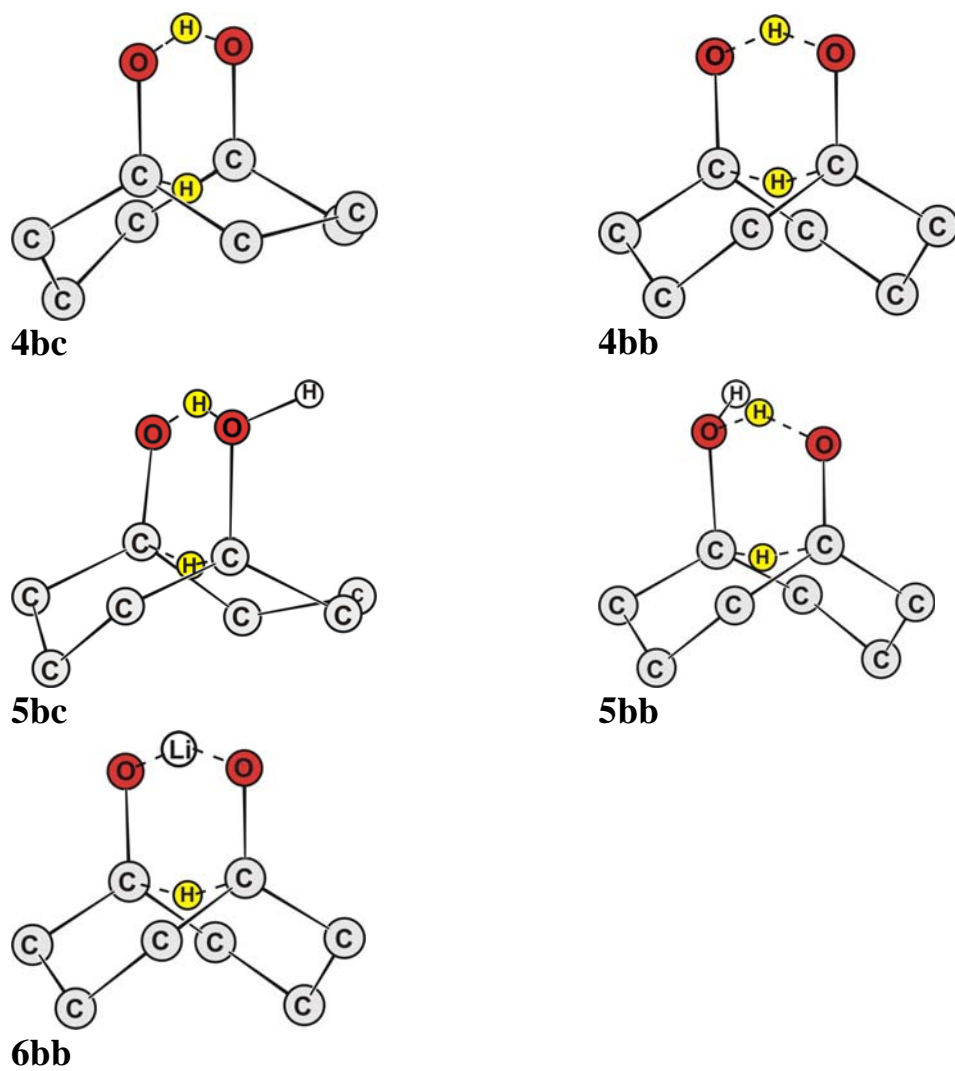


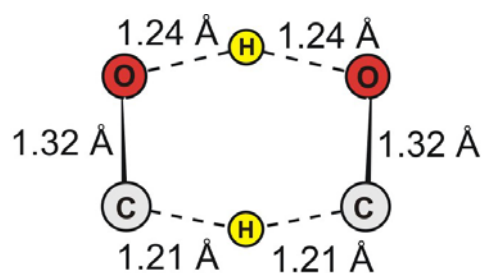
Electronic Supplementary Information

Transannular 1,5-Hydride Shift in 5-Hydroxycyclooctanone: An Experimental and Theoretical Investigation

Paul Rademacher\* and Parveen Choudhary Mohr

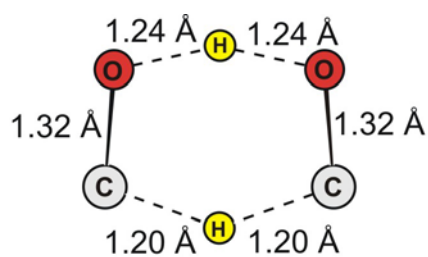


**Figure 1** Transition states for 1,5-hydride shift reaction



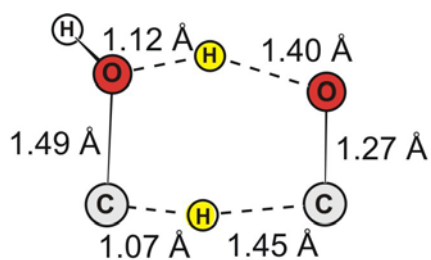
$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 158^\circ$$

