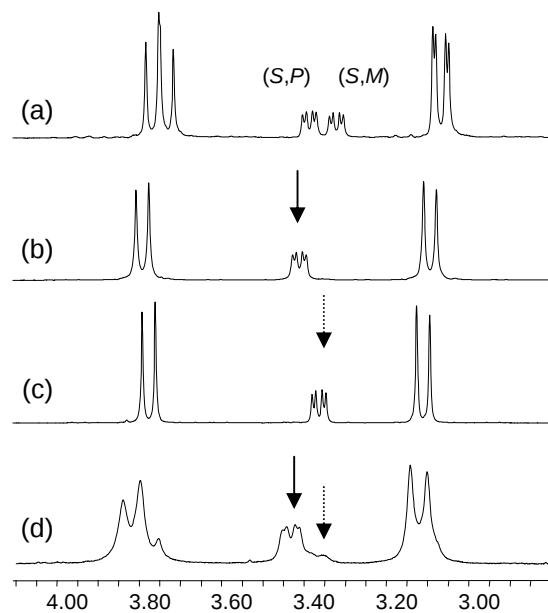


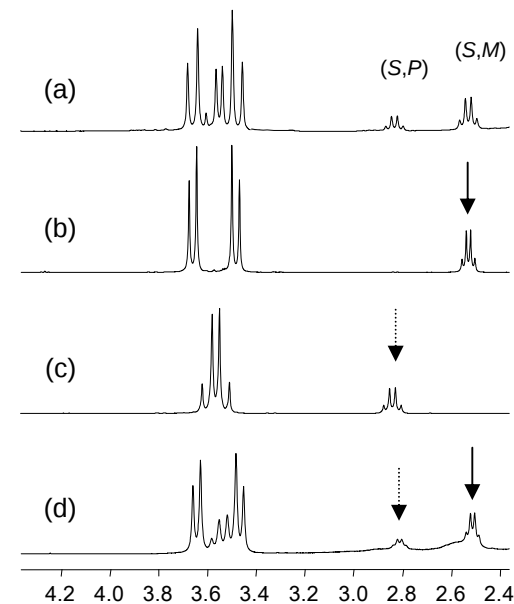
Ratios: (a) right after the synthesis; (b) first eluted (major) diastereomer; (c) second eluted (minor) diastereomer; (d) after 20 h at 80 °C (benzene)



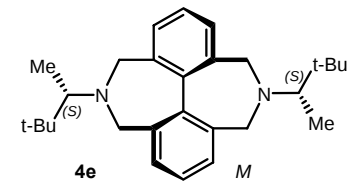
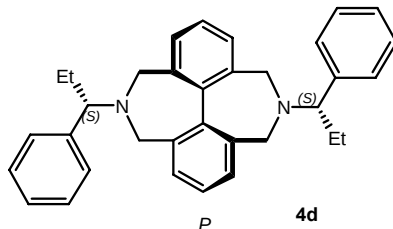
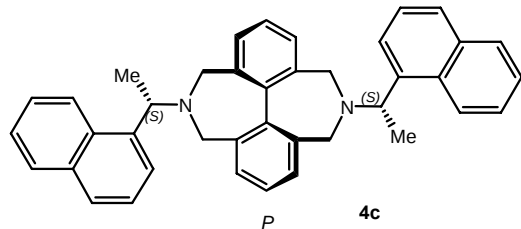
d.r. *(S,P)*:*(S,M)* : (a) 1.4 : 1; (b) > 49 : 1;
(c) > 1 : 49; (d) 4.3 : 1



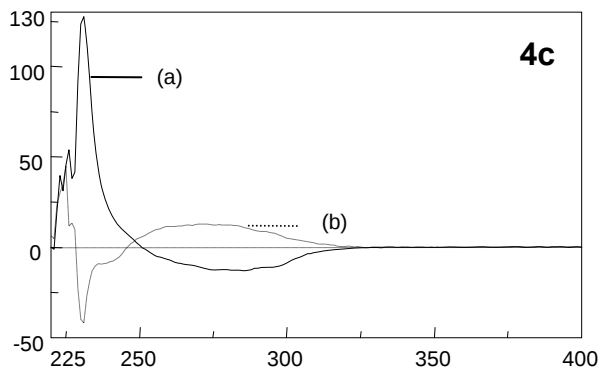
d.r. *(S,P)*:*(S,M)* : (a) 1.1 : 1; (b) > 49 : 1;
(c) > 1 : 49; (d) 4.2 : 1



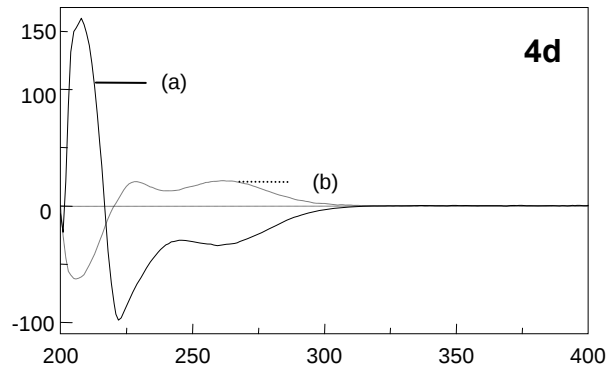
d.r. *(S,M)*:*(S,P)* : (a) 2.4 : 1; (b) > 49 : 1;
(c) > 1 : 49; (d) 2.5 : 1



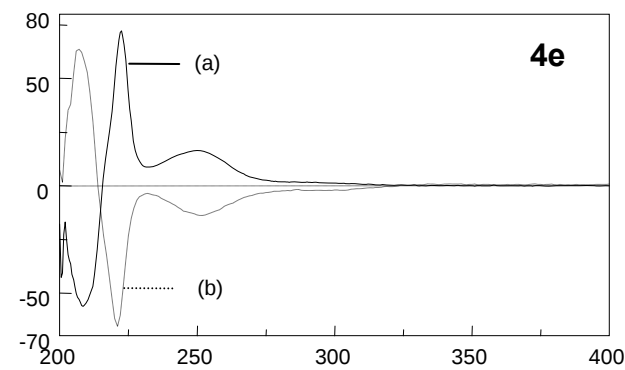
(a, plain) first eluted (major) diastereomer and (b, dashed) second eluted (minor) diastereomer



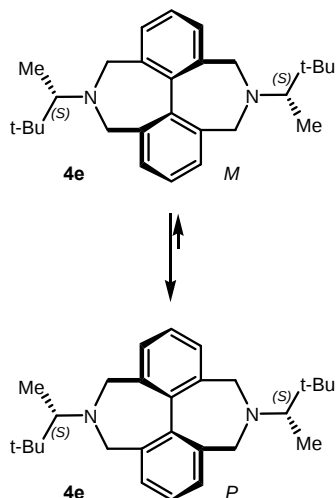
(c 1.10⁻⁵ M, CH₂Cl₂)



