Supplementary Material (ESI) for Organic & Biomolecular Chemistry This journal is (c) The Royal Society of Chemistry 2008 ELECTRONIC SUPPLEMENTARY INFORMATION

Propensity for local folding induced by the urea fragment

in short chain oligomers

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Page S3, Figure S1: Influence of the concentration and the solvent on the N-H stretching absorption in compound V series.

Page S3, Figure S2: Influence of DMSO- d_6 content in CDCl₃/DMSO- d_6 mixtures on the N²H proton resonances for **VI.1** and **V.1a**.

Page S4, Figure S3: Influence of DMSO- d_6 content in CDCl₃/DMSO- d_6 mixtures on the NH proton resonances for II.3 and II.4 .

Page S4, Figure S4: NOESY correlations demonstrating the cis, trans conformation of the N, N'-disubstituted urea fragment for **III.1** in CDCl₃.

Page S5, Figure S5: formulae of model compounds I.M and V.M used for DFT calculations.

Page S5, Figure S6: Average dimension of the N, N'-disubstituted urea fragment in ureidopeptides of type **C** and **D** as measured by X-ray diffraction.

Page S5, Figure S7: Urea-urea H-bond stabilizing the helical structure in oligoureas of the $H-(NH-CHR-CH_2-NH)_n-CO-NH_2$ type (stereoview), and also present as a minor conformer in triurea **VII**.

Page S6, Figure S8: Nomenclature used for the description of residues in compounds VI-VIII.

Page S7, Table S1: Calculated energetics, backbone torsion angles and hydrogen bond parameters for urea-peptide derivatives containing a gem-diamino residue (type A)

S1

This journal is (c) The Royal Society of Chemistry 2008 **Page S8, Table S2**: Calculated energetics, backbone torsion angles and hydrogen bond parameters for urea-peptide derivatives having two sp3 carbons in the main chain (type **B**).

Page S9, Table S3: Crystal data and structure refinement

Pages S10-S13, Table S4-S12: NMR Chemical Shifts for VI.2, VII and VIII in DMSO- d_6 and CDCl₃/DMSO- d_6 (70:30) + TMS



Figure S1. Influence of the concentration and the solvent on the N-H stretching absorption in compound V series. NH stretch region FT-IR data for compound V.1b in CCl_4 at 2, 0.2 and 0.04 mM(left) and in CH_2Cl_2 at 2 mM (right).



Figure S2. Influence of DMSO- d_6 content in CDCl₃/DMSO- d_6 mixtures on the N²H proton resonances for VI.1 (heavy line) and V.1a (broken line).



Figure S3. Influence of DMSO- d_6 content in CDCl₃/DMSO- d_6 mixtures on the NH proton resonances for II.3 (left) and II.4 (right).



Figure S4. NOESY correlations demonstrating the cis, trans conformation of the N, N'-disubstituted urea fragment for III.1 in CDCl₃.



V.M

Figure S5: formulae of model compounds I.M and V.M used for DFT calculations.



Figure S6. Average dimensions of the N, N'-disubstituted urea fragment.



Figure **S7.** Urea-urea H-bond stabilizing the helical structure in oligoureas of the H-(NH-CHR-CH₂-NH)_n-CO-NH₂ type (stereoview), and also populated in tetraurea **VIII**.

RAS N B HRe H^{Si}

Figure S8. Nomenclature used for the description of residues in compounds VI-VIII

Table S1. Calculated energetics, backbone torsion angles and hydrogen bond parameters for ureapeptide derivatives containing a gem-diamino residue.

