

## Electronic Supplementary Information for: Cyclodextrin-alloxazin conjugates for organocatalytic enantioselective sulfoxidations

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# 1 Characterizations of synthesized compounds

## 1.1 NMR spectra of compounds **4** and **5a**

The NMR spectra of **4** were measured on Bruker AVANCE-600 instrument ( $^1\text{H}$  at 600.13 MHz and  $^{13}\text{C}$  at 150.9 MHz) in DMSO. The NMR spectra of **5a** were measured on Bruker AVANCE-600 instrument ( $^1\text{H}$  at 600.13 MHz and  $^{13}\text{C}$  at 150.9 MHz) with a cryo-probe in  $\text{D}_2\text{O}$  at 37 °C. Spectra are referenced to dioxane (added as internal standard) using  $\delta_{\text{H}}(\text{dioxane}) = 3.75$  ppm and  $\delta_{\text{C}}(\text{dioxane}) = 69.3$  ppm.

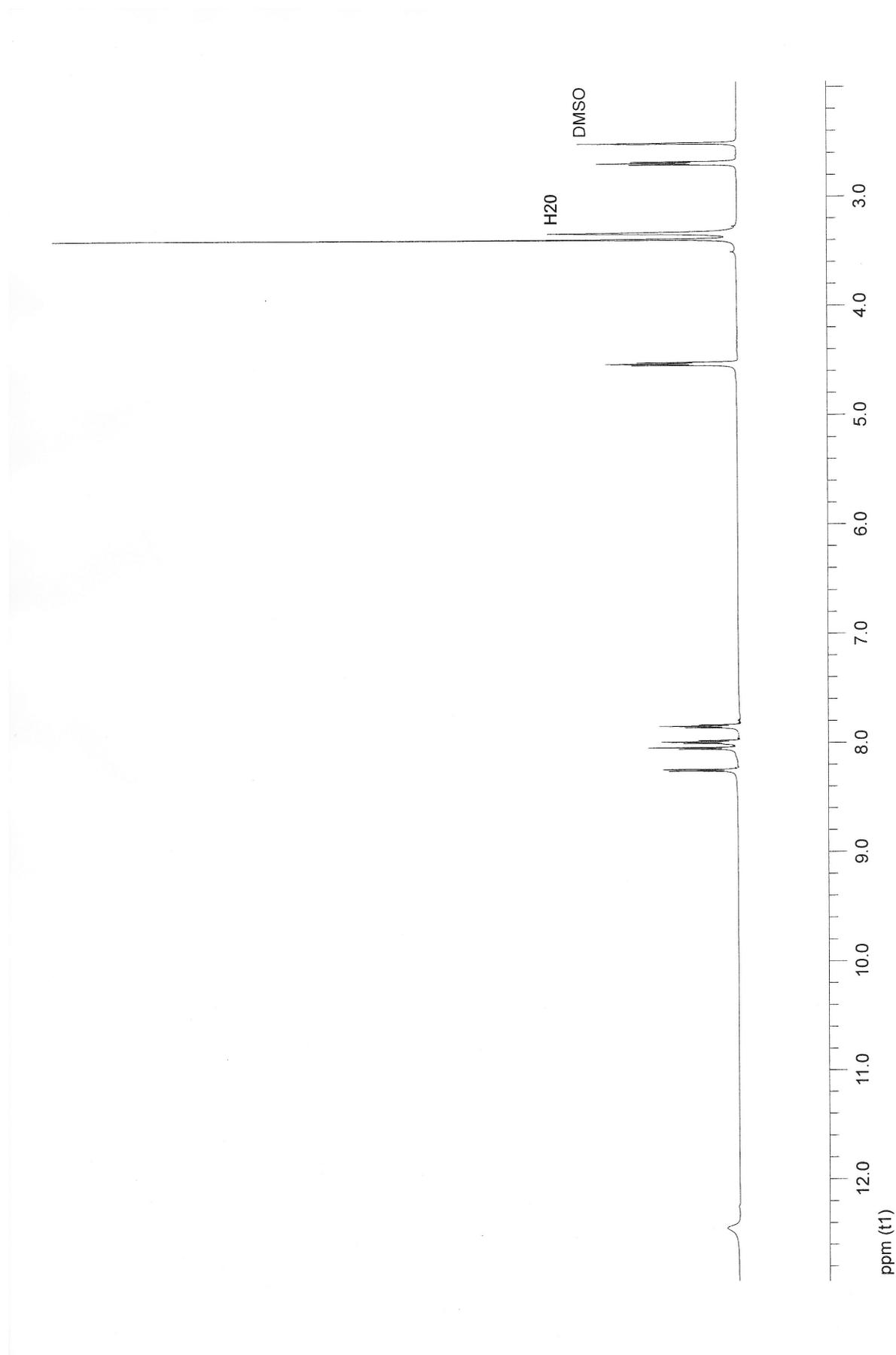


Figure S1:  $^1\text{H-NMR}$  spectrum of compound 4

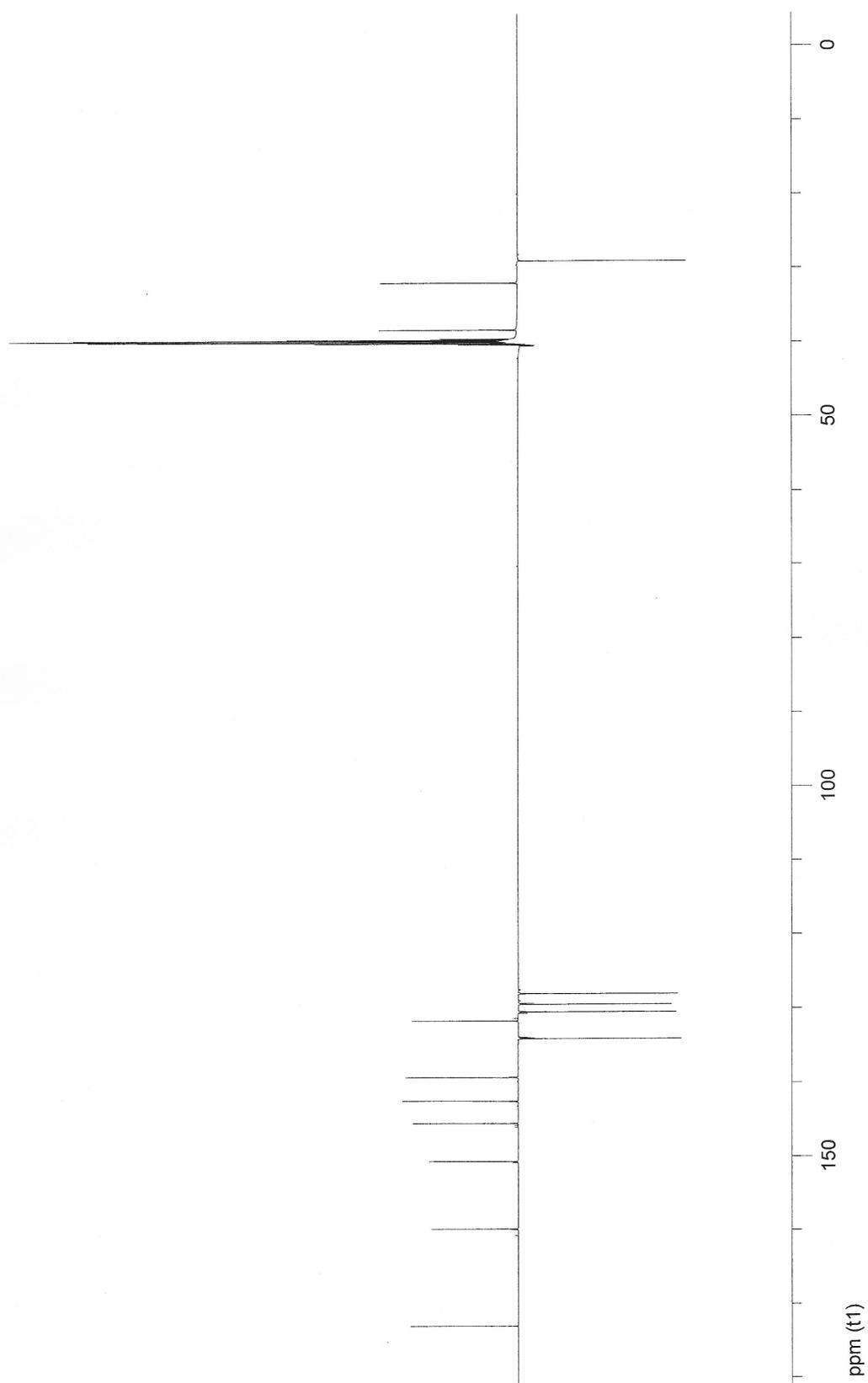


Figure S2:  $^{13}\text{C}$ -NMR spectrum of compound 4

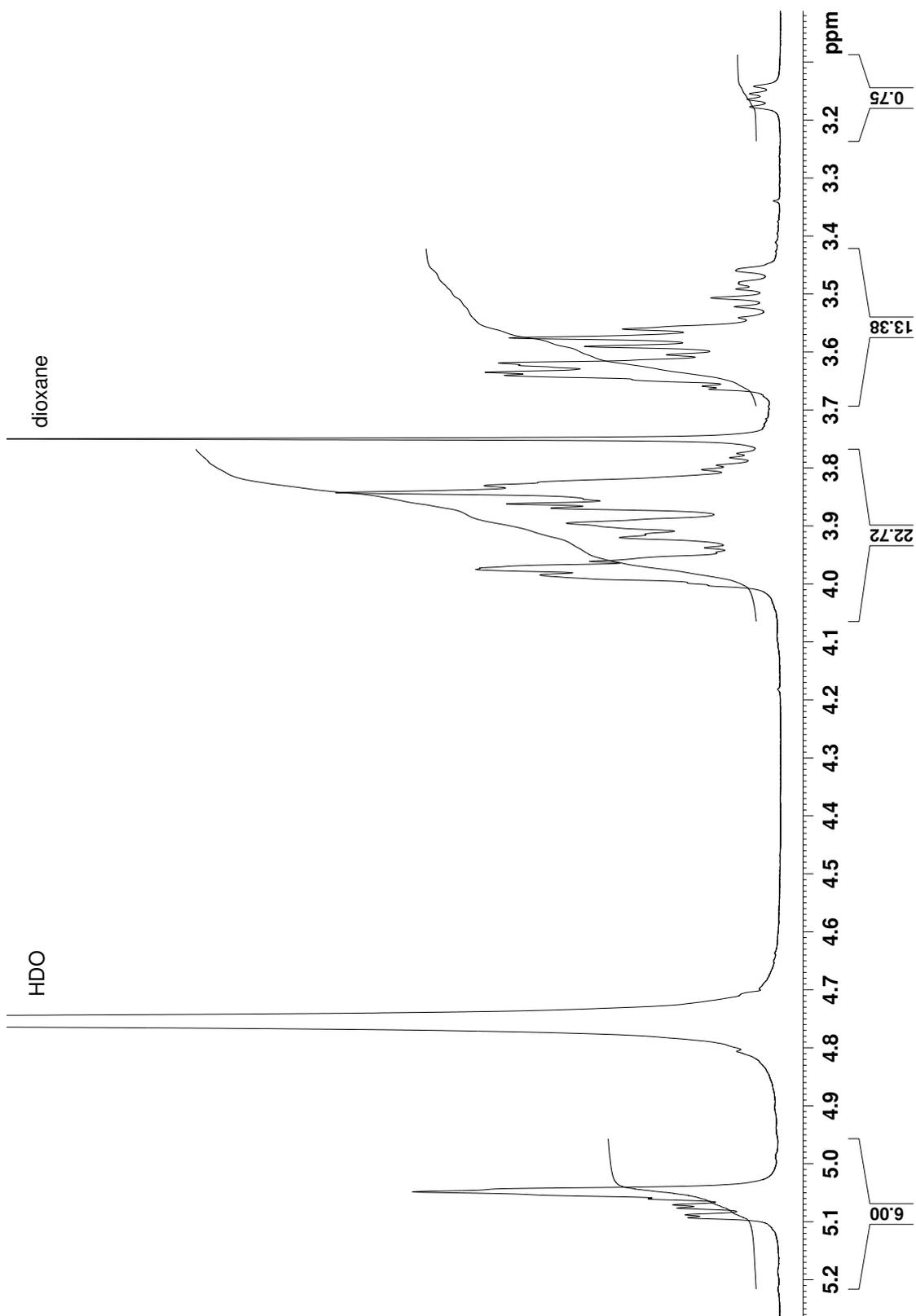


Figure S3:  $^1\text{H-NMR}$  spectrum of compound **5a**

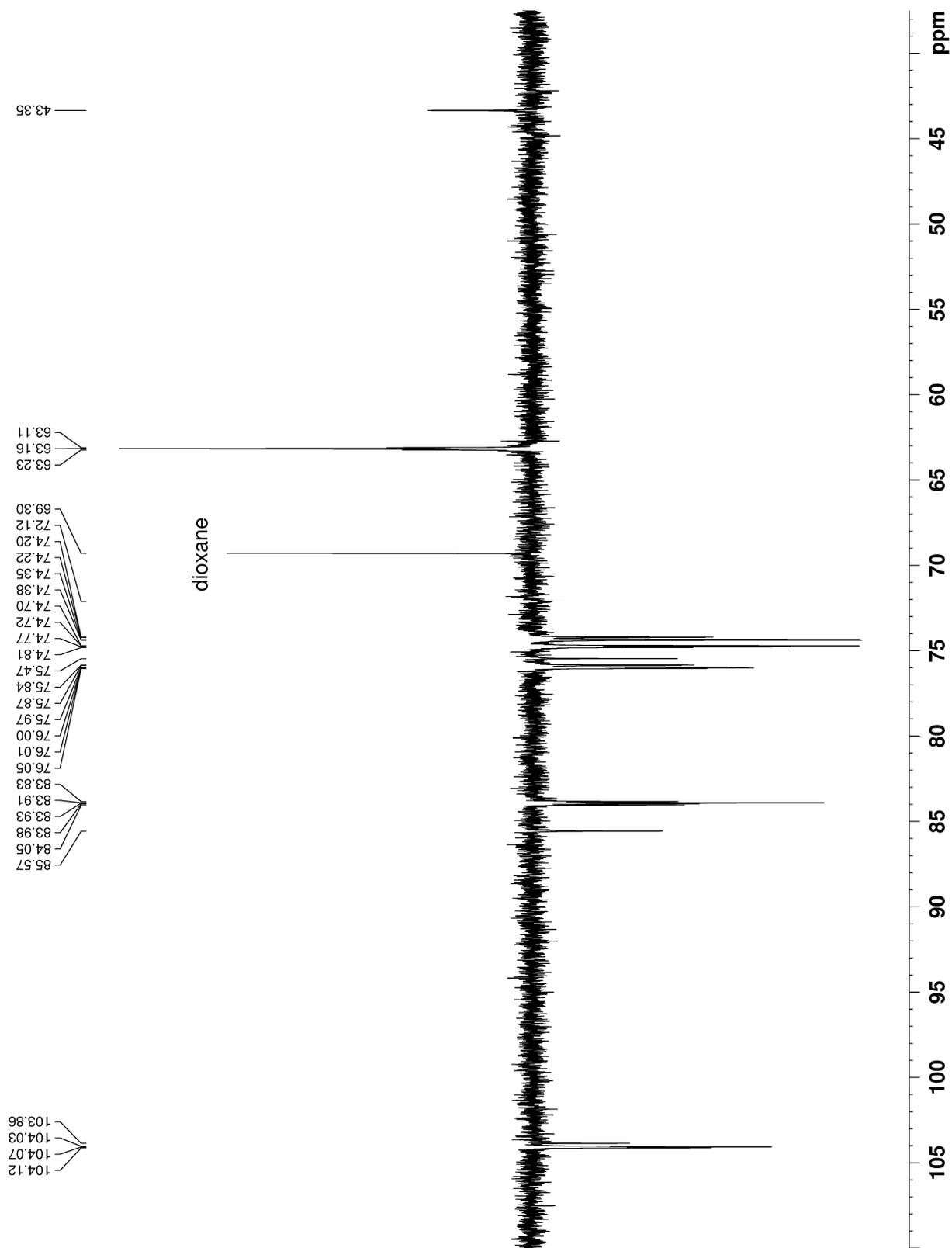


Figure S4: <sup>13</sup>C-NMR spectrum of compound **5a**

## 1.2 NMR data of conjugates **6a**, **7a** and **7b**

The NMR spectra were measured on Bruker AVANCE-600 instrument ( $^1\text{H}$  at 600.13 MHz and  $^{13}\text{C}$  at 150.9 MHz) with a cryo-probe in  $\text{D}_2\text{O}$  at 37 °C. Spectra are referenced to dioxane (added as internal standard) using  $\delta_{\text{H}}(\text{dioxane}) = 3.75$  ppm and  $\delta_{\text{C}}(\text{dioxane}) = 69.3$  ppm. For structural assignment of proton and carbon signals combination of 2D-H,H-COSY, 2D-H,H-TOCSY, 2D-H,C-HSQC and 2D-H,C-HMBC spectra was used. The complete intraresidue and sequential assignment of cyclodextrin proton and carbon signals was achieved only for compound **7b**.

