

The synthesis of double-headed nucleosides by the CuAAC reaction and their effect in secondary nucleic acid structures

Anna S. Jørgensen, Khalil I. Shaikh, Gerald Enderlin, Elise Ivarsen, Surender Kumar and Poul Nielsen*

Supporting information

Index

Synthesis of <i>N</i> 6-benzoyl- <i>N</i> 9-propargyladenine	S2
MALDI MS-data for oligonucleotides	S3
Melting temperatures – multiple incorporations of M	S4
Selected NMR-spectra for compound 6 , 8-11 , 13-15 , 18 and 19	S5-S20

Preparation of *N*6-Benzoyl-*N*9-propargyladenine (**6**)

*N*9-propargyladenine, **10**, (123 mg, 0.71 mmol) was coevaporated twice with anhydrous pyridine and dissolved in the same solvent (4 mL). Benzoyl chloride (0.10 mL, 0.86 mmol) was added and the reaction mixture was stirred at room temperature for 17 h and then at 120 °C for 6 h. Methanol (0.2 mL) was added and the mixture was concentrated under reduced pressure. The residue was coevaporated twice with toluene and purified by silica gel column chromatography (0 – 10% MeOH in CH₂Cl₂) to give a crude product **11** as a white solid (101 mg, ~85% pure, estimated yield 43%), which was used without further purification; *R*_f 0.58 (10% MeOH in CH₂Cl₂); ¹H NMR (400 MHz, DMSO-*d*₆) δ 11.21 (br s, 1H, NH), 8.78 (s, 1H, H2), 8.55 (s, 1H, H8), 8.09 – 8.03 (m, 2H, Ph), 7.68 – 7.48 (m, 3H, Ph), 5.19 (d, 2H, *J* = 2.5 Hz, CH₂), 3.54 (t, 1H, *J* = 2.5 Hz, C≡CH); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 165.7 (C=O), 151.9, 151.7 (C4, C6), 150.3 (C2), 143.9 (C8), 133.4, 132.8, 132.4, 130.8, 129.2, 128.5, 128.4 (Ph), 125.2 (C5), 77.9 (C≡CH), 76.2(C≡CH), 32.7 (CH₂); HRMS-ESI: *m/z*: 300.0842 [MNa]⁺; calcd (C₁₅H₁₁N₅O-Na⁺): 300.0856.

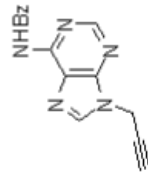
MALDI MS-data of oligonucleotides

Name	Sequence	MW calcd	MW exp
K6	5'-d(GCT CAC K CT CCC A)-3'	4021.7	4022.1
L6	5'-d(GCT CAC L CT CCC A)-3'	4030.7	4026.6
M6	5'-d(GCT CAC M CT CCC A)-3'	4049.7	4050.6
K1	3'-d(GCG K AT AAG CG)-5'	3572.4	3576.0
K2	3'-d(GCG T AK AAG CG)-5'	3572.4	3573.2
K3	5'-d(CGC ATA T KC GC)-3'	3483.4	3483.8
K4	5'-d(CGC ATA K TC GC)-3'	3483.4	3480.0
K5	5'-d(CGC A KA TTC GC)-3'	3483.4	3479.2
L1	3'-d(GCG L AT AAG CG)-5'	3581.4	3577.7
L2	3'-d(GCG T AL AAG CG)-5'	3581.4	3578.6
L3	5'-d(CGC ATA T LC GC)-3'	3492.4	3488.2
L4	5'-d(CGC ATA L TC GC)-3'	3492.4	3491.6
L5	5'-d(CGC A LA TTC GC)-3'	3492.4	3490.8
M1	3-d(GCG M AT AAG CG)-5'	3600.5	3596.0
M2	3-d(GCG T AM AAG CG)-5'	3600.5	3597.4
M3	5'-d(CGC ATA T MC GC)-3'	3511.4	3508.8
M4	5'-d(CGC ATA M TC GC)-3'	3511.4	3514.3
N1	3-d(GCG N AT AAG CG)-5'	3628	3626
N2	3-d(GCG T AN AAG CG)-5'	3628	3626
N3	5'-d(CGC ATA T NC GC)-3'	3538	3539
N4	5'-d(CGC ATA N TC GC)-3'	3538	3539
M7	3-d(GCG M AM AAG CG)-5'	3819.7	3815.1
M8	5'-d(CGC A MA M TC GC)-3'	3730.6	3732.8

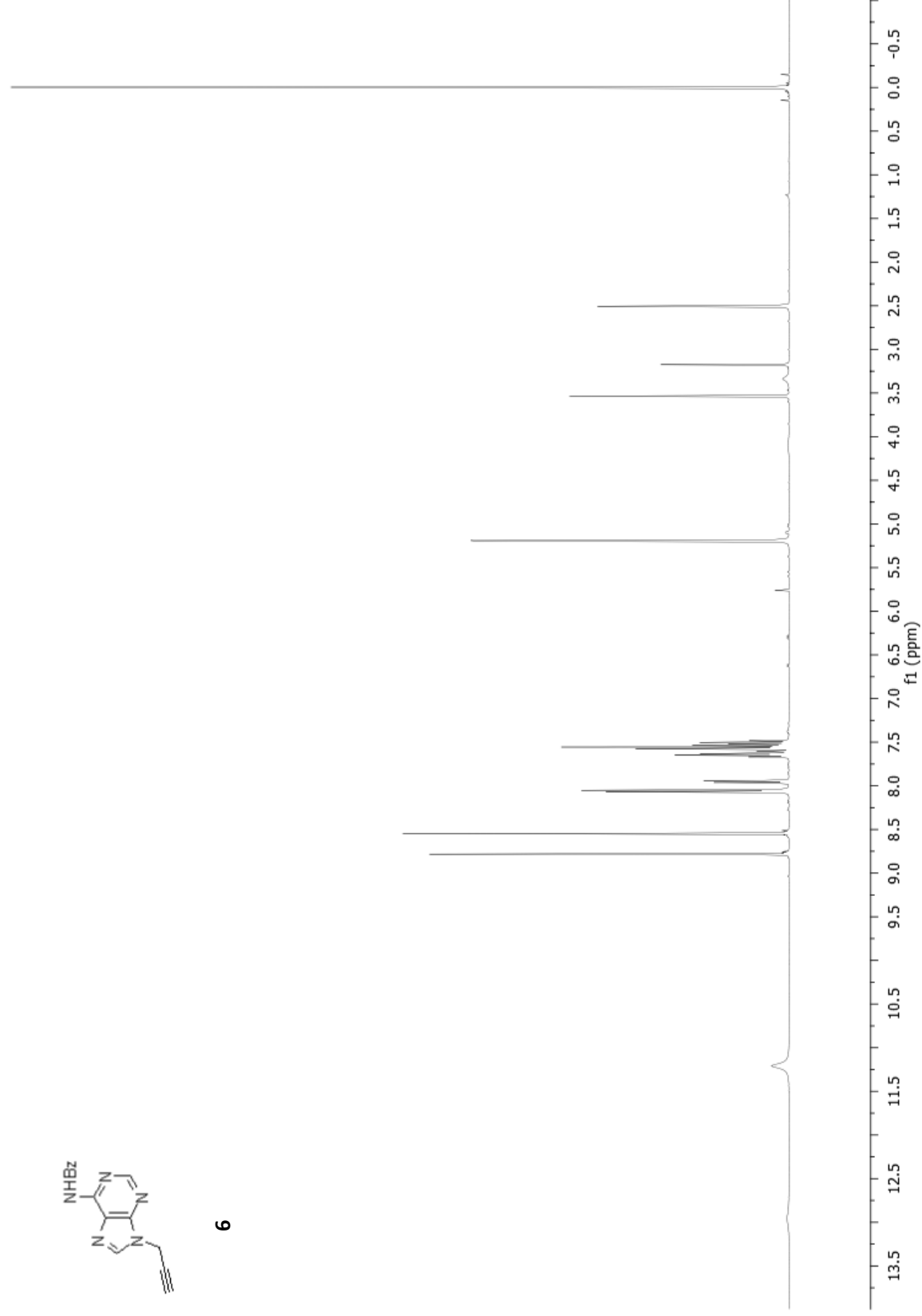
Multiple incorporations of **M** in zipper motifs