4bc



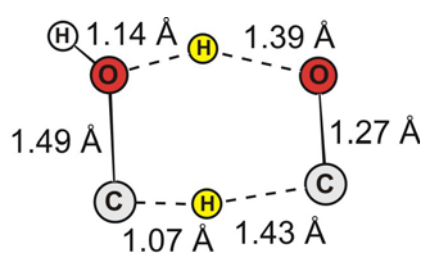
$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 161^\circ$$

4bb



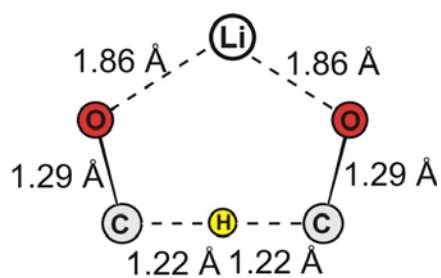
$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 164^\circ$$

5bc



$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 169^\circ$$

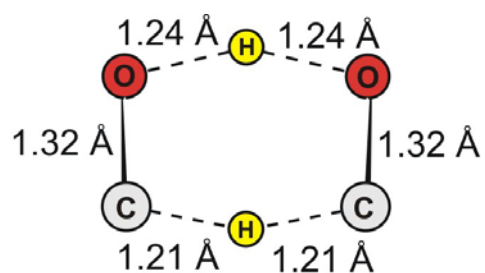
5bb



$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 177^\circ$$

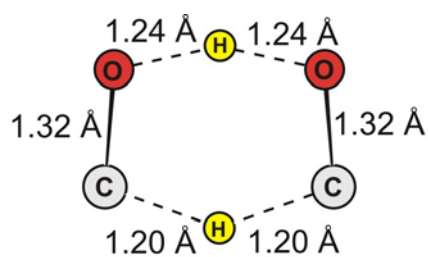
6bb

**Figure 2** Six-membered cyclic transition states in hydride shift reaction



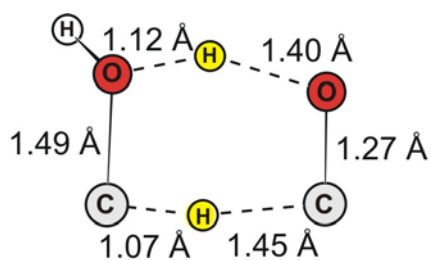
$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 158^\circ$$

4bc



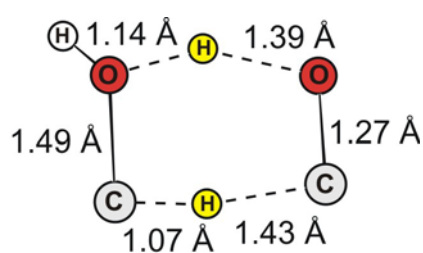
$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 161^\circ$$

4bb



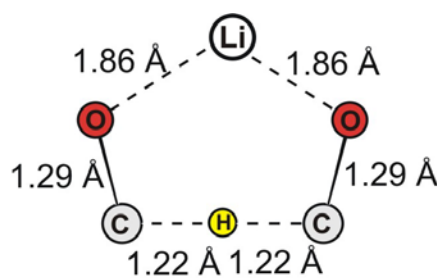
$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 164^\circ$$