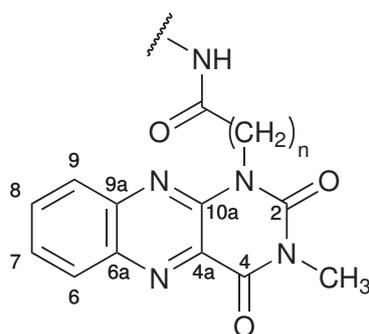


Figure S5: Numbering of alloxazine skeleton used in Table S1

Table S1: Carbon and proton NMR data of compounds **6a**, **7a** and **7b** (aglycone part) in D<sub>2</sub>O at 37 °C.

Compound	C-2	C-4	C-4a	C-6a	C-6	C-7	C-8	C-9	C-9a	C-10a	N-CH <sub>3</sub>	N-CH <sub>2</sub>	CH <sub>2</sub> -CO	C=O
<b>6a</b>	153.95	163.71	131.62	141.96	132.13	133.08	137.61	130.28	145.59	146.99	31.71	47.21	–	172.42
<b>7a</b>	153.77	163.81	131.64	141.72	132.15	132.90	137.45	130.39	145.84	147.19	31.60	42.19	36.68	176.27
<b>7b</b>	153.82	163.42	132.64	142.17	132.39	132.72	137.25	129.78	145.00	148.18	31.55	42.39	36.44	175.84

Compound	H-6	H-7	H-8	H-9	N-CH <sub>3</sub>	N-CH <sub>2</sub>	CH <sub>2</sub> -CO
<b>6a</b>	8.16 m	7.88 m	8.01 m	8.01 m	3.51 s 5.30 d (16.5)	5.00 d (16.5)	–
<b>7a</b>	8.19 bdd (8.5; 1.4)	7.89 ddd (8.5; 7.0; 1.4)	8.03 ddd (8.5; 7.0; 1.4)	8.11 bdd (8.5; 1.4)	3.49 s 4.67 m (2H)	2.75 m (2H)	
<b>7b</b>	8.36 m	8.07 m	8.18 m	8.20 m	3.55 s 5.15 ddd (14; 10; 4)	4.44 dt (14; 4; 4)	2.68 m (2H)

Table S2: Carbon and proton NMR data of compounds **6a**, **7a** and **7b** (cyclodextrin part) in D<sub>2</sub>O at 37 °C.

Compound	Glucose residue	C-1'	C-2'	C-3'	C-4'	C-5'	C-6'
<b>6a</b>		104.18	74.38	76.06	86.02	74.64	63.05
		104.13	74.35 (3)	75.99	83.83	74.61	62.96 (2)
		104.10 (2)	74.33 (2)	75.97	83.80	74.56	62.83
		104.00 (2)		75.95	83.78	74.63	62.68
				75.92	83.70	74.49	43.50
			75.69	83.60	72.68		
<b>7a</b>		104.24	74.40	76.09	85.98	74.65	63.07
		104.21 (3)	74.38 (2)	76.06	83.96	74.61	63.02
		104.10	74.34 (2)	76.01	83.91	74.58 (2)	62.89
		103.94	74.31	75.98	83.80	74.55	62.77
				75.95	83.76	73.07	62.66
			75.84	83.66		43.01	
<b>7b</b>	<b>I</b>	104.27	74.93	75.96	85.91	75.68	42.00
	<b>II</b>	104.92	74.60	75.78	83.44	75.10	62.93
	<b>III</b>	104.42	74.32	74.32	83.64	73.16	62.74
	<b>IV</b>	103.74	74.96	75.77	83.42	74.24	62.38
	<b>V</b>	105.23	74.87	76.05	82.83	74.85	63.14
	<b>VI</b>	104.55	74.51	76.29	84.24	75.63	63.00
	<b>VII</b>	104.47	74.73	75.89	82.71	73.98	61.75
		<b>H-1'</b>	<b>H-2'</b>	<b>H-3'</b>	<b>H-4'</b>	<b>H-5'</b>	<b>H-6'a+H-6'b</b>
<b>6a</b>		5.050	3.609	3.88 – 3.95	3.51 – 3.62	3.809	3.625 – 3.73 (4)
		5.032	3.619			3.70–3.75 (4)	3.99 + 3.868
		5.002	3.589 (2)			3.961	3.860 + 3.482
		4.996	3.518				
		4.965	3.565				
		4.954					
<b>7a</b>		5.026	3.543	3.84 – 3.94	3.51 – 3.60	3.63	3.51 + 3.63
		5.013	3.553			3.67	3.72 – 3.80 (3)
		4.999	3.576			3.74	3.81
		4.987	3.598 (2)			3.78	3.658 + 3.192
		4.943	3.603			3.80 (2)	
		4.897					
<b>7b</b>	<b>I</b>	5.165	3.692	3.922	3.298	3.717	4.07 + 2.70
	<b>II</b>	5.07	3.704	4.128	3.828	4.175	4.231 + 4.093
	<b>III</b>	4.77	~3.44	3.659	3.374	2.607	3.643 + 3.373
	<b>IV</b>	4.965	3.481	3.482	3.256	2.92	3.249 + 3.187
	<b>V</b>	5.19	3.744	4.152	3.711	4.508	4.168 + 3.995
	<b>VI</b>	4.91	3.625	3.995	3.659	~3.715	4.063 + 3.818
	<b>VII</b>	4.845	3.464	3.368	3.313	2.42	3.272 + 3.135