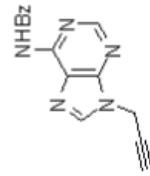
$\Delta T_m / ^\circ\text{C}^a$ [$\Delta\Delta T_m$] / $^\circ\text{C}^b$						
ON	Duplex		Zipper	ON	Duplex	
ref.	5'-d(CGC ATA TTC GC)			M4	5'-d(CGC ATA MTC GC)	-14.0
ref.	3'-d(GCG TAT AAG CG)		-1	M2	3'-d(GCG TAM AAG CG)	[-1.5]
	5'-d(CGC ATA TTC GC)	-6.0	-2	M3	5'-d(CGC ATA TMC GC)	-14.0
M1	3'-d(GCG MAT AAG CG)			M2	3'-d(GCG TAM AAG CG)	[-1.5]
	5'-d(CGC ATA TTC GC)	-6.0	-3	M4	5'-d(CGC ATA MTC GC)	-12.0
M2	3'-d(GCG TAM AAG CG)			M1	3'-d(GCG MAT AAG CG)	[+0.5]
	5'-d(CGC ATA TTC GC)	-12.0	-4	M3	5'-d(CGC ATA TMC GC)	-12.0
M7	3'-d(GCG MAM AAG CG)			M1	3'-d(GCG MAT AAG CG)	[+0.5]
	5'-d(CGC ATA TMC GC)	-6.5	-1/-3	M4	5'-d(CGC ATA MTC GC)	-19.0
M3	3'-d(GCG TAT AAG CG)			M7	3'-d(GCG MAM AAG CG)	[-0.5]
	5'-d(CGC ATA MTC GC)	-6.5	-1/-3	M8	5'-d(CGC AMA MTC GC)	-19.0
M4	3'-d(GCG TAT AAG CG)			M1	3'-d(GCG MAT AAG CG)	[+1.0]
	5'-d(CGC AMA MTC GC)	-14.0	-2/-4	M3	5'-d(CGC ATA TMC GC)	-21.0
M8	3'-d(GCG TAT AAG CG)			M7	3'-d(GCG MAM AAG CG)	[-3.0]
			+1/-1	M8	5'-d(CGC AMA MTC GC)	-22.5
				M2	3'-d(GCG TAM AAG CG)	[-2.5]
			+1/-1	M8	5'-d(CGC AMA MTC GC)	-30.5
				M7	3'-d(GCG MAM AAG CG)	[-4.5]

^a Differences in melting temperatures as compared to the unmodified duplex; $\Delta T_m = T_{m(\text{duplex})} - T_{m(\text{ref})}$. Melting temperatures (T_m values/ $^\circ\text{C}$) were obtained from the maxima of the first derivatives of the melting curves (A_{260} v.s. temperature) recorded in a medium salt buffer (Na_2HPO_4 (2.5 mM), NaH_2PO_4 (5 mM), NaCl (100 mM), EDTA (0.1 mM), pH 7.0) using 1.0 μM concentrations of each strand. **M** corresponds to the incorporation of the amidite **15**. ^b Differences in melting temperatures as compared to singly modified duplexes; $\Delta\Delta T_m = \Delta T_{m(x,y)} - (\Delta T_{m(x,\text{ref})} + \Delta T_{m(\text{ref},y)})$.

