

## Supporting information

# Hydrogen Bonding Probes of Phenol –OH Groups: Shielding Ranges, Solvent Effects and Temperature Coefficients of $^1\text{H}$ NMR Shieldings and –OH Diffusion Coefficients

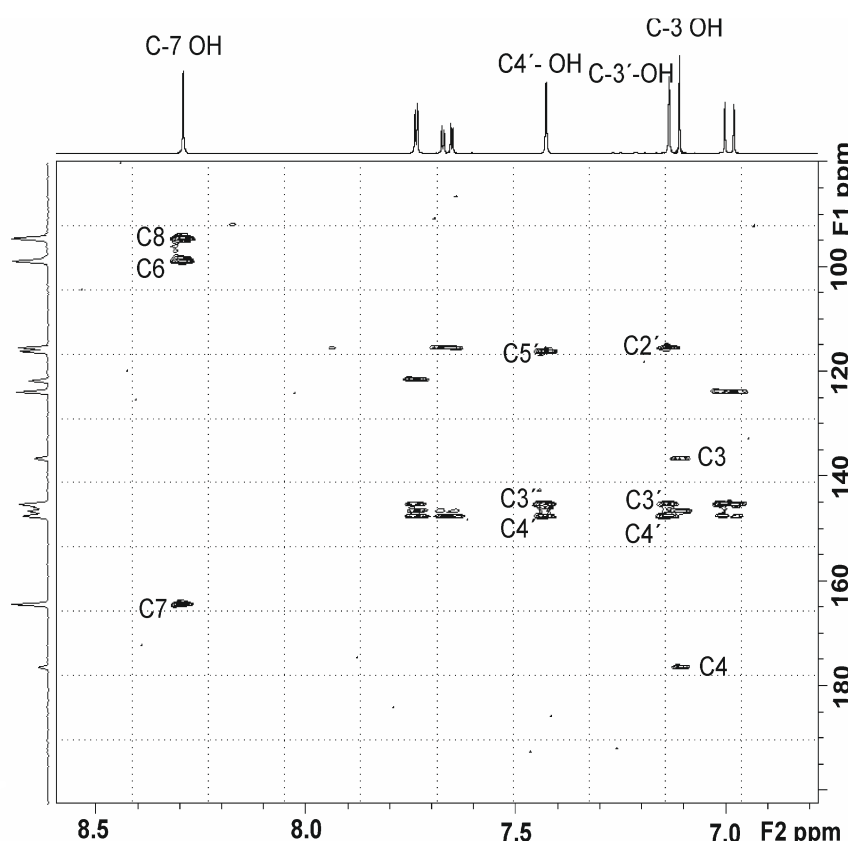
