# **Electronic Supporting Informations**

### Syntheses, Optical and Intramolecular Magnetic Properties of Monoand Di-Nitronyl Nitroxide and Oxoverdazyl Radicals Appended to 2,6-Bispyrazolylpyridine Core

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#### **GENERAL EXPERIMENTAL PROCEDURES:**

Synthesis of tert-Butyl-2-isopropenylhydrazenecarboxylate (16). Tertbutoxy carbazate (2.0 g, 15.1 mmol) was added to dry acetone (10 ml) at rt and stir for 12 h. The resulting mixture was evaporated to get white precipitate. Yield 2.0 g (73%). <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub> 27°C),  $\delta_{\rm H} = 1.38$ (9H, s), 1.94 (6H, s), 7.0 (1H, s) ppm.



Synthesis of tert-Butyl-2-isopropylhydrazenecarboxylate (17). Sodium borohydride (1 g, 26.5



mmol) was added to (16) (1.5 g, 5.29 mmol) in ethanol (150 ml) and the mixture was refluxed for 6 h. The resulting mixture was evaporated, and 60 ml of water and 23.3 ml of 1 M HCl aq. was added to the residue. The crude product was extracted with dichloromethane and purified by

recrystallization from dichloromethane/ hexane. Yield: 1.0 g (66%) <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub> 27°C), 1.07(6H, d, J=2.8), 1.38(9H, s,tert-butyl), 2.0(1H, t), 2.97(1H, sept), 1.0 (1H, d) ppm.

**2,4-diisopropylcarbono-hydrazide bis-hydrochloride**: Synthesized from **17** as per the reported procedure by Parè et al. [Ref.: Parè, E. C.; Brook, D. J. R.; Brieger, A.; Badik, M.; Schinke, M. *Org. Biomol. Chem.*, **2005**, *3*, 4258-4261.]

**2,3-diamino-2,3-dimethylbutane:** Synthesized from 2,3-dinitro-2,3-dimethylbutane as per the reported procedure by Ray et al. [Ref. Hirel, C.; Vostrikova, K. E.; Pècaut, J.; Ovcharenko, V. I.; Rey, P. *Chem. -Eur. J.* **200**1, *7*, 2008].





### ESR Spectroscopic Analysis:



**FIGURE S 1.** ESR spectra ( $\Delta ms = \pm 1$ ) of **1** and **2** at (c =  $10^{-3}$  M in toluene) 298 – 133 K



**FIGURE S 2.** ESR spectra ( $\Delta ms = \pm 1$ ) of **4** and **5** at (c = 10<sup>-3</sup> M in toluene) 273 – 133 K.



FIGURE S3. Atomic spin density distribution for biradicals 3 and 6 (DFT; B3LYP /6–31+G(d)).

**TABLE S1.** Singlet-Triplet splitting ( $\Delta E_{ST}$ ) (DFT; B3LYP /6–31+G(d); The values are in kJ/mol.)

| Molecule geometry                           | Compound 3 | Compound 6 |
|---|------------|------------|
| $\theta_1 = \theta_2 = 0^\circ$             | 126.2135   | 110.8310   |
| $\theta_1 = 0^\circ, \ \theta_2 = -10$      | 123.0737   | 110.9310   |
| $\theta_1 = 10^\circ, \theta_2 = -10^\circ$ | 120.5419   | 110.6955   |
| $\theta_1 = 20^\circ, \theta_2 = -20^\circ$ | 120.9216   | 109.9480   |
| $\theta_1 = 10^\circ, \theta_2 = 0^\circ$   | 120.7462   | 111.0150   |