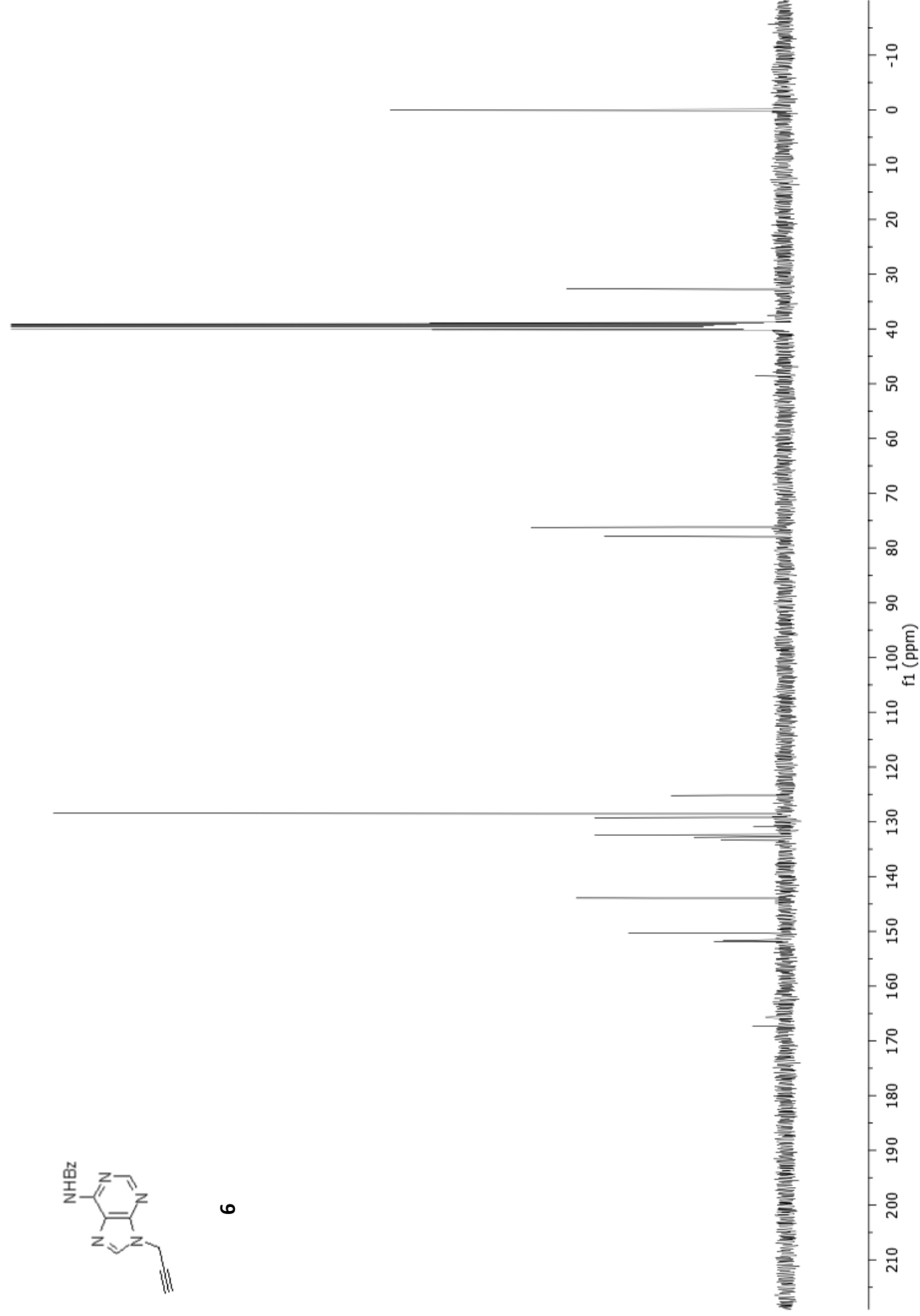


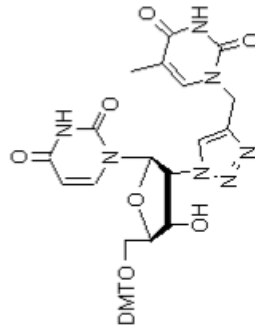
6



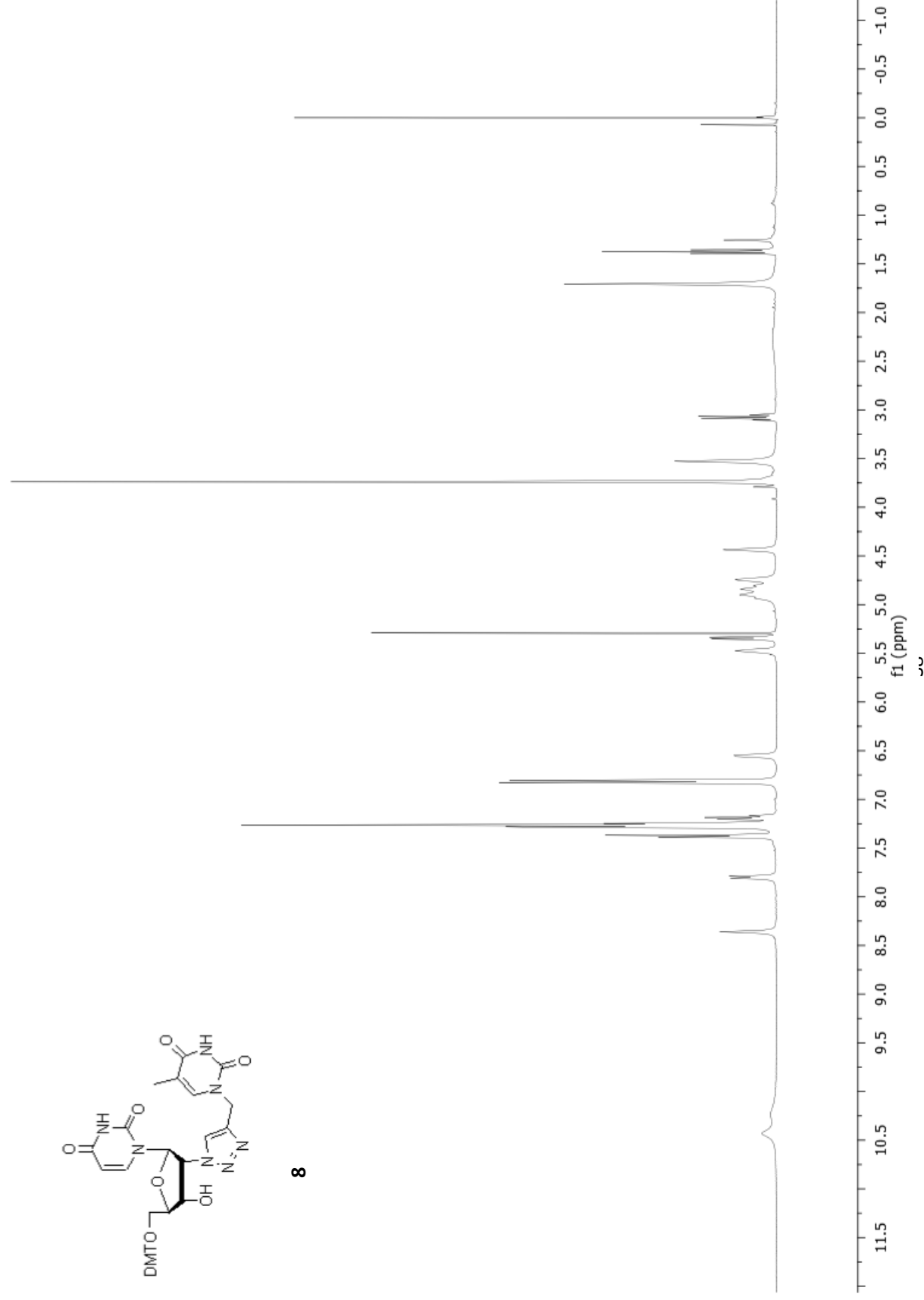


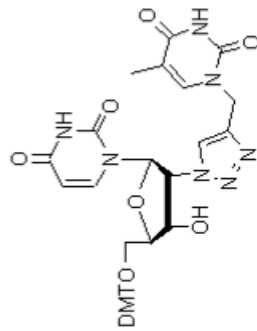
6



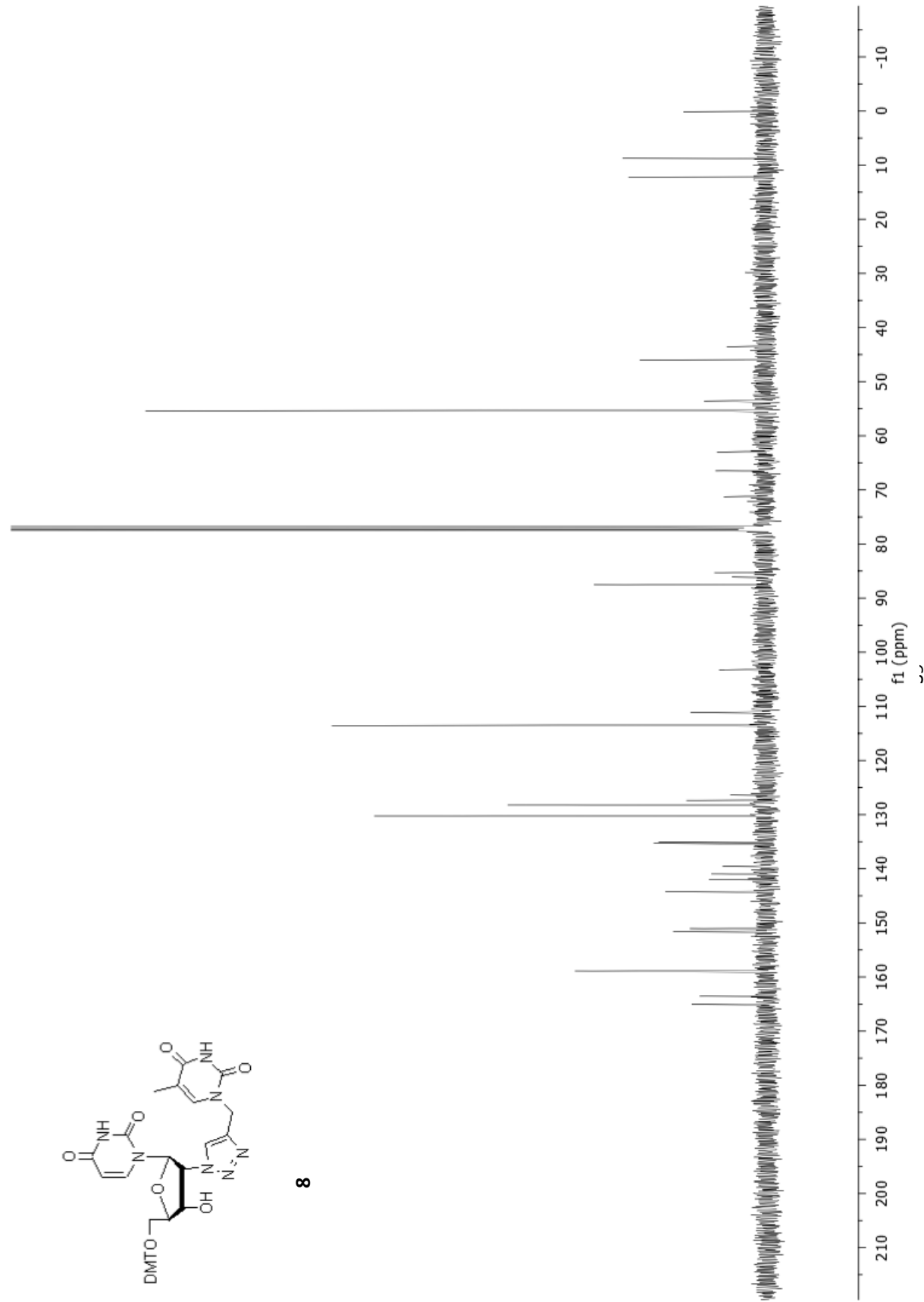