5bc



$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 169^\circ$$

5bb



$$\angle \text{C}\cdots\text{H}\cdots\text{C} = 177^\circ$$

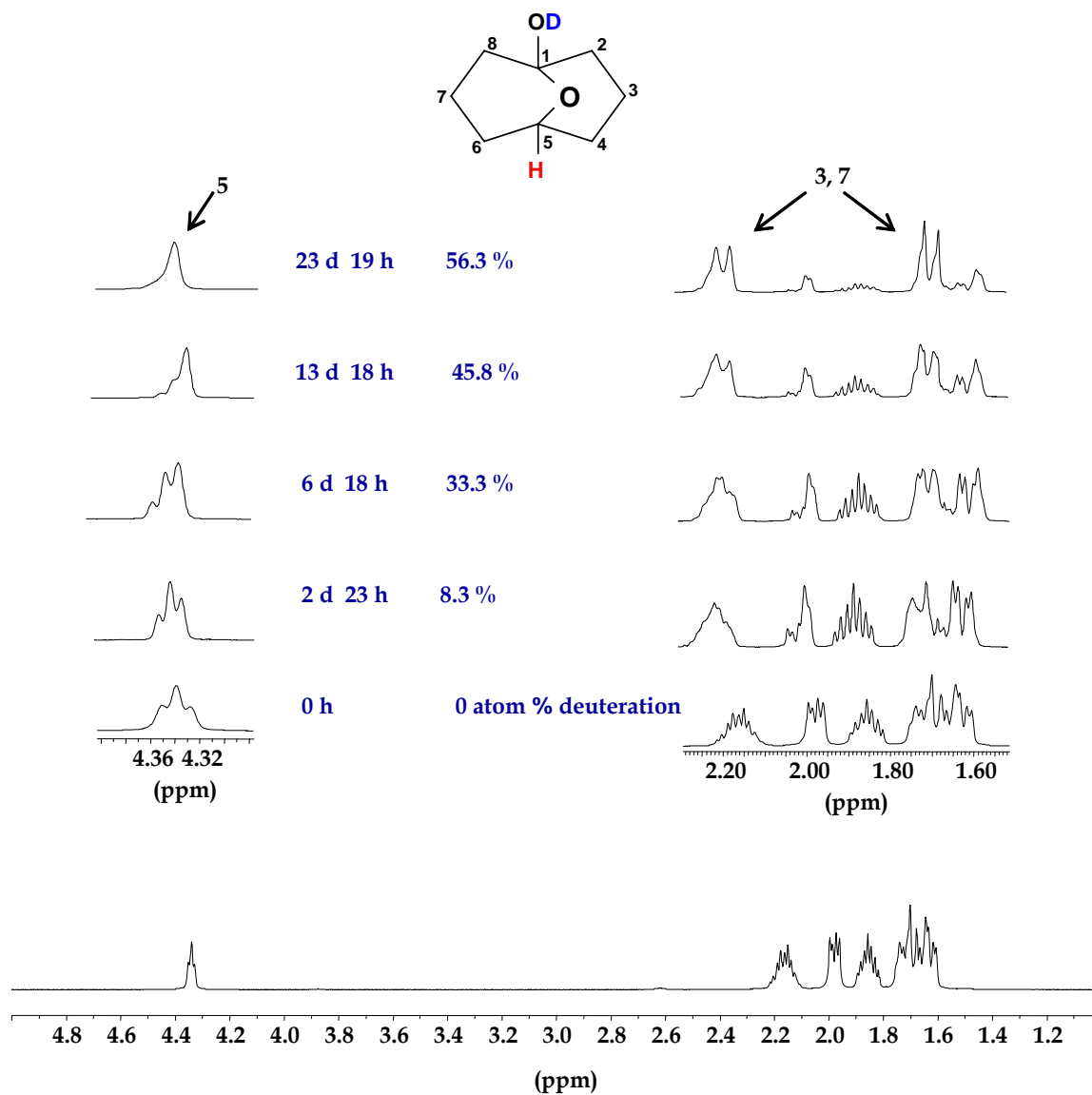
6bb

**Figure 3** Natural charges on atoms in the six-membered cyclic transition states in hydride shift reaction

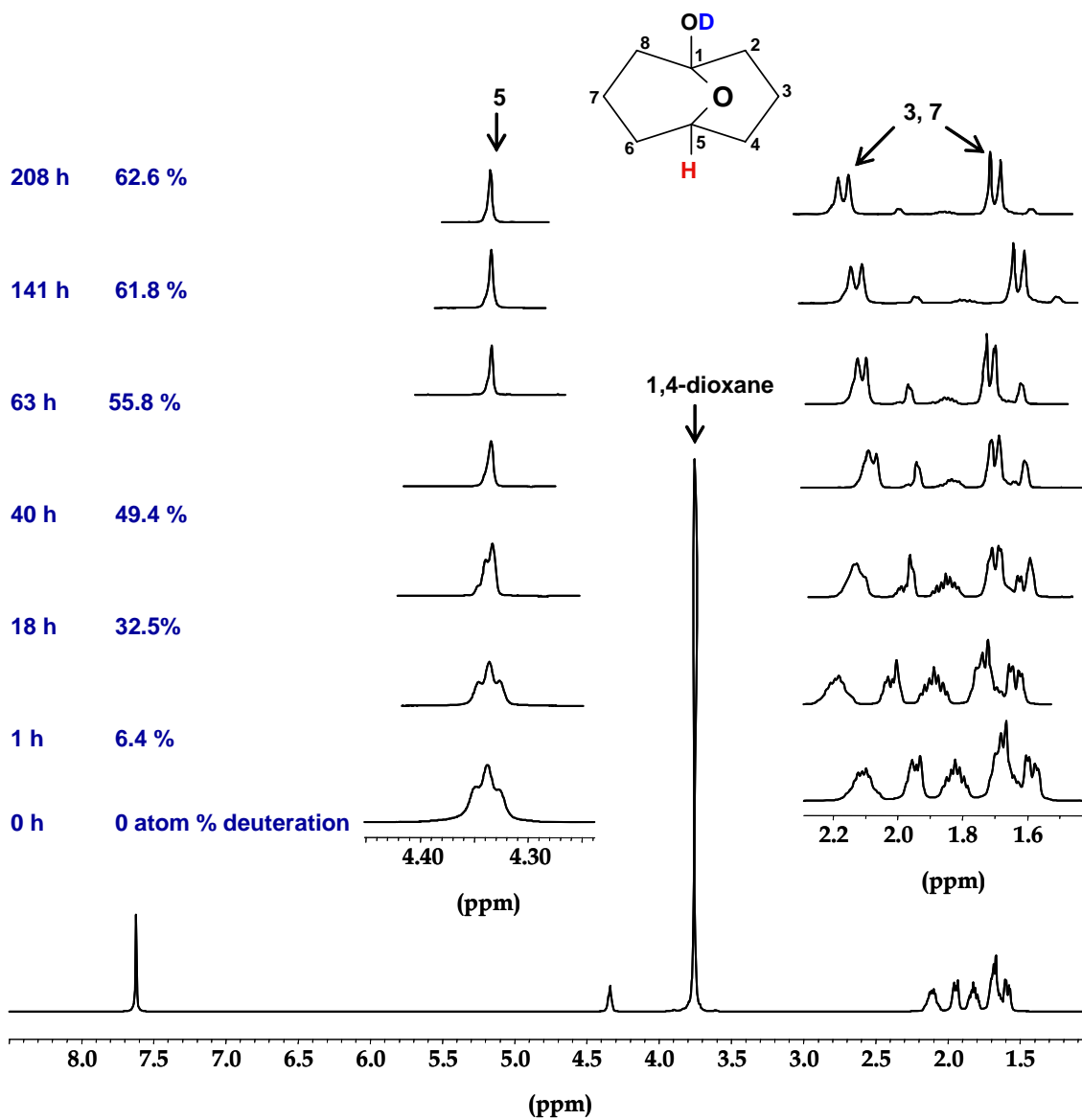
**Table ESI-1** H/D-exchange in **1HA** at ambient temperature studied by  $^1\text{H}$  NMR spectroscopy