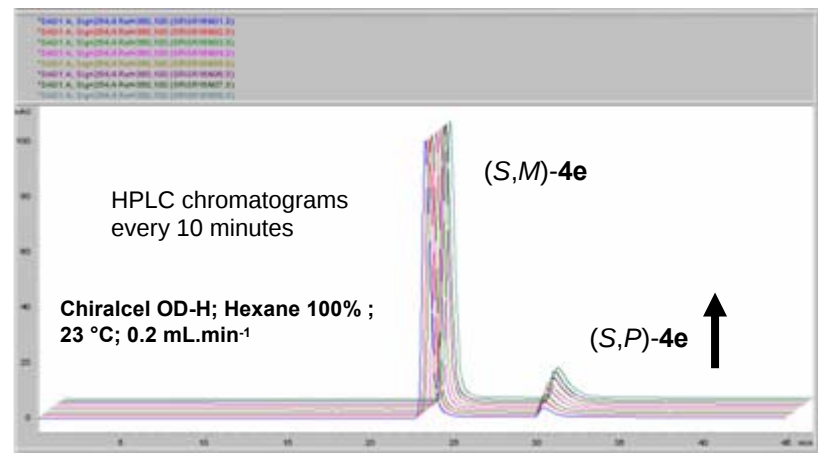
(c 1.10⁻⁵ M, MeOH)



(c 1.10⁻⁵ M, MeOH)



■ Experimental monitoring (HPLC)
of the isomerisation (60 °C)



■ First-Order kinetics : $\ln ([a]/[a_0]) = -k t$

[a] is the concentration of the major diastereomer
k is the kinetic constant and easily determined by the slope of the equation

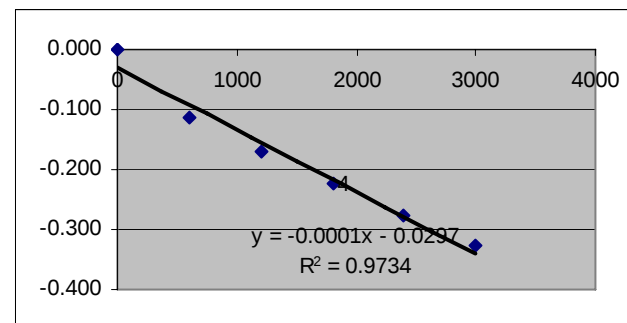
Température:

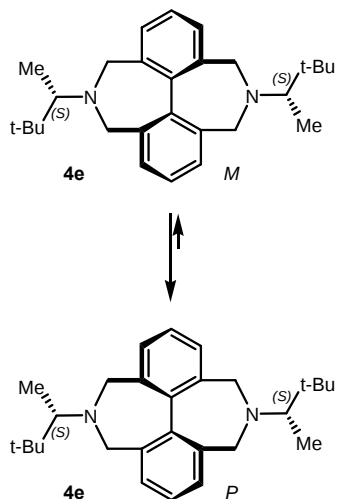
60 °C
333.15 K

slope = -1.038E-04
DG = 107.308 kJ/mol
t_{1/2} = 6.68E+03 s
R-square value = 0.97339

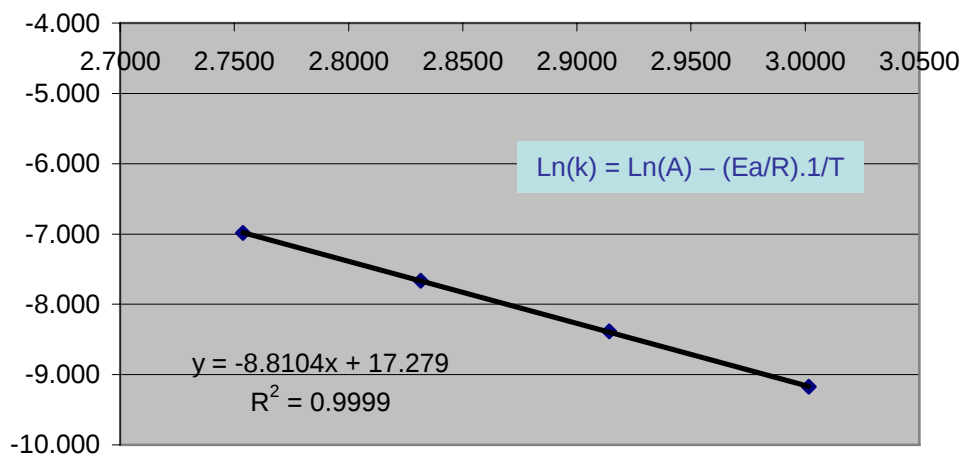
1.86 h

Entrée	t [min]	t [s]	Maj	Min	d.e.	[a]/[a ₀]	ln ([a]/[a ₀])
0	0	0	100.00	0.00	100.00	1.00	0.000
1	10	600	94.60	5.40	89.20	0.89	-0.114
2	20	1200	92.16	7.84	84.32	0.84	-0.171
3	30	1800	90.02	9.98	80.04	0.80	-0.223
4	40	2400	87.93	12.07	75.86	0.76	-0.276
5	50	3000	86.01	13.99	72.02	0.72	-0.328





	T [K]	1000/T	k[s ⁻¹]	ln(k)
60 °C	333	3.002	1.04E-04	-9.173
70 °C	343	2.914	2.28E-04	-8.386
80 °C	353	2.832	4.68E-04	-7.667
90 °C	363	2.754	9.25E-04	-6.986

**Arrhenius Equation:**

$$k = A \cdot \exp(-E_a/RT)$$

$$\ln(k) = \ln(A) - (E_a/R) \cdot 1/T$$

Eyring Equation:

$$\Delta H^\ddagger = E_a - RT$$

$$\Delta S^\ddagger = R[\ln(h \cdot A/k_B \cdot T) - 1]$$

$$\Delta G^\ddagger = \Delta H^\ddagger - T\Delta S^\ddagger$$

ln(k)= f(1000/T)

$$\text{slope} = -8.810 \quad (-E_a/R)$$

$$\text{intercept} = 17.279 \quad (\ln(A))$$

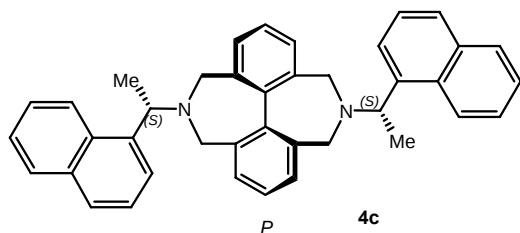
$$T = 20 \text{ }^\circ\text{C}$$

$$E_a = 73.479 \text{ kJ/mol}$$

$$A = 3.19E+07 \text{ s}^{-1}$$

$$R^2 = 0.999$$

$\Delta H^\ddagger = E_a - RT$	16.994	kcal/mol
$\Delta S^\ddagger = R[\ln(h \cdot A/k_B \cdot T) - 1]$	-26.178	cal/(mol.K)
$\Delta G^\ddagger$	24.668	kcal/mol