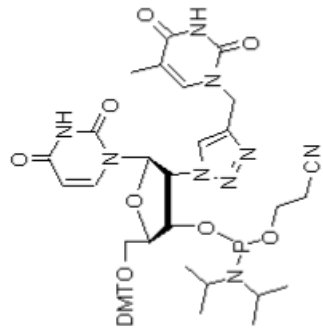
8



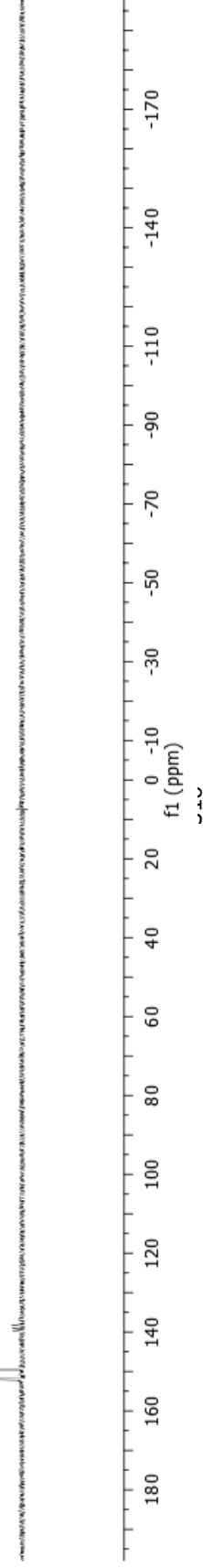


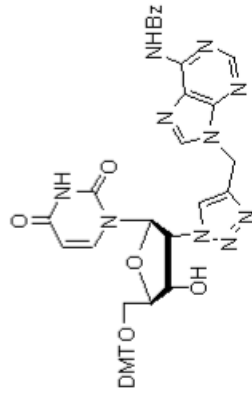
8



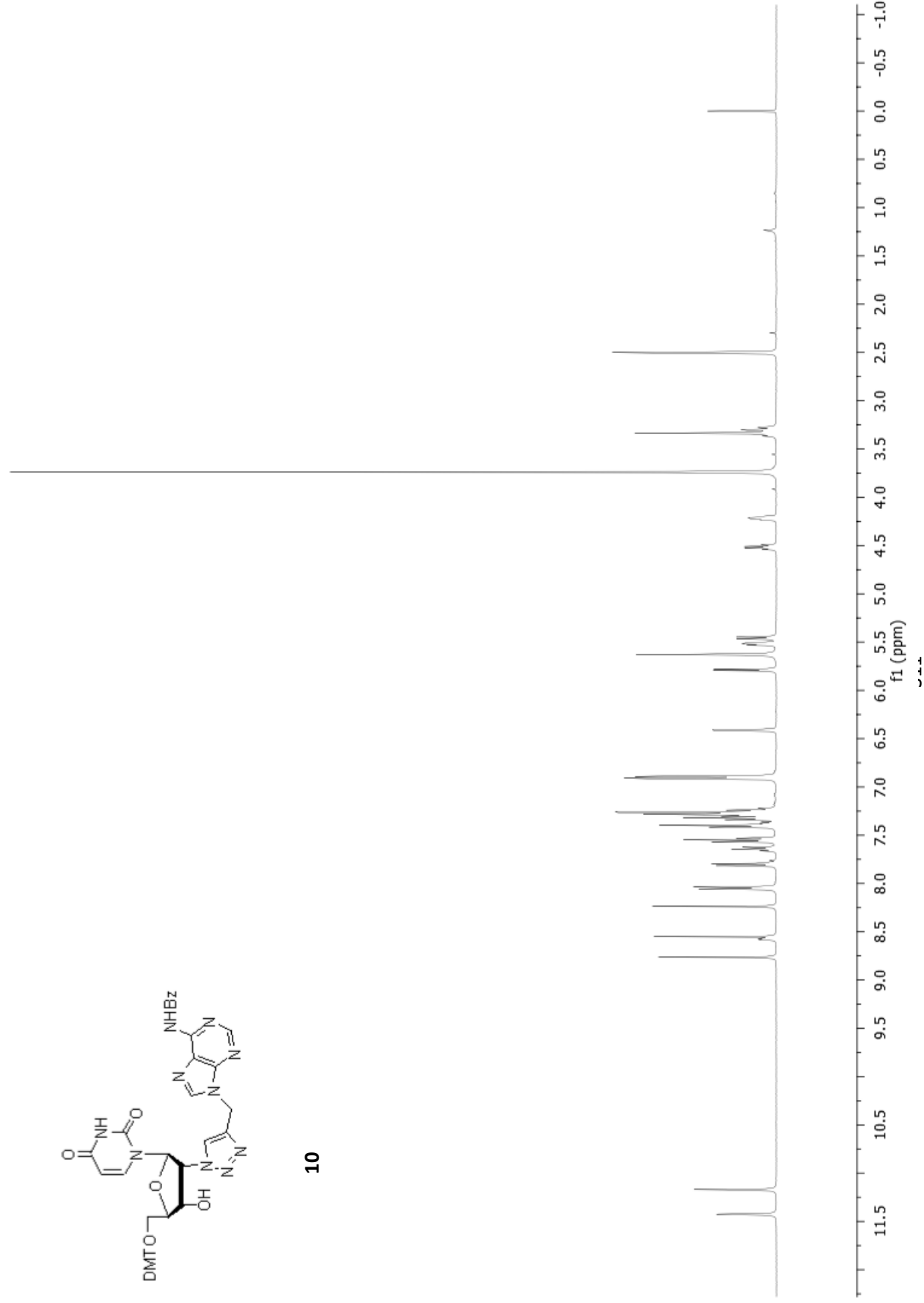