	u-I	u'-I	C6A-I	C6A'-I	C6B-I	C6B′-I	C6C-I	C6D-I	C6D′-I	C8-I	C8′-I
						I.M					
$\Delta ext{E}^{[a]}$			4.0	0.8	4.2	0.9	5.8	5.3	1.9	5.7	0.0
Δ G° ^[a]			3.1	0.3	3.6	0.1	5.6	4.6	1.3	5.8	0.0
φ ^r ^[b]			-55.5	53.3	-177.2	-75.0	55.6	-177.0	-81.4	75.4	-84.6
Ψ ^[b]			170.9	74.8	57.7	-55.1	71.2	56.9	-53.6	-100.3	108.1
r_{hb} (H•••O)			2.099	2.326	2.206	2.486	2.385	2.188	2.402	1.878	1.948
θ(N- H•••O) ^[b]			129.6	116.6	122.9	106.4	116.6	123.5	110.0	168.8	167.5
						I.3a					
$\Delta extsf{E}^{[a]}$	4.4	4.2	3.9	3.8	4.4	3.3	5.3	5.3	3.4	6.6	0.0
Δ.G° ^[a]	3.3	2.8	3.5	2.6	3.8	2.5	5.0	4.9	3.5	7.6	0.0
•	-161.5	-83.4	-52.6	52.8	-173.4	-93.3	53.9	-173.7	-102.4	70.0	-78.6
$\mathbf{\hat{\Psi}}^{\text{\tiny{[b]}}}$	112.8	154.7	168.3	75.4	60.5	-52.4	132.4	56.9	-57.2	-108.6	112.3
, γ _{hb} c] (H●●●O) [[]	-		2.004	2.438	2.079	2.269	2.390	2.036	2.134	1.898	1.998
θ (N- H•••O) ^[b]	_		133.7	112.8	126.9	116.2	115.3	128.8	120.7	166.5	164.8

[a] in kcal/mol

[b] in degrees [c] in Å

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Table S2. Calculated energetics, backbone torsion angles and hydrogen bond parameters for ureapeptide derivatives having two sp³ carbons in the main chain.

	u-V	C7A-V	C7B-V	C7C-V	C7D-V	//-V	C9-V
				v.1	м		
$\Delta extsf{E}^{ extsf{a}}$	0.3	-1.9	-3.0	1.6	-2.2	-2.7	0.0
$\Delta_{}$ G° ^[a]	2.5	-4.2	-4.4	-0.5	-3.4	-3.5	0.0
• ^[b]	-79.1	-74.9	-158.2	-90.1	-158.2	-86.9	-97.3
$v^{[b]}$	170.9	78.4	-71.8	60.7	-71.6	48.8	46.9
$\Psi^{\text{[b]}}$	107.5	116.3	82.0	167.3	82.6	-102.2	87.8
$r_{hh}(H \bullet \bullet \circ)^{[c]}$	-	1.954	1.976	2.144	1.968	2.310	1.965
θ (N-	-	148.9	150.5	143.0	150.7	128.5	164.2
H●●●O) ^[b]							
				v.2	a		
$\Delta extsf{E}^{ extsf{a}}$	-1.7	-1.5	-3.3	1.2	-2.4	-2.0	0.0
$\Delta_{}$ G° ^[a]	-2.7	-3.4	-3.3	-0.5	-2.2	-1.9	0.0
$\Phi_{[p]}$	-77.6	-80.1	-160.6	-89.2	-160.8	-84.8	-94.3
$v^{[b]}$	169.6	75.0	-62.5	60.5	-61.8	41.7	46.7
$\Psi^{\text{\tiny{[b]}}}$	104.4	125.6	84.5	172.2	85.2	-105.3	89.1
$r_{\rm hb} (\mathrm{H} \bullet \bullet \circ \mathrm{O})^{[\mathrm{c}]}$	-	2.030	2.048	2.189	2.020	2.279	1.993
θ (N-	-	150.1	148.4	141.5	149.1	132.2	161.9
H●●●O) ^[b]							

[a] in kcal/mol

[b] in degrees [c] in Å

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Table S3. Crystal data and structure refinement