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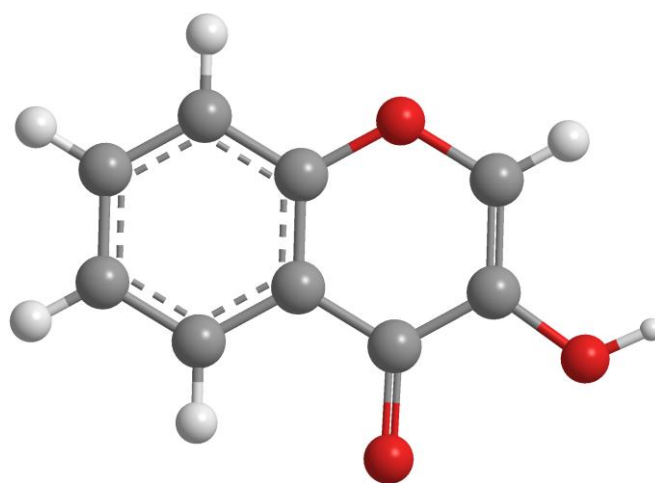
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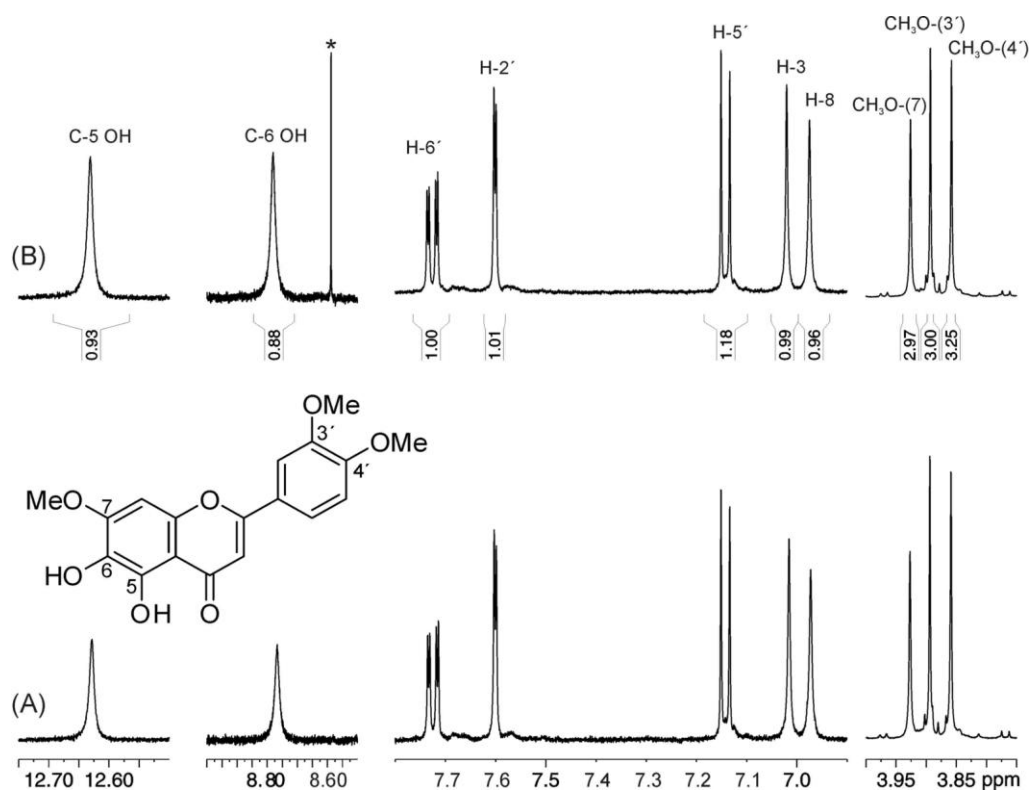
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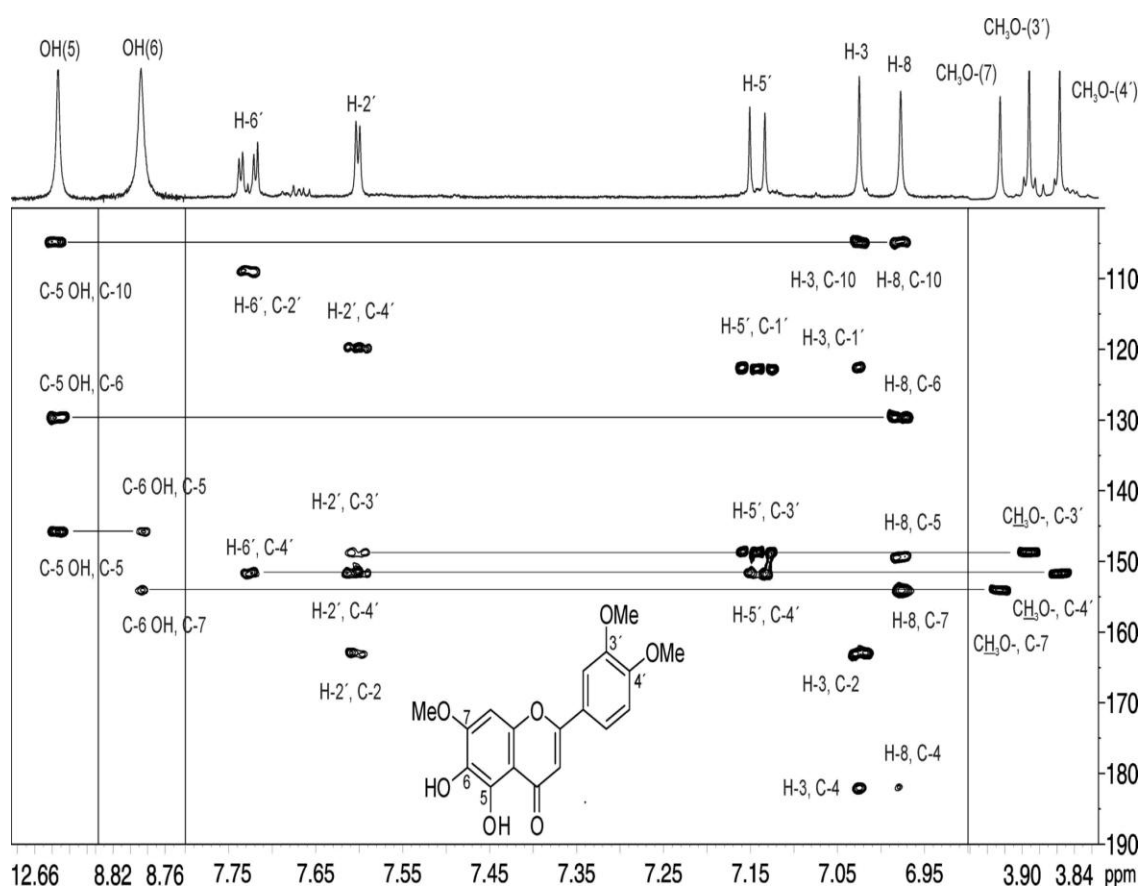
**FIGURE S1.** 2D  $^1\text{H}$  -  $^{13}\text{C}$  HMBC NMR spectrum (500 MHz) of quercetin, **1**, in  $\text{CD}_3\text{CN}$ , concentration 5 mM, with the presence of 2  $\mu\text{L}$  picric acid, 8 mM in  $\text{CD}_3\text{CN}$  (number of scans = 12, experimental time = 3 h).