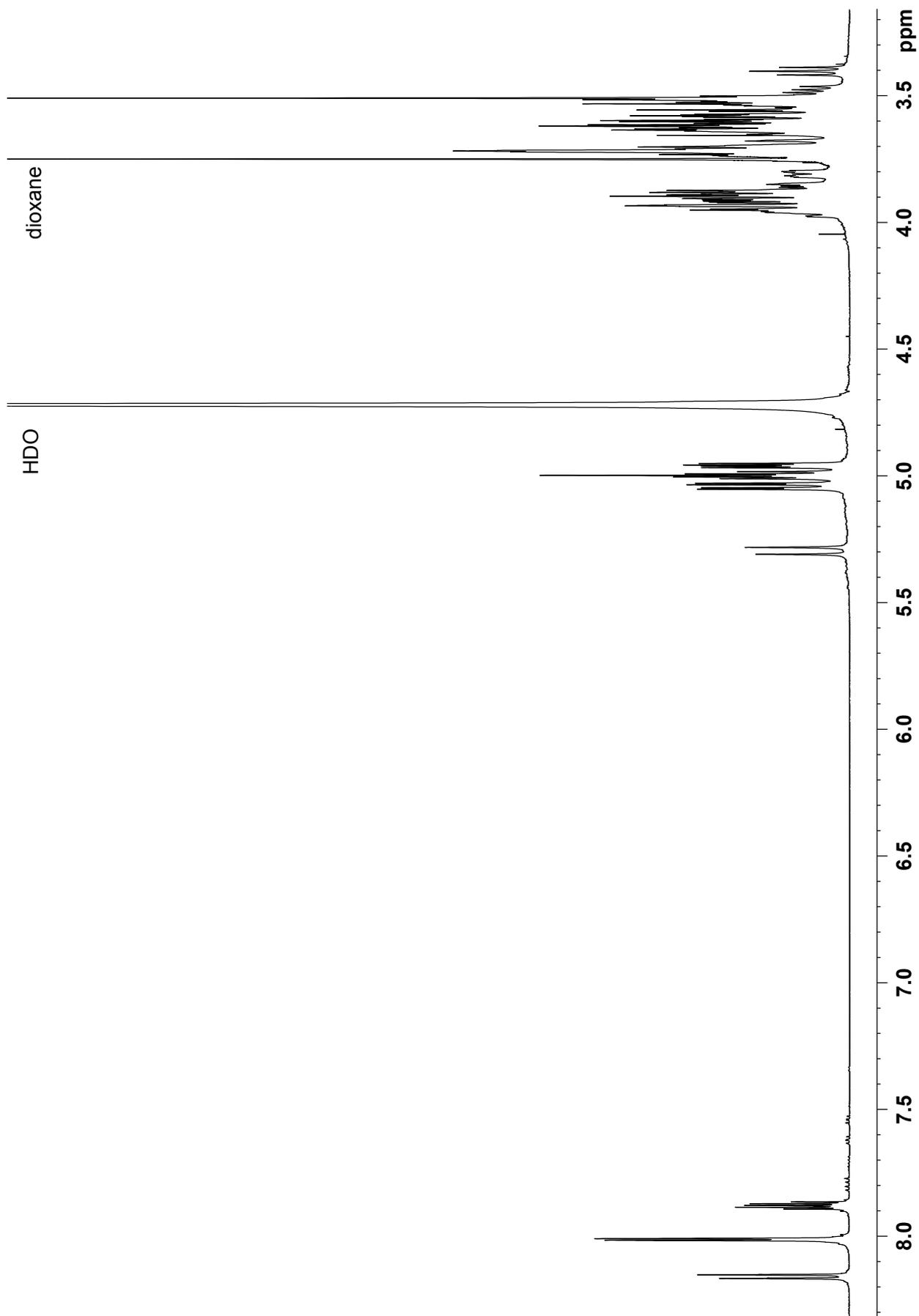


Figure S6: <sup>1</sup>H-NMR spectrum of conjugate 6a

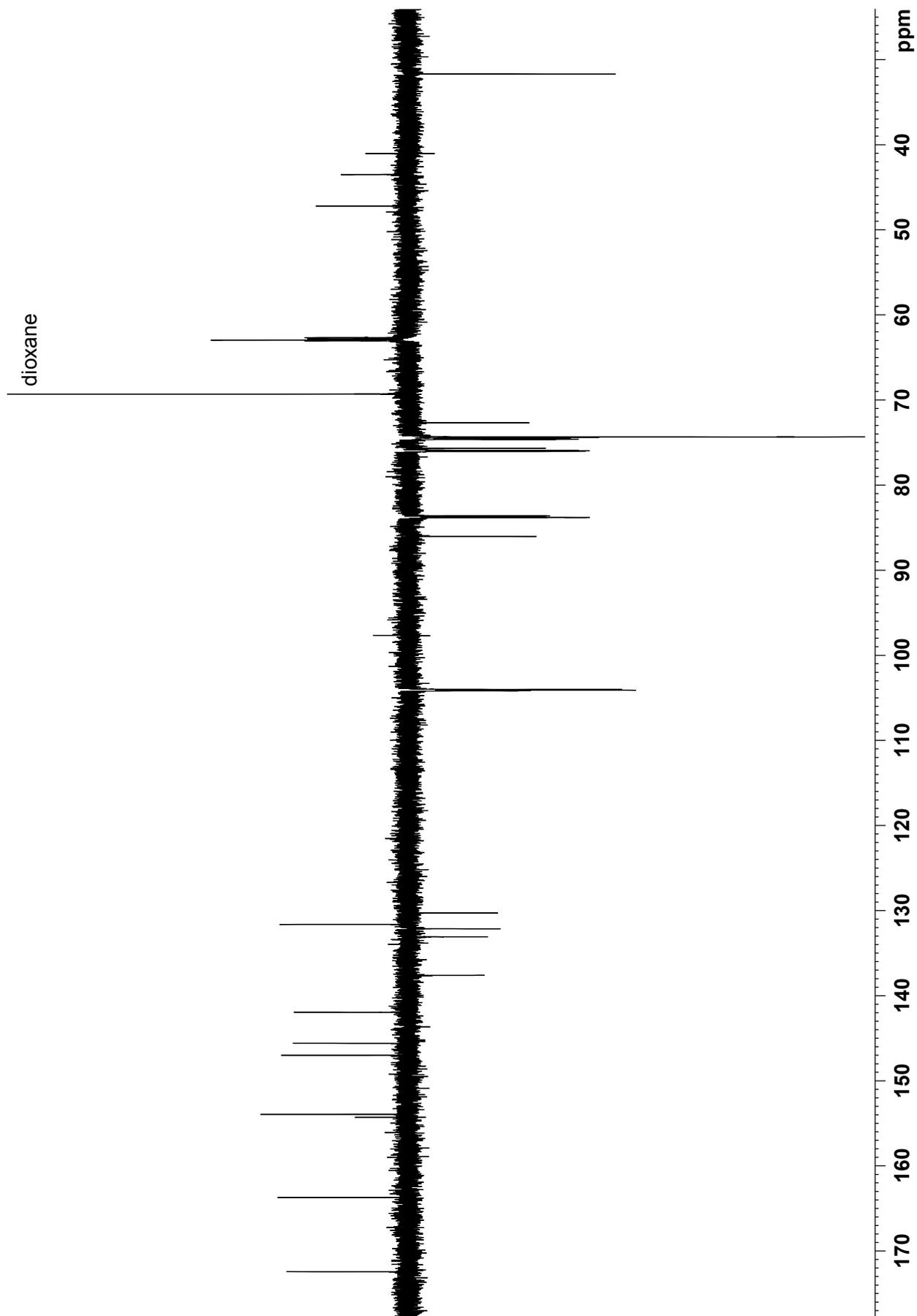


Figure S7:  $^{13}\text{C}$ -NMR spectrum of conjugate **6a**

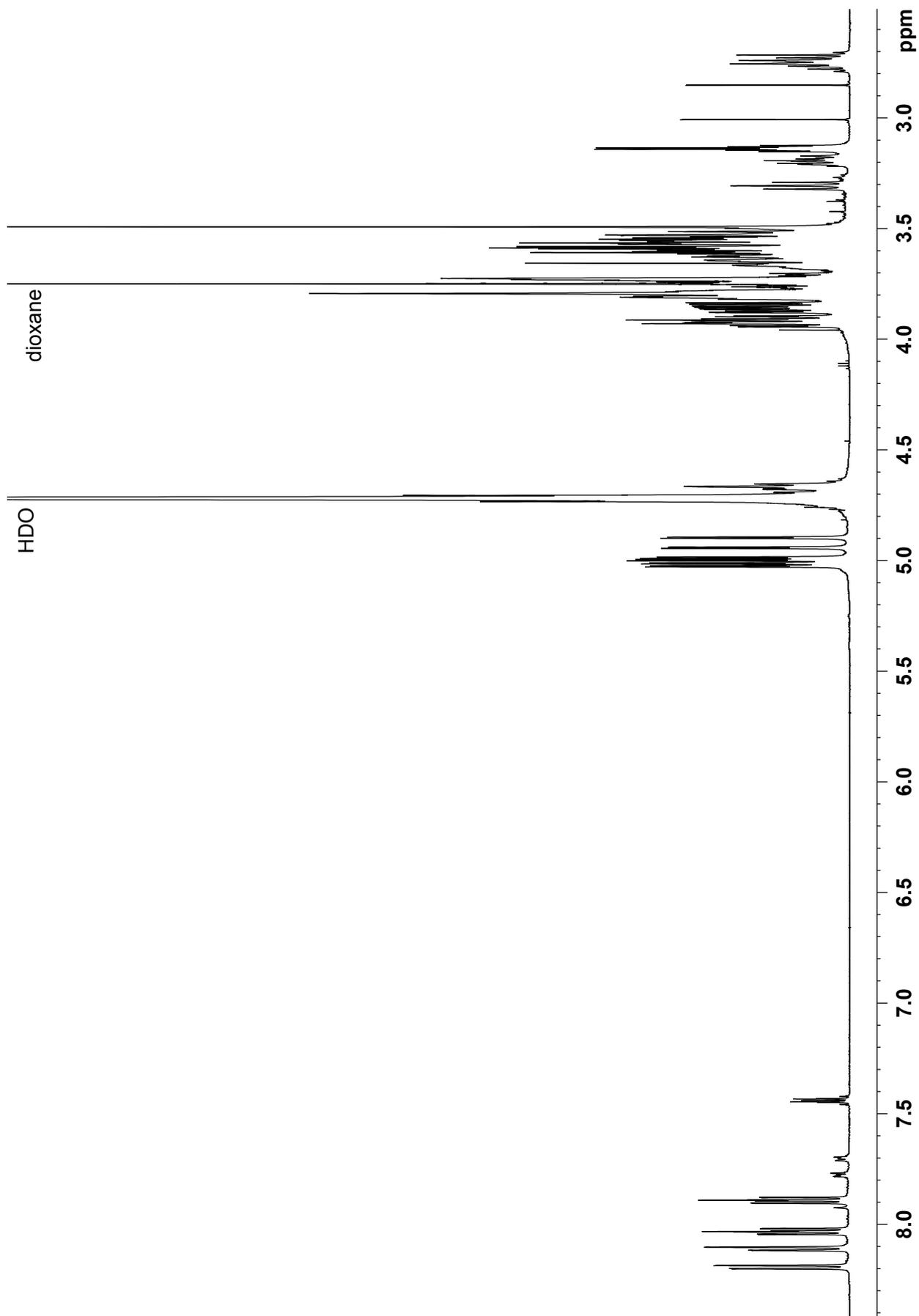


Figure S8: <sup>1</sup>H-NMR spectrum of conjugate 7a

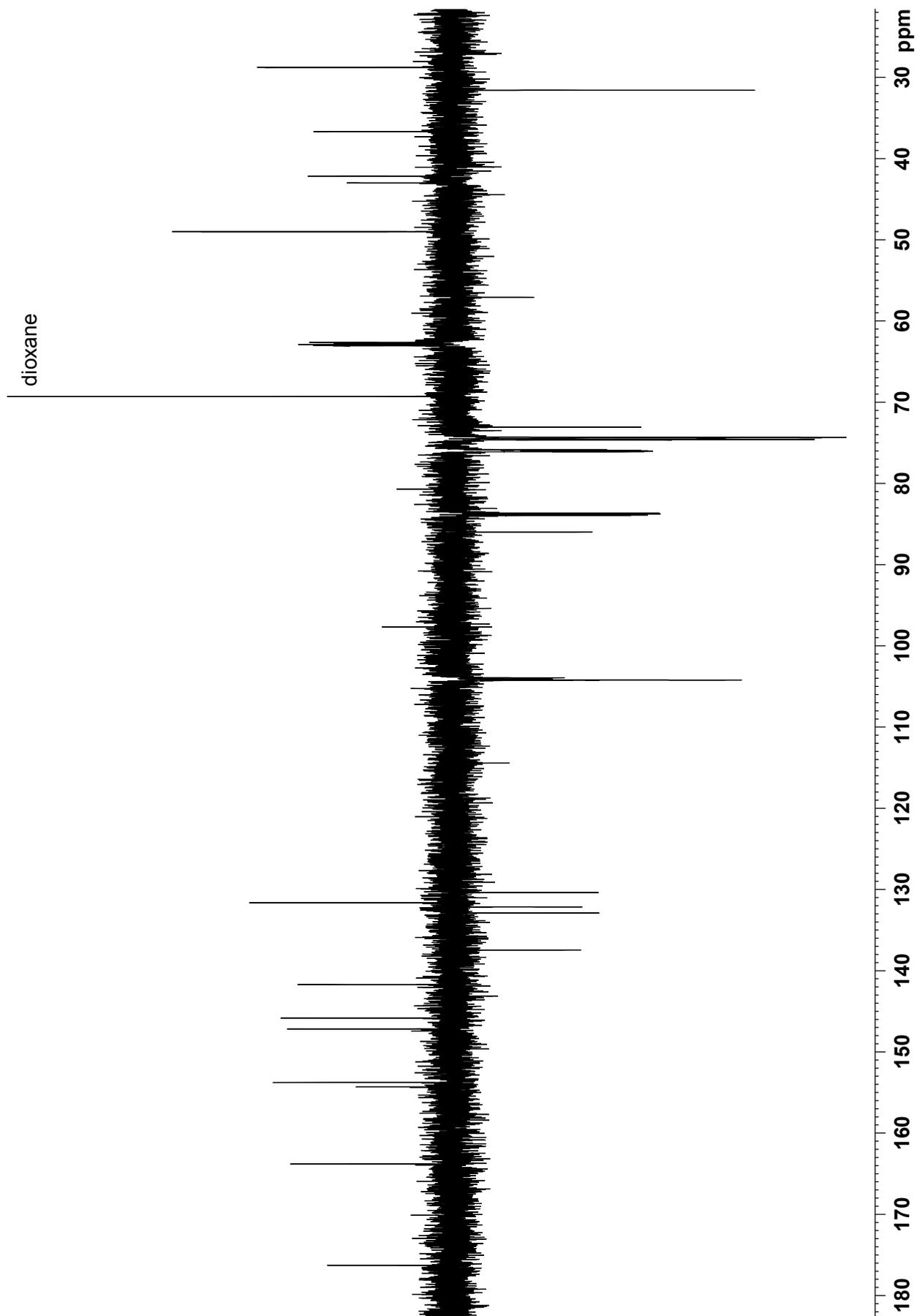


Figure S9:  $^{13}\text{C}$ -NMR spectrum of conjugate **7a**

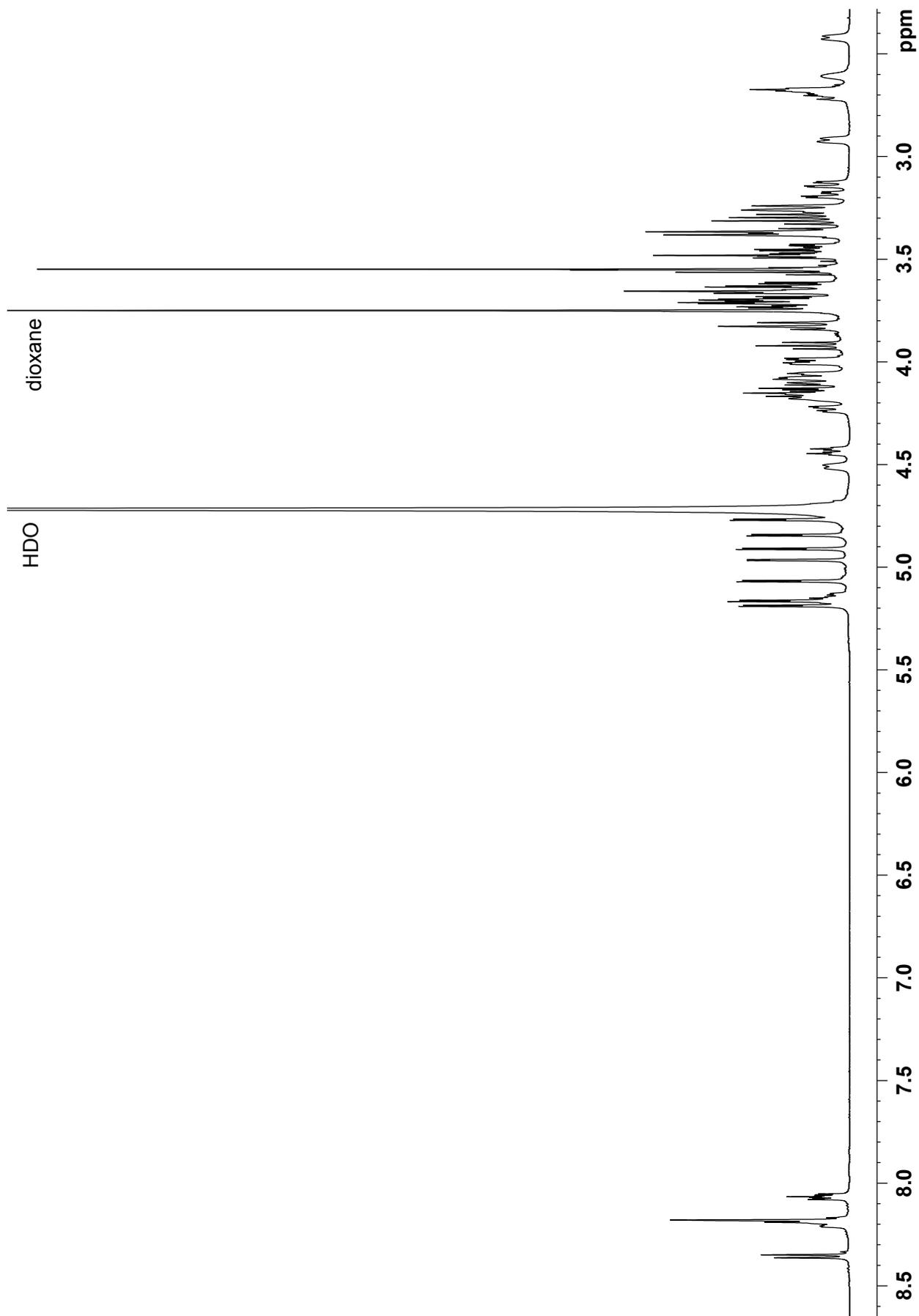


Figure S10: <sup>1</sup>H-NMR spectrum of conjugate 7b

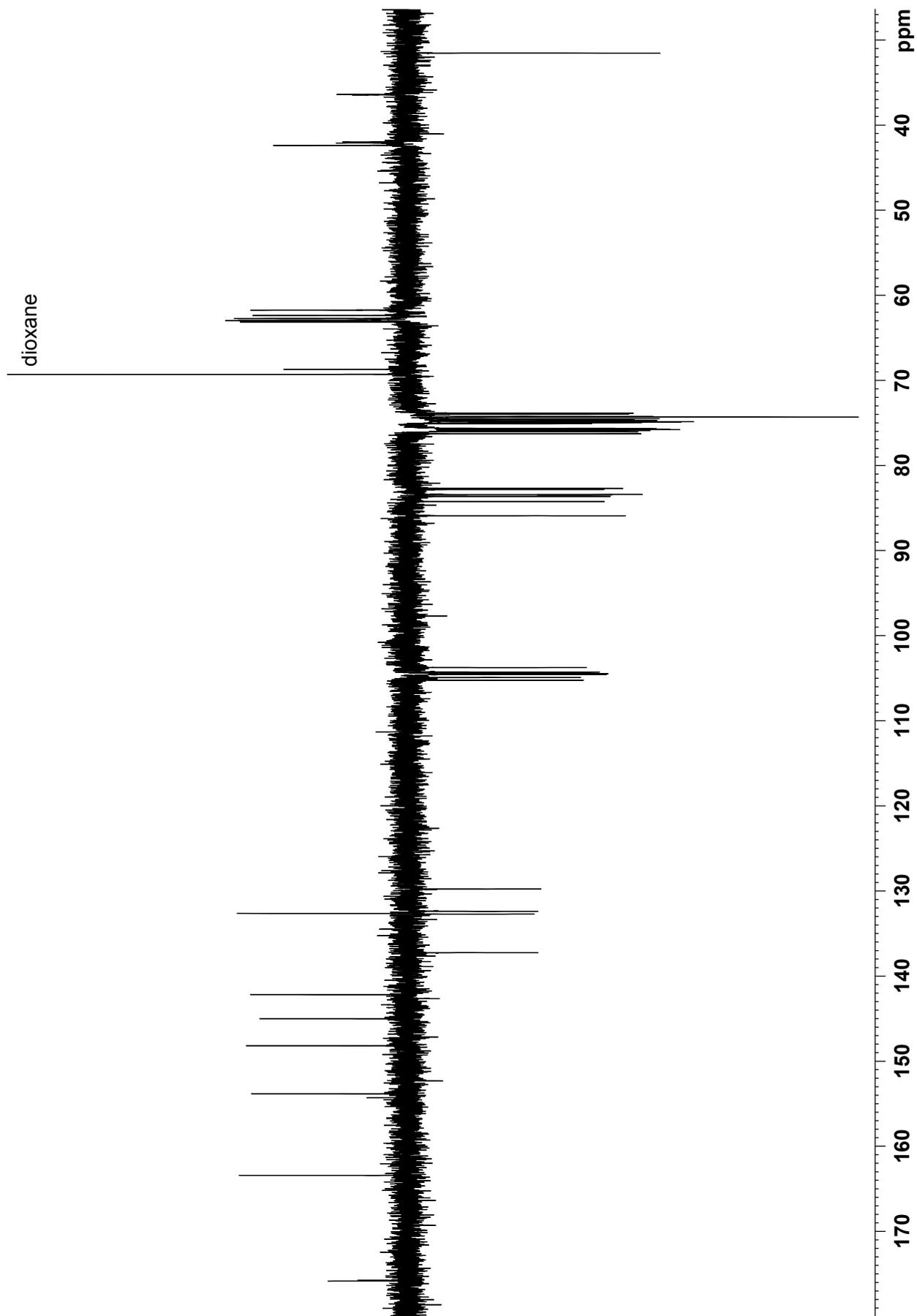


Figure S11:  $^{13}\text{C}$ -NMR spectrum of conjugate 7b

### 1.3 HR-MS spectra of flavinium salts 1a, 2a and 2b

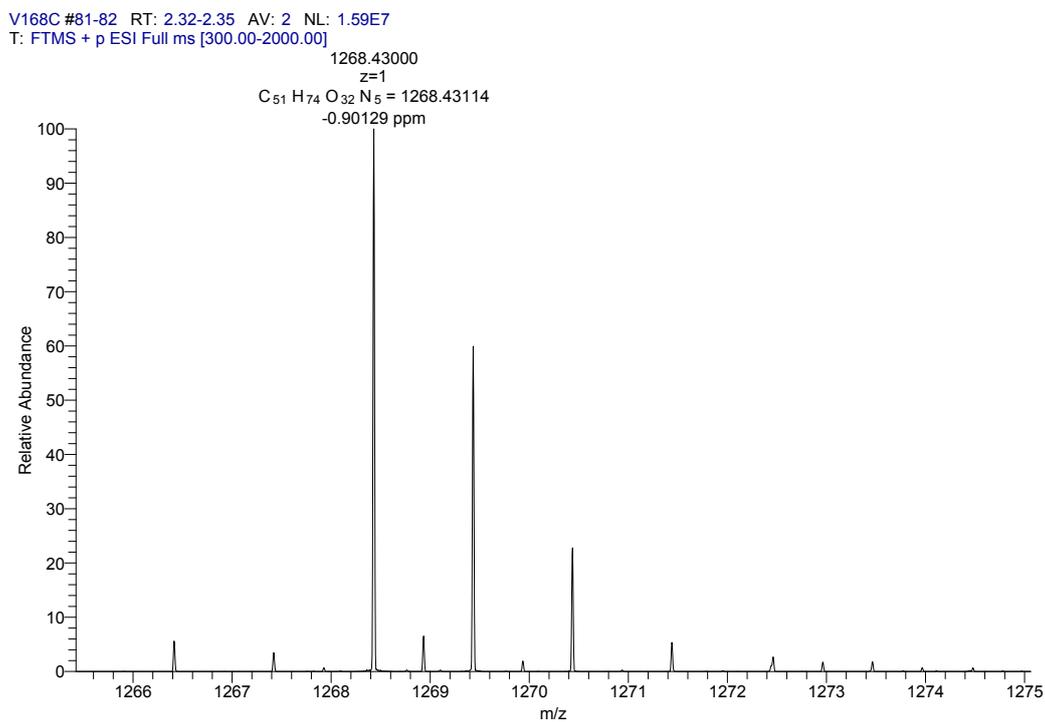


Figure S12: High resolution mass spectrum of compound **1a**

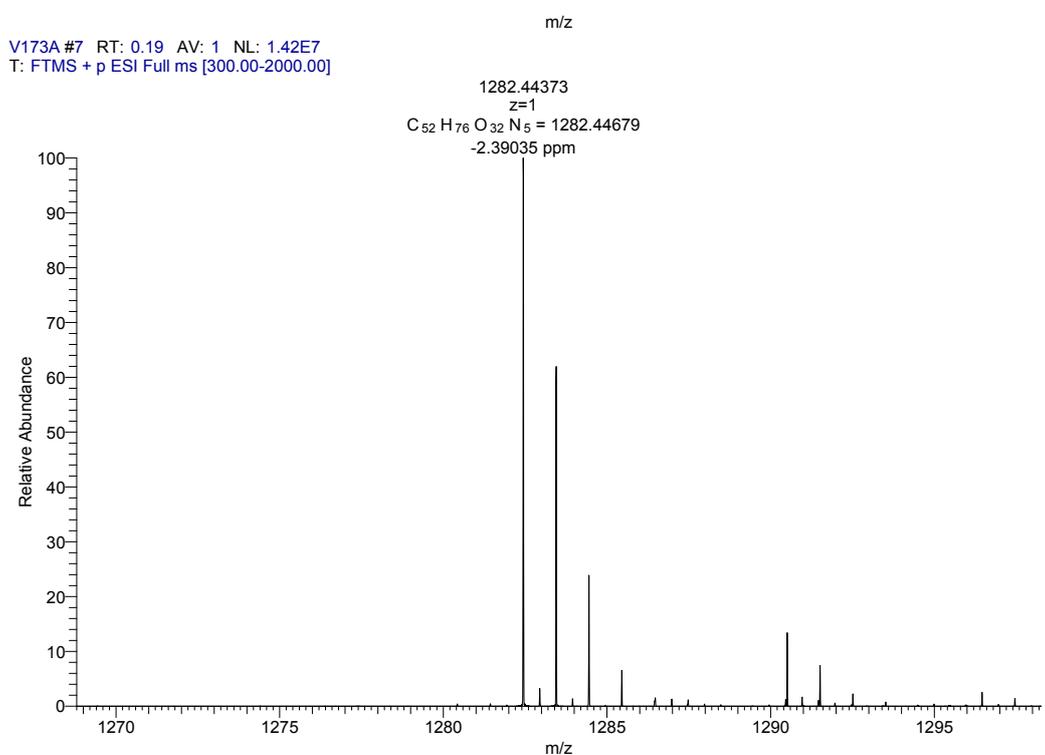


Figure S13: High resolution mass spectrum of compound **2a**

V154B #148-149 RT: 4.30-4.32 AV: 2 NL: 1.96E7  
T: FTMS + p ESI Full ms [300.00-2000.00]

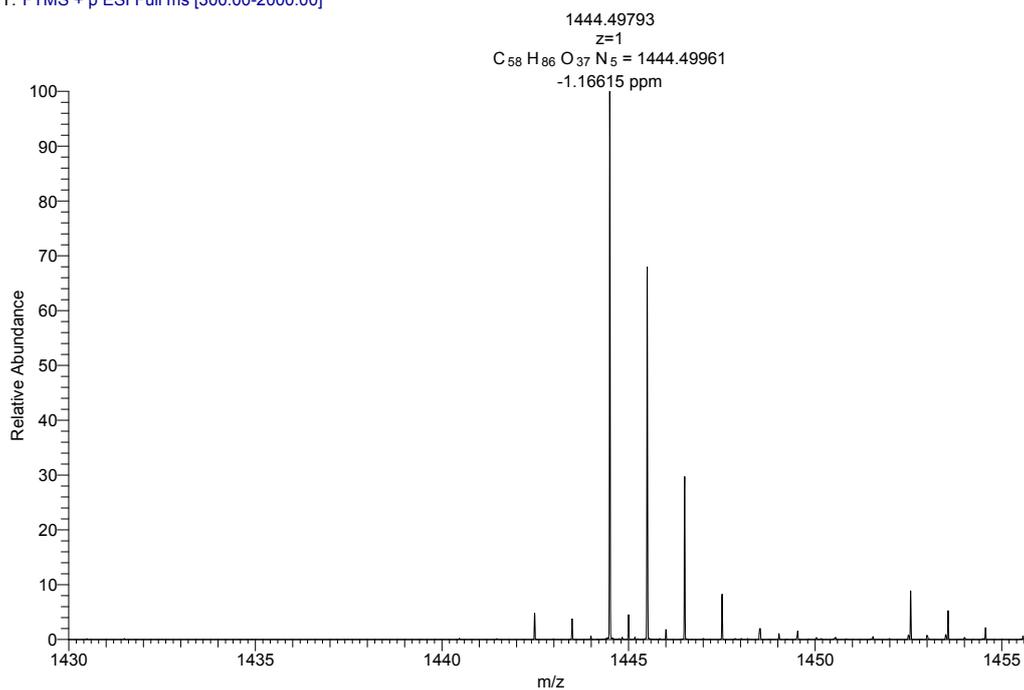


Figure S14: High resolution mass spectrum of compound **2b**

## 1.4 HR-MS spectra of conjugates **6a**, **7a** and **7b**

190111servis 28 #44-51 RT: 1.21-1.40 AV: 8 NL: 3.15E6  
T: FTMS + p ESI Full ms [200.00-2000.00]

