
Supplementary Information

4,4,9,9-Tetraphenyl Pyrroloindolizine: A Structural Analogue of Calix[2]pyrrole

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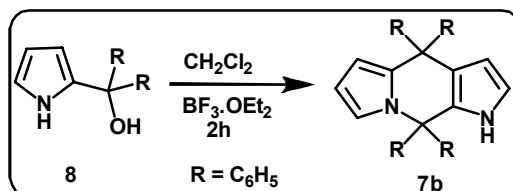
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1. General Information

All the chemicals were of the best grade commercially available and are used without further purification. NMR spectra were recorded on a Bruker Avance Bruker AMX 500 spectrophotometer with CDCl_3 as solvent. Chemical shifts are reported as δ in units of parts per million (ppm) relative to TMS. FAB mass spectra were obtained on a JEOL SX-120/DA6000 spectrometer using argon (6 KV, 10 mA) as the FAB gas. FT-IR spectra were recorded on a Nicolet Impact 400D Infrared spectrophotometer. Melting points were determined on a Buchi melting point apparatus and are uncorrected. The single crystal X-ray diffraction data was collected on a Bruker AXS Kappa Apex 2 CCD diffractometer at 293(2) K for **7b**.

2. Synthesis of 7b:



To the stirred solution of 2-(diphenylhydroxymethyl)pyrrole, **8** (450 mg, 3.58 mmol) in dry dichloromethane (250 mL) under argon atmosphere at room temperature, $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (0.04 mL, 0.36 mmol) was added slowly and allowed to stir for 2 h. The reaction mixture was then quenched with aq. NaOH solution and extracted with dichloromethane. The organic layer was then washed with water and brine solution. After drying over anhydrous Na_2SO_4 , the solvent was evaporated under vacuum. The dark residue obtained was purified by silica gel column chromatography (100 – 200 mesh). The colorless fraction eluted with hexane as the eluent was identified as **7b**. The white solid was recrystallized from CH_2Cl_2 and petroleum ether in 20% yield.

Spectral data of **7b**: m.p: 263°C ; ^1H NMR (500 MHz, CDCl_3 , 298 K): δ = 7.60 (brs, NH, 1H), 7.16 (m, CH-phenyl, 6H), 7.06 (s, CH-phenyl, 10H), 6.92 (d, J = 8.5 Hz, CH-phenyl, 4H), 6.68 (t, J = 5.5 Hz, CH-Pyrrole, 1H), 6.42 (m, CH-pyrrole, 1H), 6.13 (t, J = 6.5 Hz, CH-pyrrole, 1H), 5.96 (t, J = 5.5 Hz, CH-pyrrole, 1H), 5.77 (m, CH-pyrrole, 1H); ^{13}C NMR (125 MHz, CDCl_3 , 298 K): δ = 147.42, 142.70, 139.02, 130.64, 129.20, 128.61, 127.99, 127.58, 127.19, 125.59, 125.13, 118.69, 117.96, 107.87, 67.97, 51.86; FT-IR: (KBr, cm^{-1}) = 3408, 3053, 3024, 2961, 2922, 1597, 1486, 1445, 1266, 1181, 1069, 1028, 874, 752, 522, 507; FAB MS (m/z): Calcd. for $\text{C}_{34}\text{H}_{26}\text{N}_2$ 462.21; Observed 464.29 ($M+2$); Anal. Calcd. for $\text{C}_{34}\text{H}_{26}\text{N}_2$: C, 88.28; H, 5.67; N, 6.06; Found: C, 88.26; H, 5.65; 6.03.