Arrhenius Equation:

$$k = A \cdot \exp(-E_a/RT)$$

$$\ln(k) = \ln(A) - (E_a/R) \cdot 1/T$$

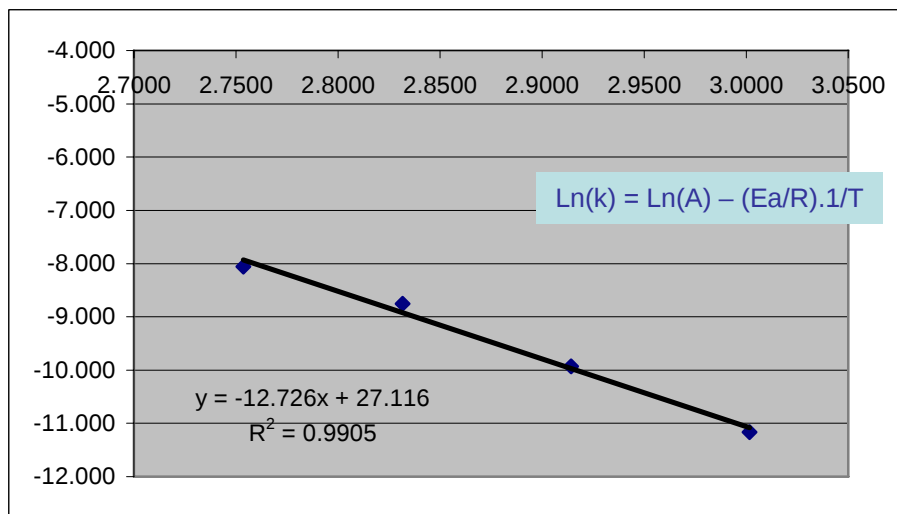
Eyring Equation:

$$\Delta H^\ddagger = E_a - RT$$

$$\Delta S^\ddagger = R[\ln(h \cdot A/k_B \cdot T) - 1]$$

$$\Delta G^\ddagger = \Delta H^\ddagger - T\Delta S^\ddagger$$

T [K]	1000/T	k [s ⁻¹]	ln(k)
333	3.002	1.42E-05	-11.164
343	2.914	4.87E-05	-9.929
353	2.832	0.000158	-8.754
363	2.754	0.000316	-8.059



$\ln(k) = f(1000/T)$

slope=	-12.726	(- E_a/R)
intercept=	27.116	($\ln(A)$)

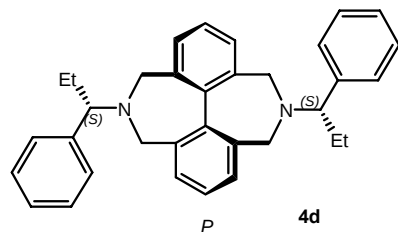
T 20 °C

E_a 106.138 kJ/mol

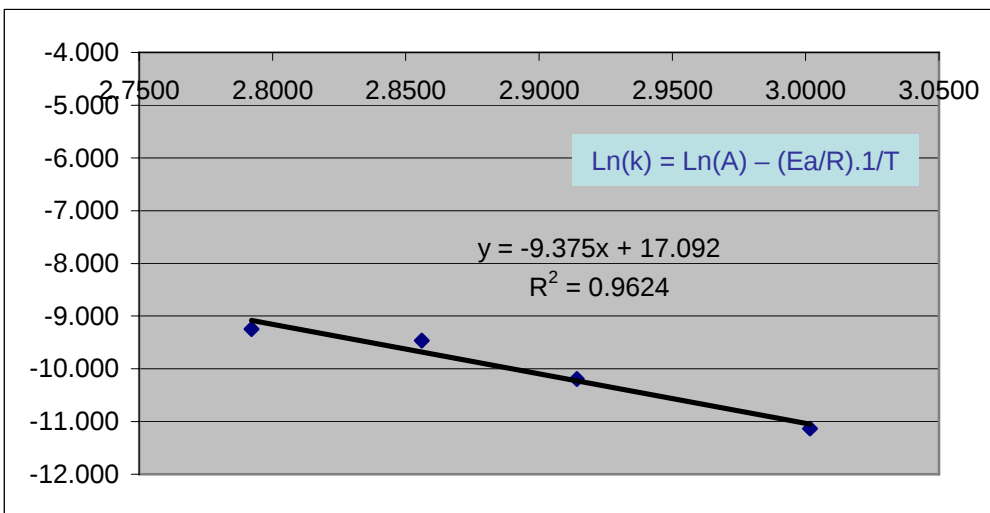
A 5.97E+11 s⁻¹

$R^2 = 0.991$

$\Delta H^\ddagger = E_a - RT$	24.807	kcal/mol
$\Delta S^\ddagger = R[\ln(h \cdot A/k_B \cdot T) - 1]$	-6.614	cal/(mol.K)
$\Delta G^\ddagger$	26.746	kcal/mol



T [K]	1000/T	k [s ⁻¹]	ln(k)
333	3.002	1.46E-05	-11.133
343	2.914	3.74E-05	-10.194
350	2.856	7.74E-05	-9.466
358	2.792	9.63E-05	-9.248

**Arrhenius Equation:**

$$k = A \cdot \exp(-E_a/RT)$$

$$\text{Ln}(k) = \text{Ln}(A) - (E_a/R) \cdot 1/T$$

Eyring Equation:

$$\Delta H^\ddagger = E_a - RT$$

$$\Delta S^\ddagger = R[\text{Ln}(h \cdot A / k_B \cdot T) - 1]$$

$$\Delta G^\ddagger = \Delta H^\ddagger - T\Delta S^\ddagger$$

ln(k)= f(1000/T)

$$\text{slope} = -9.374959$$

$$(- E_a/R)$$

$$\text{intercept} = 17.09232$$

$$(\text{Ln}(A))$$

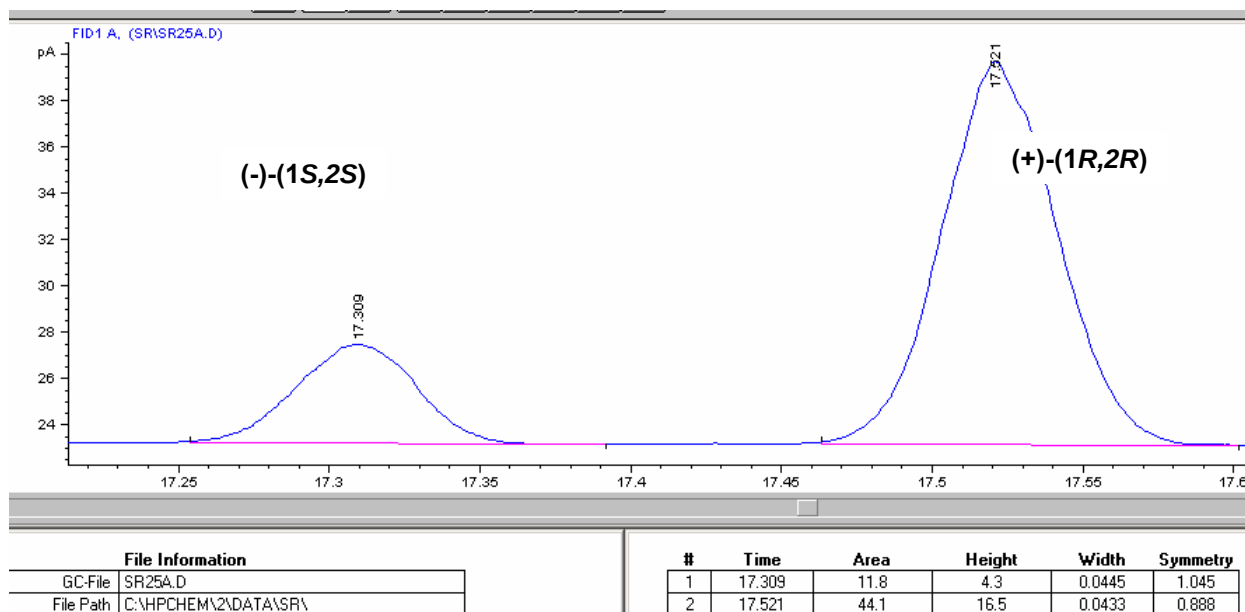
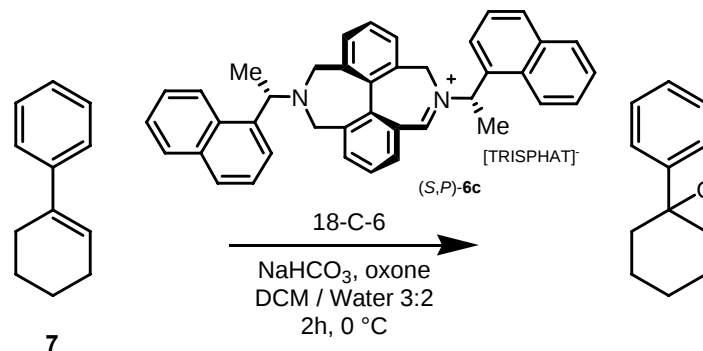
$$T \quad 20 \text{ } ^\circ\text{C}$$

