# **Electronic Supplementary Information (ESI)**

Novel tryptophan metabolites, chromoazepinone A, B and C, produced by a blocked mutant of *Chromobacterium violaceum*, the biosynthetic implications and the biological activity of chromoazepinone A and B

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## 1. Spectroscopic data of the methyl ester of chromoazepinone A (5)



Fig. S1. EIMS spectrum

Fig. S2. <sup>1</sup>H NMR spectrum of Me ester of **5** inDMSO-*d*<sub>6</sub>





Fig. S3. <sup>13</sup>C NMR spectrum of Me ester of **5** inDMSO- $d_6$ 

Fig. S4. UV-Visible spectrum of Me ester of 5 in MeOH



Fig. S5. IR Spectrum of Me ester of 5



Fig. S6. Analyses and assignments of NMR data and its other physical data



Compound **5**   $[\alpha]_D^{20} = + 56.0 (c=0.16, EtOH);$ UV(MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 272 (4.30), 289 (4.31), 331 shoulder (3.89), 428 (3.70) nm Me ester of Compound **5** UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 270 (4.14), 289 (4.23), 425 (3.66) nm; IR (KBr)  $\nu_{max}$  1690, 1640, 1518, 1260, 740 cm<sup>-1</sup> HREIMS of Me ester of **5**  m/z M<sup>+</sup> 371.1272 (calcd for C<sub>22</sub>H<sub>17</sub>O<sub>3</sub>N<sub>3</sub>, 371.1270)

Methyl ester of Compound  ${\bf 5}$ 

400 MHz, DMSO-d <sub>6</sub>	The solvent peak: $\delta_{\rm H}$ =2.49, $\delta_{\rm C}$ =39.50
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Pos	ition	$^{1}\mathrm{H}$	<sup>13</sup> C	Position	<sup>1</sup> H	<sup>13</sup> C	Positio	on <sup>1</sup> H	<sup>13</sup> C	Position	1 <sup>I</sup> H	<sup>13</sup> C	Position	$^{1}\mathrm{H}$	<sup>13</sup> C
1	7	.09 (s)	102.8 (d)	5a	_	133.6(s)	<b>9</b> 7.	14 (t, J=7.9 Hz	) 120.2(d)	2'	6.46 (d, J=1.5 Hz)	123.6(d)	<b>6'</b> 7.0	06 (t, J=8.0 Hz)	121.4(d)
2			130.8 (s)	6	11.99 (br s)		10 7.	90 (d, J=7.9 H	z) 120.6(d)	3'	—	108.8(s)	<b>7'</b> 7.2	8 (d, J=8.0 Hz)	111.5(d)
3	7.0	7 (d, J=6.1	Hz) —	6a		137.5(s)	10a		125.4(s)	3'a		125.9(s)	7'a		136.0(s)
4	5.46	(d, J=6.1 l	Hz) 56.97 (d)	<b>7</b> 7.47	(d, J=7.9 Hz	z) <b>112.5(d)</b>	10b		120.4(s)	4' 7	7.81 (d, J=8.0 Hz)	118.8(d)	11	—	165.4(s)
5			182.5 (s)	8 7.37 (	(t, J=7.9 Hz)	126.4(d)	1'	10.75 (brs)		5'	6.99 (t, J=8.0 Hz)	118.8(d)	12	3.75 (3H, s)	52.40(q)

# 2. Spectroscopic data of chromoazepinone B (6)

Fig. S7. EIMS spectrum



El Mass Spectrum of compound 6

Fig. S8. <sup>1</sup>H-NMR spectrum in DMSO  $d_6$ 





Fig. S9. <sup>13</sup>C-NMR spectrum in DMSO  $d_6$ 

Fig. S10. UV-visible spectrum of **6** and its change as a function of pH.