Table S2. Crystal data and structure refinement for 1 (CCDC: 768145)

| Empirical formula                     | $\mathbf{C}_{24}\mathbf{H}_{24}\mathbf{N}_{7}\mathbf{O}_{2}$ |
|---------------------------------------|--|
| Formula weight                        | 442.50   |
| Temperature                           | 100(2) K   |
| Wavelength                            | 0.71073 A  |
| Crystal system, space group           | Monoclinic, P2(1)/c  |
|                                       |  |
| Unit cell dimensions                  | a = 14.584(2) A alpha = 90 deg.                              |
|                                       | b = 10.5609(14) A beta = 109.665(2) deg.                     |
|                                       | c = 15.240(2) A gamma = 90 deg.                              |
|                                       | 2210 100 83  |
| Volume                                | 2210.4(5) A <sup>3</sup>                                     |
| Z, Calculated density                 | 4, 1.330 Mg/m <sup>3</sup>                                   |
| Absorption coefficient                | 0.089 mm <sup>-1</sup>                                       |
| F(000)                                | 932  |
| Crystal size                          | 0.36 x 0.22 x 0.16 mm  |
| Theta range for data collection       | 1.48 to 26.04 deg.   |
| Limiting indices                      | -17<=h<=18, -12<=k<=13, -18<=l<=18                           |
| <b>Reflections collected / unique</b> | 22106 / 4347 [R(int) = 0.0980]                               |
| <b>Completeness to theta = 26.04</b>  | 99.8 %   |
| Absorption correction                 | Semi-empirical from equivalents                              |
| Max. and min. transmission            | 0.9859 and 0.9686  |
| Refinement method                     | Full-matrix least-squares on F <sub>2</sub>                  |
| Data / restraints / parameters        | 4347 / 0 / 302   |
| Goodness-of-fit on F^2                | 1.042  |
| Final R indices [I>2sigma(I)]         | R1 = 0.0651, wR2 = 0.1373                                    |
|                                       |  |
| R indices (all data)                  | R1 = 0.1084, wR2 = 0.1542                                    |
| Largest diff. peak and hole           | 0.321 and -0.332 e.Å <sup>-3</sup>                           |

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**FIGURE S6:** Comparative <sup>1</sup>H-NMR spectra of radical precursors **10** and **13** obtained from the corresponding aldehyde **7**.

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**FIGURE S7:** Comparative <sup>1</sup>H-NMR spectra of radical precursors **11** and **14** obtained from the corresponding aldehyde **8**.

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Sample ID:

Analysis type:

# **S12**

### FLASH EA 1112 SERIES CHN REPORT SCHOOL OF CHEMISTRY UNIVERSITY OF HYDERABAD

CHO





| Element Name                   | Element %                 | Ret. Time               |
|--------------------------------|---------------------------|-------------------------|
| Nitrogen<br>Carbon<br>Hydrogen | 22. 45<br>68. 65<br>4. 09 | 0. 78<br>1. 18<br>3. 77 |
|                                |                           |                         |

PHO

CBh





#### LCMS-2010A DATA REPORT SCHOOL OF CHEMISTRY UNIVERSITY OF HYDERABAD



User Sample Inj. Volume Data Name Method Name : Admin : PHM7 : 5.000 : C:\LCMSsolution\User\Data\PHM7-APCI-POS1.qld : C:\LCMSsolution\User\Method\esi.qlm





OHCYCHO

Cosh









| Element Name | Element % | Ret. Time |
|--------------|-----------|-----------|
| Nitrogen     | 23. 85    | 0. 73     |
| Carbon       | 69. 58    | 1. 16     |
| Hydrogen     | 6. 51     | 3. 97     |
|              |           |           |

Bh





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100 90 80 70 60 50 40 30 20 10 0



-10 -20 -30

-40 -50 -60 -70 -80

ppm

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**S26** 

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#### FLASH EA 1112 SERIES CHN REPORT SCHOOL OF CHEMISTRY UNIVERSITY OF HYDERABAD

rodical

**S45** 

Method filename:<br/>Sample ID:D:\Program Files\Thermo Finnigan\Eager 300 for EA1112\DATA\Sys\_data\_ex<br/>PHE-3 (# 8)<br/>UnkNown<br/>UNK-21062009-1.dat<br/>1.056



| Element Name | Element % | Ret. Time |
|--------------|-----------|-----------|
| Nitrogen     | 21. 32    | 0. 79     |
| Carbon       | 62. 15    | 1. 20     |
| Hydrogen     | 5. 95     | 3. 92     |

osh