$$E_a \quad 78.187 \text{ kJ/mol}$$

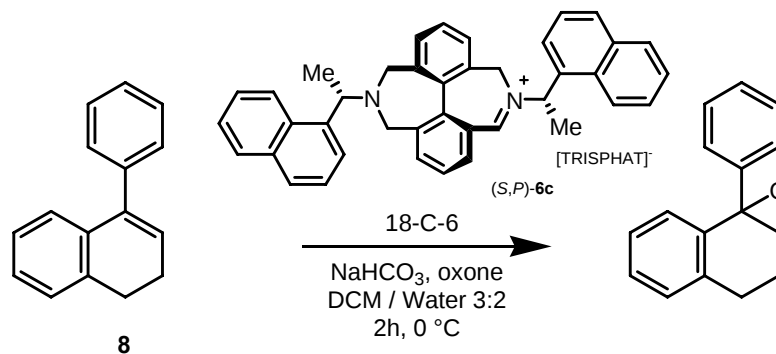
$$R^2 = 0.962$$

$$A \quad 2.64\text{E}+07 \text{ s}^{-1}$$

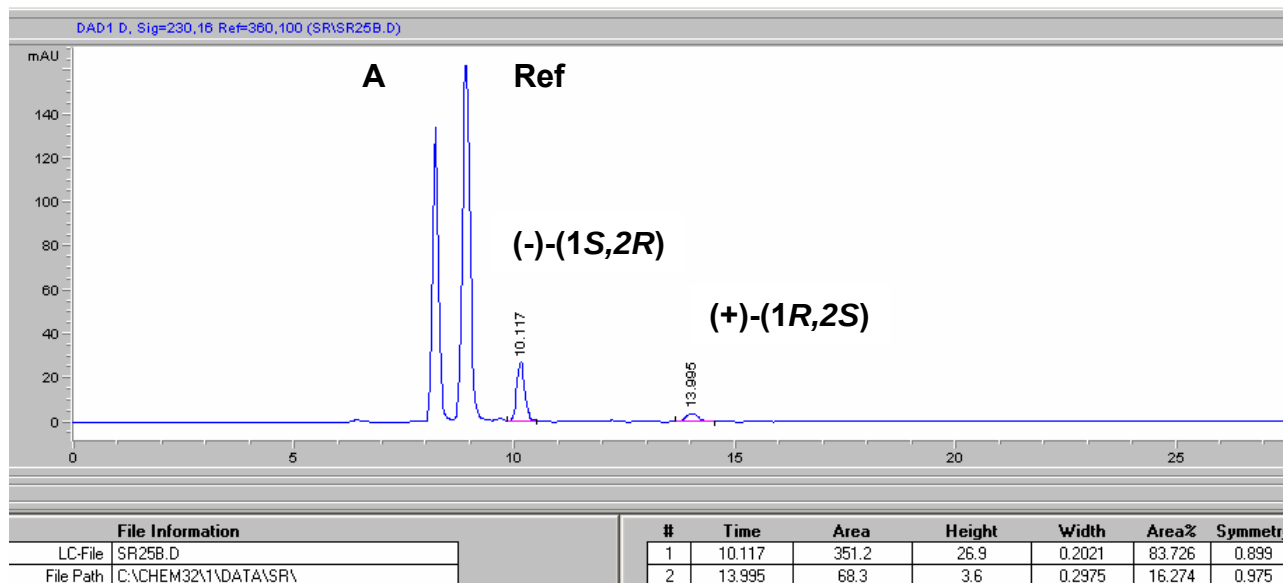
$\Delta H^\ddagger = E_a - RT$	18.120	kcal/mol
$\Delta S^\ddagger = R[\text{Ln}(h \cdot A / k_B \cdot T) - 1]$	-26.550	cal/(mol.K)
$\Delta G^\ddagger$	25.903	kcal/mol



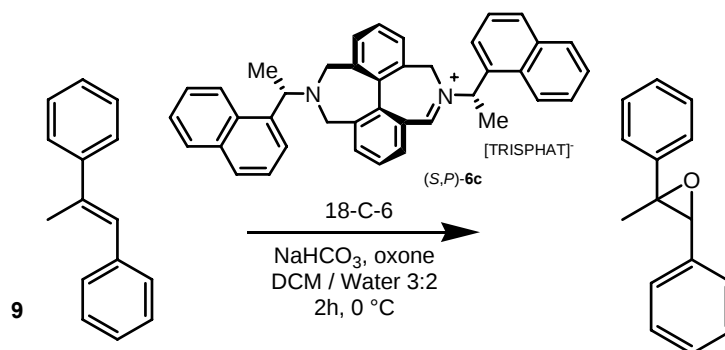
Determined by CSP-GC (Chiraldex Hydrodex β -3P), T_{inj} 250 °C, $P = 0.842$ bar;
 Conditions: 80 °C, 5 min, then progression to 180 °C in 20 min, then 180 °C for 5 min.