Acid or Base M	Time h	Deuteration %	<b>1HA</b> <sup>a</sup> M	$k^b \text{ sec}^{-1}$
3.4 DCl	0	0.0	0.64	$8.95 \times 10^{-7}$
	71	8.3	0.56	
	162	33.3	0.32	
	330	45.8	0.20	
	571	56.3	0.10	
7.9 DCl	0	0.0	0.79	$5.12 \times 10^{-6}$
	1	6.4	0.71	
	18	32.5	0.41	
	40	49.4	0.21	
	63	55.8	0.13	
	141	61.8	0.06	
7.9 DCl <sup>c</sup>	0	0.0	0.77	$3.73 \times 10^{-3}$
	0.17	59.9	0.08	
12 DCl	0	9.2	0.57	$1.33 \times 10^{-4}$
	1	50.2	0.16	
	4	59.8	0.07	
0.65 NaOD	0	12.9	0.54	$6.45 \times 10^{-5}$
	1	34.2	0.33	
	4	47.0	0.20	
	22	53.6	0.13	
1.30 NaOD	0	14.1	0.51	$1.02 \times 10^{-4}$
	1	41.3	0.25	
	4	55.8	0.11	
	23	56.6	0.10	

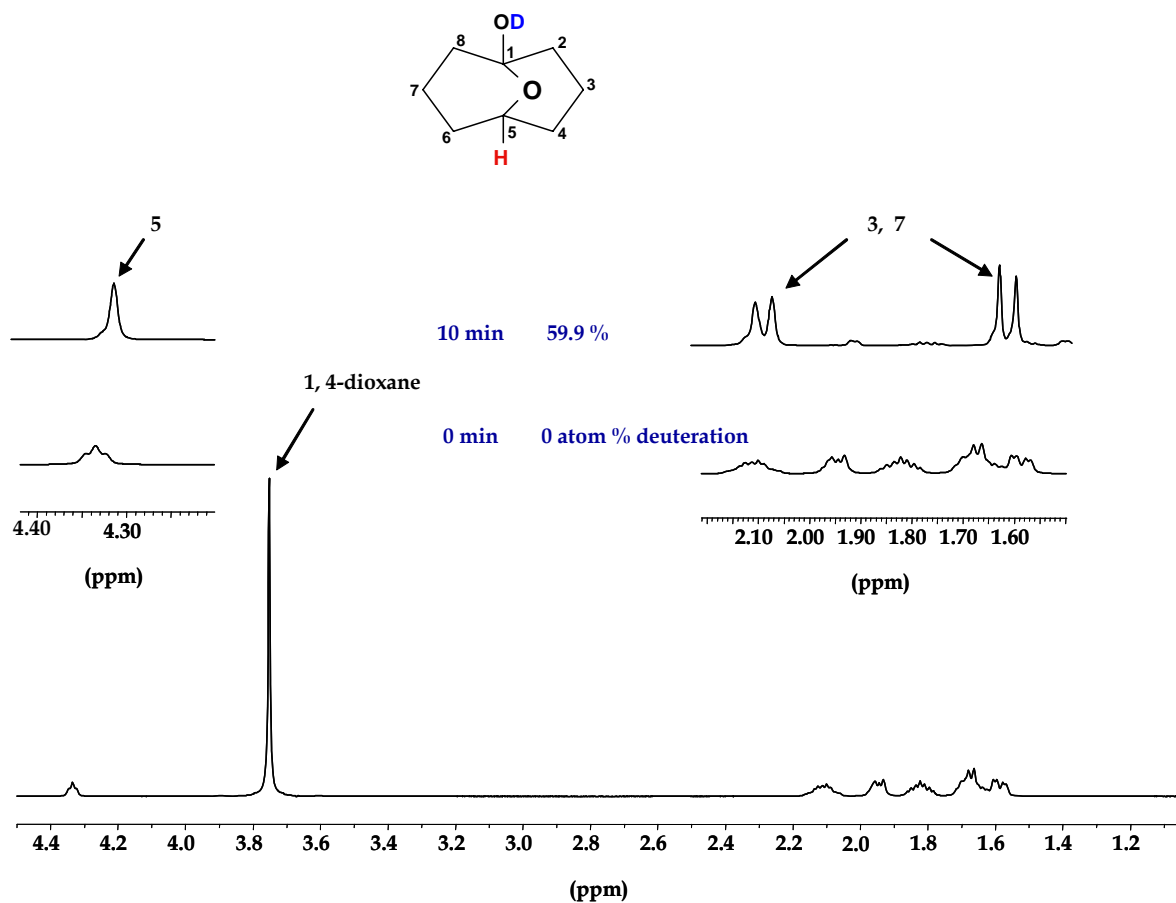
<sup>a</sup> Concentration of undeuterated **1HA**, determined from Degree of Deuteration assuming that at 66.66 % **1HA** = 0.00 M. <sup>b</sup> Rate constant. <sup>c</sup> 100 °C.