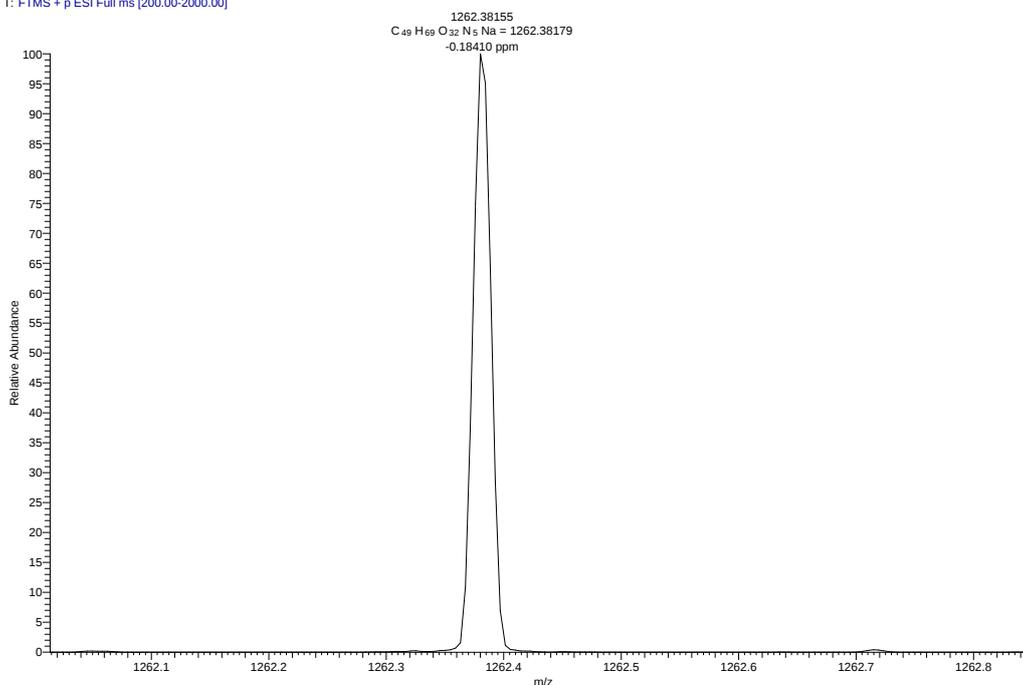
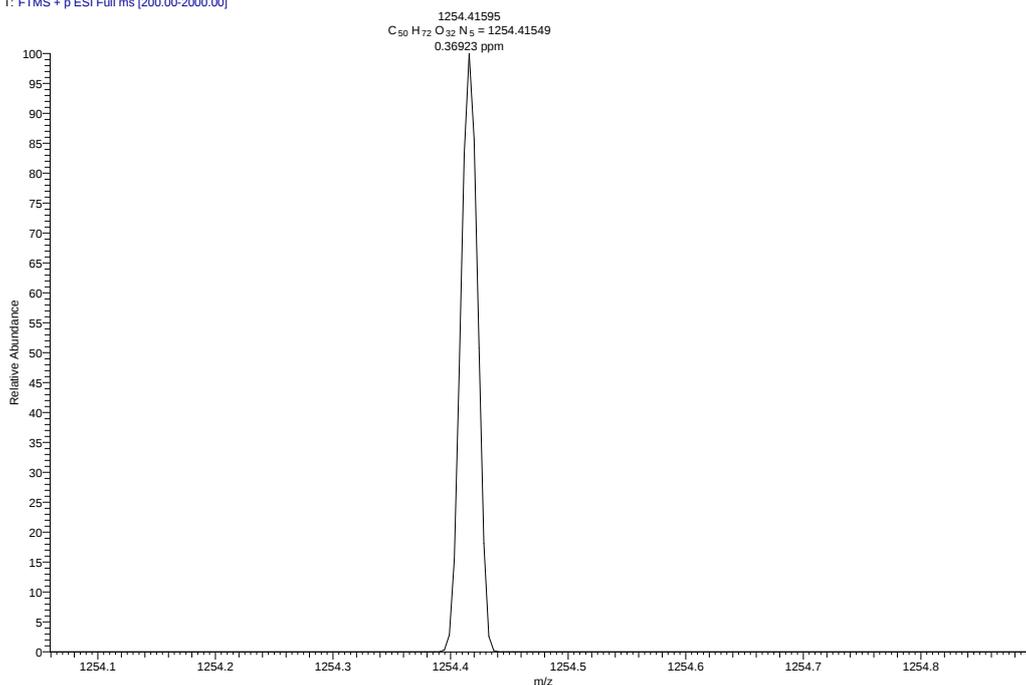


Figure S15: High resolution mass spectrum of compound **6a**

190111servis\_27 #44-51 RT: 1.21-1.41 AV: 8 NL: 8.89E4  
T: FTMS + p ESI Full ms [200.00-2000.00]



190111servis\_27 #44-51 RT: 1.21-1.41 AV: 8 NL: 2.28E6  
T: FTMS + p ESI Full ms [200.00-2000.00]

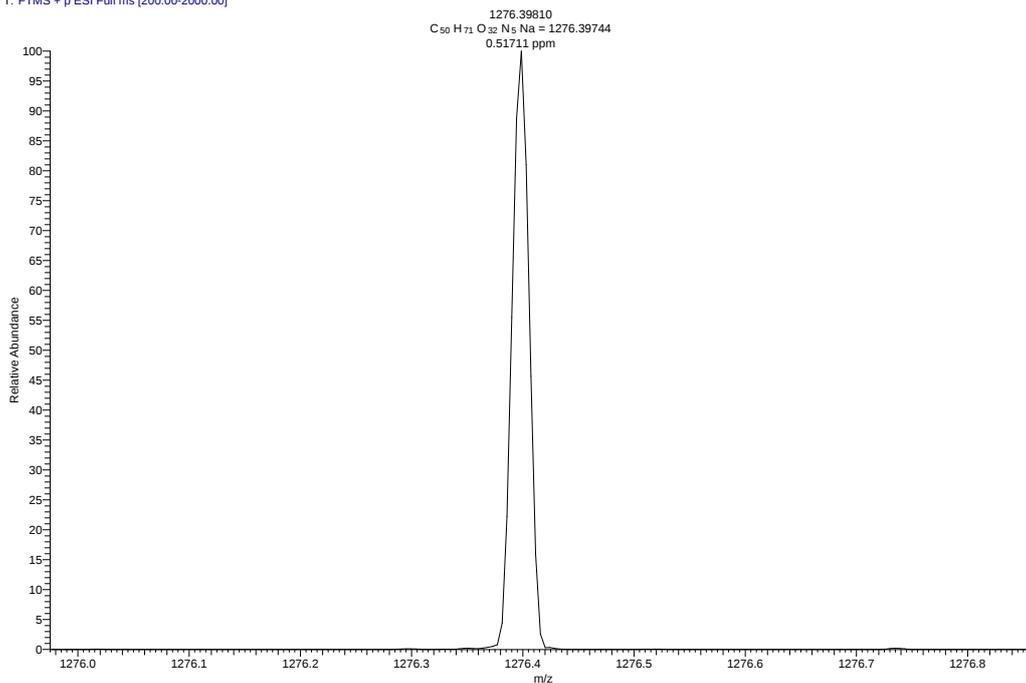


Figure S16: High resolution mass spectra of compound **7a**

060510servis\_2 #44-48 RT: 1.23-1.35 AV: 5 NL: 3.69E6  
T: FTMS + p ESI Full ms [200.00-2000.00]

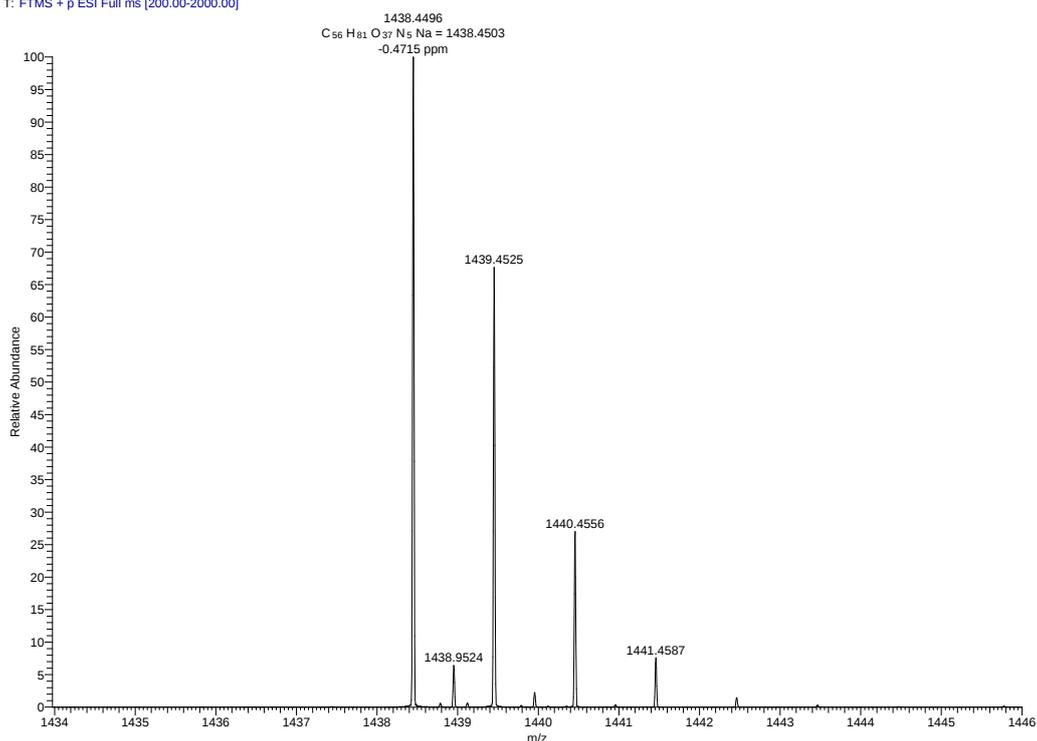


Figure S17: High resolution mass spectrum of compound **7b**

## 1.5 UV-VIS spectra of conjugates **1a**, **2a** and **2b**

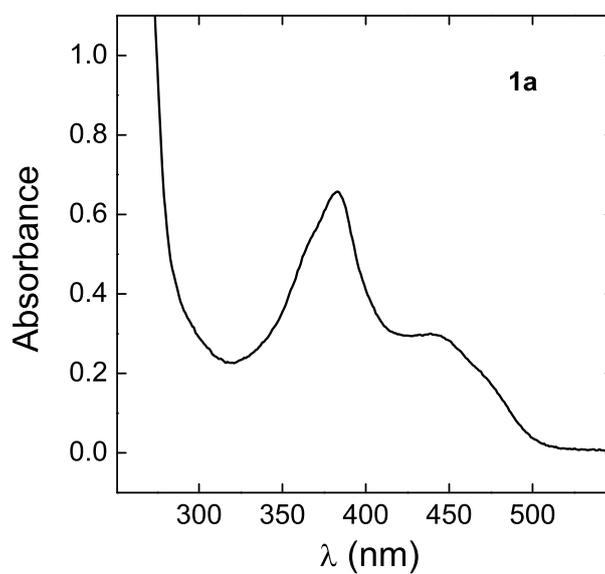


Figure S18: UV-VIS spectrum of compound **1a**. Phosphate buffer (pH 2),  $c(\mathbf{1a}) = 5 \times 10^{-5}\text{M}$

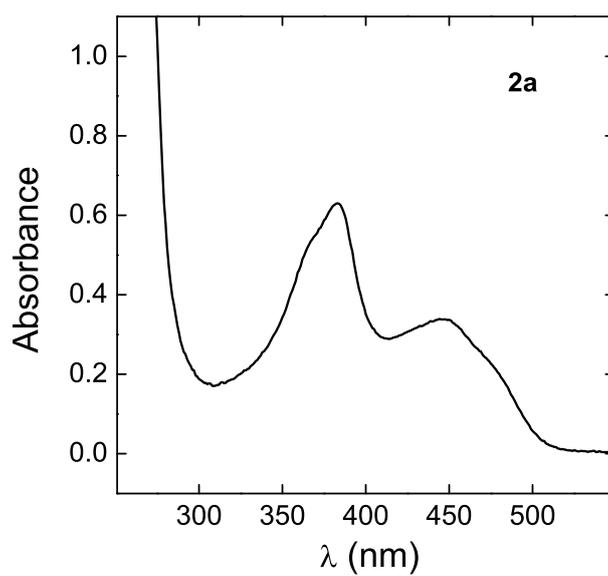


Figure S19: UV-VIS spectrum of compound **2a**. Phosphate buffer (pH 2),  $c(\mathbf{2a}) = 5 \times 10^{-5}\text{M}$

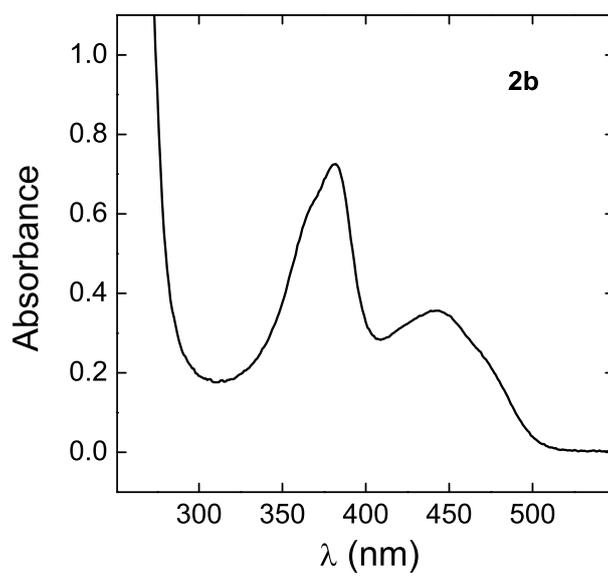


Figure S20: UV-VIS spectrum of compound **2b**. Phosphate buffer (pH 2),  $c(\mathbf{2b}) = 5 \times 10^{-5}\text{M}$

## 1.6 UV-VIS spectra of conjugates **6a**, **7a** and **7b**

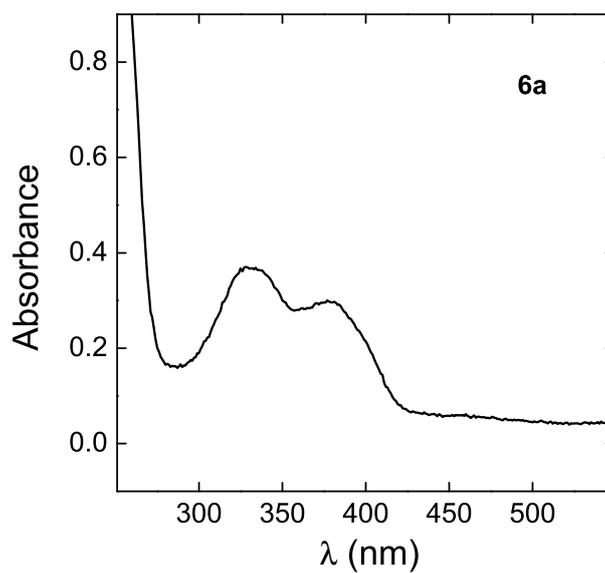


Figure S21: UV-VIS spectrum of compound **6a**. Phosphate buffer (pH 2),  $c(\mathbf{6a}) = 5 \times 10^{-5}$  M

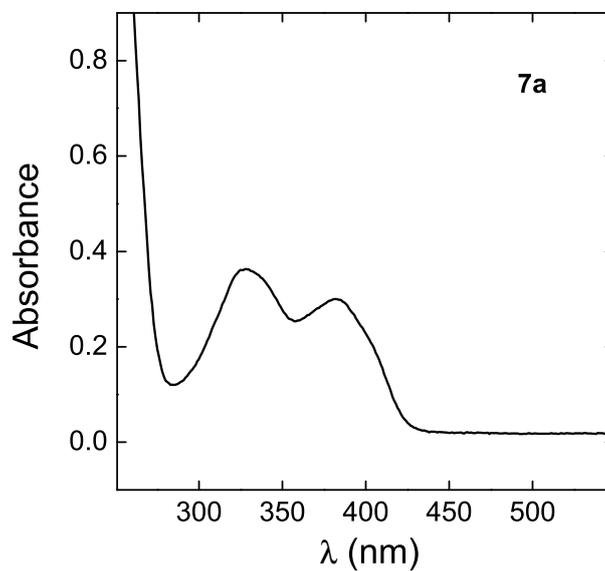


Figure S22: UV-VIS spectrum of compound **2a**. Phosphate buffer (pH 2),  $c(\mathbf{7a}) = 5 \times 10^{-5}$  M

