Electronic Supplementary Information

Fluorescent Detection of Methylmercury by Desulfurization Reaction of Rhodamine Hydrazide Derivatives

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General synthetic materials and spectroscopic methods.

Silica gel 60 (230–400 mesh, Merck) was used for column chromatography. Analytical thin layer chromatography was performed using Merck 60 F₂₅₄ silica gel (precoated sheets, 0.25 mm thick). All reagents and solvents for reactions were used as received with the following exceptions. All other chemicals used were purchased from Sigma-Aldrich and were used as received. Nuclear magnetic resonance (NMR) spectra were recorded in CDCl₃ unless otherwise stated, with tetramethylsilane (TMS) as internal reference at ambient temperature, mainly on a Bruker Avance II-400 Fourier Transform Spectrometer operating at 400 MHz for ¹H and at 100.6 MHz for ¹³C. Mass spectra were recorded on a ZQ-4000 LC-MS and QUATTRO LC Triple Quadrupole Tandem mass spectrometer for both low resolution and high resolution mass spectra. The pH was recorded by HI-8014 instrument (HANNA). Melting points were measured on a Z289078 (Sigma-Aldrich) microscope and were uncorrected. Infrared absorption spectra were recorded as a solution in CH₂Cl₂ on an Avatar 360 FT-IR spectrophotometer. Fluorescence emission spectra were obtained using a Hitachi F-4500 spectrofluorimeter linked to a Pentium PC running SpectraCalc software package. The slit width was 2.5 nm for both excitation and emission. The photon multiplier voltage was 400 V. A circulating PBS buffer/DMSO bath was used during all experiments to regulate the temperature at 25.0 ± 0.1 °C. Samples were contained in 10.0 nm path length quartz cuvettes (3.5 mL volume). Upon excitation at 500 nm, the emission spectra were integrated over the range 510-650 nm. All measurements were conducted at least in triplicate. We used a TE2000 fluorescence microscope for the fluorescence imaging experiments and a dissecting microscope (Stemi 2000-C, ZAISS) for the microscopic imaging experiments.

Synthesis of rhodamine thiosemicarbazides.



The rhodamine 6G hydrazide **3** (200 mg, 0.47 mmol) in DMF (1.5 mL) was added to a solution of phenyl isothiocyanate (0.1 mL, 0.65 mmol) in DMF (1.5 mL). The reaction mixture was stirred for 6 h at room temperature. After the solvent was evaporated under reduced pressure, the crude product was column chromatographed on silica-gel (elution with hexanes/EtOAc/CH₂Cl₂ = 4:1:1) to give the 236 mg (90%) of **1**: R_f = 0.5 (silica gel, hexanes/EtOAc = 1:1); mp 150-152 °C; ¹H NMR (250 MHz, CDCl₃) δ = 8.06-8.03 (m, 1H), 7.66-7.57 (m, 3H), 7.26-7.22 (m, 3H), 3.26-3.17 (m, 4H), 1.84 (s, 6H), 1.32 (t, *J* = 7.1 Hz, 6H); ¹³C NMR (62.9 MHz, CDCl₃) δ = 182.2, 166.9, 152.5, 150.8, 148.4, 138.1, 134.1, 129.5, 129.0, 128.2, 128.1, 125.8, 125.2, 124.4, 123.2, 118.9, 104.4, 95.9, 67.0, 37.9, 16.1, 13.7; IR (film, cm⁻¹) 3326, 2964, 2960, 2950, 2356, 2351, 1713, 1620, 1517, 1430, 1424, 1347, 1274, 1212, 1089, 1016; HRMS (FAB) *m*/*z* calcd for C₃₃H₃₃N₅O₂S [(M+H)⁺] 564.2423; found 564.2433.

Compound 4: $R_f = 0.7$ (silica gel, hexanes/EtOAc = 1:1); mp 222–224 °C; ¹H NMR (400 MHz, CDCl₃) $\delta = 8.02-7.98$ (m, 4H), 7.69 (t, J = 7.2 Hz, 1H), 7.61 (t, J = 7.6 Hz, 1H), 7.37 (d, J = 5.2 Hz, 2H), 7.25 (s, 1H), 6.37 (s, 2H), 6.30 (s, 2H), 3.59 (bs, 2H), 3.18–3.16 (m, 4H), 1.87 (s, 6H), 1.29 (t, J = 7.2 Hz, 6H); ¹³C NMR (100.6 MHz, CDCl₃) $\delta = 181.9$, 167.8, 152.9, 150.3, 148.3, 144.1, 144.0, 134.8, 129.3, 128.9, 127.4, 125.0, 124.2, 124.0, 122.8, 118.7, 104.3, 97.2, 68.0, 38.3, 16.9, 14.8; IR (film, cm⁻¹) 3421, 3317, 2967, 2924, 2863, 1714, 1619, 1593, 1515, 1467, 1420, 1329, 1269, 1212, 1165, 1113, 1083, 1014; ESI-MS *m*/*z* calcd for C₃₃H₃₂N₆O₄S [(M + Na)⁺] 631; found 631.