3. Spectral analyses of 7b:

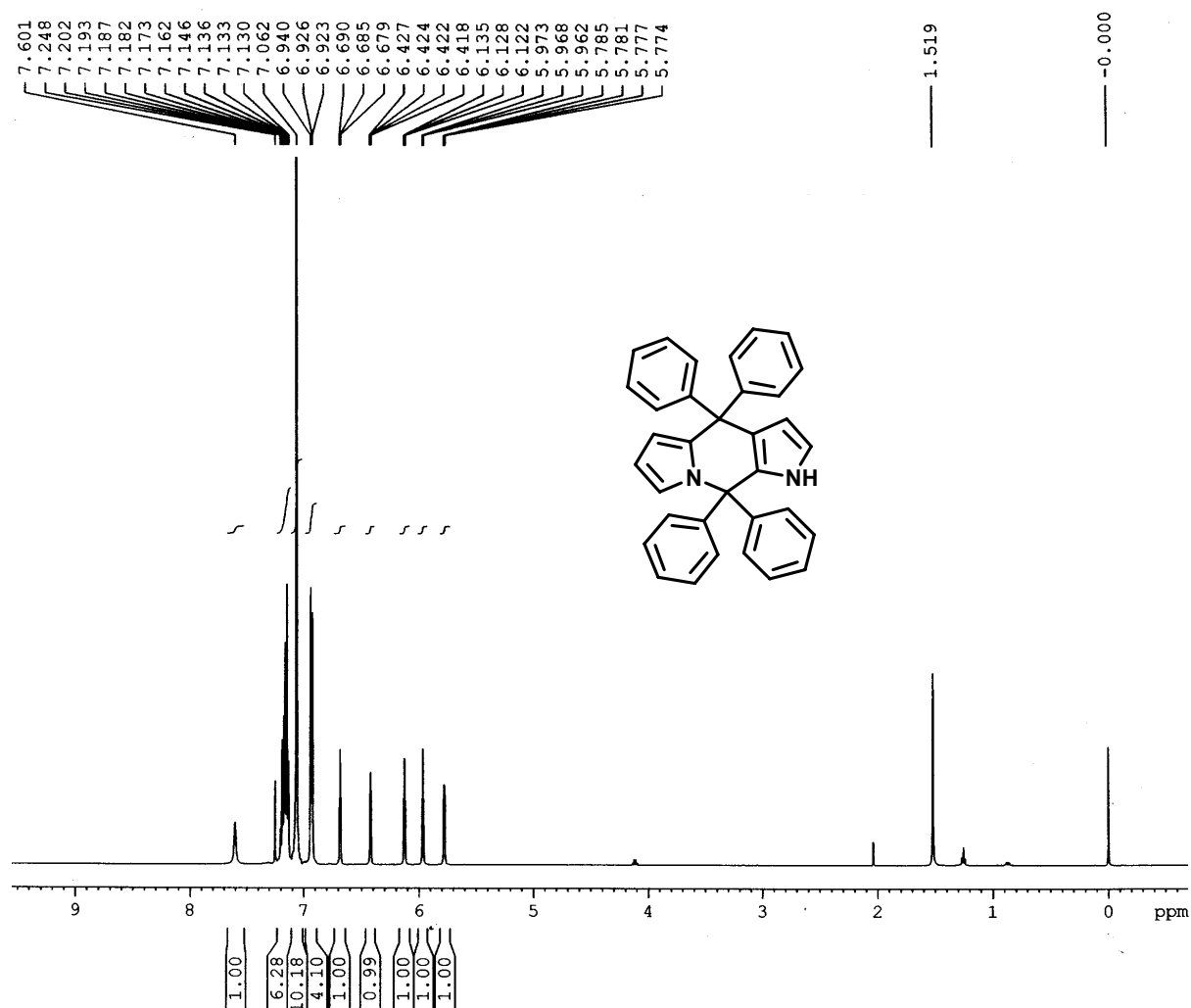


Figure S1. ¹H-NMR Spectrum of **7b** in CDCl₃

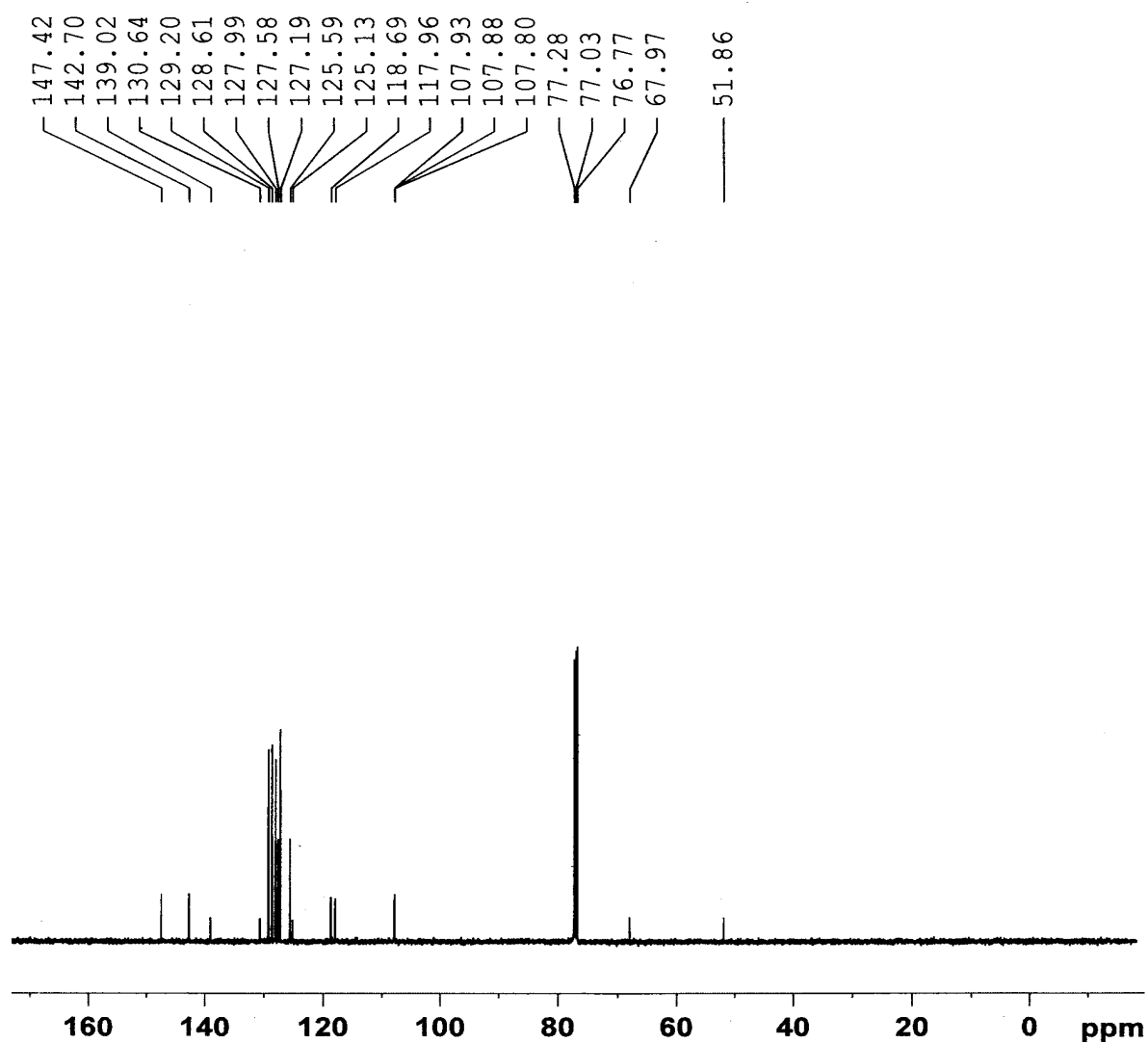