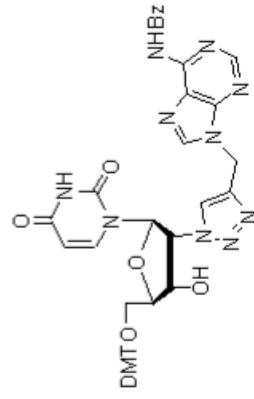
9



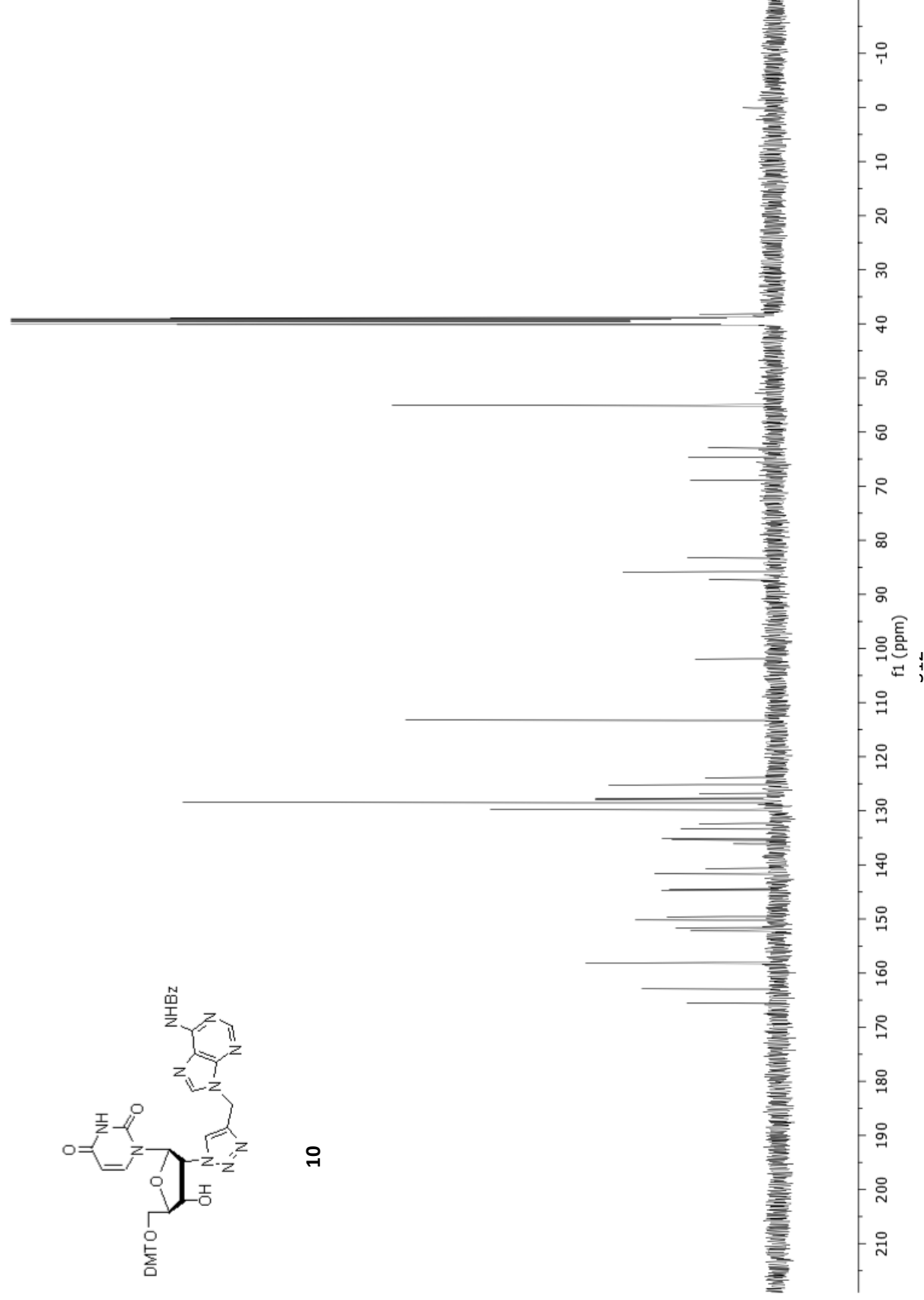


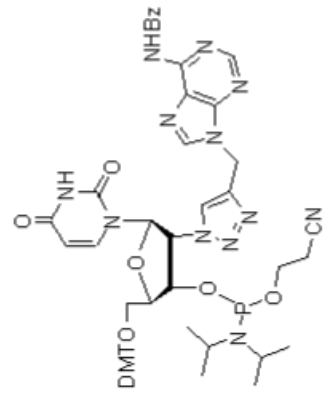
10



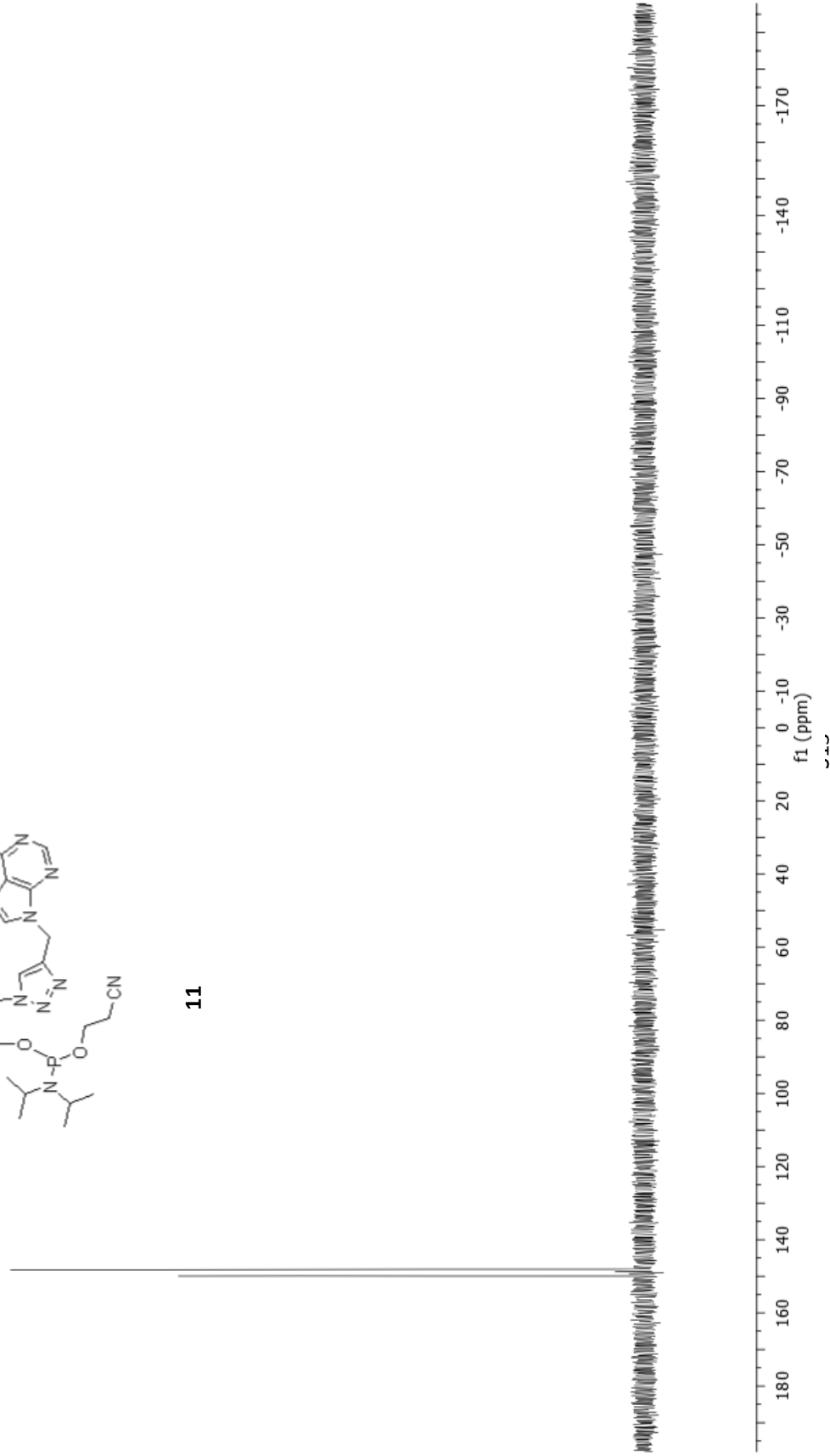


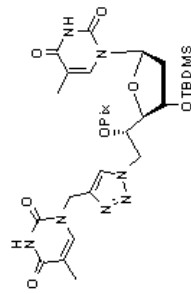