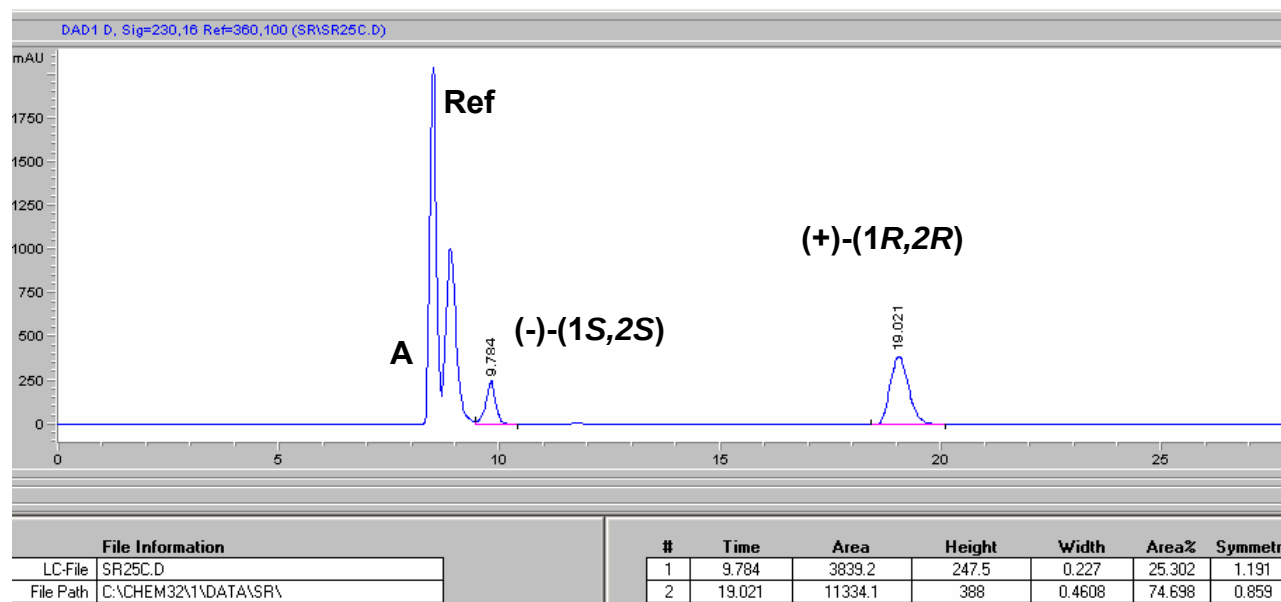
A: Alkene, Ref: naphthalene



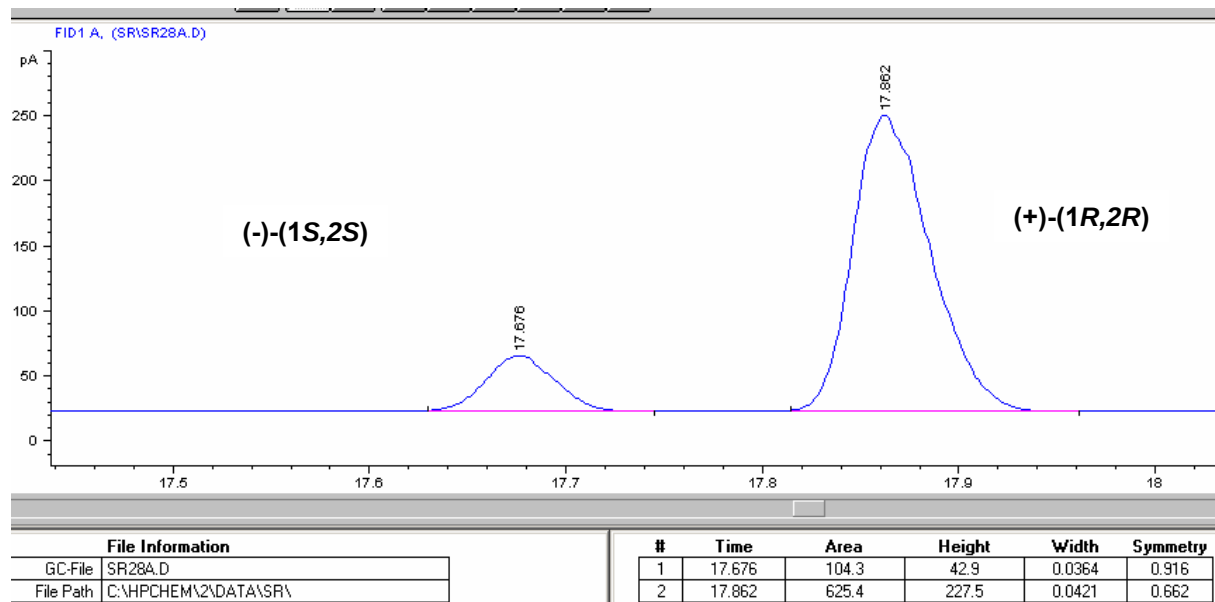
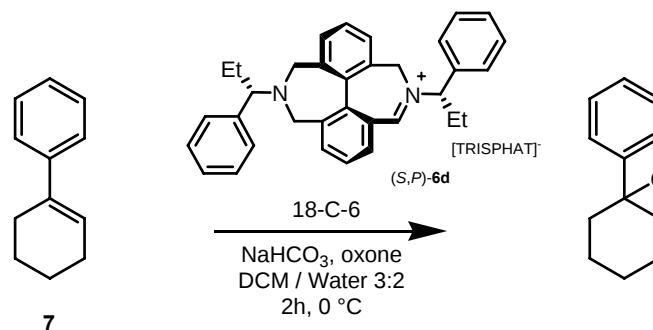
Determined by CSP-HPLC (Chiralcel OD-H); Hexane / iPrOH 95 : 5 ; 0.5 mL.min⁻¹



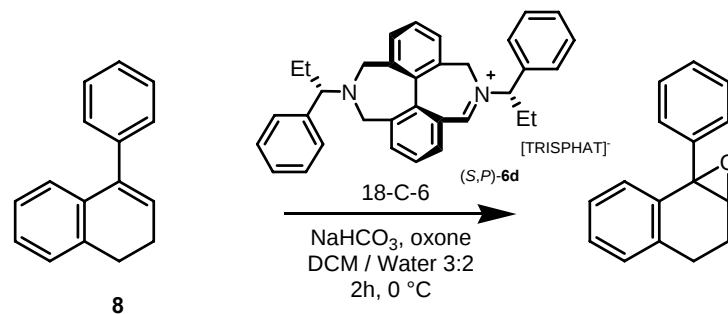
A: Alkene, Ref: naphthalene



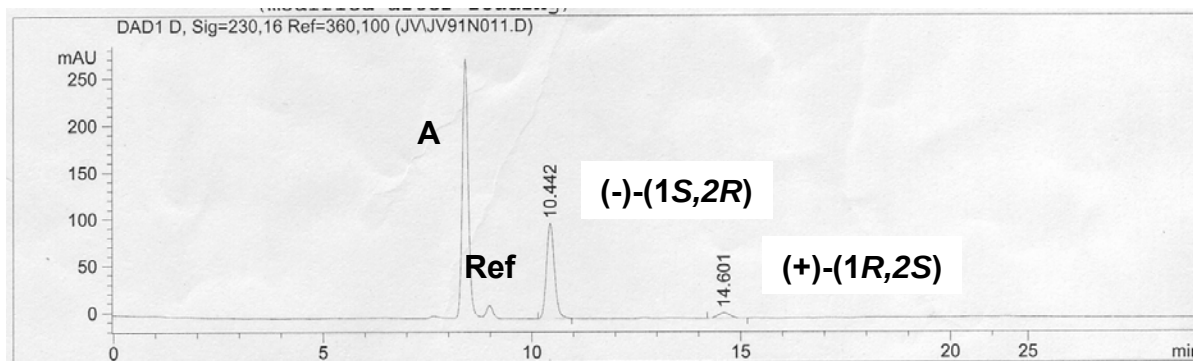
Determined by CSP-HPLC (Chiralcel OD-H); Hexane / iPrOH 95 : 5 ; 0.5 mL.min⁻¹



Determined by CSP-GC (Chiraldex Hydrodex β-3P), T_{inj} 250 °C, P = 0.842 bar;
 Conditions: 80 °C, 5 min, then progression to 180 °C in 20 min, then 180 °C for 5 min.