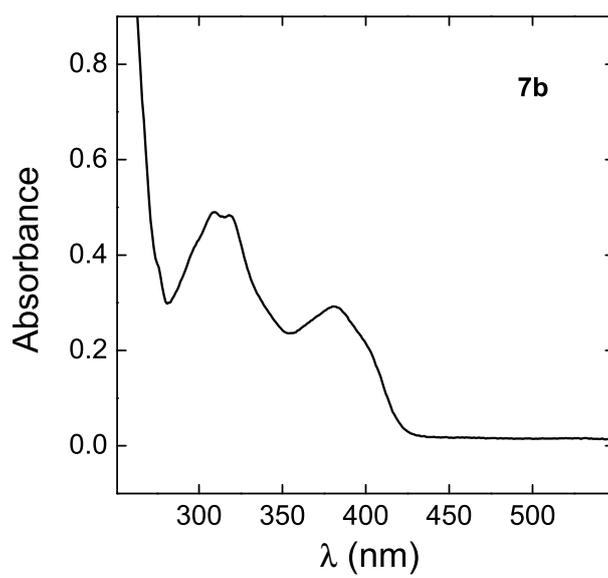


Figure S23: UV-VIS spectrum of compound **2b**. Phosphate buffer (pH 2),  $c(\mathbf{7b}) = 5 \times 10^{-5}$  M

## 2 Determination of ee by $^1\text{H-NMR}$ or chiral HPLC

All catalytic oxidations of sulfides were performed in the same manner as described by us.<sup>1</sup> Product of catalytic reaction was isolated by extraction with  $\text{CCl}_4$  ( $2 \times 500 \mu\text{L}$ ) and then with  $\text{CDCl}_3$  ( $500 \mu\text{L}$ ). The conversion of catalytic oxidations was determined as the ratio of peaks of the respective *S*-methyl groups in  $^1\text{H-NMR}$  spectra. The enantiomeric excesses of *n*-alkyl methyl sulfoxides were determined by  $^1\text{H-NMR}$  using (*R*)-(-)-3,5-dinitro-*N*-(1-phenylethyl)-benzamide as a shift reagent.<sup>2</sup>  $^1\text{H-NMR}$  shifts may slightly vary according to the composition of the  $\text{CDCl}_3$ - $\text{CCl}_4$  mixture. In cases of aryl-, benzyl-, *tert*-butyl- and cyclohexyl- methyl sulfoxides, enantiomeric excess was determined by HPLC equipped with a chiral column. In addition, a set of samples representing all alkyl methyl sulfoxide structures was also analyzed by HPLC method using both UV and optical rotation (Chiralyser).

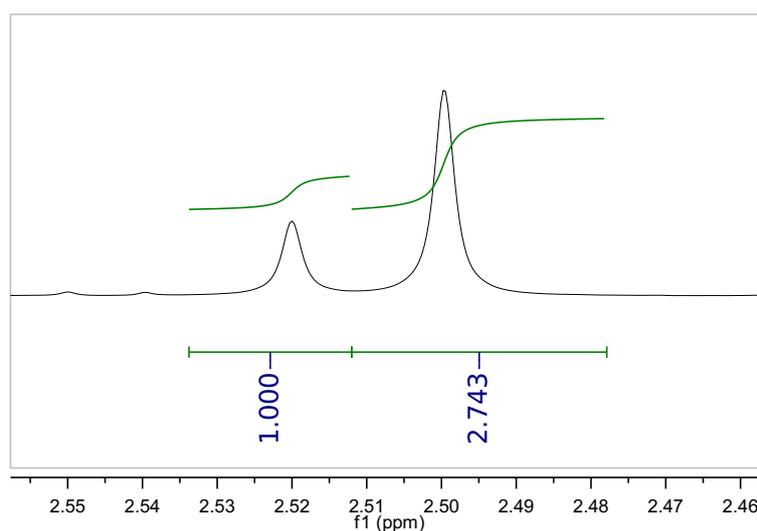


Figure S24: Part of the  $^1\text{H-NMR}$  spectrum of product of hexyl(methyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 1, Table 1, main text).

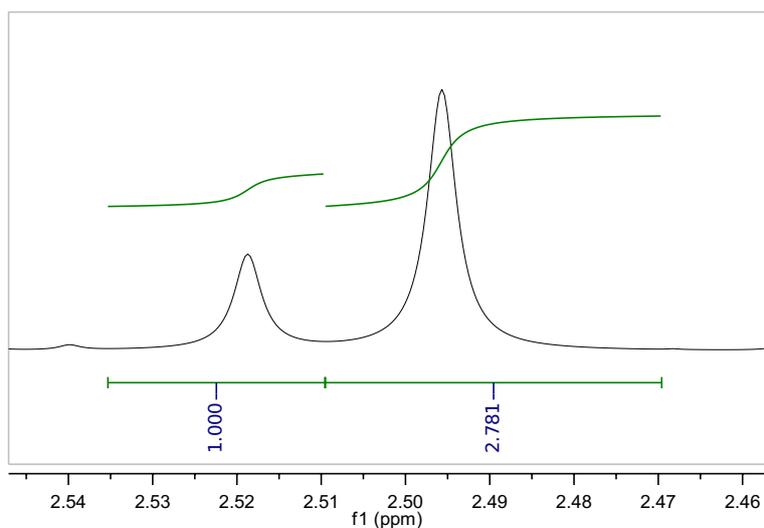


Figure S25: Part of the <sup>1</sup>H-NMR spectrum of product of hexyl(methyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 2, Table 1, main text).

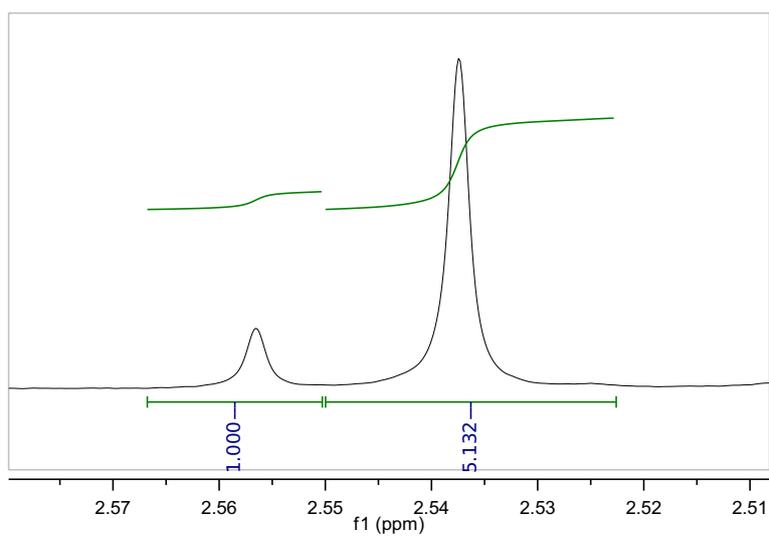
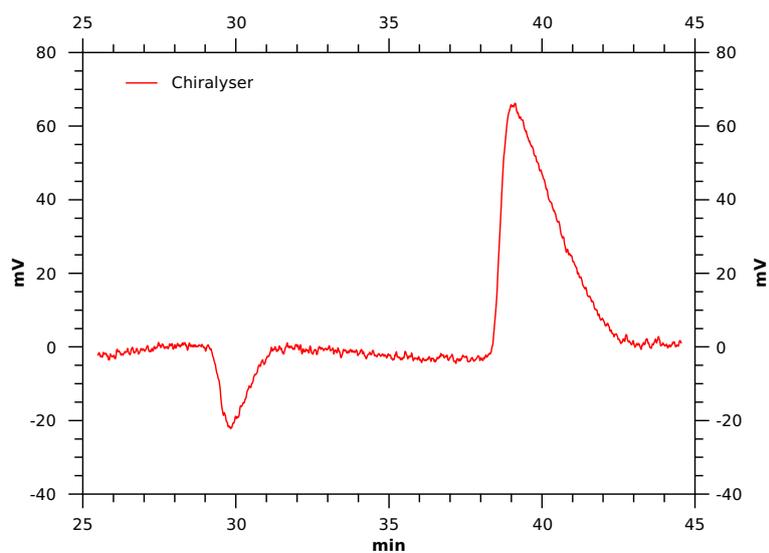


Figure S26: Part of the <sup>1</sup>H-NMR spectrum of product of methyl(octyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 3, Table 1, main text).



	Reten. Time [min]	Start Time [min]	End Time [min]	Area [%]	W05 [min]
1	29.833	28.893	31.673	14.9	1.07
2	39.113	37.593	43.24	85.1	1.95

Figure S27: Part of the HPLC chromatogram of product of methyl(octyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 3, Table 1, main text). Conditions: **Daicel® Chiralpak AS-H** column, heptane/*i*-PrOH (17:3), 1 mL/min.

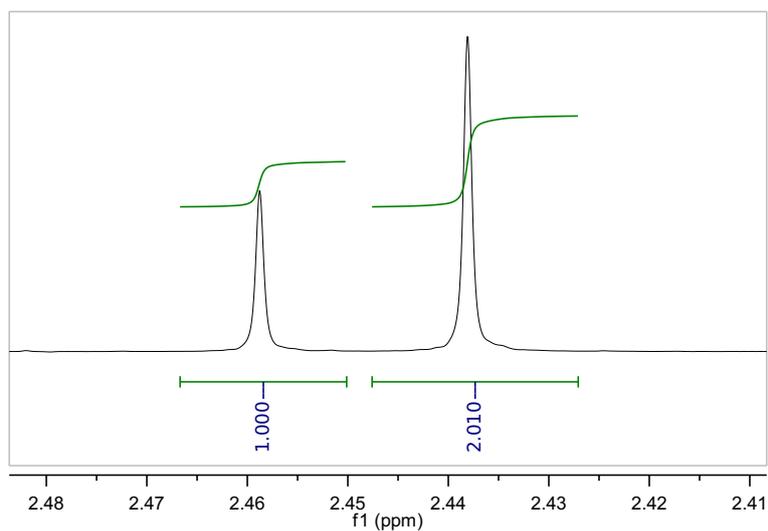
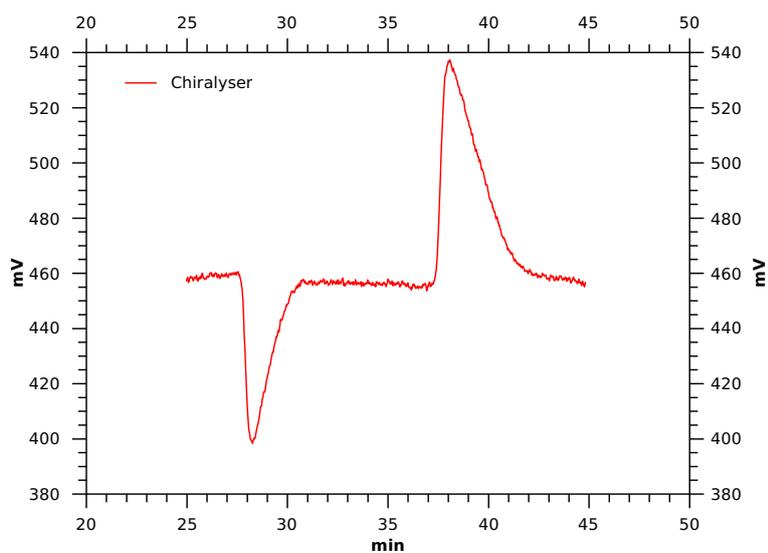


Figure S28: Part of the  $^1\text{H}$ -NMR spectrum of product of methyl(octyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 4, Table 1, main text).



	Reten. Time [min]	Start Time [min]	End Time [min]	Area [%]	W05 [min]
1	28.260	27.540	31.187	31.4	1.30
2	38.067	36.920	43.020	68.6	2.16

Figure S29: Part of the HPLC chromatogram of product of methyl(octyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 4, Table 1, main text). Conditions: **Daicel® Chiralpak AS-H** column, heptane/*i*-PrOH (17:3), 1 mL/min.

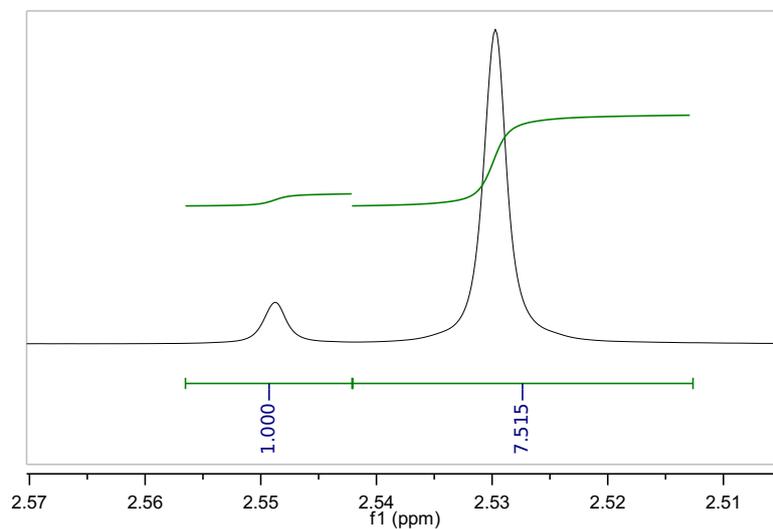
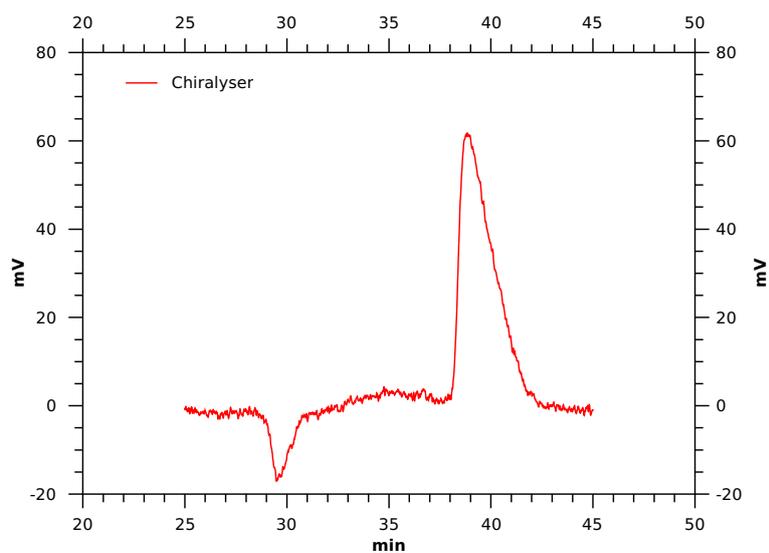


Figure S30: Part of the  $^1\text{H}$ -NMR spectrum of product of methyl(octyl)sulfide oxidation catalyzed by **1a** (5 mol%), (Entry 5, Table 1, main text).



	Reten. Time [min]	Start Time [min]	End Time [min]	Area [%]	W05 [min]
1	29.487	28.440	31.100	12.2	1.12
2	38.847	37.620	44.180	87.8	1.77

Figure S31: Part of the HPLC chromatogram of product of methyl(octyl)sulfide oxidation catalyzed by **1a** (5 mol%), (Entry 5, Table 1, main text). Conditions: **Daicel® Chiralpak AS-H** column, heptane/*i*-PrOH (17:3), 1 mL/min.

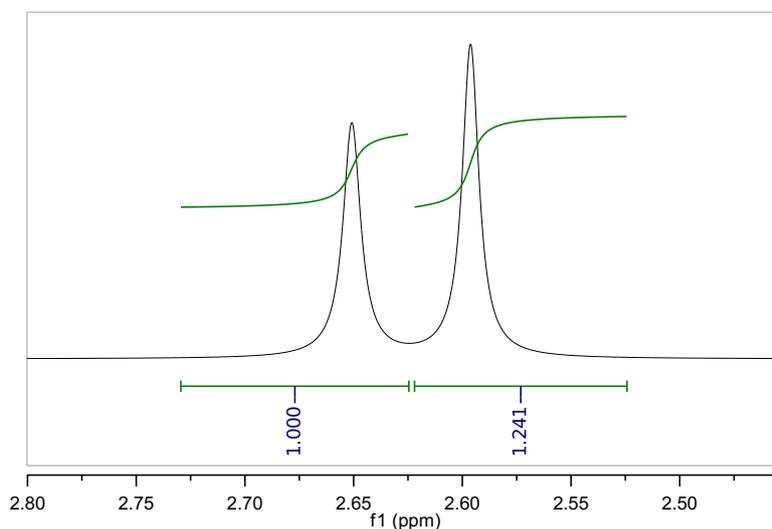


Figure S32: Part of the  $^1\text{H}$ -NMR spectrum of product of decyl(methyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 6, Table 1, main text).