**SCHEME S1.** Structural CSD search query.



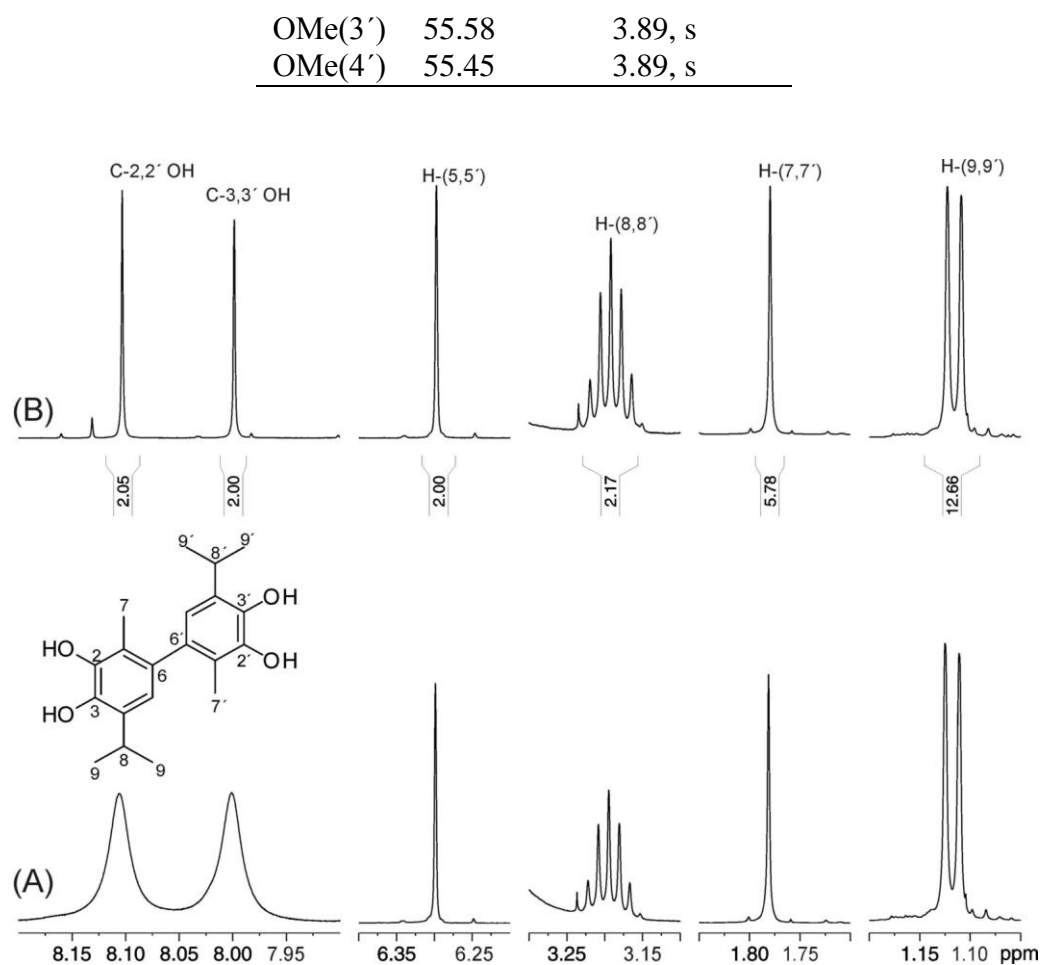
**FIGURE S2.** Selected regions of 500 MHz  $^1\text{H}$  NMR of compound, **10**, in 0.5 mL  $\text{DMSO}-d_6$  ( $T = 288\text{ K}$ , number of scans = 256, experimental time = 34 min). (A) without the addition of picric acid, (B) after the addition of 3  $\mu\text{L}$ , 2 mM in  $\text{DMSO}-d_6$ , solution of picric acid. The peak with the asterisk denotes the absorption of picric acid.