10



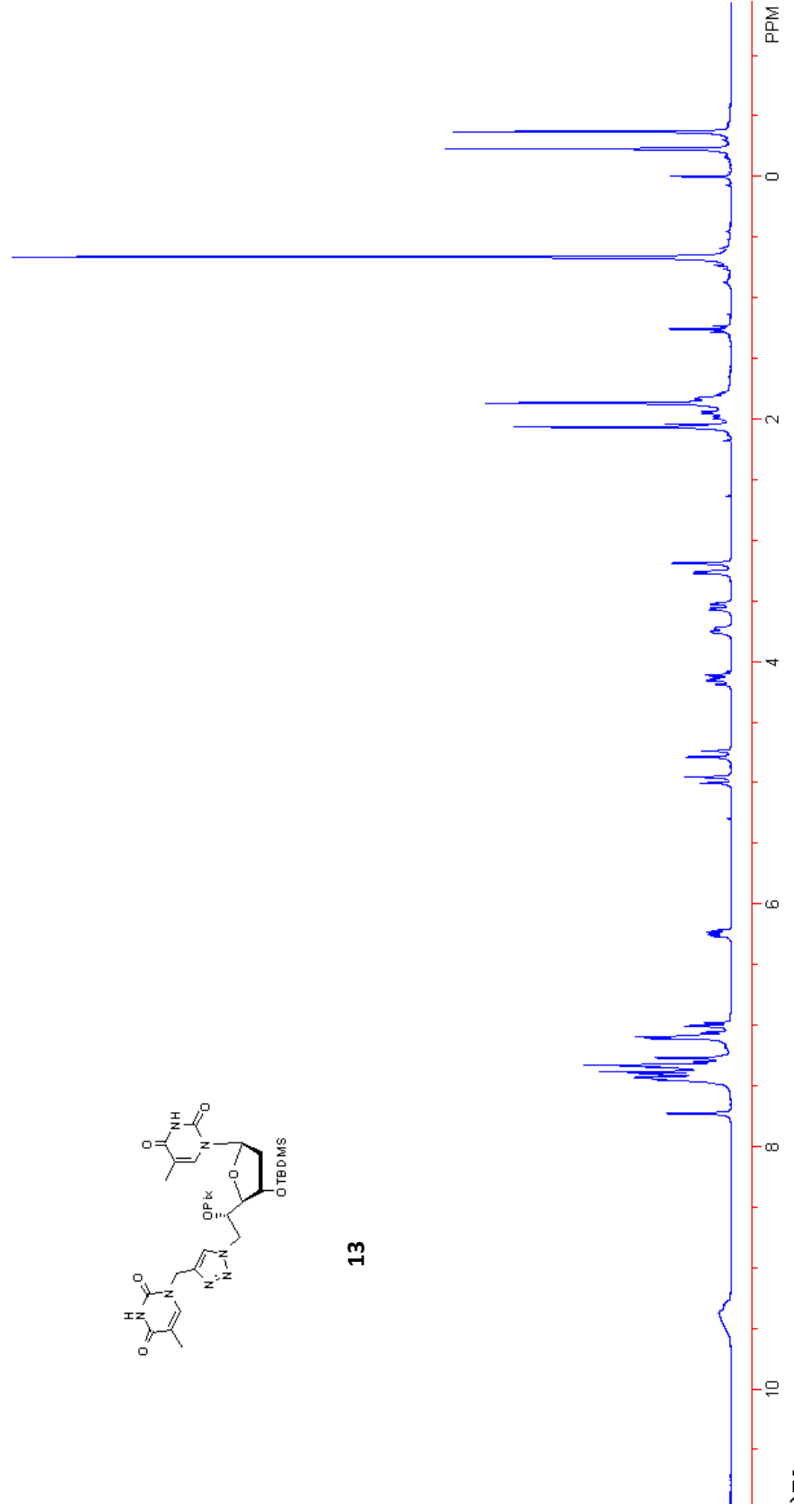


11

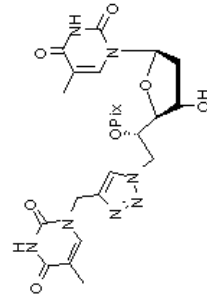




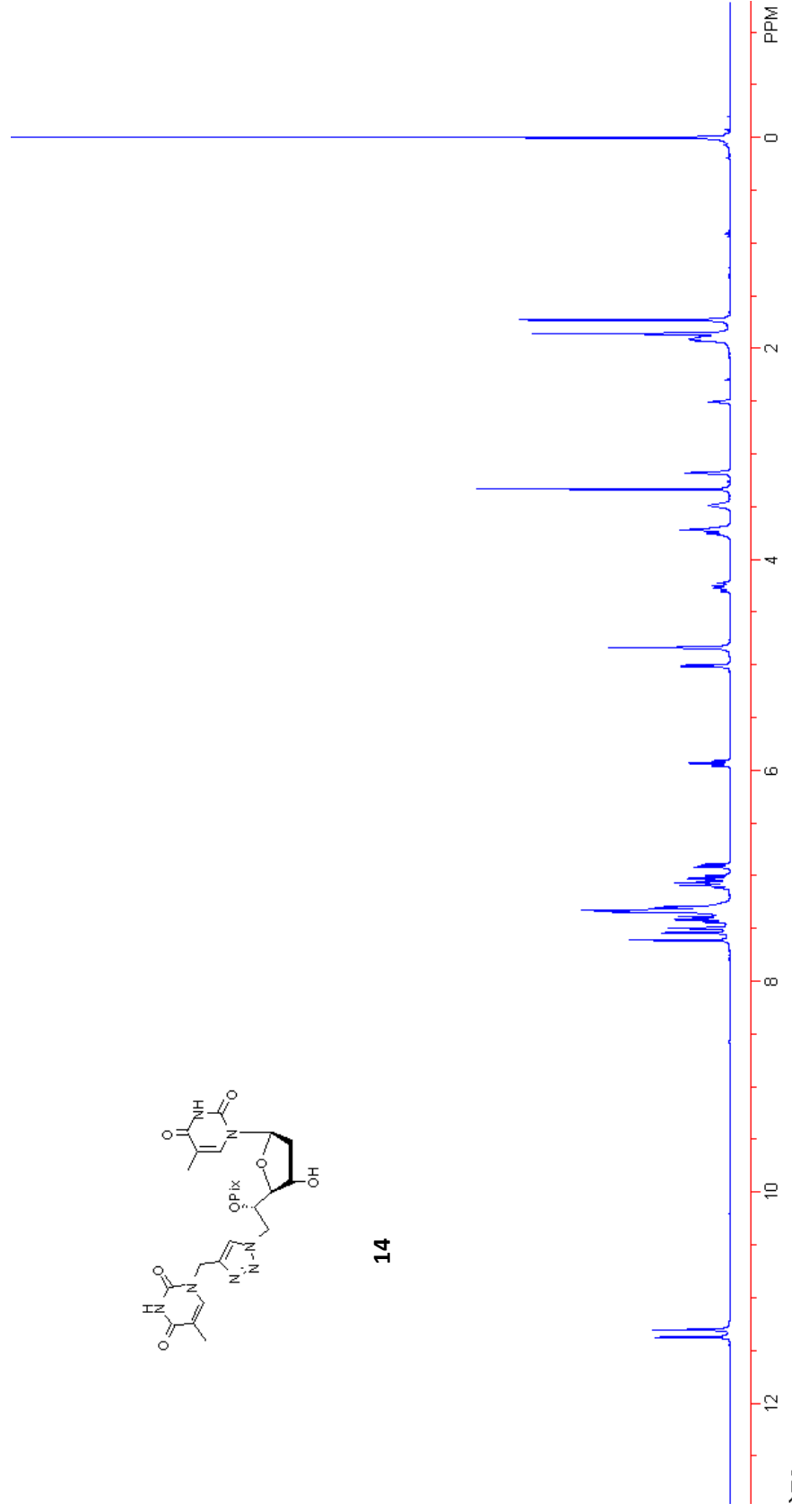
13

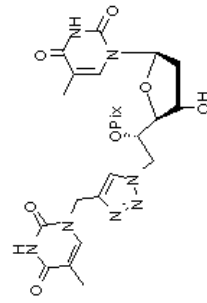


S14