Figure S2. ^{13}C -NMR Spectrum of **7b** in CDCl_3

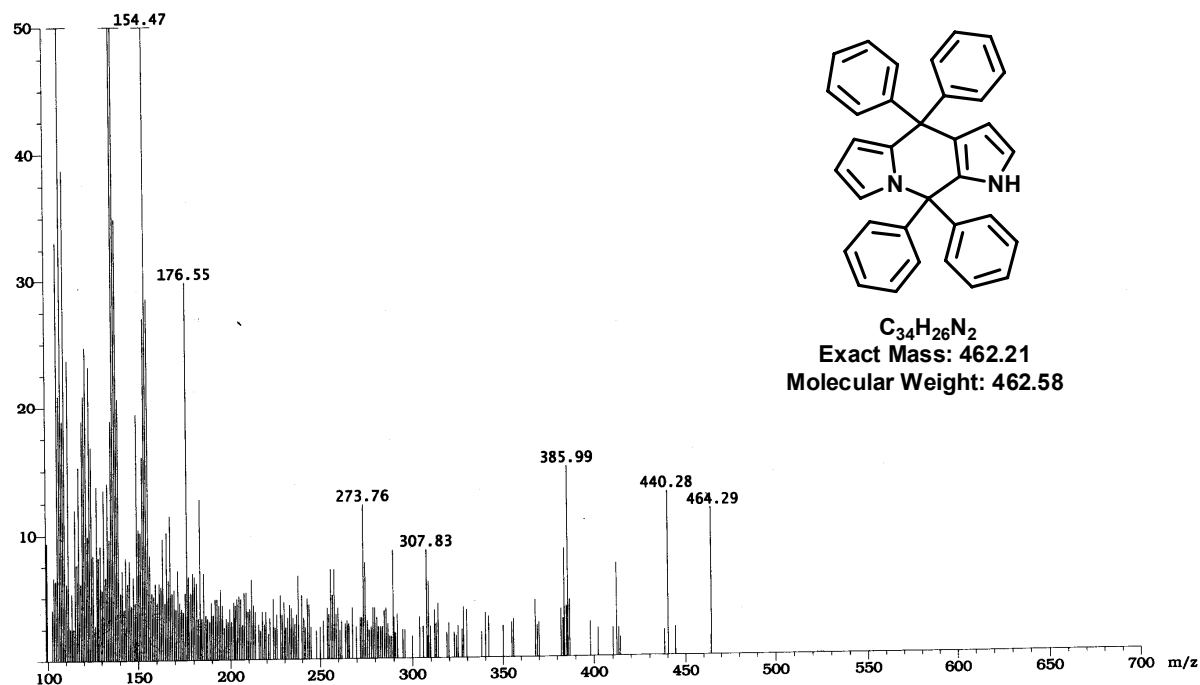


Figure S3. FAB MS spectrum of **7b**

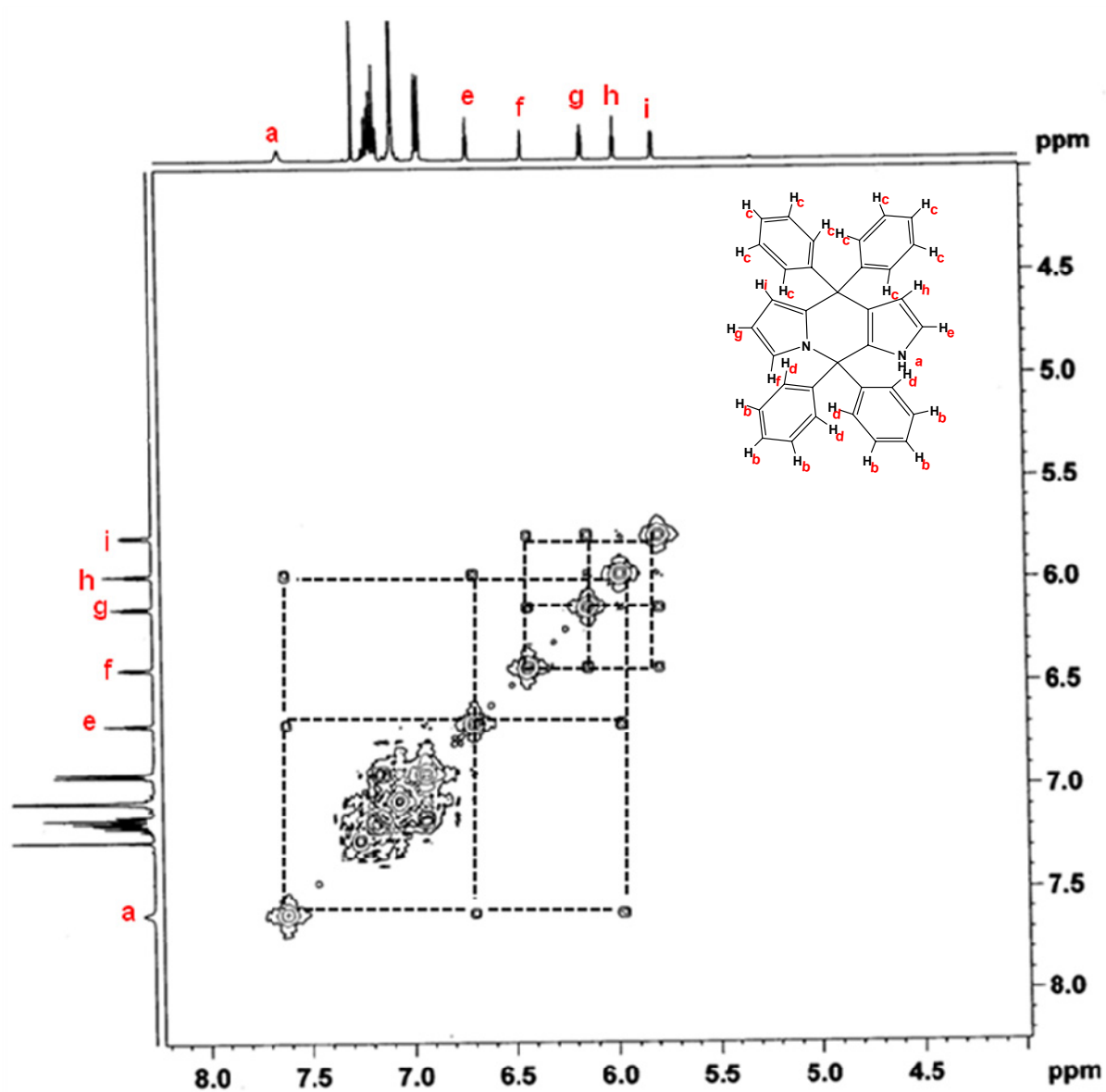