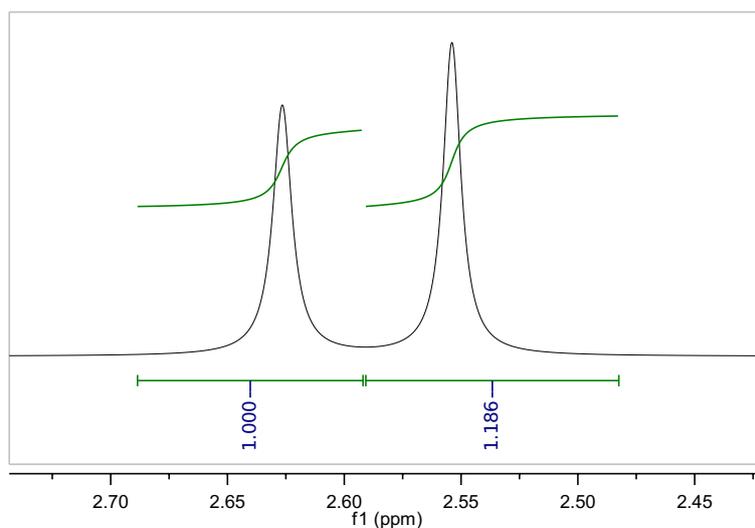


Figure S33: Part of the <sup>1</sup>H-NMR spectrum of product of decyl(methyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 7, Table 1, main text).

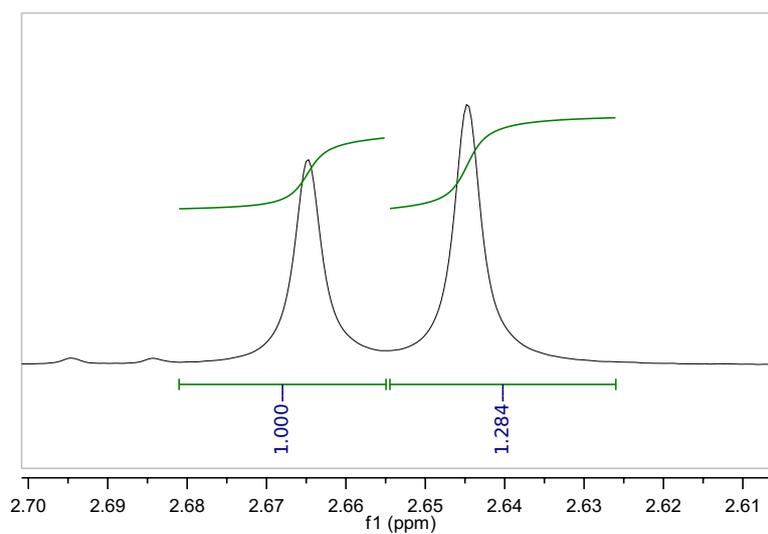


Figure S34: Part of the <sup>1</sup>H-NMR spectrum of product of dodecyl(methyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 8, Table 1, main text).

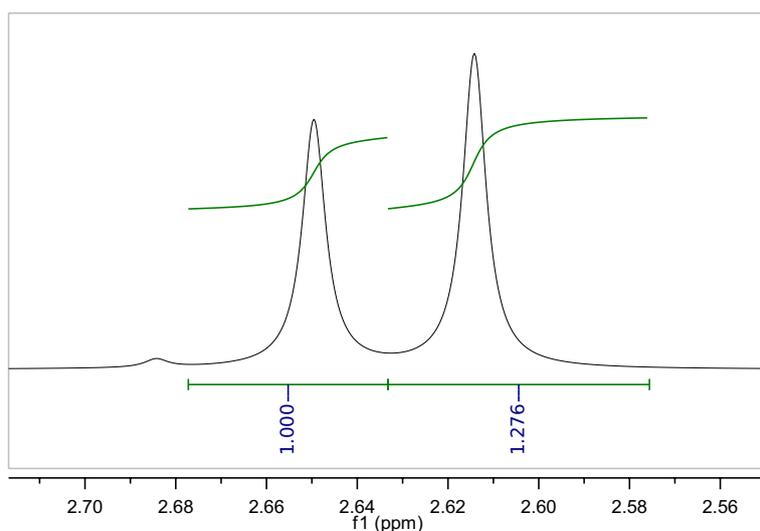


Figure S35: Part of the <sup>1</sup>H-NMR spectrum of product of dodecyl(methyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 9, Table 1, main text).

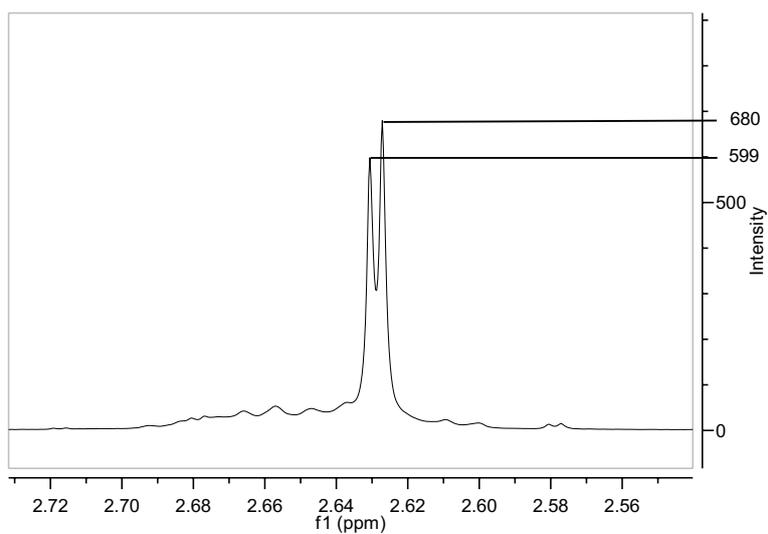


Figure S36: Part of the <sup>1</sup>H-NMR spectrum of product of cyclohexyl(methyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 11, Table 1, main text). Enantiomeric excess was estimated from the differences of peak's height.

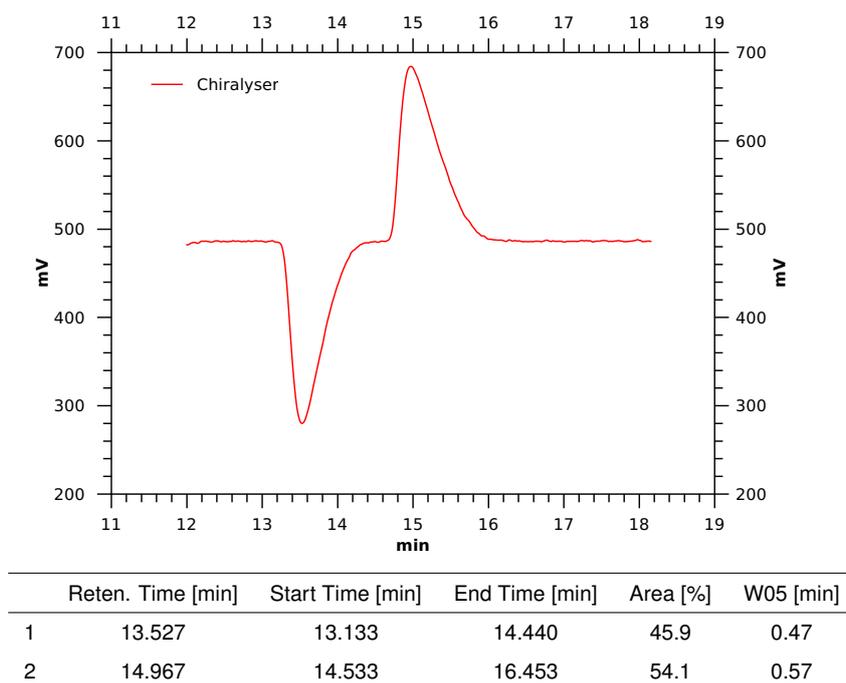


Figure S37: Part of the HPLC chromatogram of product of cyclohexyl(methyl)sulfide oxidation catalyzed by **1a** (1 mol%), (Entry 12, Table 1, main text). Conditions: **Phenomenex® Lux 5u Amylose-2** column, heptane/*i*-PrOH (17:3), 1 mL/min.

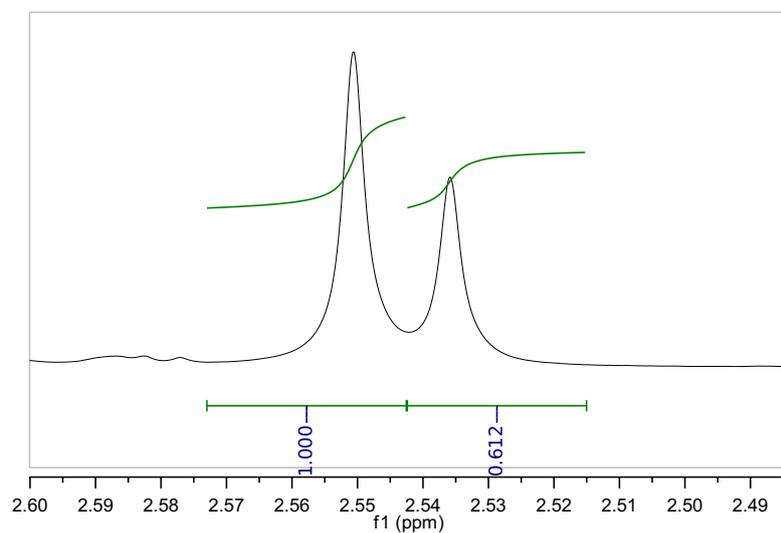


Figure S38: Part of the  $^1\text{H}$ -NMR spectrum of product of hexyl(methyl)sulfide oxidation catalyzed by **2a** (1 mol%), (Entry 1, Table 2, main text).

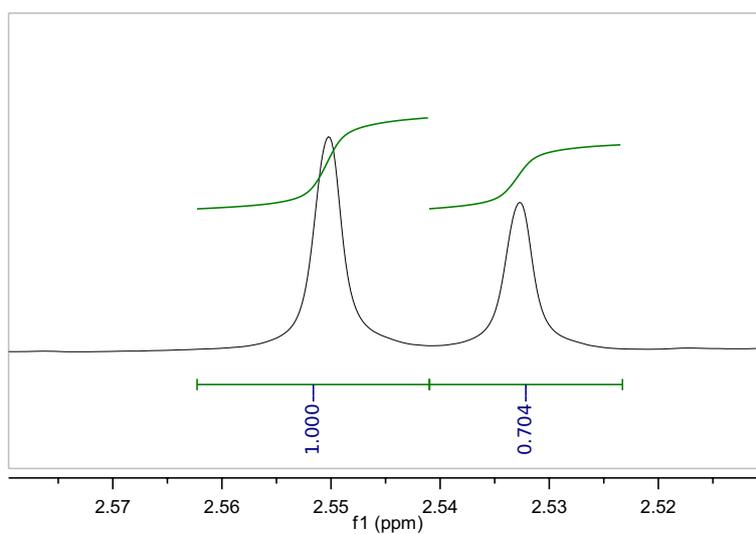


Figure S39: Part of the <sup>1</sup>H-NMR spectrum of product of methyl(octyl)sulfide oxidation catalyzed by **2a** (1 mol%), (Entry 2, Table 2, main text).

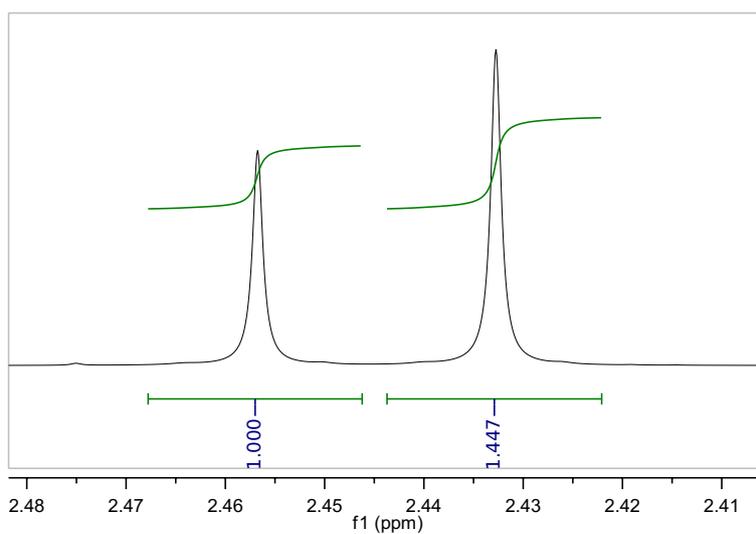
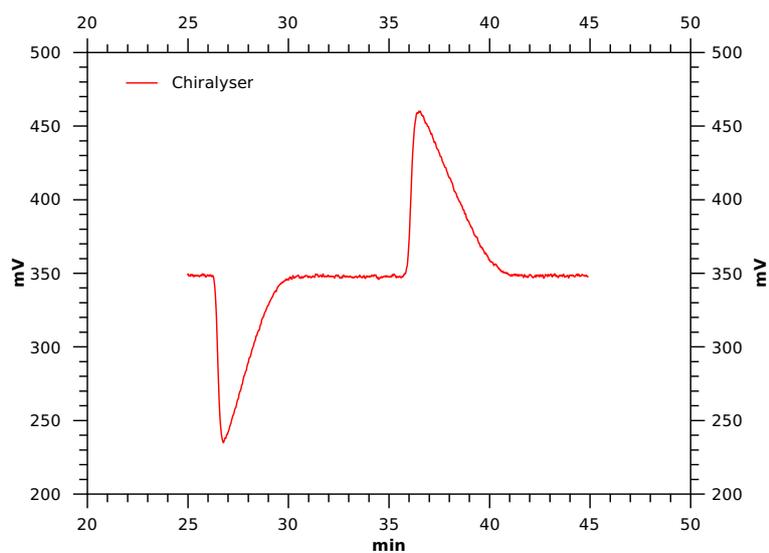


Figure S40: Part of the <sup>1</sup>H-NMR spectrum of product of methyl(octyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 2, Table 3, main text).



	Reten. Time [min]	Start Time [min]	End Time [min]	Area [%]	W05 [min]
1	26.760	25.733	30.220	41.6	1.57
2	36.487	35.353	41.680	58.4	2.27

Figure S41: Part of the HPLC chromatogram of product of methyl(octyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 2, Table 3, main text). Conditions: **Daicel® Chiralpak AS-H** column, heptane/*i*-PrOH (17:3), 1 mL/min.

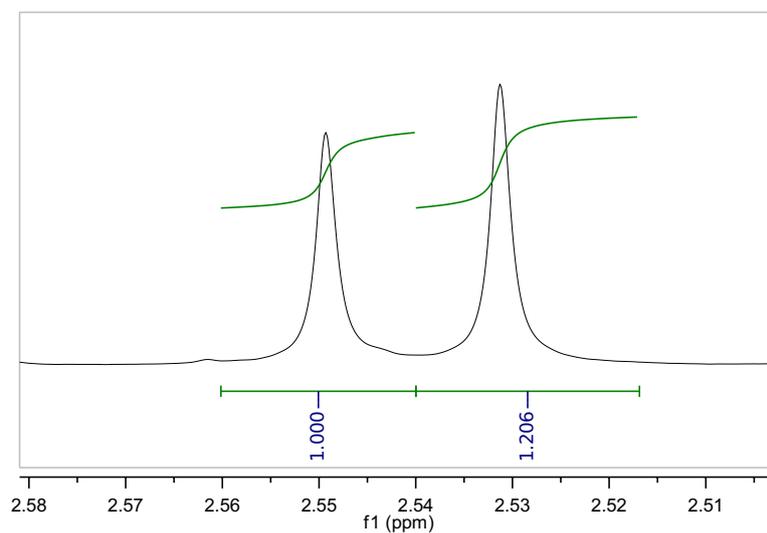
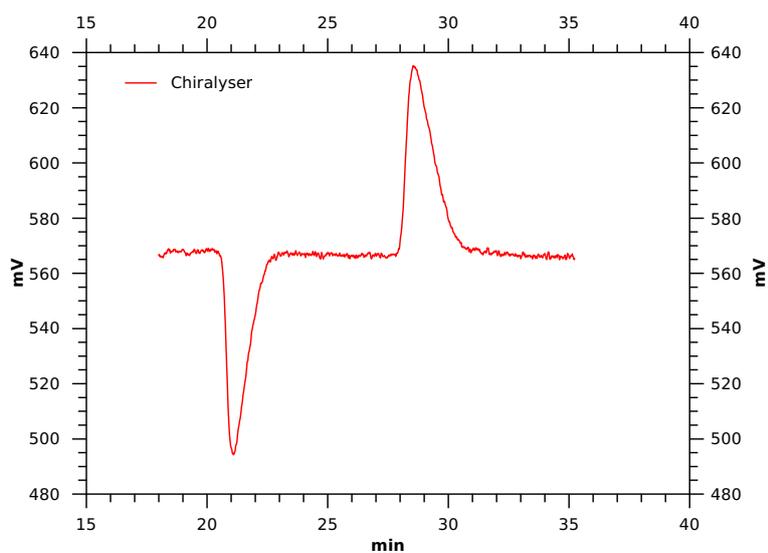


Figure S42: Part of the  $^1\text{H}$ -NMR spectrum of product of decyl(methyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 3, Table 3, main text).



	Reten. Time [min]	Start Time [min]	End Time [min]	Area [%]	W05 [min]
1	21.100	20.213	22.960	44.9	0.94
2	28.540	27.427	31.433	55.1	1.23

Figure S43: Part of the HPLC chromatogram of product of decyl(methyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 3, Table 3, main text). Conditions: **Daicel® Chiralpak AS-H** column, heptane/*i*-PrOH (17:3), 1 mL/min.