**Figure ESI-1**  $^1\text{H}$ -NMR spectra for **1HA** undergoing H/D exchange in 3.4 M DCl in  $\text{D}_2\text{O}$  at ambient temperature

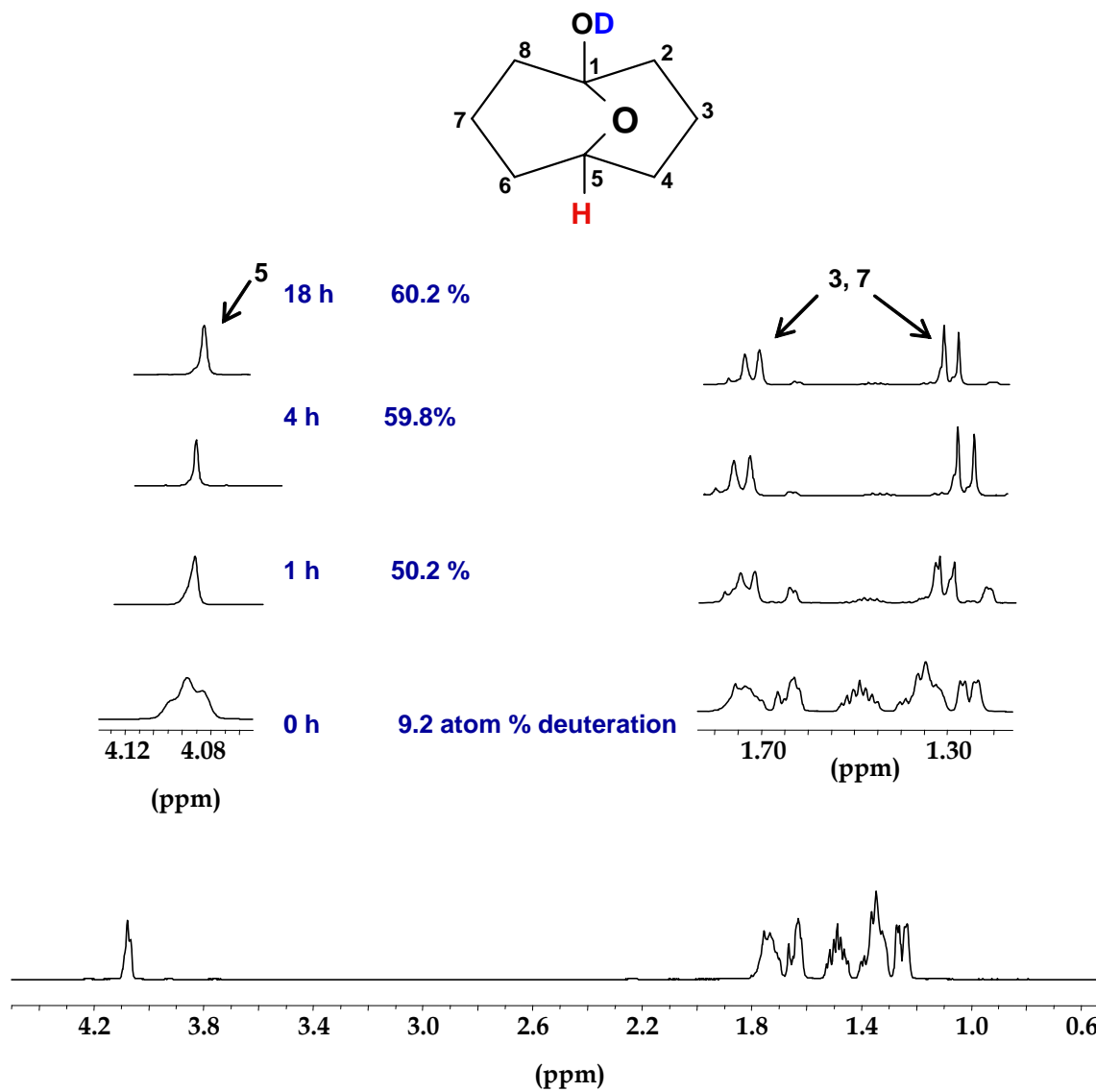


**Figure ESI-2**  $^1\text{H-NMR}$  spectra for **1HA** undergoing H/D exchange in 7.9 M