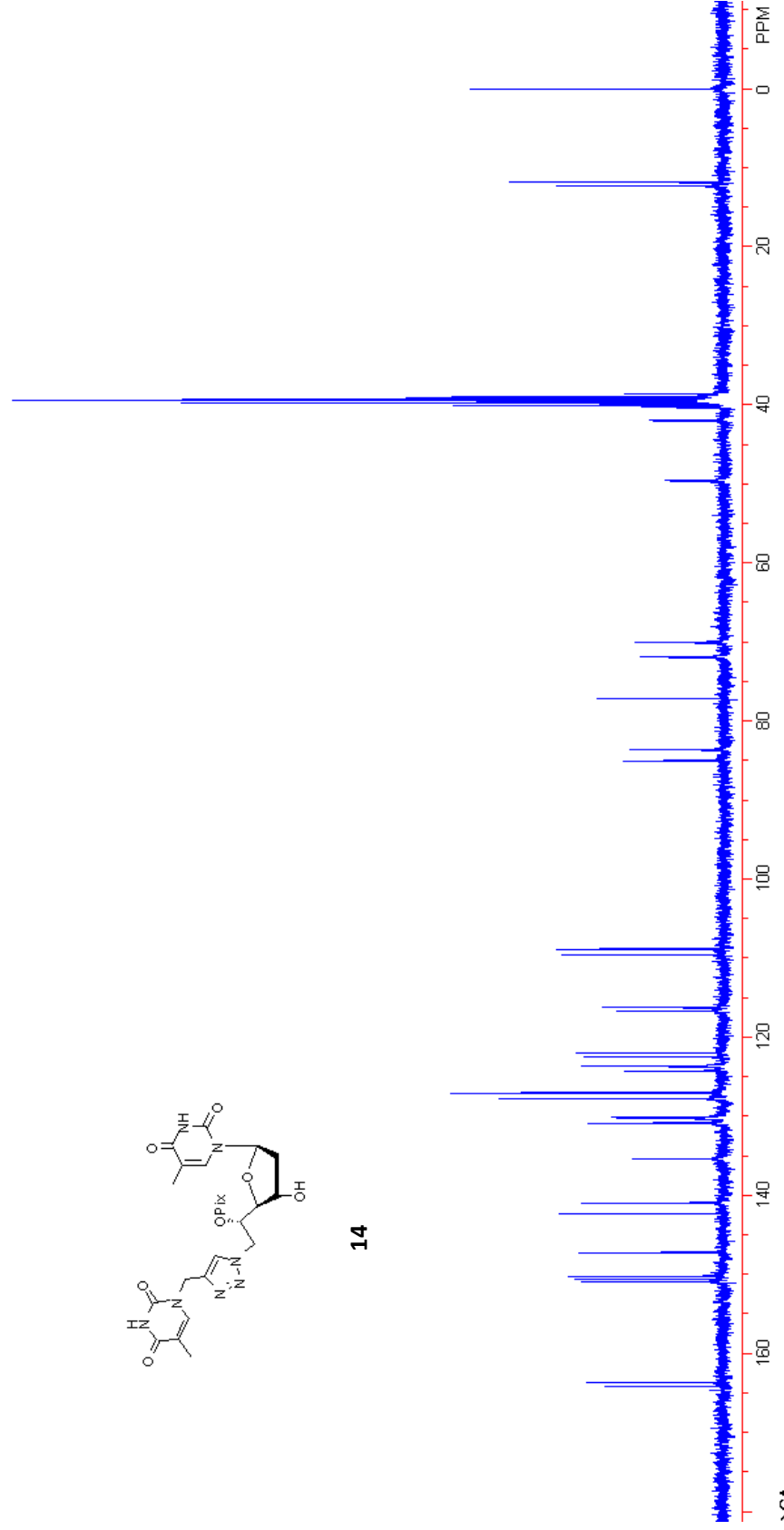


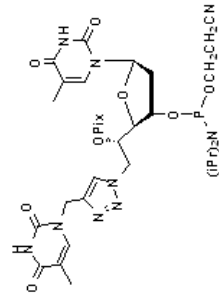
14



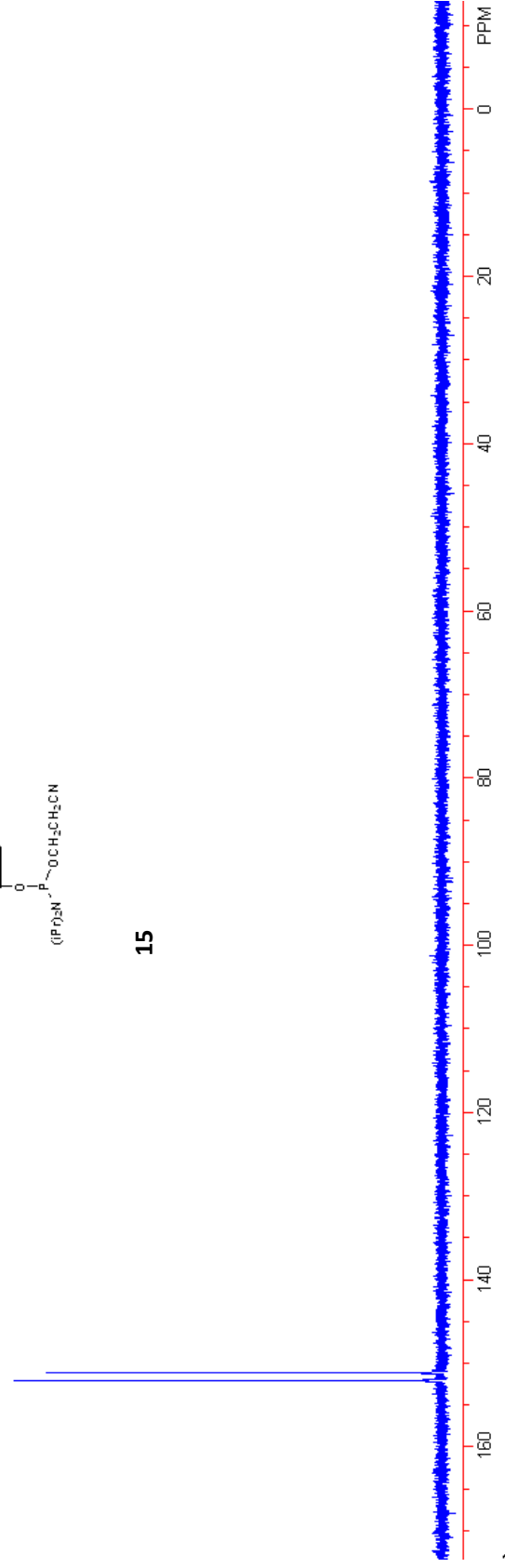


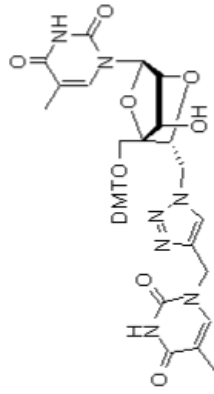
14



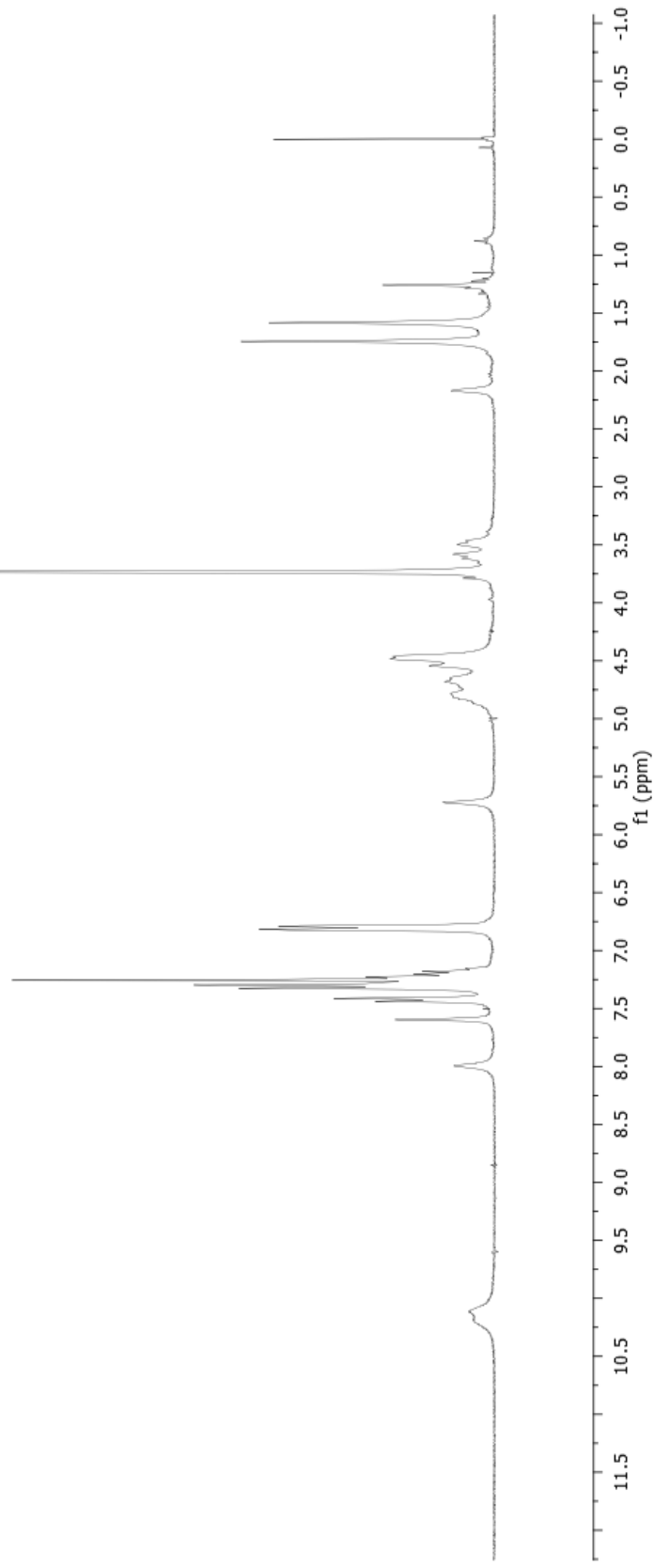