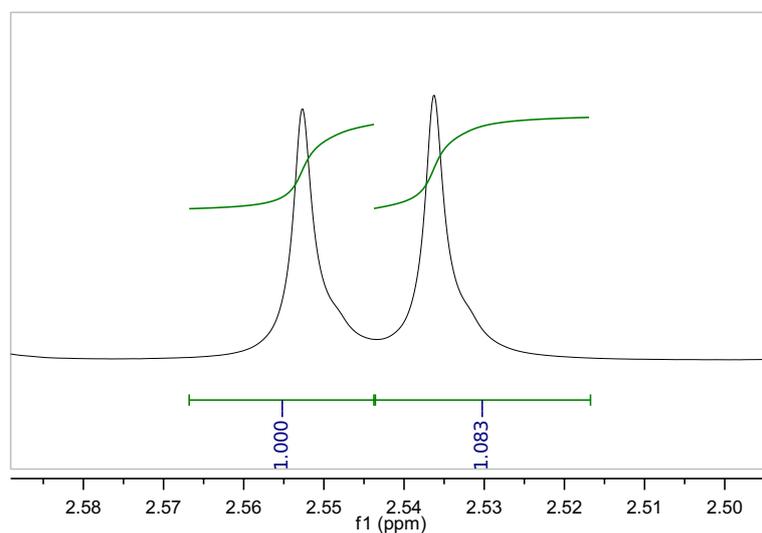


Figure S44: Part of the  $^1\text{H}$ -NMR spectrum of product of dodecyl(methyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 4, Table 3, main text).

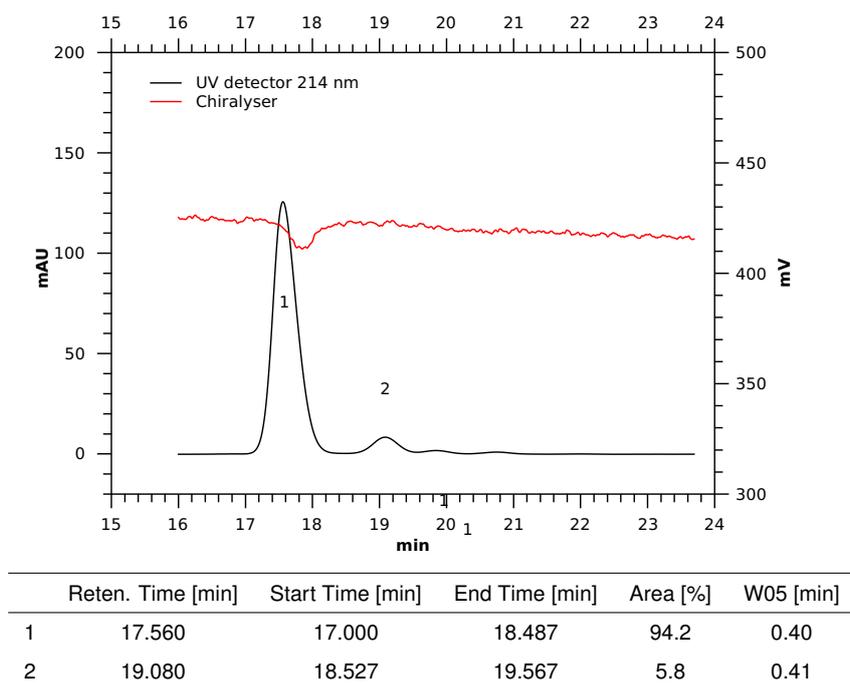


Figure S45: Part of the HPLC chromatogram of product of *tert*-butyl(methyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 5, Table 3, main text). Conditions: **Phenomenex® Lux 5u Cellulose-4** column, heptane/*i*-PrOH (9:1), 1 mL/min.

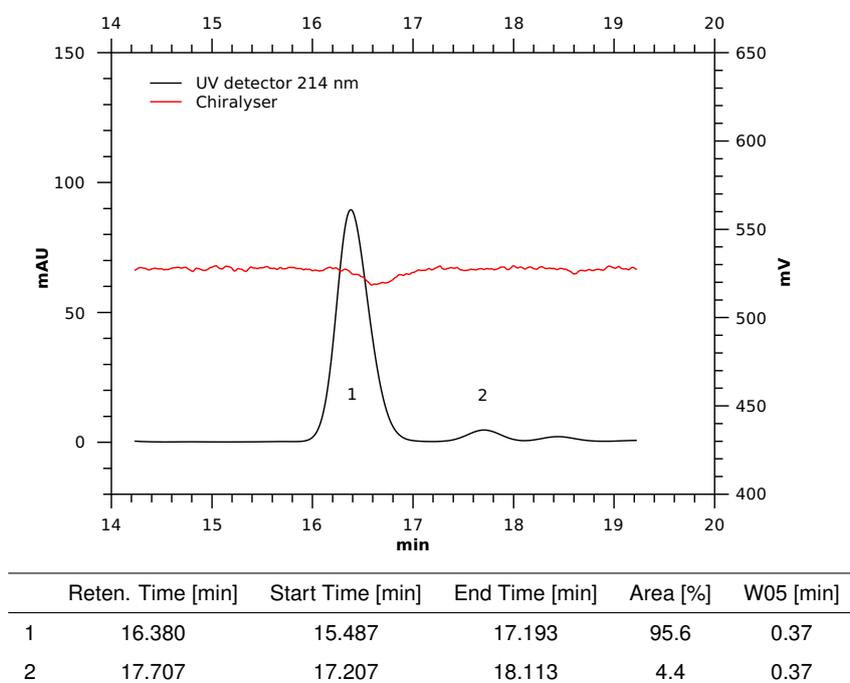


Figure S46: Part of the HPLC chromatogram of product of *tert*-butyl(methyl)sulfide oxidation catalyzed by **1b** (5 mol%), (Entry 6, Table 3, main text). Conditions: **Phenomenex® Lux 5u Cellulose-4** column, heptane/*i*-PrOH (9:1), 1 mL/min.

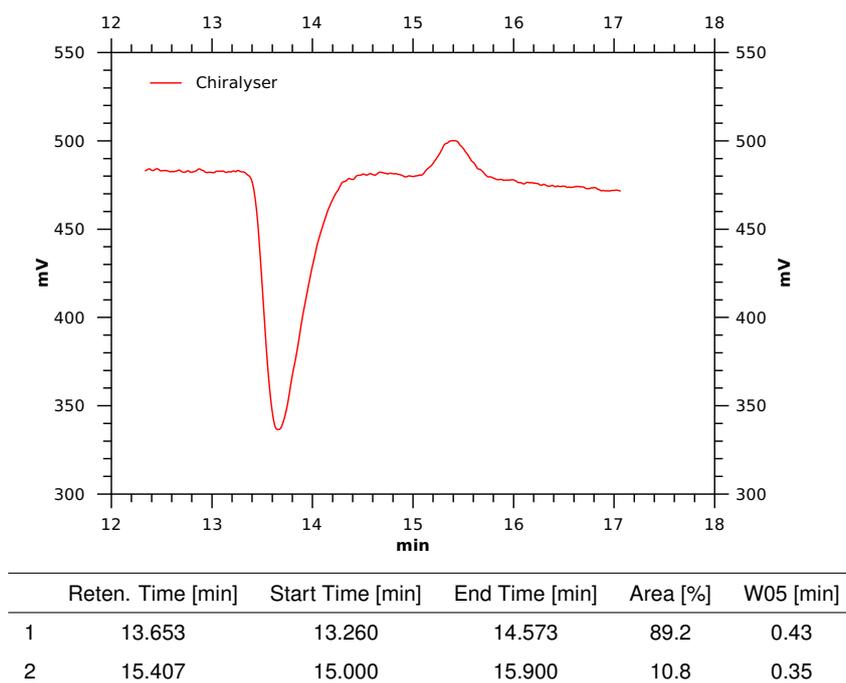


Figure S47: Part of the HPLC chromatogram of product of cyclohexyl(methyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 7, Table 3, main text). Conditions: **Phenomenex® Lux 5u Amylose-2** column, heptane/*i*-PrOH (17:3), 1 mL/min.

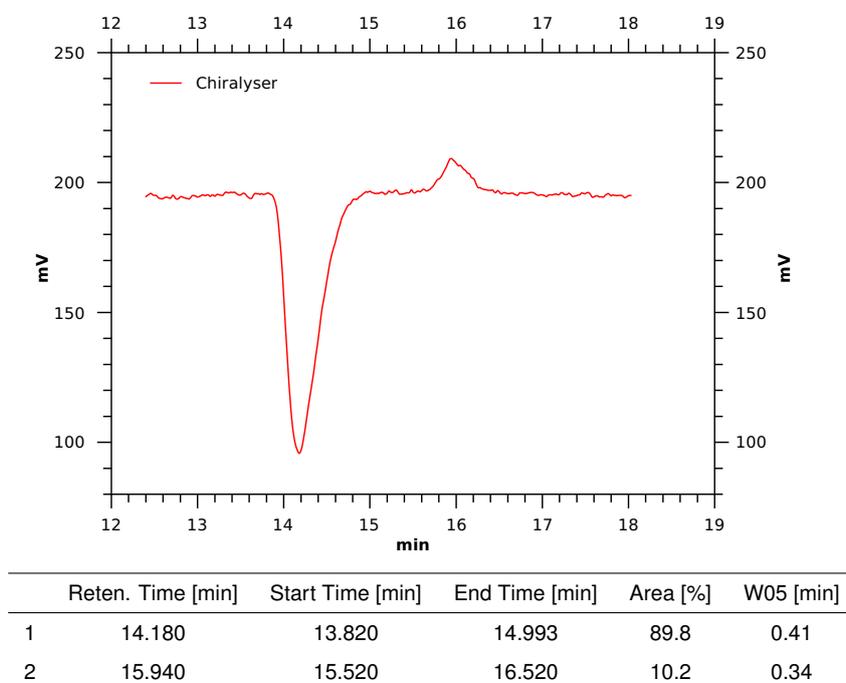


Figure S48: Part of the HPLC chromatogram of product of cyclohexyl(methyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 8, Table 3, main text). Conditions: **Phenomenex® Lux 5u Amylose-2** column, heptane/*i*-PrOH (17:3), 1 mL/min.

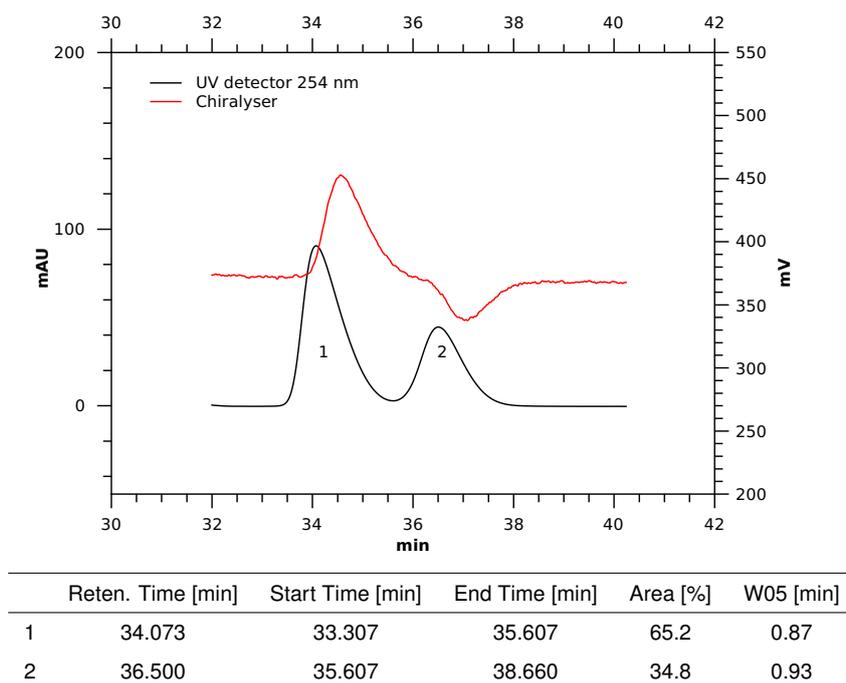


Figure S49: Part of the HPLC chromatogram of product of benzyl(methyl)sulfide oxidation catalyzed by **1b** (1 mol%), (Entry 9, Table 3, main text). Conditions: **Knauer® Eurocel 01 5 $\mu$ m** column, heptane/*i*-PrOH (17:3), 0.5 mL/min.

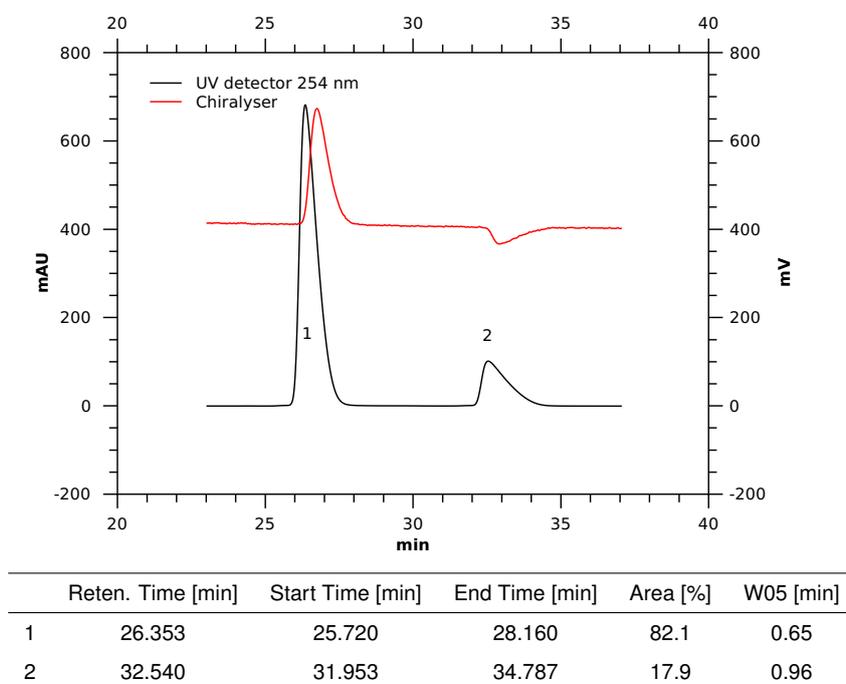


Figure S50: Part of the HPLC chromatogram of product of thioanisole oxidation catalyzed by **1b** (1 mol%), (Entry 11, Table 3, main text). Conditions: **Knauer® Eurocel 01 5 $\mu$ m** column, heptane/*i*-PrOH (9:1), 0.65 mL/min.

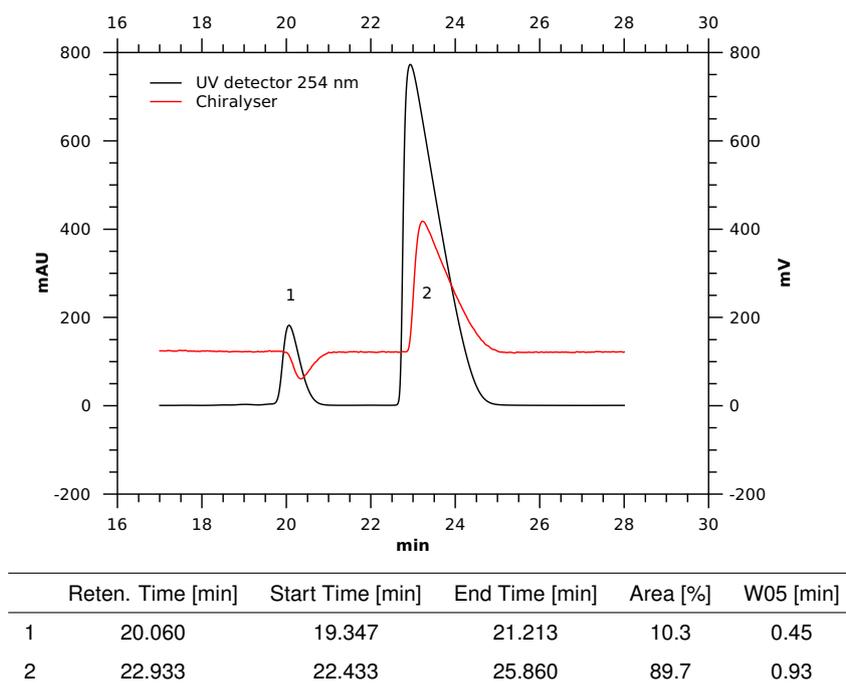


Figure S51: Part of the HPLC chromatogram of product of 4-methylthioanisole oxidation catalyzed by **1b** (1 mol%), (Entry 13, Table 3, main text). Conditions: **Phenomenex® Lux 5u Amylose-2** column, heptane/*i*-PrOH (17:3), 1 mL/min.