Figure S4. ^1H - ^1H COSY Spectrum of **7b** in CDCl_3

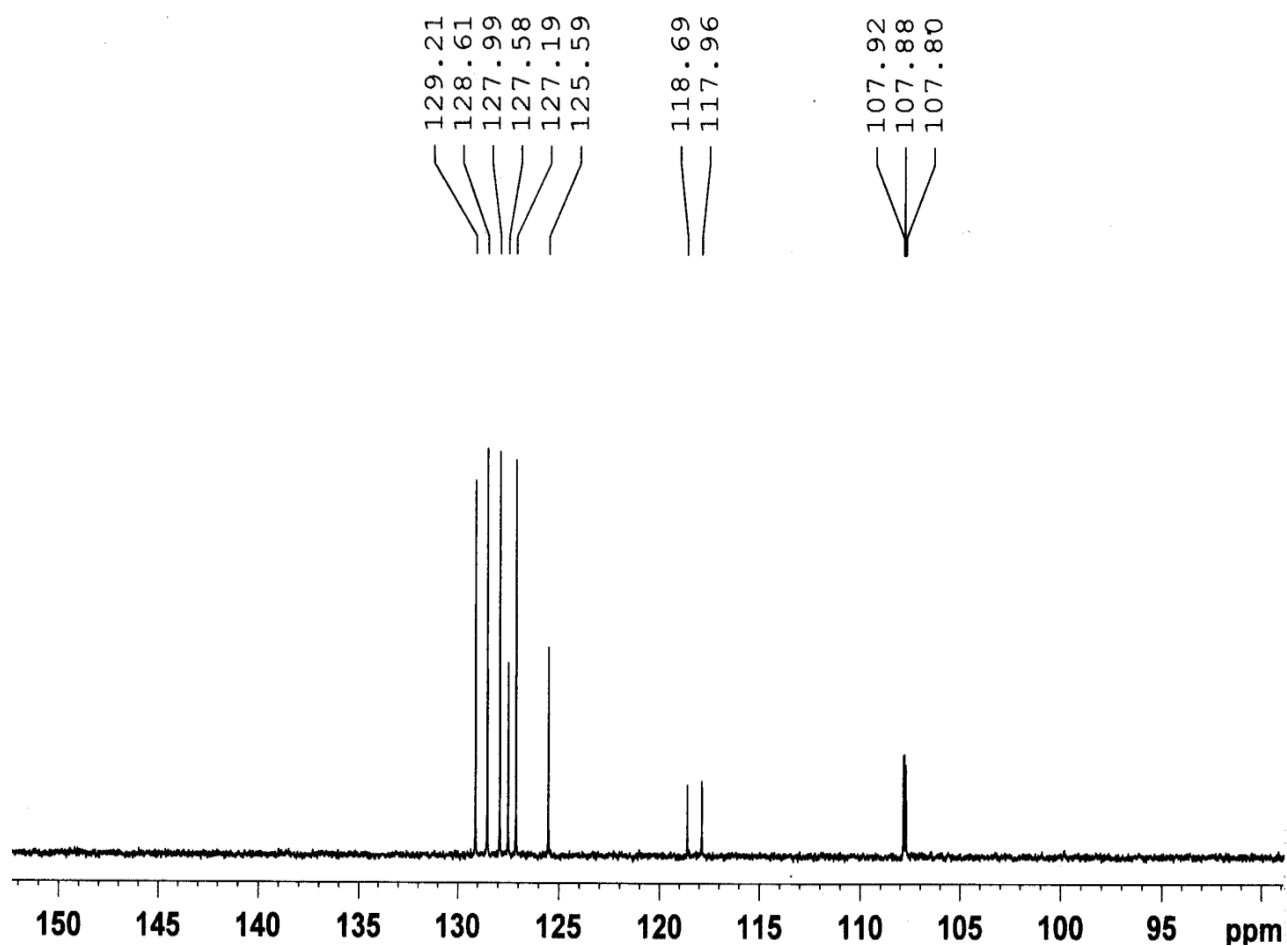


Figure S5. DEPT 45 Spectrum of **7b** in CDCl₃

4. Single crystal X-ray structure and analysis of **7b**:

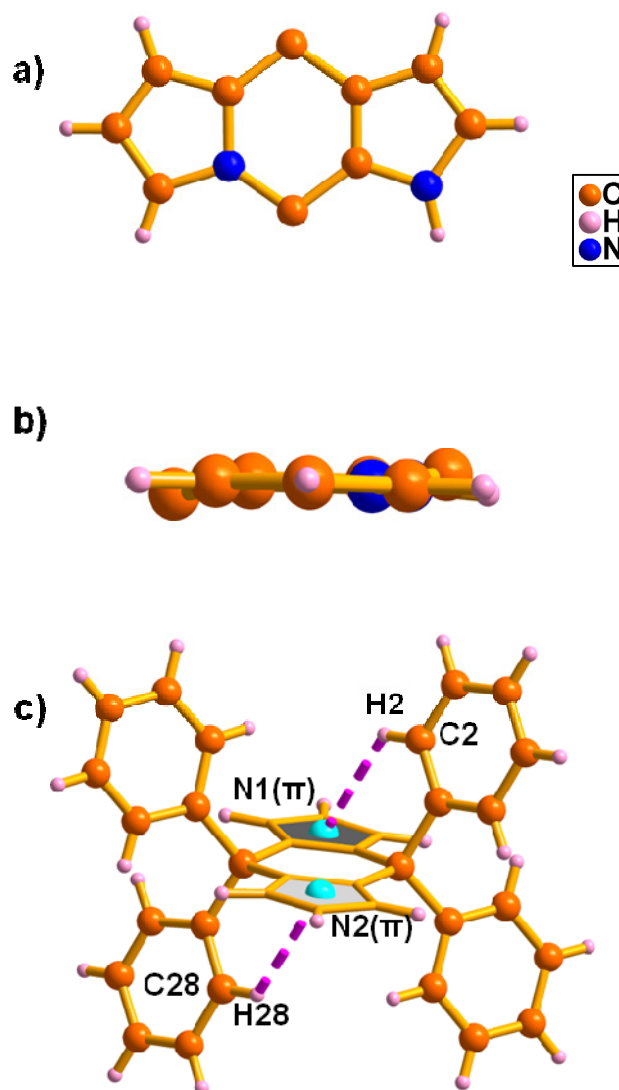


Figure S6. Single crystal X-ray structure of **7b**. a) top view, b) side view and c) top view with intramolecular hydrogen bonding interactions. The *meso*-diaryl groups in (a and b) are omitted for clarity. The distances and angles are C28-H28...N1: 2.83 Å, 127° and C2-H2...N2: 2.81 Å, 127°.