	I.1b	I.3c	I.4b	II.1'a	II.2	V.la	V.1c	VI.2
formula	$C_{14}H_{29}N_{3}O_{3}$	$C_{17}H_{27}N_{3}O_{4}$	$C_{13}H_{25}N_{3}O_{3}$	$C_{17}H_{32}N_4O_3$	$C_{_{26}}H_{_{40}}N_{_4}O_{_6}$	$C_{16}H_{25}N_{3}O_{3}$	$C_{_{17}}H_{_{27}}N_{_{3}}O_{_{3}}$	$C_{_{19}}H_{_{24}}N_{_4}O_{_2}$
Mr	287.4	337.42	271.36	340.47	504.62	307.39	321.42	340.42
Wavelength λ (Å)	1.5406	0.71073	0.71073	0.71073	0.71073	0.71073	1.5406	1.5406
crystal system	monoclinic	monoclinic	monoclinic	monoclinic	orthorhombic	monoclinic	orthorhombic	triclinic
space group	P2(1)/c	<i>P</i> 2(1)	P2(1)/c	<i>P</i> 2(1)	P2(1)2(1)2(1)	<i>C</i> 2	P2(1)2(1)2(1)	<i>P</i> 1
a (Å)	12.169(6)	10.1826(8)	9.4284(2)	10.1850(12)	9.3903(7)	20.565(5)	5.241(1)	4.675(2)
b (Å)	9.397(3)	9.5473(7)	15.8240(3)	9.4470(7)	16.069(1)	5.223(1)	17.883(3)	9.497(2)
<i>c</i> (Å)	16.467(2)	10.2926(11)	21.0020(5)	10.571(2)	19.446(2)	17.230(4)	19.356(2)	10.809(3)
α∙•deg•	90	90	90	90	90	90	90	76.78(2)
β ••deg•	103.86(3)	103.133(3)	98.536(1)	103.216(4)	90	105.921(8)	90	81.59(3)
γ••deg•	90	90	90	90	90	90	90	77.55(2)
Z / Z'	4 / 1	2 / 1	8 / 2	2 / 1	4 / 1	4 / 1	4 / 1	1 / 1
volume $(Å^3)$	1828(1)	974.4(2)	3098.7(1)	990.2(2)	2934.3(4)	1779.7(7)	1814.1(5)	453.8(3)
$D_{_{calc}}$ (g cm ⁻³)	1.044	1.15	1.163	1.142	1.142	1.147	1.177	1.246
μ (mm ⁻¹)	0.593	0.082	0.083	0.079	0.081	0.08	0.656	0.666
2θ scan range \bullet deg \bullet	7.48- 129.88	5.08-46.38	4.36-51.0	7.66-50.04	5.02-41.32	4.12-46.46	6.72-139.66	8.44- 139.58
range <i>h</i>	0 to 14	-11 to 11	-11 to 11	-12 to 12	-9 to 9	-22 to 22	0 to 6	-5 to 5
range k	0 to 11	-9 to 9	-19 to 19	-11 to 11	-15 to 15	-5 to 5	0 to 21	-11 to 11
range <i>l</i> reflns	-19 to 18	-11 to 11	-25 to 25	-12 to 12	-19 to 19	-18 to 18	0 to 23	-13 to 13
collected	3249	4334	11307	15160	7886	4290	2026	3348
unique reflns	3097	1461	5770	1857	1660	1410	2026	1723
R _{int} R1, wR2 [<i>I</i> >	0.09 0.0656, 0.1825	0.042 0.0642, 0.1654	0.044 0.0517, 0.128	0.043 0.045, 0.1102	0.079 0.0441, 0.0809	0.096 0.0856, 0.2327	0 0.0503, 0.1307	0.0518 0.0461, 0.1039



Table S4.	¹ H-NMR	Chemical	Shifts	(in ppm)	of	VI.2	in	DMSO-d	at 🕯	298K	(400	MHz))
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N <i>H</i> Bn	NH	Ν'Η	N <i>H</i> Me	lpha CH	$\Delta \delta$ (^{δ} CH)	$^{\beta}$ CH	CH_2 Ph (Bn)	$^{\gamma}$ CH	CH_3
6.38	6.00	5.90	5.85	3.07, 2.96	0.11	3.79	4.17	2.72, 2.62	2.54

¹H-NMR Chemical Shifts (in ppm) of **VI.2** in $CDCl_3/DMSO-d_6$ (70:30) at 298K (400 MHz) Table S5. $^{\alpha}$ CH $^{\beta}$ CH $\Delta \delta$ (^{α}CH) CH_2Ph (Bn) $^{\gamma}$ CH N*H*Bn NH Ν'Η N*H*Me CH_3 5.81 5.83 5.70 3.16, 2.96 0.20 3.92 4.27 2.77 (d) 2.66 6.22