A: Alkene, Ref: naphthalene

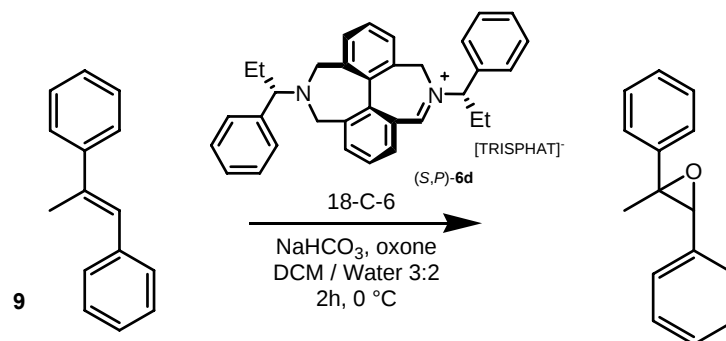


Signal 1: DAD1 D, Sig=230,16 Ref=360,100

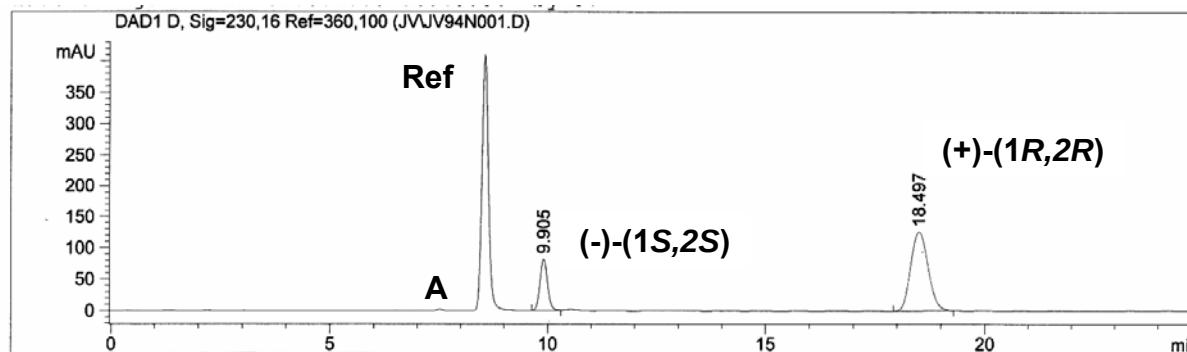
Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	10.442	BB	0.2070	1347.62756	101.43116	92.2391
2	14.601	PP	0.3161	113.38863	5.69863	7.7609

Totals : 1461.01620 107.12979

Determined by CSP-HPLC (Chiralcel OD-H); Hexane / iPrOH 95 : 5 ; $0.5 \text{ mL} \cdot \text{min}^{-1}$



A: Alkene, Ref: naphthalene

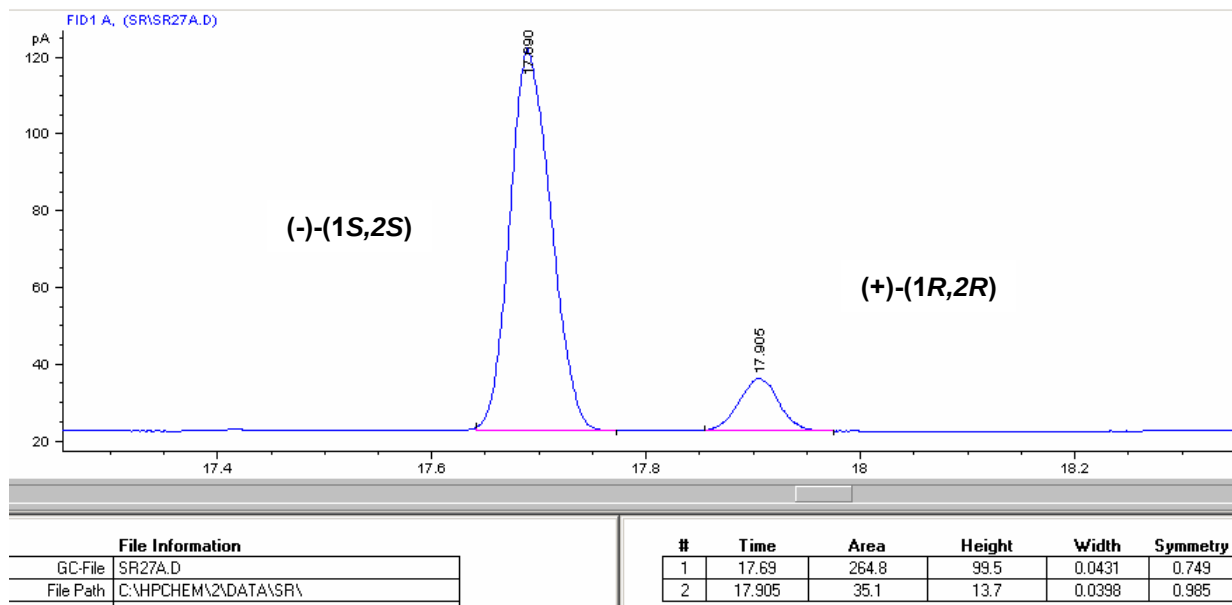
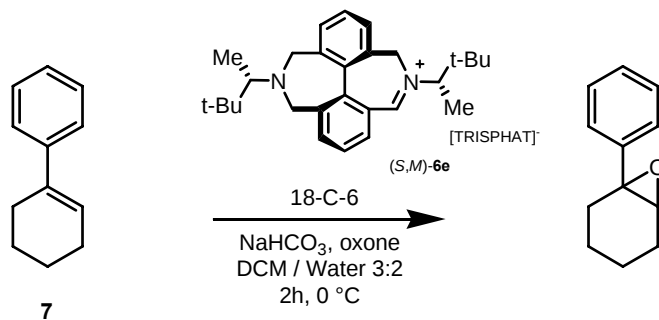


Signal 1: DAD1 D, Sig=230,16 Ref=360,100

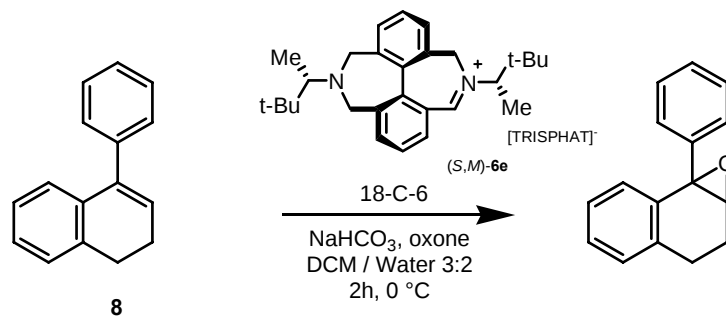
Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	9.905	BB	0.1896	994.71185	81.93610	22.3244
2	18.497	PB	0.4368	3460.99536	125.86764	77.6756