Compound 5: $R_f = 0.5$ (silica gel, hexanes/EtOAc = 1:1); mp 210–212 °C; ¹H NMR (400 MHz, CDCl₃) $\delta = 8.04$ (d, J = 8.4 Hz, 1H), 7.67–7.60 (m, 2H), 7.42 (s, 1H), 7.27–7.23 (m, 1H), 6.89 (s, 1H), 6.87 (d, J = 8.8 Hz, 2H), 6.71 (d, J = 8.8 Hz, 2H), 6.41 (s, 2H), 6.25 (s, 2H), 3.75 (s, 3H), 3.58 (bs, 2H), 1.79 (s, 6H), 1.34 (t, J = 7.2 Hz, 6H); ¹³C NMR (100.6 MHz, CDCl₃) $\delta = 183.0$, 167.3, 157.8, 152.8, 150.6, 148.2, 134.5, 130.6, 129.2, 129.1, 126.8, 124.9, 123.9, 118.7, 113.7, 104.5, 97.3,

67.5, 55.5, 38.4, 16.9, 14.8; IR (film, cm⁻¹) 3326, 2958, 2924, 2863, 1722, 1679, 1614, 1519, 1467, 1420, 1355, 1273, 1243, 1217, 1161, 1178, 1087; ESI-MS *m*/*z* calcd for $C_{34}H_{35}N_5O_3S$ [(M + Na)⁺] 616; found 616.

Imaging of HeLa cells incubated with methylmercury and 1: HeLa cells were seeded in a 6-well plate at a density of 1×10^5 cells per well in culture media. After 24 h, the cells were incubated with 1 (20 μ M) in culture media for 30 min at 37 °C. After washing with PBS to remove the remaining probe, the cells were further treated with CH₃Hg⁺ (5–20 μ M) in culture media for 10 min. The cells incubated under these conditions were imaged by fluorescence microscope (TE2000, Nikon).

Imaging of zebrafish incubated with methylmercury and 1: Zebrafish was kept at 28 °C and maintained at optimal breeding conditions. For mating, male and female zebrafish was maintained in one tank at 28 °C on a 12 h light/12 h dark cycle and then the spawning of eggs were triggered by giving light stimulation in the morning. Almost all the eggs were fertilized immediately. The 4-day old zebrafish was maintained in E3 embryo media (15 mM NaCl, 0.5 mM KCl, 1 mM MgSO₄, 1 mM CaCl₂, 0.15 mM KH₂PO₄, 0.05 mM Na₂HPO₄, 0.7 mM NaHCO₃, 10–5% methylene blue; pH 7.5).¹⁹ The 4-day old zebrafish was incubated with chemosensor **1** (20 μ M) in E3 embryo media for 30 min at 28 °C. After washing with PBS to remove the remaining chemosensors, the zebrafish was further treated with CH₃Hg⁺ (20 μ M) for 10 min at 28 °C. The zebrafish was imaged by a fluorescence microscope (TE2000, Nikon).

Real-time monitoring of methylmercury uptake in HeLa cells and A549 cells: HeLa cells and A549 cells were seeded in a 96-well plate at a density of 10^4 cells per well in culture media. After 24 h, the culture media were replaced with fresh media and the cells were incubated with **1** (20 μ M) in culture media for 30 min. After washing with PBS to remove the remaining sensors, various concentrations of CH₃Hg⁺ (0–200 μ M) were added to the cells in culture media, and fluorescence intensity was determined using a fluorescent microplate reader (SpectraMax GeminiEM, Molecular Devices).

Fluorescence intensity and color changes after additions of Hg^{2+} and $\mathrm{CH}_{3}\mathrm{Hg}^{+}$.

A solution of **1** (10 μ M) was prepared in spectroscopic PBS-buffer (DMSO 1%) at pH 7.4. A solution of **1** (2.0 mL) was placed in a quartz cell (10.0 mm width) and the fluorescence spectrum was recorded (excitation at 500 nm, emission at 560 nm, 25 °C).



A) Fluorescence spectra of **1** (10 μ M) upon addition of CH₃Hg⁺ and Hg²⁺ in PBS-buffer (DMSO 1%) at pH 7.4 (excitation at 500 nm, emission at 560 nm, at 25 °C): in the presence of the following metal ions: 1, only; 2, Hg²⁺ 1.0 equiv; 3, CH₃Hg⁺ 1.0 equiv; 4, CH₃Hg⁺ 10.0 equiv; 5, Fe²⁺; 6, Pb²⁺; 7, Cu²⁺; 8, Mg²⁺; 9, Ag⁺; 10, Ca²⁺; 11, Ni²⁺; 12, Co²⁺; 13, Zn²⁺; 14, Cd²⁺; 15, Mn²⁺; 16, Cr²⁺; 17, Ba²⁺; 18, Na⁺; 19, K⁺; 20, Li⁺ (addition of 20.0 equiv, respectively). B) Time dependent fluorescence intensity changes of **1** (10 μ M) with CH₃Hg⁺ and Hg²⁺ in PBS-buffer (DMSO 1%) at pH 7.4 (excitation at 500 nm, emission at 560 nm). C) Color changes of **1** (10 μ M) upon additions of 1-Hg²⁺ (1.0 equiv), 2-CH₃Hg⁺ (1.0 equiv), 3-CH₃Hg⁺ (2.0 equiv), 4-CH₃Hg⁺ (3.0 equiv), and 5-CH₃Hg⁺ (10.0 equiv) in water (DMSO 1%) at 25 °C (from left to right, respectively).



Nanomolar detection of methylmercury in HeLa cells and zebrafish.

a), d) Microscopic images of HeLa cells and zebrafish treated with 1 (20 μ M) in the absence of CH₃Hg⁺ (control). b), e) Fluorescence images of HeLa cells and zebrafish treated with 1 (20 μ M) in the absence of CH₃Hg⁺ (control). c), f) Fluorescence images of HeLa cells and zebrafish treated with both CH₃Hg⁺ (300 nM, 100 nM) and 1 (20 μ M).



Compound 5¹H-NMR data (in CDCl₃).

