Supporting Information

Colour-Responsive Fluorescent Oxy Radical Sensors

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S1. Cyclic voltammogram for **Q1** in dry CH_3CN (0.2 M TBATFB) at a glassy carbon electrode at 50 mV s⁻¹ using a silver wire reference.



S2. Cyclic voltammogram for **Q2** in dry CH_3CN (0.2 M TBATFB) at a glassy carbon electrode at 50 mV s⁻¹ using a silver wire reference.



S3. Cyclic voltammogram for **Q3** in dry CH_3CN (0.2 M TBATFB) at a glassy carbon electrode at 50 mV s⁻¹ using a silver wire reference.



S4. Cyclic voltammogram for **Q4** in dry CH_3CN (0.2 M TBATFB) at a glassy carbon electrode at 50 mV s⁻¹ using a silver wire reference.



S5. Cyclic voltammogram for **Q5** in dry CH_3CN (0.2 M TBATFB) at a glassy carbon electrode at 50 mV s⁻¹ using a silver wire reference.



S6. ORTEP for X-ray determined structure of 14B (Q2). Thermal ellipsoids are shown at 15% probability.



S7. Computer calculated structures for **Q1** (top) and **Q2** (bottom) using DFT (B3LYP) and the 6-311G basis set.



S8. Absorption spectra recorded for **Q1** (black), **Q2** (red), **Q3** (blue), **Q4** (green) and **Q5** (cyano) in MeCN.

Solvent	n ^a	ε	λ_{ABS} / nm	λ_{EM} / nm	SS / cm ⁻¹	λ_{tot} / eV
THF ^c	1.405	7.58	439	606	6277	0.39
EA ^d	1.372	6.02	439	606	6277	0.39
DCM ^e	1.431	9.1	446	632	6599	0.41 (0.47) ^j
MeTHF ^f	1.403	6.97	441	598	5954	0.37
Bu ₂ O	1.399	3.1	446	585	5328	0.33
DMF ^g	1.431	36.7	430	640	7631	0.47
MeCN	1.344	37.5	433	644	7567	0.47
CHX ^h	1.426	2.02	467	561	3588	0.22
Toluene	1.493	2.4	447	595	5564	0.34
Pentane	1.358	2.1	438	556	4845	0.30
CCl ₄	1.460	2.2	459	587	4750	0.29

Table 2. Photophysical properties collected for Q2 in a range of solvents.

^arefractive index, ^bdielectric constant, ^cTetrahydrofuran, ^dethyl acetate, ^edichloromethane,

^fmethyltetrahydrofuran, ^gN,N-dimethylformamide, ^hcyclohexane. ^jcalculated by $\lambda_{tot} = \lambda_{ABS} - \Delta E$



S9. Absorption and emission profiles recorded for **Q1** in MeCN (top) and cyclohexane (bottom).



S10. Absorption and emission profiles recorded for Q4 in cyclohexane (top) and MeCN (bottom).



S11. Absorption and emission profiles recorded for Q5 in cyclohexane (top) and MeCN (bottom)



Figure S1. ¹H NMR spectrum of compound 8B.

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Figure S2. ¹H NMR spectrum of compound **8B** (8.5–7.0 ppm).



Figure S3. ¹H NMR spectrum of compound **8B** (2.5–0.0 ppm).







Figure S6. ¹H NMR spectrum of compound 9A.



Figure S7. ¹H NMR spectrum of compound **9A** (2.5–0.0 ppm).





Figure S9. ¹³C NMR spectrum of compound 9A (45–0.0 ppm).





Figure S11. ¹H NMR spectrum of compound 13A.



Figure S12. ¹H NMR spectrum of compound **13A** (2.2–0.0 ppm).



Figure S13. ¹H NMR spectrum of compound **13A** (8.5–5.5 ppm).



Figure S14. ¹³C NMR spectrum of compound 13A.



Figure S15. ¹³C NMR spectrum of compound 13A (60–0.0 ppm).



Figure S16. ¹H NMR spectrum of compound Q1.



Figure S17. ¹H NMR spectrum of compound Q1 (8.8–4.6 ppm).





Figure S19. ¹³C NMR spectrum of compound Q1.



Figure S20. ¹³C NMR spectrum of compound Q1 (80–0.0 ppm).





Figure S22. ¹H NMR spectrum of compound Q2.



Figure S23. ¹H NMR spectrum of compound **Q2** (8.5–7.1 ppm).



Figure S24. ¹H NMR spectrum of compound Q2 (5.0–0.0 ppm).



Figure S25. ¹³C NMR spectrum of compound Q2.



Figure S26. ¹³C NMR spectrum of compound Q2 (160–115 ppm).



Figure S27. ¹³C NMR spectrum of compound Q2 (42–21 ppm).



Figure S28. ¹H NMR spectrum of compound Q3.



Figure S29. ¹H NMR spectrum of compound **Q3** (1.55–1.0 ppm).



Figure S30. ¹H NMR spectrum of compound **Q3** (5.5–2.0 ppm).

12.4

20.6

44.2

20.9

13.9

13.9

24.3

42.7

24.4

11.2

4.2

12.4

11.7

28.5

20.1

15.0 25.1

ppm



INDEX	FREQUENCY	PPM	HEIGHT
1	394.4	0.657	38.9
2	386.8	0.645	103.7
3	384.4	0.641	5.0
4	3/9./	0.633	31.0
5	326.3	0.544	4.0
6	318.7	0.531	7.2
7	314.6	0.524	7.1
8	310.5	0.518	10.0
9	306.3	0.511	7.0
10	302.2	0.504	5.8
11	58.1	0.097	7.4



Figure S31. ¹H NMR spectrum of compound **Q3** (0.8–0.0 ppm).



Figure S32. ¹³C NMR spectrum of compound Q3.



Figure S33. ¹H NMR spectrum of compound Q4.





Figure S35. ¹H NMR spectrum of compound **Q4** (5.0–0.0 ppm).







Figure S38. ¹³C NMR spectrum of compound Q4 (60–0.0 ppm).



Figure S39. ¹³C NMR spectrum of compound Q4 (100–50 ppm).



Figure S40. ¹H NMR spectrum of compound Q5.



Figure S41. ¹H NMR spectrum of compound **Q5** (9.0–7.0 ppm).



Figure S42. ¹H NMR spectrum of compound **Q5** (5.0–0.0 ppm).



Figure S43. ¹³C NMR spectrum of compound Q5.



Figure S44. ¹³C NMR spectrum of compound Q5 (160–115 ppm).



Figure S45. ¹³C NMR spectrum of compound Q5 (60–0.0 ppm).