Totals : 4455.70721 207.80373

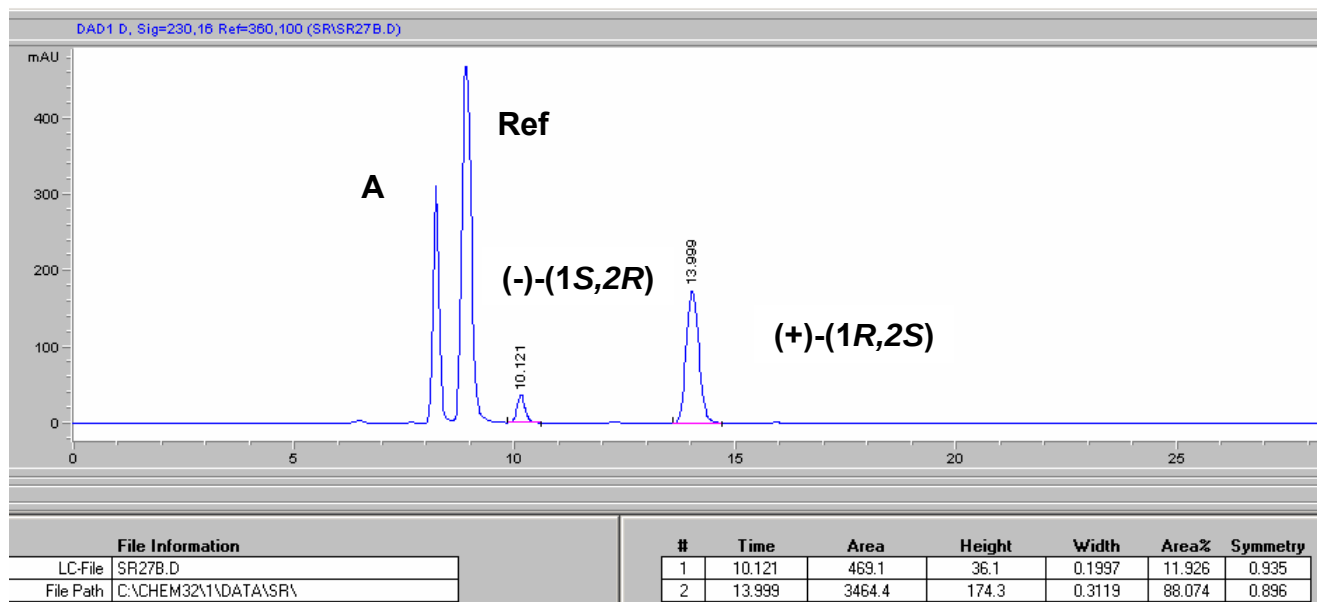
Determined by CSP-HPLC (Chiralcel OD-H); Hexane / iPrOH 95 : 5 ; 0.5 mL.min⁻¹



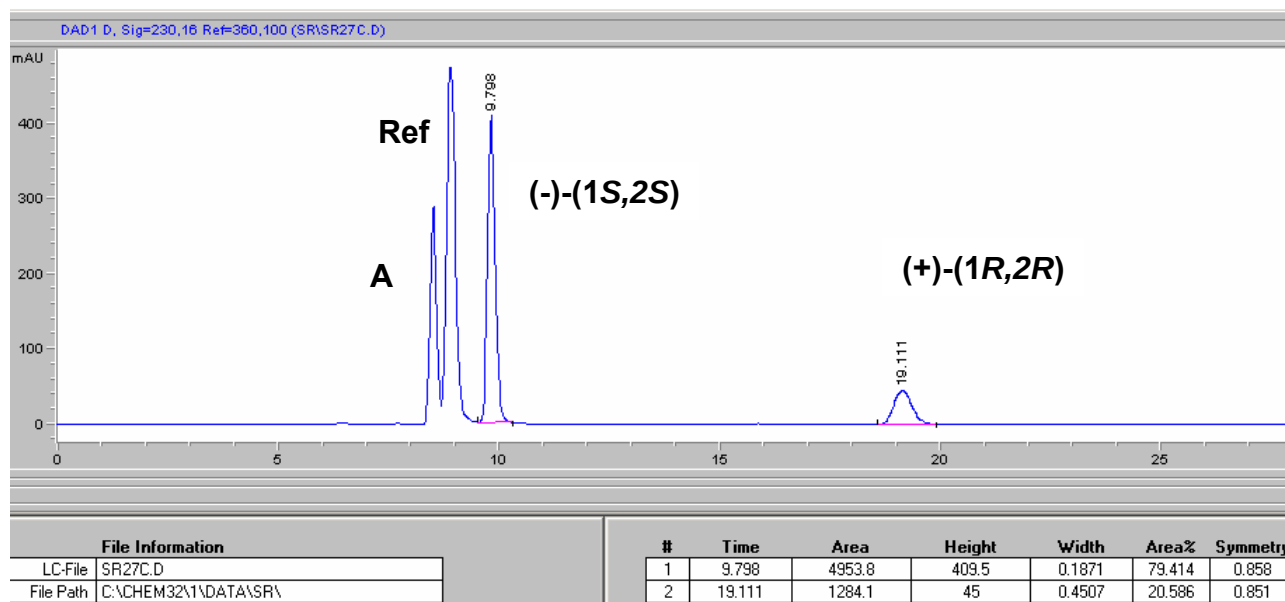
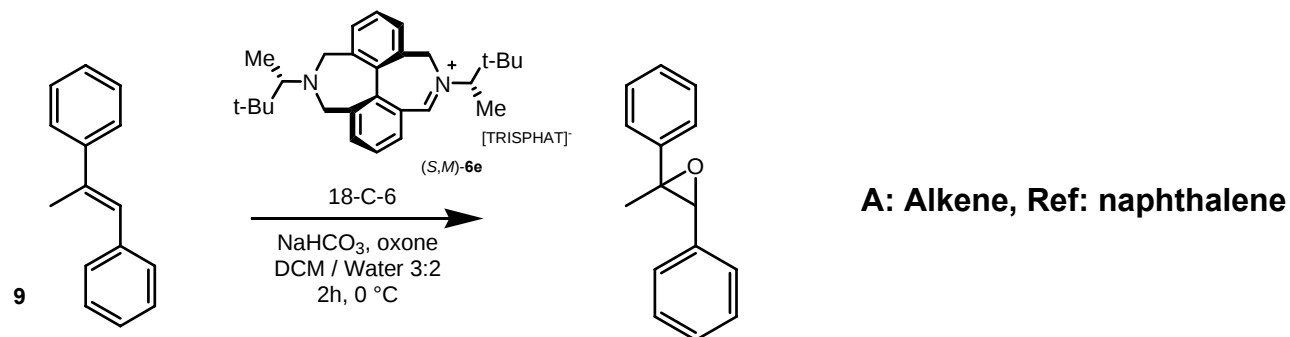
Determined by CSP-GC (Chiraldex Hydrodex β -3P), T_{inj} 250 °C, $P = 0.842$ bar;
 Conditions: 80 °C, 5 min, then progression to 180 °C in 20 min, then 180 °C for 5 min.