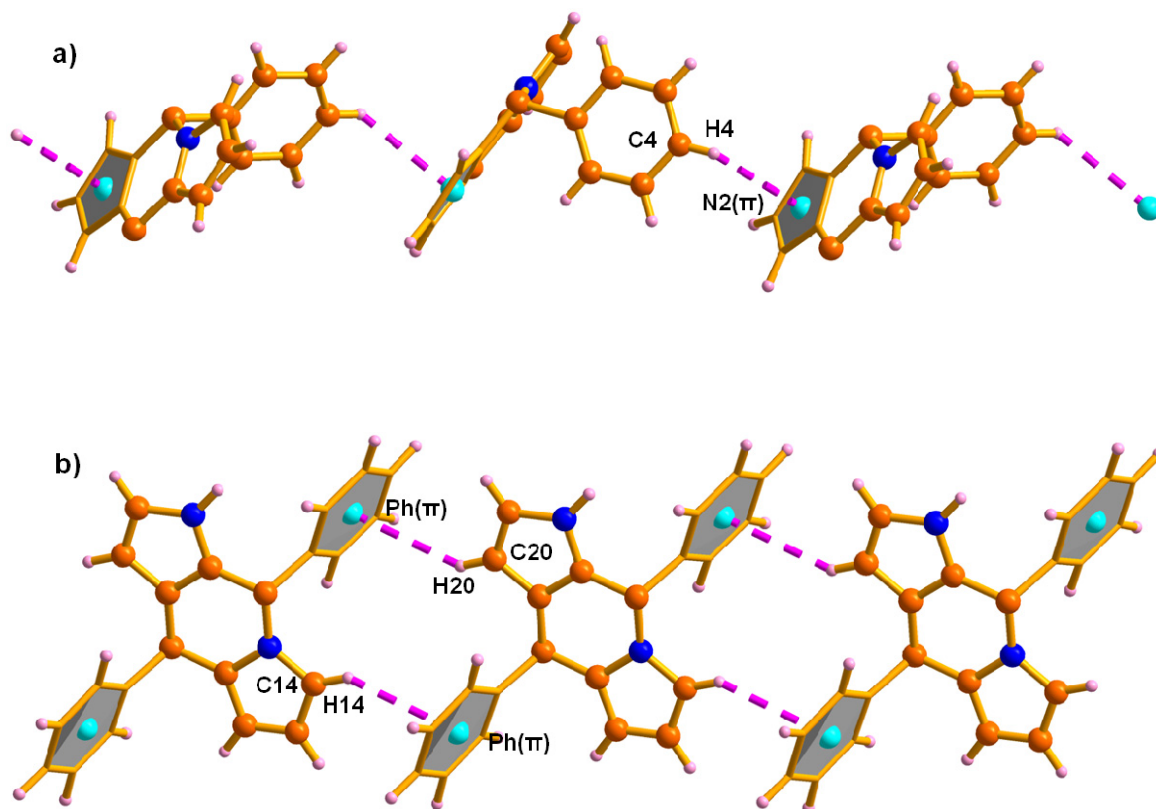


Figure S7. Single crystal X-ray analysis of **7b**. a) and b) are 1-D arrays. The intermolecular hydrogen bonding interactions are formed in a) between one of the *meso*-phenyl rings and α,β - linked pyrrolic π -cloud; b) between the pyrrolic β -CH and one of the *meso*-phenyl rings π -cloud and N, α pyrrolic α -CH and one of the *meso*-phenyl rings π -cloud. The distances and angles in (a and b) are: C4-H4...N2(π): 2.92 Å, 129°; C20-H20...Ph(π): 2.99 Å, 142° and C14-H14...Ph(π): 2.98 Å, 143°. The groups which are not involved in hydrogen bonding interactions are omitted for clarity.