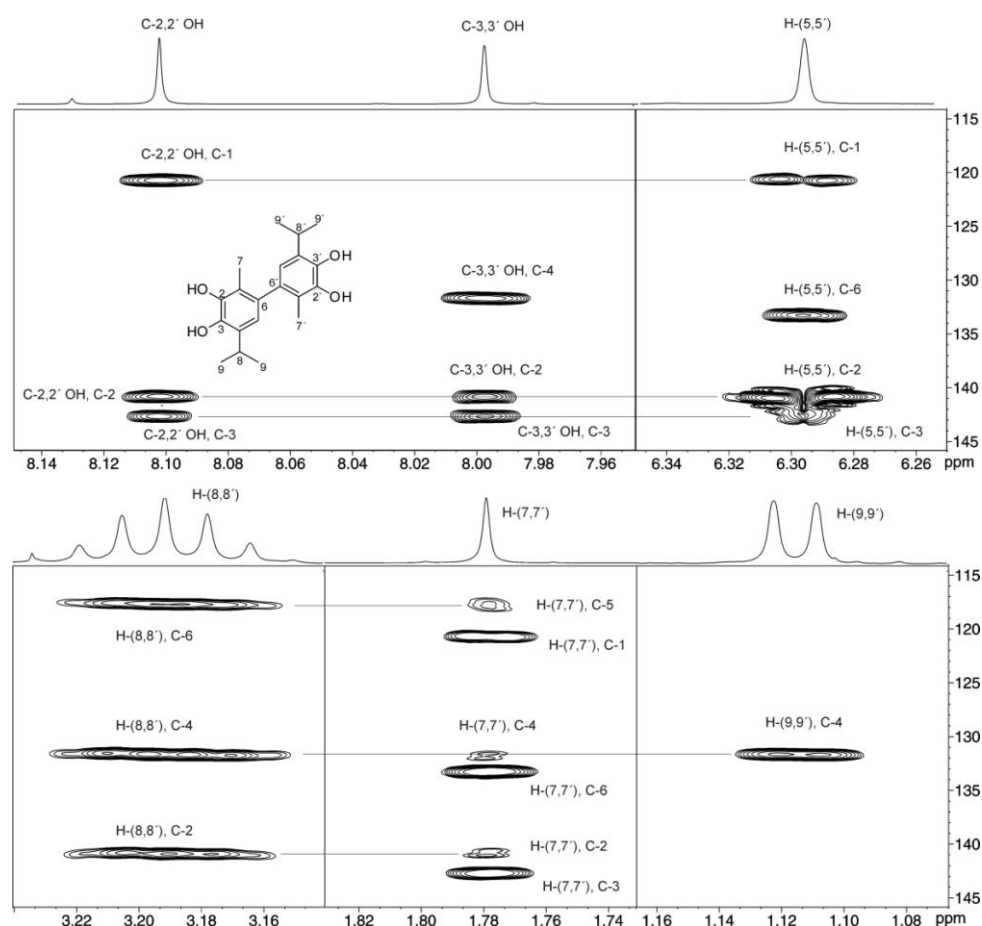
**FIGURE S3.** Selected regions of a 500 MHz 2D  $^1\text{H}$ - $^{13}\text{C}$  HMBC NMR ( $T = 288\text{K}$ , number of scans = 120, experimental time = 16 h) of compound, **10**, in 0.5 mL  $\text{DMSO}-d_6$ , after the addition of 3  $\mu\text{L}$ , 2 mM in  $\text{DMSO}-d_6$ , solution of picric acid. The common cross-peaks of -OH groups, C-5 OH and C-6 OH, and of the methoxy groups, are demonstrated.

**TABLE S1.** NMR Spectroscopic Data (500 MHz,  $\text{DMSO}-d_6$ ) for compound **10**.

notation	$\delta_{\text{C}}$	$\delta_{\text{H}}$ ( $J$ in Hz)
2	163.2	—
3	103.2	7.02, s
4	181.98	—
5	145.69	12.63, s
6	129.62	8.79, s
7	154.04	—
8	90.99	6.97, s
9	149.62	—
10	104.74	—
OMe(7)	56.04	3.93, s
1'	122.5	—
2'	108.91	7.60, d (2.2)
3'	148.64	—
4'	151.64	—
5'	111.3	7.14, d (8.7)
6'	119.7	7.72, dd (2.1, 8.6)