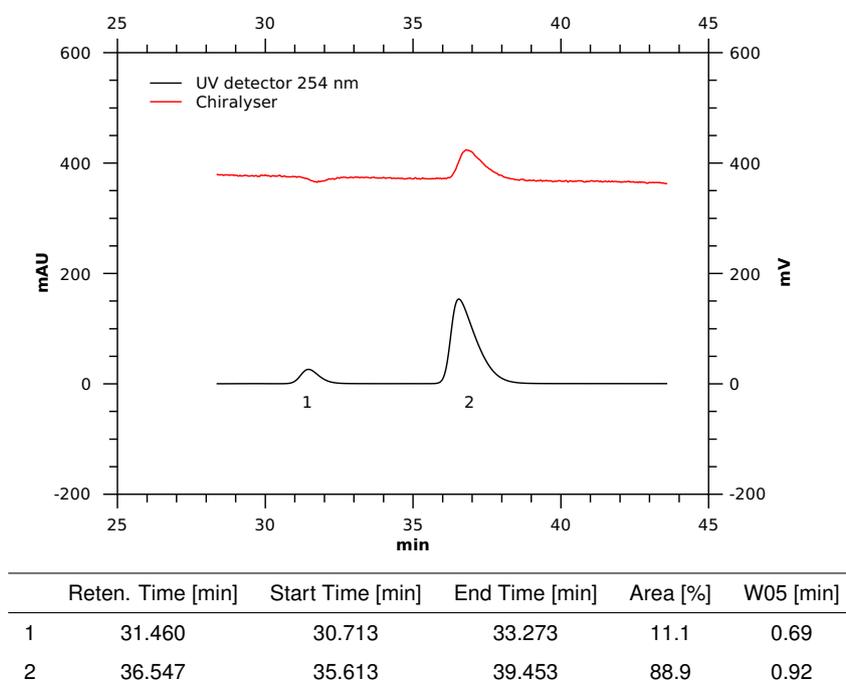


Figure S52: Part of the HPLC chromatogram of product of 4-methylthioanisole oxidation catalyzed by **1b** (0.3 mol%), (Entry 14, Table 3, main text). Conditions: **Phenomenex® Lux 5u Amylose-2** column, heptane/*i*-PrOH (9:1), 1 mL/min.

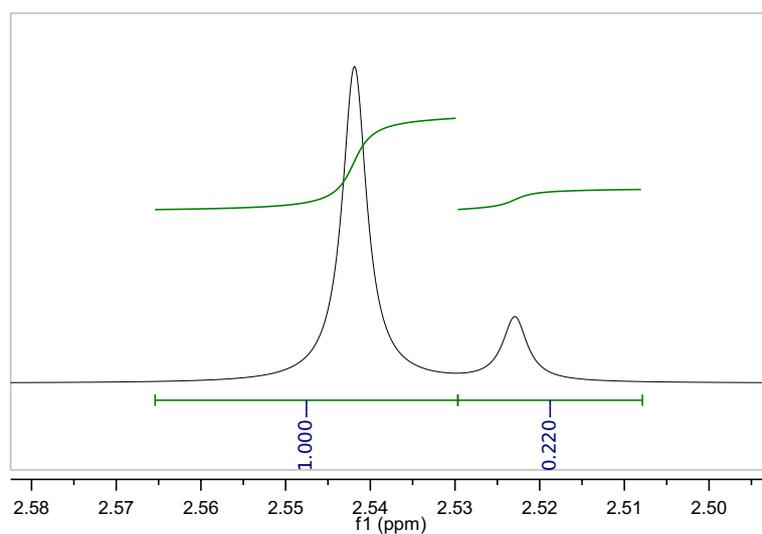
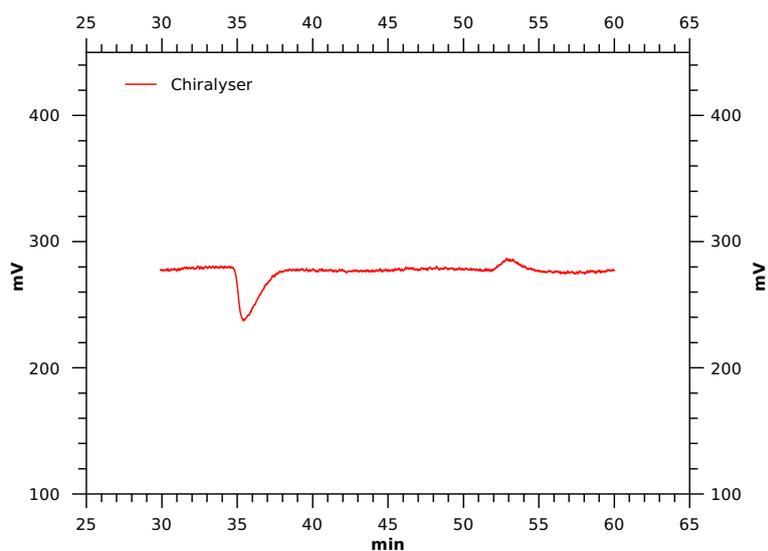


Figure S53: Part of the  $^1\text{H}$ -NMR spectrum of product of hexyl(octyl)sulfide oxidation catalyzed by **2b** (1 mol%), (Entry 1, Table 4, main text).



	Reten. Time [min]	Start Time [min]	End Time [min]	Area [%]	W05 [min]
1	35.380	34.440	38.487	80.4	1.51
2	52.853	51.847	55.467	19.6	1.31

Figure S54: Part of the HPLC chromatogram of product of hexyl(octyl)sulfide oxidation catalyzed by **2b** (1 mol%), (Entry 1, Table 4, main text). Conditions: **Daicel® Chiralpak AS-H** column, heptane/*i*-PrOH (17:3), 1 mL/min.

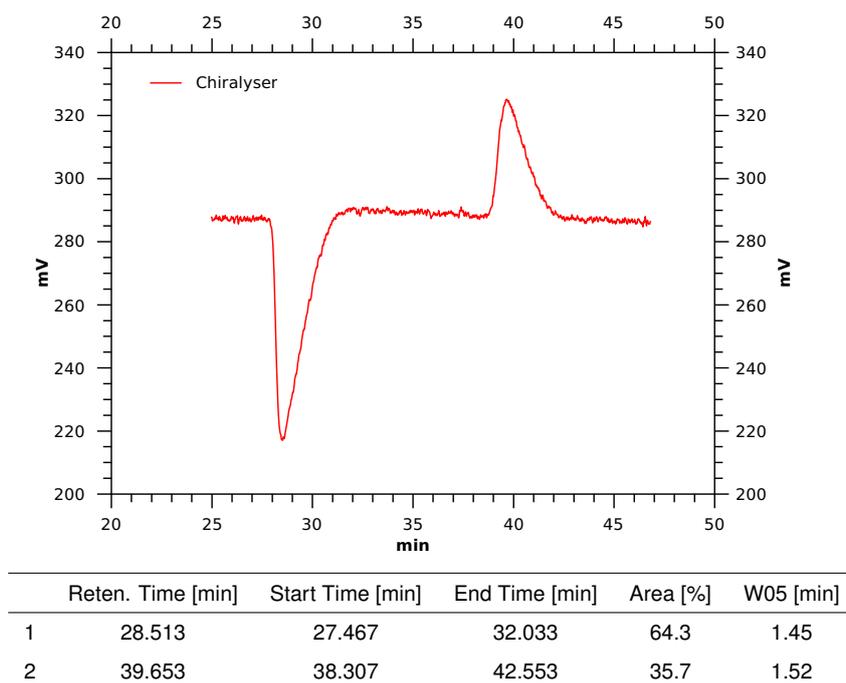


Figure S55: Part of the HPLC chromatogram of product of methyl(octyl)sulfide oxidation catalyzed by **2b** (1 mol%), (Entry 2, Table 4, main text). Conditions: **Daicel® Chiralpak AS-H** column, heptane/*i*-PrOH (17:3), 1 mL/min.

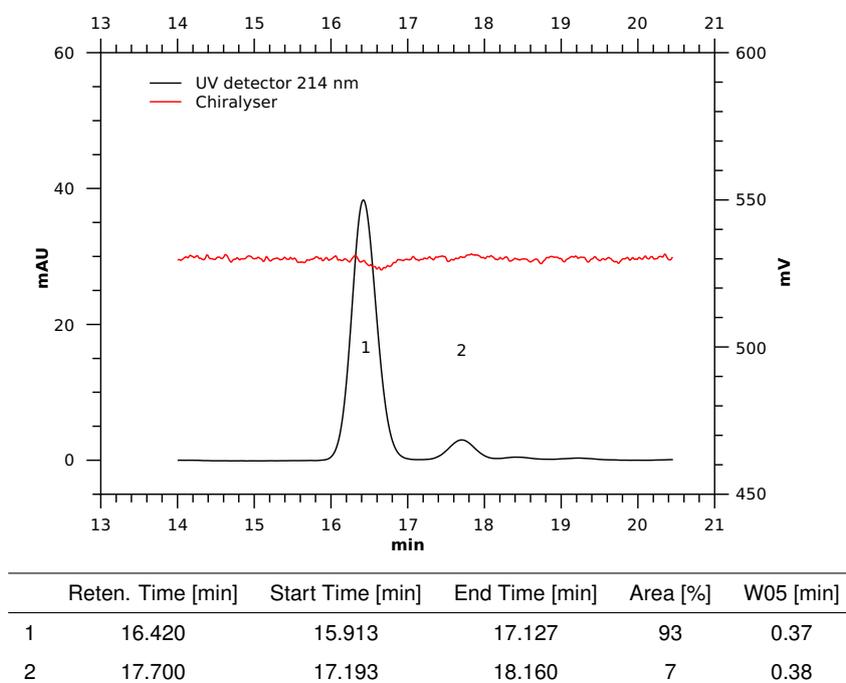


Figure S56: Part of the HPLC chromatogram of product of *tert*-butyl(methyl)sulfide oxidation catalyzed by **2b** (1 mol%), (Entry 5, Table 4, main text). Conditions: **Phenomenex® Lux 5u Cellulose-4** column, heptane/*i*-PrOH (9:1), 1 mL/min.

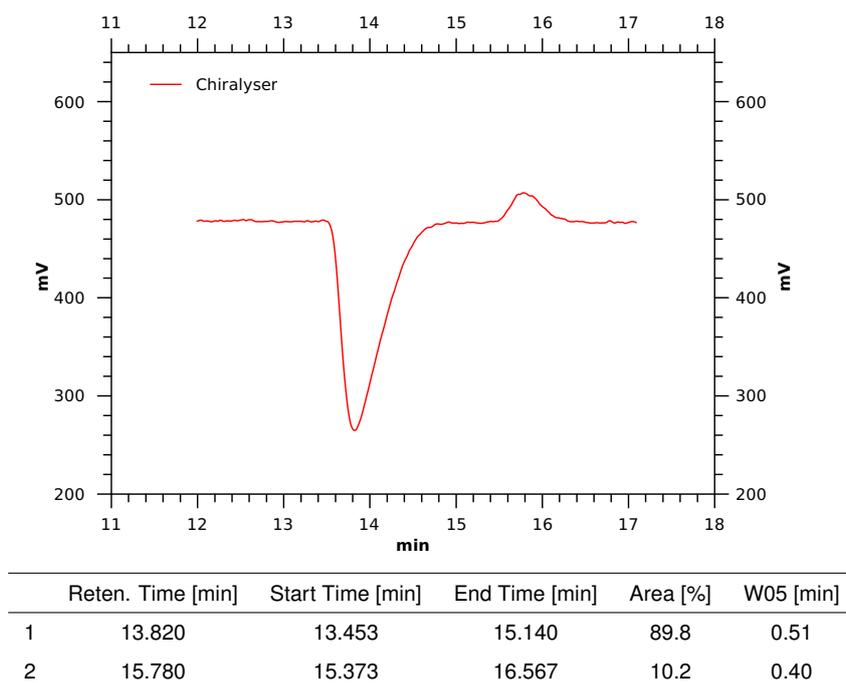


Figure S57: Part of the HPLC chromatogram of product of cyclohexyl(methyl)sulfide oxidation catalyzed by **2b** (1 mol%), (Entry 6, Table 4, main text). Conditions: **Phenomenex® Lux 5u Amylose-2** column, heptane/*i*-PrOH (17:3), 1 mL/min.

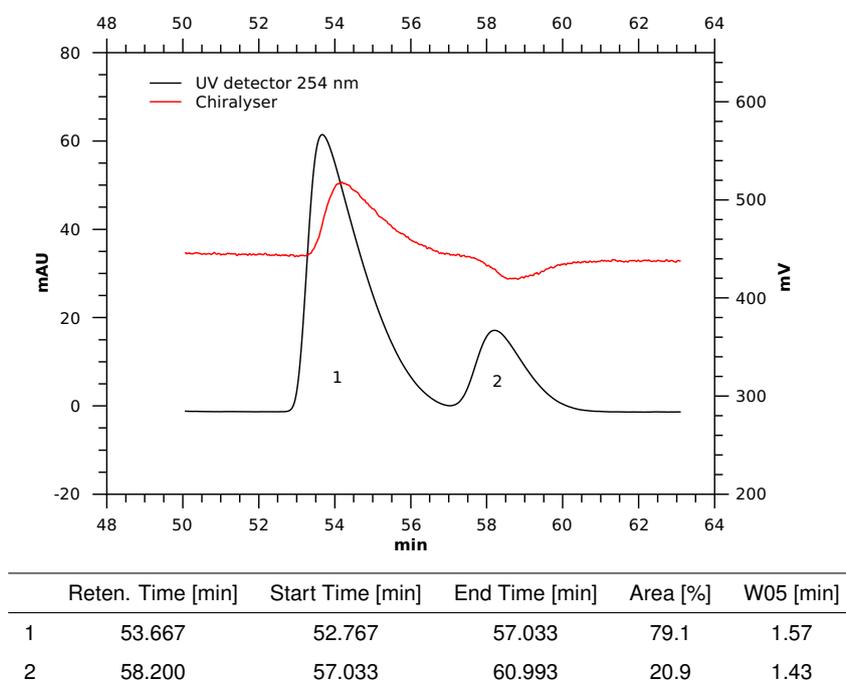
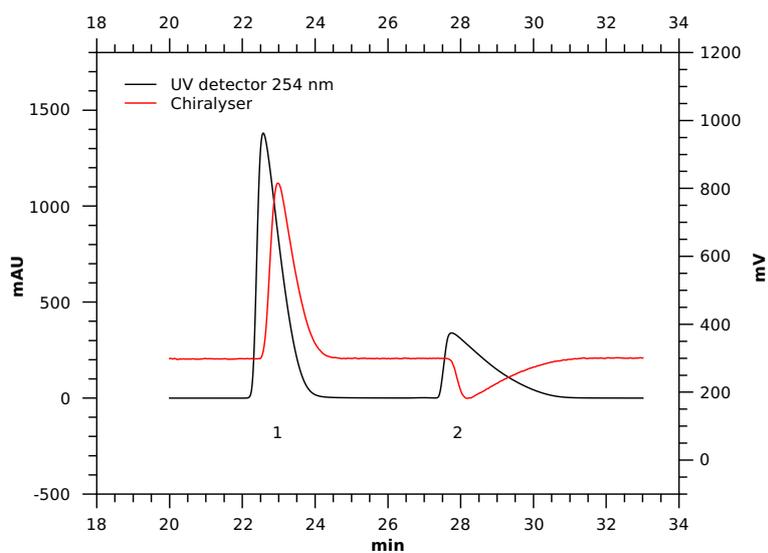
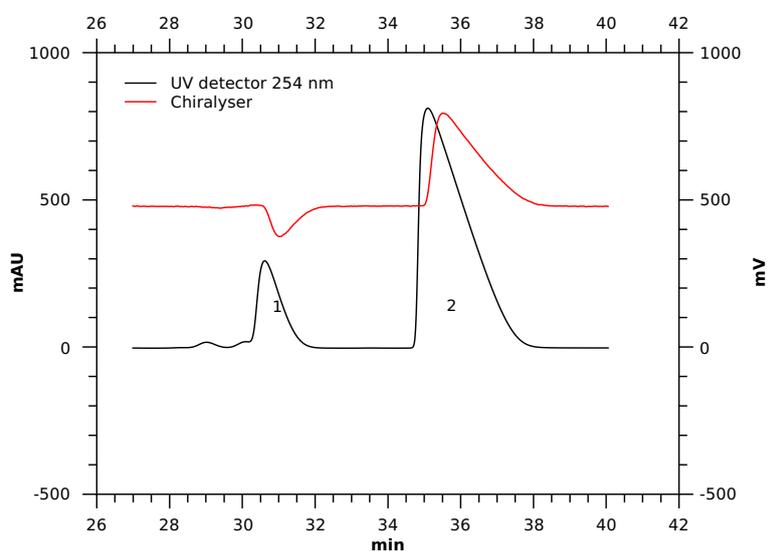


Figure S58: Part of the HPLC chromatogram of product of benzyl(methyl)sulfide oxidation catalyzed by **2b** (1 mol%), (Entry 7, Table 4, main text). Conditions: **Knauer® Eurocel 01 5µm** column, heptane/*i*-PrOH (9:1), 0.65 mL/min.



	Reten. Time [min]	Start Time [min]	End Time [min]	Area [%]	W05 [min]
1	22.573	22.033	24.800	67.8	0.70
2	27.747	27.300	31.373	32.2	1.32

Figure S59: Part of the HPLC chromatogram of product of thioanisole oxidation catalyzed by **2b** (1 mol%), (Entry 8, Table 4, main text). Conditions: **Knauer® Eurocel 01 5 $\mu$ m** column, heptane/*i*-PrOH (9:1), 0.65 mL/min.



	Reten. Time [min]	Start Time [min]	End Time [min]	Area [%]	W05 [min]
1	30.613	30.167	32.040	15.5	0.71
2	35.093	34.473	39.087	84.5	1.44

Figure S60: Part of the HPLC chromatogram of product of 4-methylthioanisole oxidation catalyzed by **2b** (1 mol%), (Entry 9, Table 4, main text). Conditions: **Phenomenex® Lux 5u Amylose-2** column, heptane/*i*-PrOH (9:1), 0.65 mL/min.

## References

- [1] V. Mojr, V. Herzig, M. Buděšínský, R. Cibulka and T. Kraus, *Chem. Commun.*, 2010, **46**, 7599–7601.
- [2] M. Deshmukh, E. Duňach, S. Juge and H. B. Kagan, *Tetrahedron Letters*, 1984, **25**, 3467–3470.