15

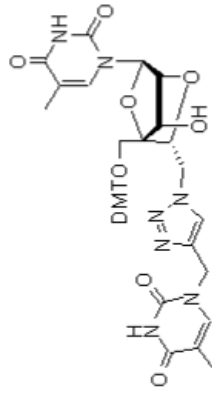




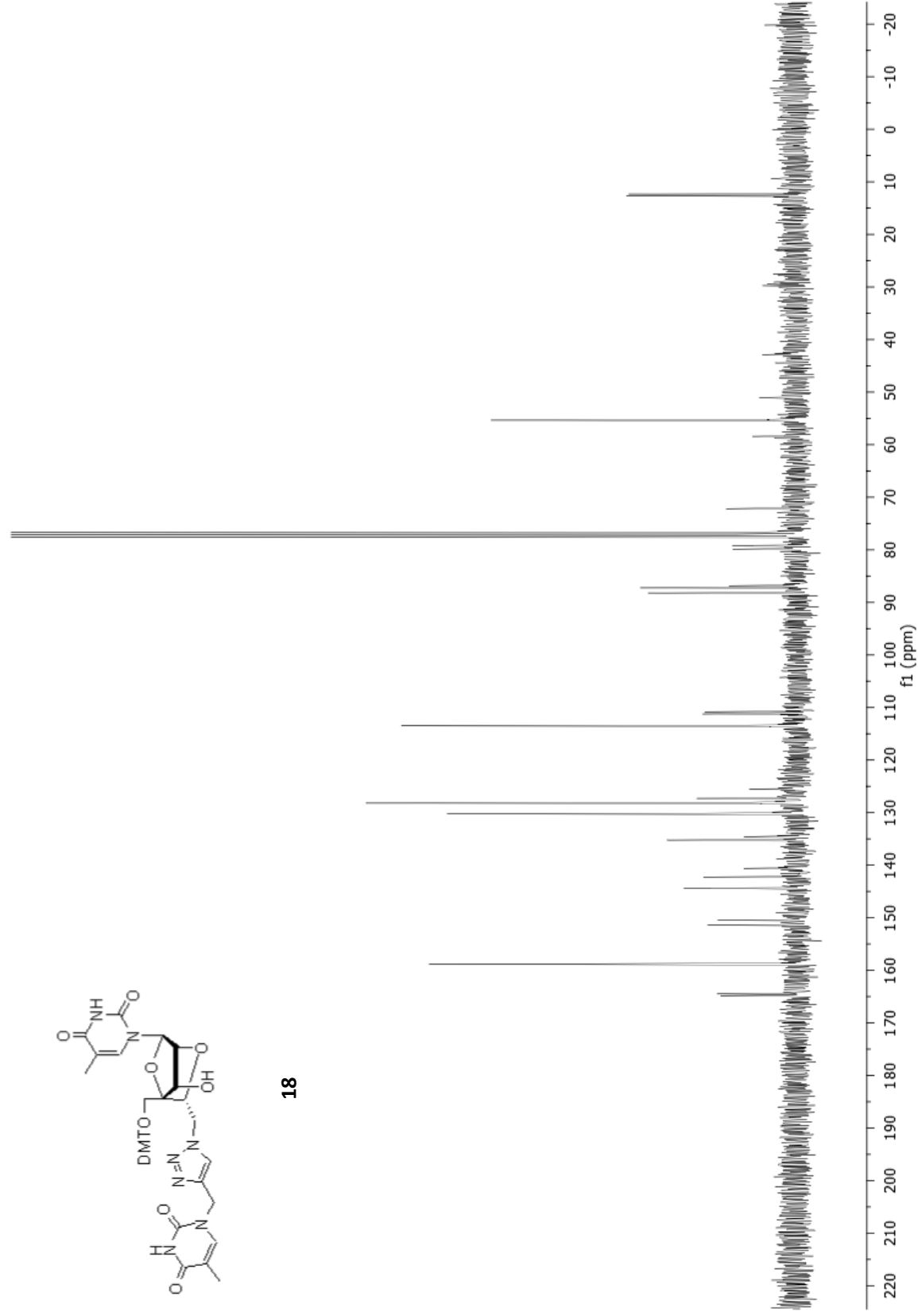
18



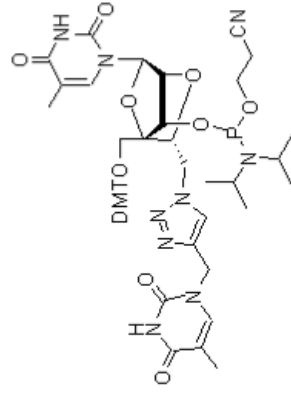
S19



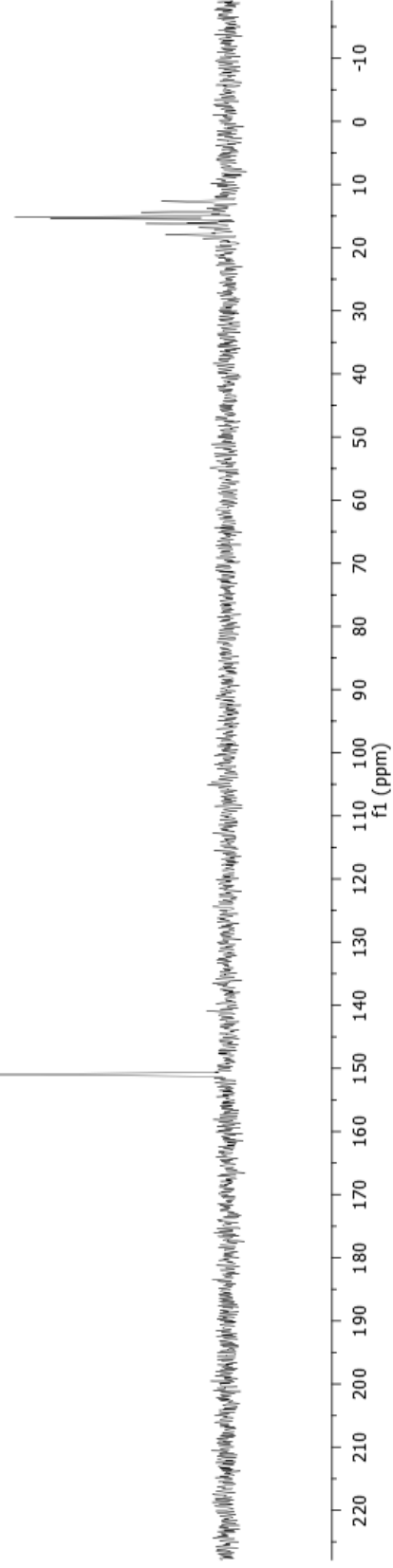
18



S20



19



S21