Fig. S11. IR spectrum (KBr tablet) of 6

COSY



Fig. S12. Analyses and assignments of NMR data and its other physical data



 $\begin{array}{l} \text{UV(MeOH): pH 2.0, } \lambda_{\max} \ (\log \ \varepsilon) \ 282 \ (4.18), \ 322 \ (4.20), \ 422 \ (4.15), \ 550 \ (3.95) \ \text{nm; pH 7 \& 12, } \lambda_{\max} \ (\log \ \varepsilon) \ 282 \ (4.26), \ 322 \ (4.32), \ 405 \ (4.21) \ \text{nm.} \ \text{HREIMS } m/z \ \text{M}^+ \ 355.0930 \ (\text{calcd for } C_2 | H_{13} O_3 N_3, \ 355.0957) \end{array}$ The solvent peak:  $\delta_{\rm H}$ =2.49,  $\delta_{\rm C}$ =39.50 400 MHz, DMSO-d<sub>6</sub>

Positior	<sup>1</sup> H	<sup>13</sup> C	Position	$^{1}\mathrm{H}$	<sup>13</sup> C	Position	1 <sup>1</sup> H	<sup>13</sup> C	Position	1 <sup>1</sup> H	<sup>13</sup> C	Position	$^{1}\mathrm{H}$	<sup>13</sup> C
1	8.73 (s)	119.0	5a	_	140.0	<b>9</b> <sup>7.4</sup>	0 (t, J=7.6 Hz)	122.2	2'	9.00 (s)	133.1	6'	7.22 (t, J=7.9 Hz)	122.6
2		134.8	<b>6</b> 13.1	17 (very brs)		10 8.3	9 (d, J=7.6 Hz)	121.9	3'	—	115.4	7'	7.48 (d, J=7.9 Hz)	111.6
3		_	6a -		139.0	10a	—	125.2	3'a		126.8	7'a	—	136.3
4	_	153.4	<b>7</b> 7.71	(d, J=7.6 Hz)	113.2	10b		122.4	4'	9.38 (d, J=7.9 Hz)	124.1	11	—	168.5
5	—	170.6	<b>8</b> 7.62	(t, J=7.6 Hz)	128.6	1'	11.80 (s)	—	5'	7.17(t, J=7.9 Hz)	121.1			

### 3. Spectroscopic data of methyl ester of chromoazepinone B (6)

Fig. S13. EIMS spectrum



Fig. S14. <sup>1</sup>H NMR spectrum in DMSO  $d_6$ 



Fig. S15. <sup>13</sup>C NMR spectrum in DMSO  $d_6$ 



Fig. S16. UV-visible spectrum dissolved in MeOH (8.67x10<sup>-5</sup>M)



S10

#### IR spectrum of compound 6-Me ester; Not Measured

Fig. S17. Analyses and assignments of NMR data and its physical data



### 4. Spectroscopic data of the methyl ester of chromoazepinone C (7)

#### Fig. S18. EIMS spectrum







Fig. S20. <sup>13</sup>C NMR spectrum of compound 7-Me ester in DMSO  $d_6$ 







Fig. S22. IR spectrum of compound 7- Me ester (KBr tablet)



#### Fig. S23. Analyses and assignments of NMR data of -Me ester of 7 and its physical data



 $\begin{array}{l} \mbox{Compound 7} \\ \left[ \alpha \right]_D{}^{20} = +\ 26.6\ (c{=}0.15,\ EtOH); \\ \mbox{Me ester of Compound 7} \\ \mbox{UV (MeOH)} \ \lambda_{max}\ (\log\ e)\ 280\ (4.04),\ 285 \\ (3.99),\ 347\ (3.93)\ nm;\ IR\ (KBr)\ \nu_{max}\ 1702, \\ 1640,\ 1620,\ 1258,\ 740\ cm^{-1} \end{array}$ 

HREIMS of Me ester of **7** *m/z* M<sup>+</sup> 371.1275 (calcd for C<sub>22</sub>H<sub>17</sub>O<sub>3</sub>N<sub>3</sub>, 371.1270)