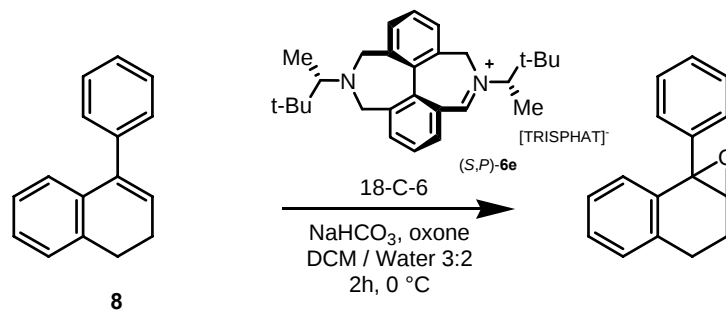


A: Alkene, Ref: naphthalene

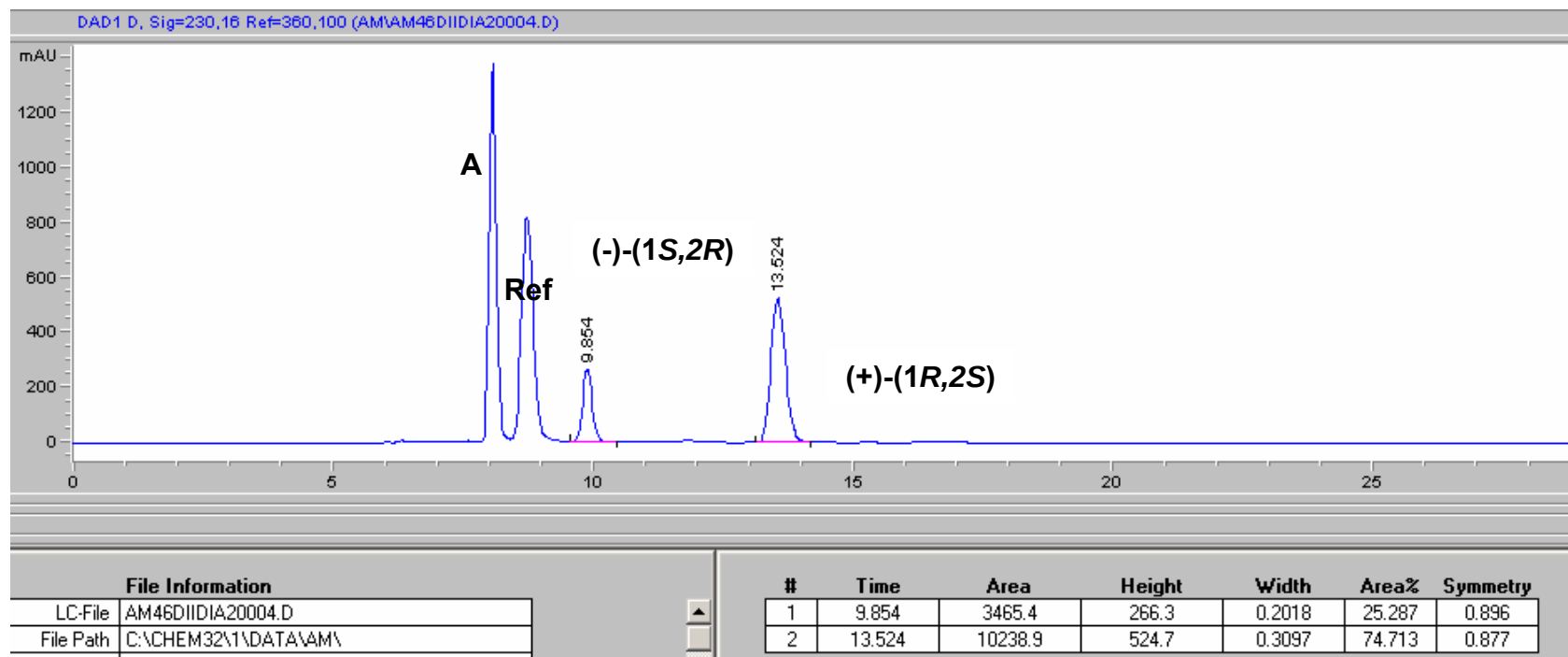


Determined by CSP-HPLC (Chiralcel OD-H); Hexane / iPrOH 95 : 5 ; 0.5 mL.min⁻¹

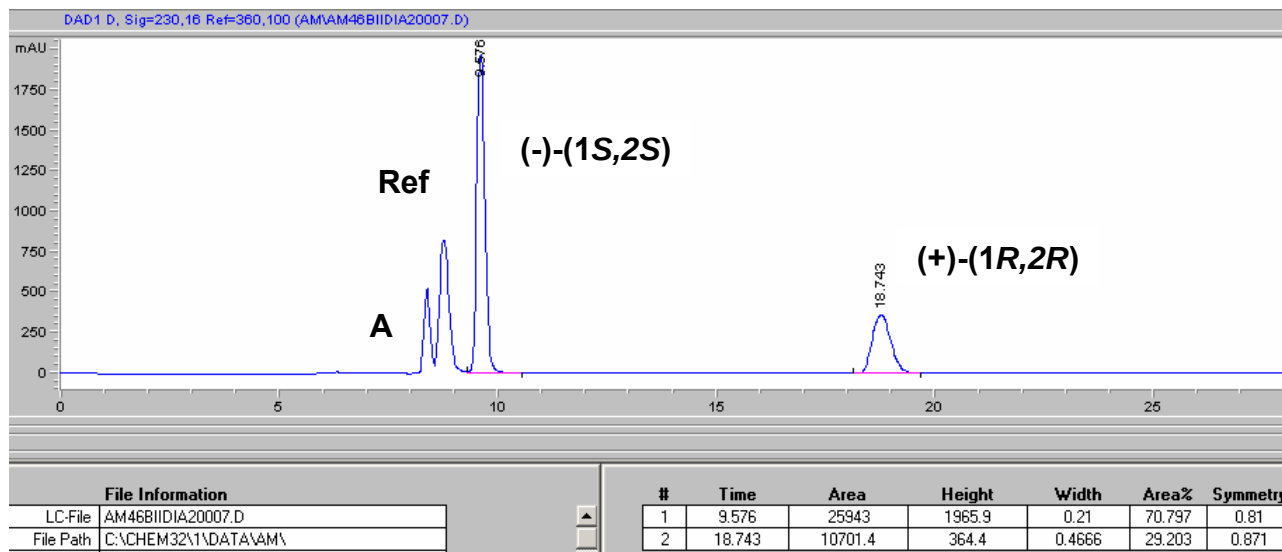
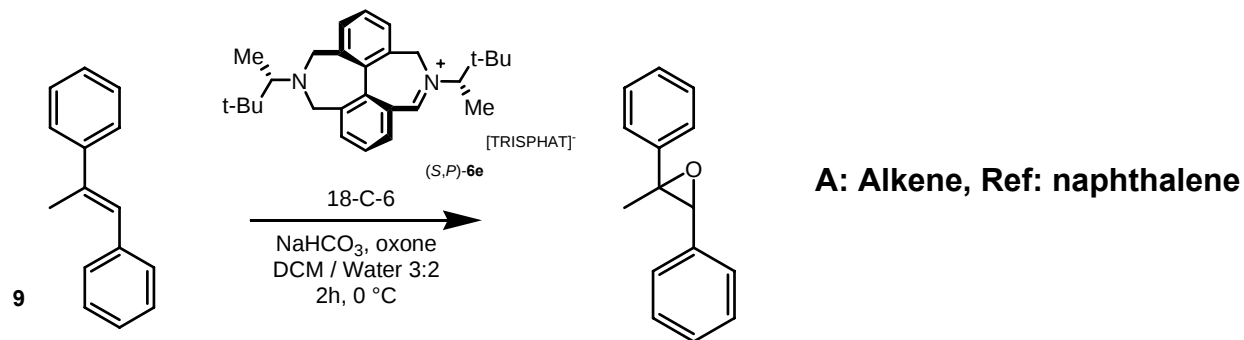
Determined by CSP-HPLC (Chiralcel OD-H); Hexane / iPrOH 95 : 5 ; 0.5 mL.min⁻¹



A: Alkene, Ref: naphthalene



Determined by CSP-HPLC (Chiralcel OD-H); Hexane / iPrOH 95 : 5 ; 0.5 mL.min⁻¹



Determined by CSP-HPLC (Chiralcel OD-H); Hexane / iPrOH 95 : 5 ; 0.5 mL.min⁻¹