Table S6. Evolution of the NH resonance (ppm) for **VI.2** in DMSO- d_6 and CDCl₃/DMSO- d_6 (70:30) at 298K

Δδ	(N <i>H</i> Bn)	$\Delta\delta$ (NH)	∆δ (NH′)	∆δ (N <i>H</i> Me)
	0.16	0.19	0.07	0.15

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Table S7. ¹H-NMR Chemical Shifts (in ppm) of VII in DMSO- d_6 at 298K (500 MHz)

Residue	N <i>H</i> Bn	NH	Ν'Η	N <i>H</i> Me	lpha CH	Δδ (^α CH)	^β CH	CH_2 Ph (Bn)	ΥCH	CH_3
1	-	5.94	5.92	5.86	3.13, 2.85	0.28	3.82	-	3.74	2.49
2	6.41	5.94	6.01	-	3.10, 2.86	0.24	3.88	4.28, 4.20	3.81	-

Table S8. ¹H-NMR Chemical Shifts (in ppm) of **VII** CDCl₃/DMSO- d_6 (70:30) at 298K (500 MHz)

Residue	NHBn	NH	Ν'Η	N <i>H</i> Me	°СН	∆ð (°CH)	^P CH	CH ₂ Ph (Bn)	'CH	CH ₃
1	-	5.83	5.80	5.73	3.30, 2.90	0.40	4.00	-	2.74	2.58
2	6.28	5.81	5.88	-	3.30, 2.86	0.44	3.92	4.31, 4.22	2.73	-

Table S9. Evolution of the NH resonance (ppm) for **VI.2** in DMSO- d_6 and CDCl₃/DMSO-d6 (70:30) at 298K

Residue $\Delta \delta$	(N <i>H</i> Bn)	$\Delta\delta$ (NH)	$\Delta \delta$ (NH')	∆δ (N <i>H</i> Me)
1	-	0.11	0.12	0.13
2	0.13	0.13	0.13	-



Table S10. ¹H-NMR Chemical Shifts (in ppm) of **VIII** in DMSO- d_6 at 298K (500 MHz)

Residue	N <i>H</i> Bn	NH	N′H	N <i>H</i> Me	lpha CH	Δδ (^α CH)	^β CH	CH ₂ Ph (Bn)	^ү СН	CH_3
1	_	5.93	6.02	5.96	3.27, 2.73	0.54	3.71	-	2.55, 2.43	2.55
2	-	5.93	5.84	-	3.24, 2.47	0.77	3.81	-	2.64, 2.55	-
3	6.47	6.02	6.13	_	3.20, 2.73	0.47	3.95	4.22, 4.11	2.68, 2.55	-
Table S1	1 . ¹ H-N	MR Chem	ical Shi	fts (in	ppm) of VII	I CDCl ₃ /D	MSO- <i>d6</i>	(70:30) at 2	298K (400 MHz)
Residue	N <i>H</i> Bn	NH	Ν'Η	N <i>H</i> Me	lpha CH	$\Delta \delta$ (^{α} CH)	^β CH	CH_2 Ph (Bn)	ΎСН	CH_3
1	-	5.91	6.07	6.01	3.59, 2.71	0.88	3.89	-	2.55, 2.48	2.70
2	_	5.68	5.77	-	3.53, 2.24	1.29	4.06	-	2.67, 2.60	-
3	6.37	5.85	6.19	_	3.46, 2.56	0.90	4.18	4.31, 4.16	2.72, 2.56	_

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Table S12. Evolution of the NH resonance (ppm) for **VIII** in DMSO-d6 and CDCl₃/DMSO- d_6 (70:30) at 298K

Residue A	∆δ (N <i>H</i> Bn)	$\Delta\delta$ (NH)	Δδ (NH′)	∆ð (N <i>H</i> Me)
1	_	0.02	-0.05	-0.05
2	_	0.25	0.07	_
3	0.10	0.17	-0.06	_