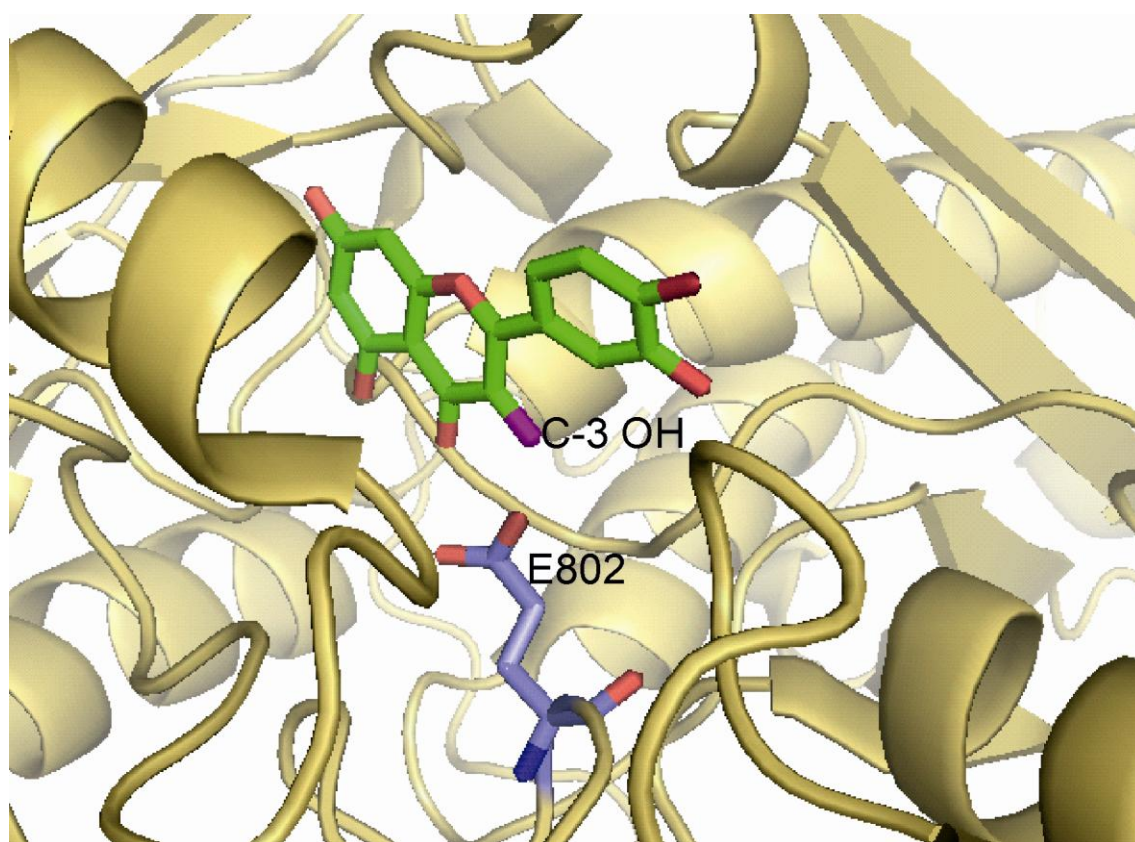
**FIGURE S4.** Selected regions of 500 MHz  $^1\text{H}$  NMR of compound, **26**, in 0.5 mL  $\text{DMSO-}d_6$  ( $T = 289\text{ K}$ , number of scans = 128, experimental time = 17 min). (A) without the addition of picric acid, (B) after the addition of 13  $\mu\text{L}$ , 10 mM in  $\text{DMSO-}d_6$ , solution of picric acid.



**FIGURE S5.** Selected areas of 500 MHz 2D  $^1\text{H}$ - $^{13}\text{C}$  HMBC NMR ( $T = 289\text{ K}$ , number of scans = 76, experimental time = 10 h) of compound, **26**, in 0.5 mL  $\text{DMSO-}d_6$ , with the addition of 13  $\mu\text{L}$ , 10 mM in  $\text{DMSO-}d_6$ , solution of picric acid. The common cross-peaks between the OH groups are indicated.

**TABLE S2.** NMR Spectroscopic Data (500 MHz,  $\text{DMSO-}d_6$ ) for compound **26**.

notation	$\delta_{\text{C}}$	$\delta_{\text{H}}$ ( $J$ in Hz)
1, 1'	120.65	—
2, 2'	140.95	8.10, s
3, 3'	142.7	8.00, s
4, 4'	131.65	—
5, 5'	117.57	6.30, s
6, 6'	133.30	—
7, 7'	13.47	1.78, s
8, 8'	26.36	3.19, m
9, 9'	22.90	1.12, d (6.9)



**FIGURE S6.** X-ray structure of xanthine oxidase in complex with quercetin (pdbid: 3NVY). The C-3 OH of quercetin interacts with the glutamate 802 (E802) of the protein (distance between OE1 of E802 and the C-3 OH was measured to 2.07 Å).