

Solvent effects on the rate of the keto-enol interconversion of 2-nitrocyclohexanone

Gabriella Siani, Guido Angelini, Paolo De Maria, Antonella Fontana and Marco Pierini

Table of contents

Table 1. UV-determined tautomeric constants, K_T , of 2-NCH in several organic solvents S2 and their binary mixtures and selected descriptors of solvent polarity.

Table 2. Sensitivity of the selected descriptors of solvent polarity to different solute- S2 solvent interactions.

Table 3. Second order rate constants, k_{app}^{TEA} , k_1^{TEA} and k_{-1}^{TEA} for the base catalyzed S3 keto-enol interconversion of 2-NCH in aprotic solvents at different temperatures.

Table 4. Second order rate constants, k_{app}^{Pyr} , k_1^{Pyr} and k_{-1}^{Pyr} for the base catalyzed keto- S4 enol interconversion of 2-NCH in aprotic solvents and their binary mixtures at different temperatures.

Table 1. UV-determined tautomeric constants, K_T , of 2-NCH in several organic solvents and their binary mixtures and selected descriptors of solvent polarity.

Solvent	$K_{T(UV)}$	ϵ	δ_H	α	β	π^*	$E_T(30)$	T_N
C ₆ H ₁₂	6.46	2.0	8.2	0.000	0.000	0.000	30.9	0.03
CCl ₄	5.97	2.2	8.6	0.000	0.100	0.280	32.4	0.25
CHCl ₃	1.92	4.8	9.3	0.200	0.100	0.580	39.1	0.67
CH ₂ Cl ₂	0.78	8.9	9.7	0.130	0.100	0.820	40.7	0.67
CH ₃ CN -CH ₂ Cl ₂ (20:80)	0.39	13.7	10.1	0.142	0.160	0.806	41.7	0.67
CH ₃ CN -CH ₂ Cl ₂ (50:50)	0.22	21.7	10.8	0.160	0.250	0.785	43.2	0.67
CH ₃ CN - CHCl ₃ (80:20)	0.14	28.8	11.3	0.192	0.340	0.716	44.3	0.67
CH ₃ CN -CH ₂ Cl ₂ (80:20)	0.14	30.7	11.4	0.178	0.340	0.764	44.6	0.67
CH ₃ CN	0.12	35.9	11.8	0.190	0.400	0.750	45.6	0.67

Table 2. Sensitivity of the selected descriptors of solvent polarity to different solute-solvent interactions.

Descriptor	solute-solvent interactions
ϵ	pure aspecific (dipole-dipole and ion-dipole) electrostatic interactions
δ_H	cavitation energy interactions
$E_T(30)$, π^* , T_N	aspecific (dispersive and electrostatic) and specific (hydrogen bonding and/or lone pair donor-acceptor) interactions
α	H-bond donor specific interactions
β	H-bond acceptor specific interactions

Table 3. Second order rate constants, $k_{\text{app}}^{\text{TEA}}$, k_1^{TEA} and k_{-1}^{TEA} for the base catalyzed keto-enol interconversion of 2-NCH in aprotic solvents at different temperatures.

Solvent	T (°K)	$k_{\text{app}}^{\text{TEA}} (\text{M}^{-1}\text{s}^{-1})$	$k_1^{\text{TEA}} (\text{M}^{-1}\text{s}^{-1})$	$k_{-1}^{\text{TEA}} (\text{M}^{-1}\text{s}^{-1})$	$\Delta G^\# (\text{Kcal mol}^{-1})$	K_T
CHCl ₃	288.15	25.7 (± 0.7)	17.4	8.30	13.3	2.10
	293.15	29.6 (± 0.4)	19.8	9.80	13.5	2.01
	308.15	39.2 (± 0.4)	24.9	14.3	13.9	1.75
	313.15	44.0 (± 0.8)	27.5	16.5	14.0	1.66
CH ₂ Cl ₂	283.15	49.8 (± 0.9)	22.2	27.6	14.3	0.805
	288.15	61.8 (± 4.3)	27.4	34.4	14.5	0.797
	303.15	108 (± 2)	47.1	60.9	14.9	0.773
	308.15	116 (± 3)	50.3	65.7	15.1	0.765
CH ₃ CN	283.15	299 (± 5)	27.2	272	15.0	0.100
	288.15	343 (± 6)	32.0	311	15.2	0.103
	303.15	633 (± 33)	64.8	568	15.8	0.114
	308.15	752 (± 136)	78.8	673	16.0	0.117

Table 4. Second order rate constants, $k_{\text{app}}^{\text{Pyr}}$, k_1^{Pyr} and k_{-1}^{Pyr} for the base catalyzed keto-enol interconversion of 2-NCH in aprotic solvents and their binary mixtures at different temperatures.

Solvent	T (°K)	$10^{-2} k_{\text{app}}^{\text{Pyr}} (\text{M}^{-1}\text{s}^{-1})$	$10^{-2} k_1^{\text{Pyr}} (\text{M}^{-1}\text{s}^{-1})$	$10^{-2} k_{-1}^{\text{Pyr}} (\text{M}^{-1}\text{s}^{-1})$	$\Delta G^\# (\text{Kcal mol}^{-1})$	K_T
CHCl ₃	288.15	0.84 (± 0.02)	0.57	0.27	19.6	2.10
	293.15	1.1 (± 0.1)	0.70	0.40	19.8	2.01
	303.15	1.9 (± 0.1)	1.2	0.70	20.2	1.84
	308.15	2.2 (± 0.1)	1.4	0.80	20.4	1.75
	313.15	2.8 (± 0.2)	1.7	1.1	20.6	1.66
	318.15	3.4 (± 0.1)	2.1	1.3	20.8	1.58

CH ₂ Cl ₂	283.15	0.83 (\pm 0.02)	0.37	0.46	19.6	0.805
	293.15	1.6 (\pm 0.1)	0.70	0.90	19.2	0.797
	303.15	2.5 (\pm 0.1)	1.1	1.4	20.0	0.773
	308.15	3.5 (\pm 0.2)	1.5	2.0	20.1	0.765
	313.15	4.3 (\pm 0.1)	1.9	2.4	20.3	0.757
CH ₃ CN	288.15	5.8 (\pm 0.1)	0.54	5.3	18.5	0.103
	303.15	13 (\pm 1)	1.3	12	19.0	0.114
	308.15	19 (\pm 1)	2.0	17	19.1	0.117
	318.15	29 (\pm 2)	3.2	26	19.4	0.124