DCl in  $\text{D}_2\text{O}$  at ambient temperature with dioxane as standard

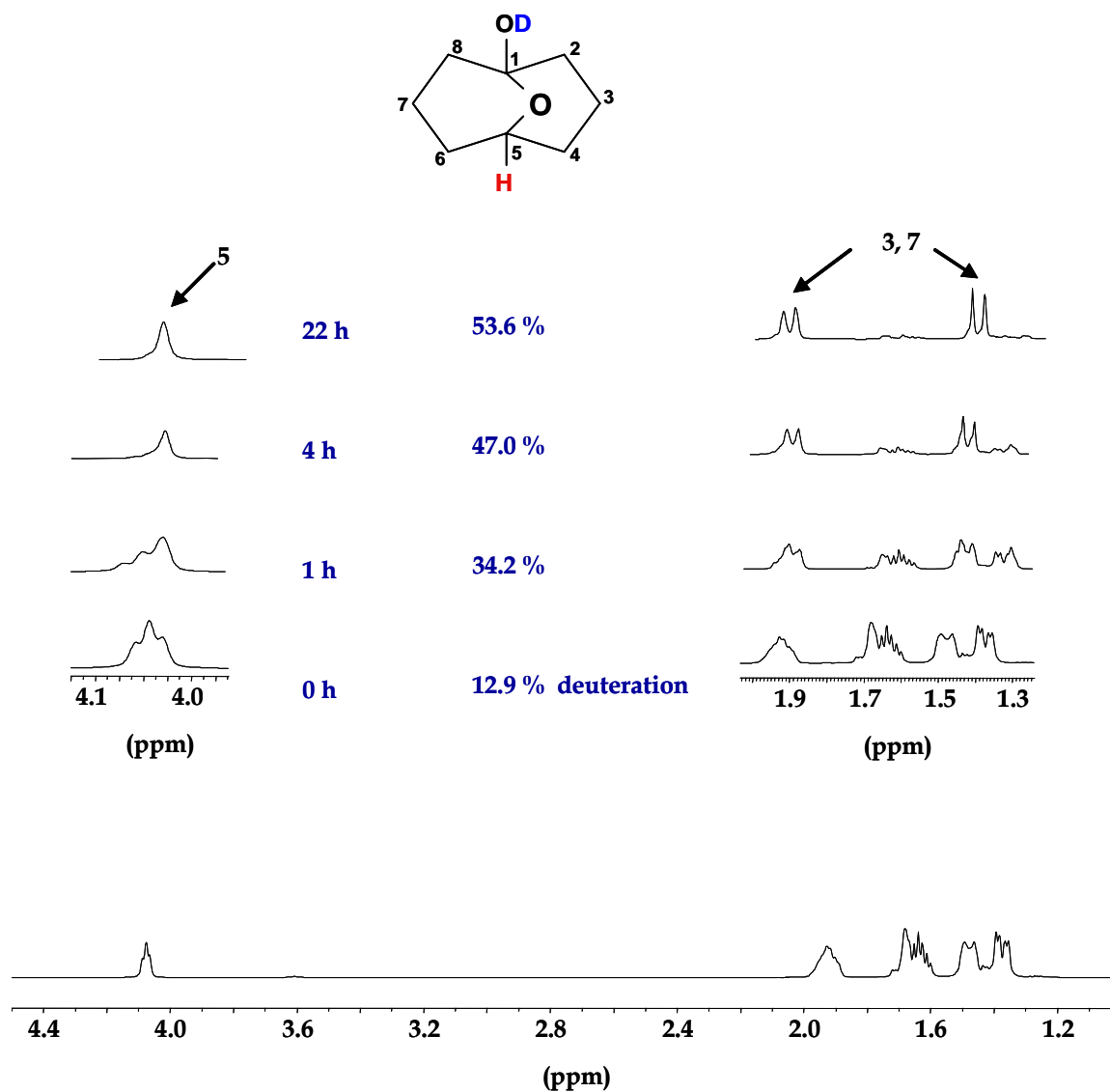


**Figure ESI-3**  $^1\text{H-NMR}$  spectra for **1HA** undergoing H/D exchange in 7.9 M DCl in  $\text{D}_2\text{O}$  at 100 °C with dioxane as standard