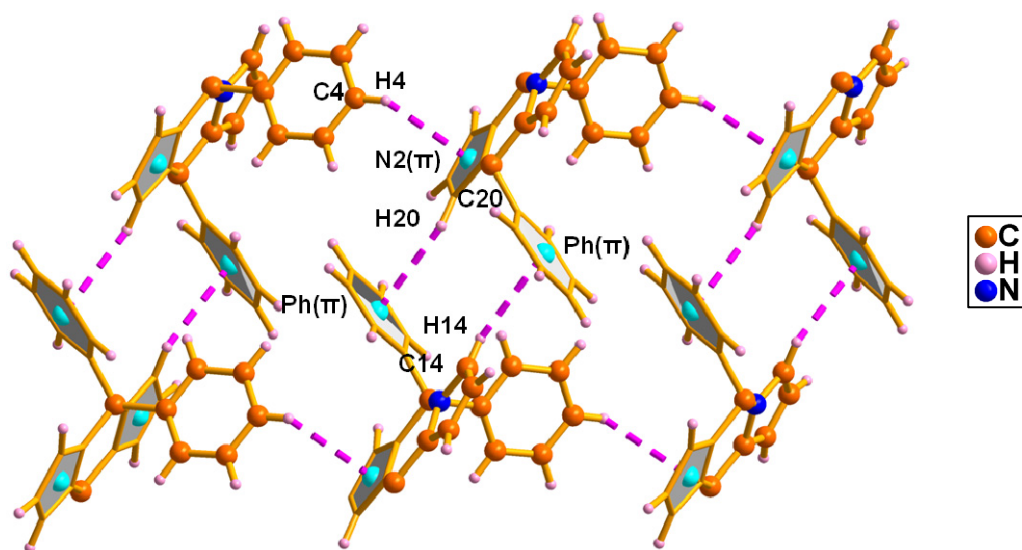


Figure S8. 2D array of **7b**. The intermolecular hydrogen bonding interactions are generated by combining the two one-dimensional interactions (*Figure S7*). The distances and angles are: C4-H4...N2(π): 2.92 Å, 129°; C20-H20...Ph(π): 2.99 Å, 142° and C14-H14...Ph(π): 2.98 Å, 143°. The groups which are not involved in hydrogen bonding interactions are omitted for clarity.

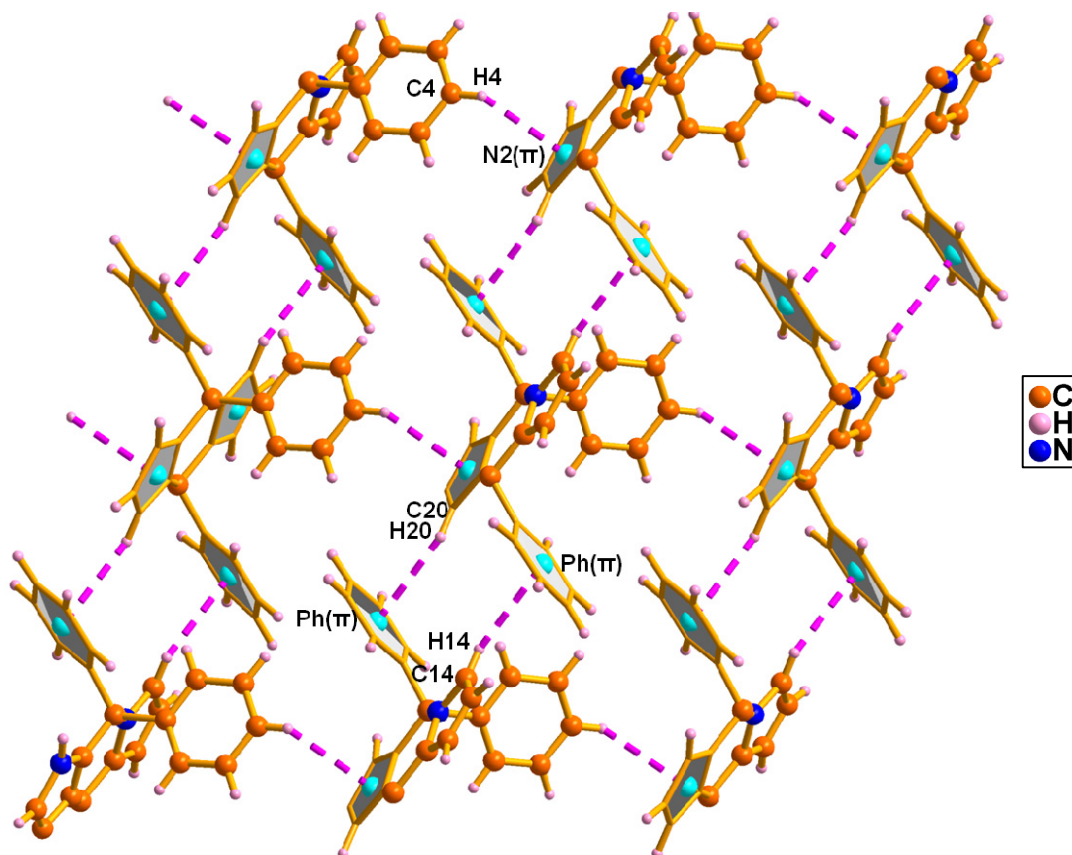


Figure S9. Supramolecular assembly of **7b**.