400 MHz, DMSO- $d_6$  The solvent peak:  $\delta_{\rm H}$ =2.49,  $\delta_{\rm C}$ =39.50

Positio	m <sup>1</sup> H	<sup>13</sup> C	Position	$^{1}\mathrm{H}$	<sup>13</sup> C	Positio	n <sup>1</sup> H	<sup>13</sup> C	Position	<sup>1</sup> H	<sup>13</sup> C	Position	$^{1}\mathrm{H}$	<sup>13</sup> C
1	7.67 (s)	115.5	5a	_	136.3 <sup><i>a</i></sup>	<b>9</b> 7	.15 (t, J=8.0 Hz)	120.4	<b>2'</b> 6.6	7 (d, <i>J</i> =1.9 Hz)	122.8	<b>6'</b> 7.0	5(t, <i>J</i> =8.0 Hz)	121.4
2	_	119.6	6 11	.86 (brs)		10 7.	75 (d, J=8.0 Hz)	118.1	3'		106.9	<b>7'</b> 7.3	l (d, J=8.0 Hz)	111.5
3	9.11 (s)		6a -		136.3	10a		126.0	3'a	—	125.7	7'a	—	<b>136.2</b> <sup><i>a</i></sup>
4	<u> </u>	166.3	<b>7</b> 7.47	(d, J=8.0 Hz)	112.0	10b		107.9	<b>4'</b> 7.5	5 (d, J=8.0 Hz)	118.2	11	—	163.8
5	5.42 (d)	45.51	<b>8</b> 7.21	(t, J=8.0 Hz)	122.4	1'	10.89 (brs)		<b>5'</b> 6.9	97 (t, <i>J</i> =8.0 Hz)	118.7	12 3	.74 (3H, s)	52.37

The chemical shifts of C-5a, C-6a and C-7'a are interchangeable due to the very close values.

# 5. Spectroscopic data of arcyriarubin A (9)

Fig. S24. EIMS spectrum of compound 10



Fig. S25. <sup>1</sup>H-NMR spectrum of compound **9** dissolved in DMSO- $d_6$ 



Fig. S26. <sup>13</sup>C-NMR spectrum of compound **9** dissolved in DMSO- $d_6$ 



Fig. S27. UV-visible spectrum of compound **9** dissolved in MeOH.



Fig. S28. IR spectrum of compound 9 (KBr)



# Fig. S29. Analyses and assignments of NMR data of compound 9 and its physical data



UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ) 276.5 (4.05), 369.0 (3.71), 458.0 (3.85) nm; IR (KBr)  $\nu_{max}$  3390, 1750, 1700, 1530, 1340, 740 cm<sup>-1</sup> HREIMS *m/z* M<sup>+</sup> 327.1020 (calcd for C<sub>20</sub>H<sub>13</sub>N<sub>3</sub>O<sub>2</sub>, 327.1008)

400 MHz, DMSO- <i>d</i> <sub>6</sub>	The solvent peak: $\delta_{\rm H}$ =2.49, $\delta_{\rm C}$ =39.50
100 11112, 211100 46	The solvent peak. of 2.19, of 59.50

Position <sup>1</sup> H <sup>13</sup> C	Position <sup>1</sup> H	<sup>13</sup> C	Position <sup>1</sup> H	<sup>13</sup> C	Position	$^{1}\mathrm{H}$	<sup>13</sup> C
<b>1 &amp; 1'</b> 11.65 (s) —	3a & 3'a —	126.0 (s)	<b>6 &amp; 6'</b> 6.99 (t, 8.0 Hz)	122.2 (d)	8 & 8'		128.3 (s)
<b>2 &amp; 2'</b> 7.73 (d, 2.5 Hz) 129.7 (d)	<b>4 &amp; 4'</b> 6.80 (d, 8.0 Hz)	121.5 (d)	<b>7 &amp; 7'</b> 7.36 (d, 8.0 Hz)	112.3 (d)	9 & 9'	—	173.6 (s)
<b>3 &amp; 3'</b> 106.2 (s)	5 & 5' 6.64 (t, 8.0 Hz)	119.9 (d)	7a & 7'a 🛛 ——	136.57 (s)	10	10.89 (s)	