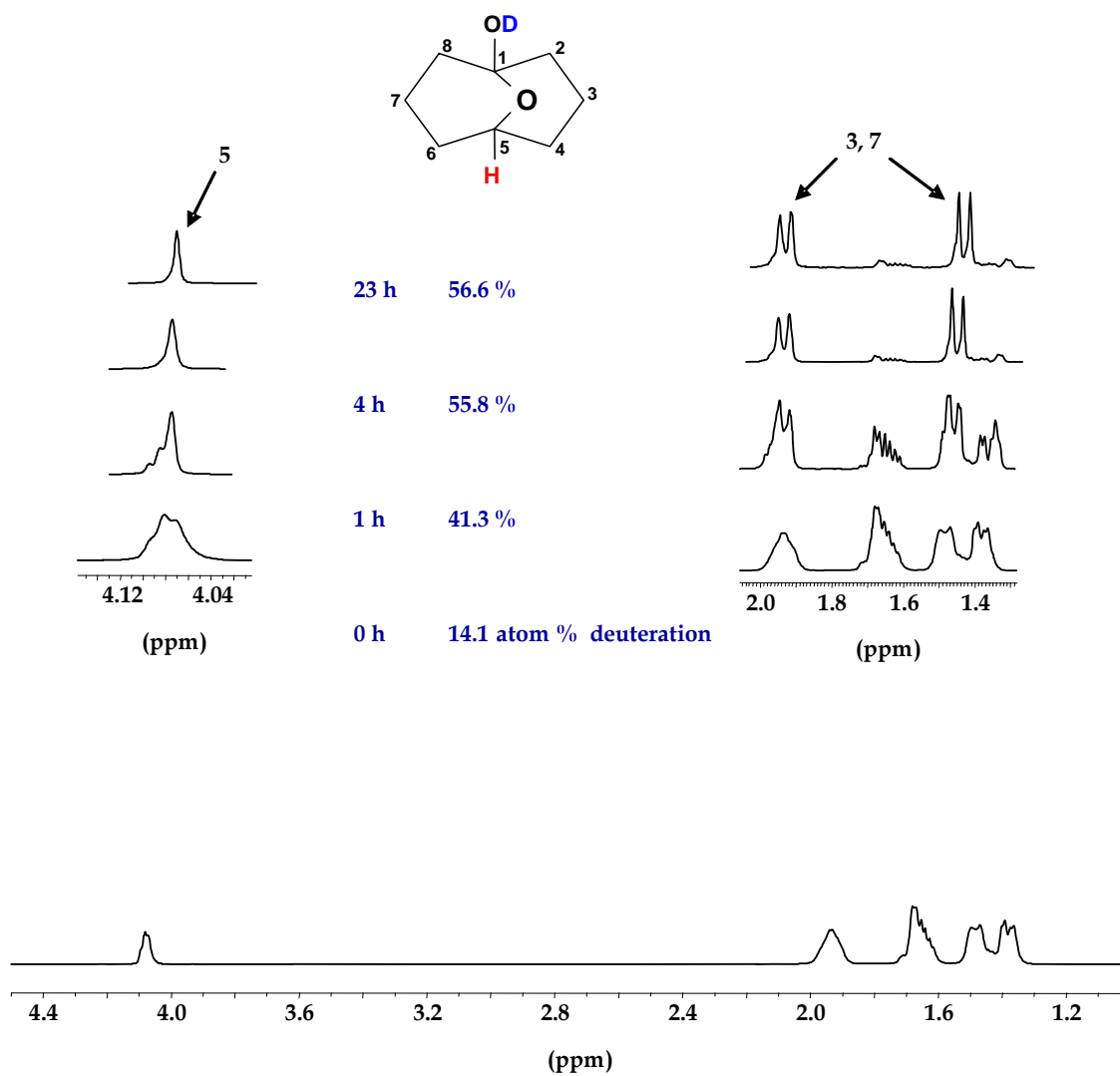




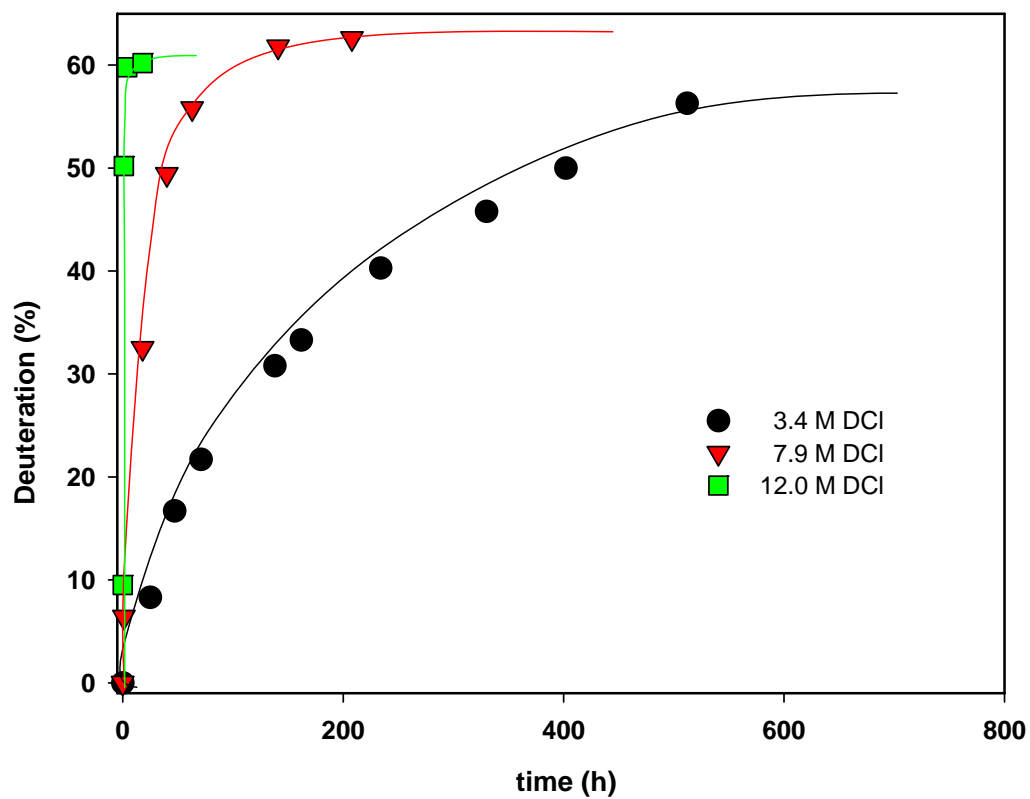
**Figure ESI-4**  $^1\text{H-NMR}$  spectra for **1HA** undergoing H/D exchange in 12.0 M DCl in  $\text{D}_2\text{O}$  at ambient temperature



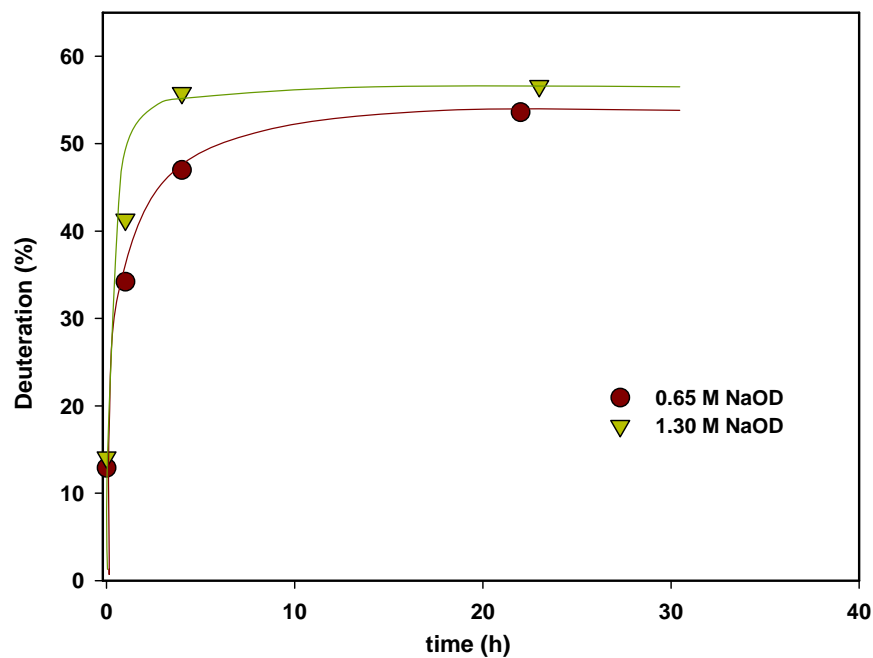
**Figure ESI-5**  $^1\text{H-NMR}$  spectra for **1HA** undergoing H/D exchange in 0.65 M NaOD in  $\text{D}_2\text{O}$  at ambient temperature



**Figure ESI-6**  $^1\text{H}$ -NMR spectra for **1HA** undergoing H/D exchange in 1.30 M NaOD in  $\text{D}_2\text{O}$  at ambient temperature



**Figure ESI-7** Deuteration of **1HA** as function of time at different concentrations of DCI in  $D_2O$  at ambient temperature



**Figure ESI-8** Deuteration of **1HA** as function of time at different concentrations of NaOD in  $D_2O$  at ambient temperature