

Electronic supplementary information

**Analysis of guest binary mixtures by *tert*-butylcalix[6]arene using host memory
of previously bound guests**

Goulnaz D. Safina ^a, Marat A. Ziganshin ^a, Aidar T. Gubaidullin ^b and Valery V. Gorbachuk ^{a,*}

^a*Kazan Federal University, A.M. Butlerov Institute of Chemistry, Russia, 420008, Kazan,
Kremlevskaya, 18, Fax: +7 843 2337416, Tel: +7 843 2337309, E-mail:*

Valery.Gorbachuk@ksu.ru

^b*A.E. Arbuzov Institute of Organic and Physical Chemistry, Akad. Arbuzova 8, 420088 Kazan,
Russia.*

Supplemental TG/DSC/MS data for clathrates of *tert*-butylcalix[6]arene with CCl_4 , CHCl_3 , C_6H_6 , *c*- C_6H_{12} and products of saturation of calixarene **1** powder with headspace of C_6H_6 /*c*- C_6H_{12} and CCl_4 / CHCl_3 mixtures.

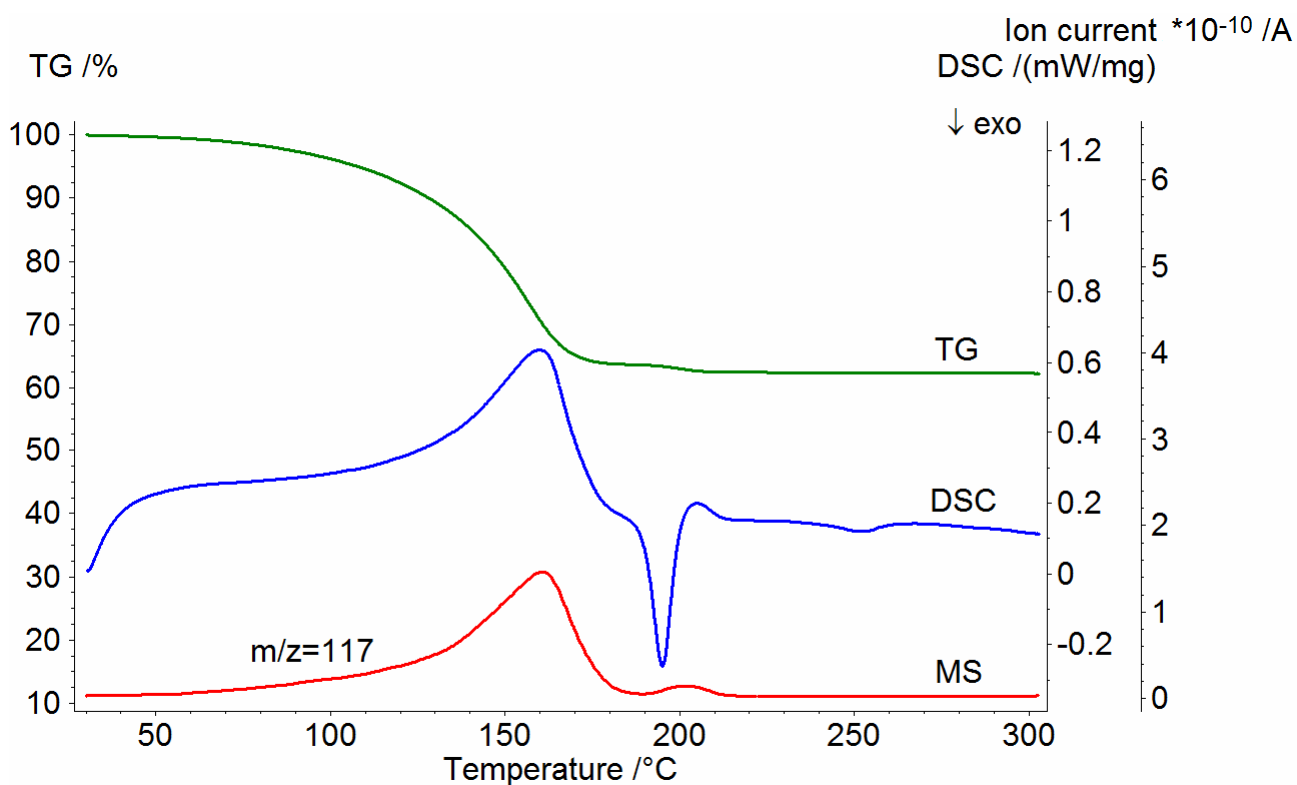


Fig. S1 The data of TG/DSC/MS analysis of *tert*-butylcalix[6]arene clathrate with CCl_4 . Heating rate 10 K min^{-1} .

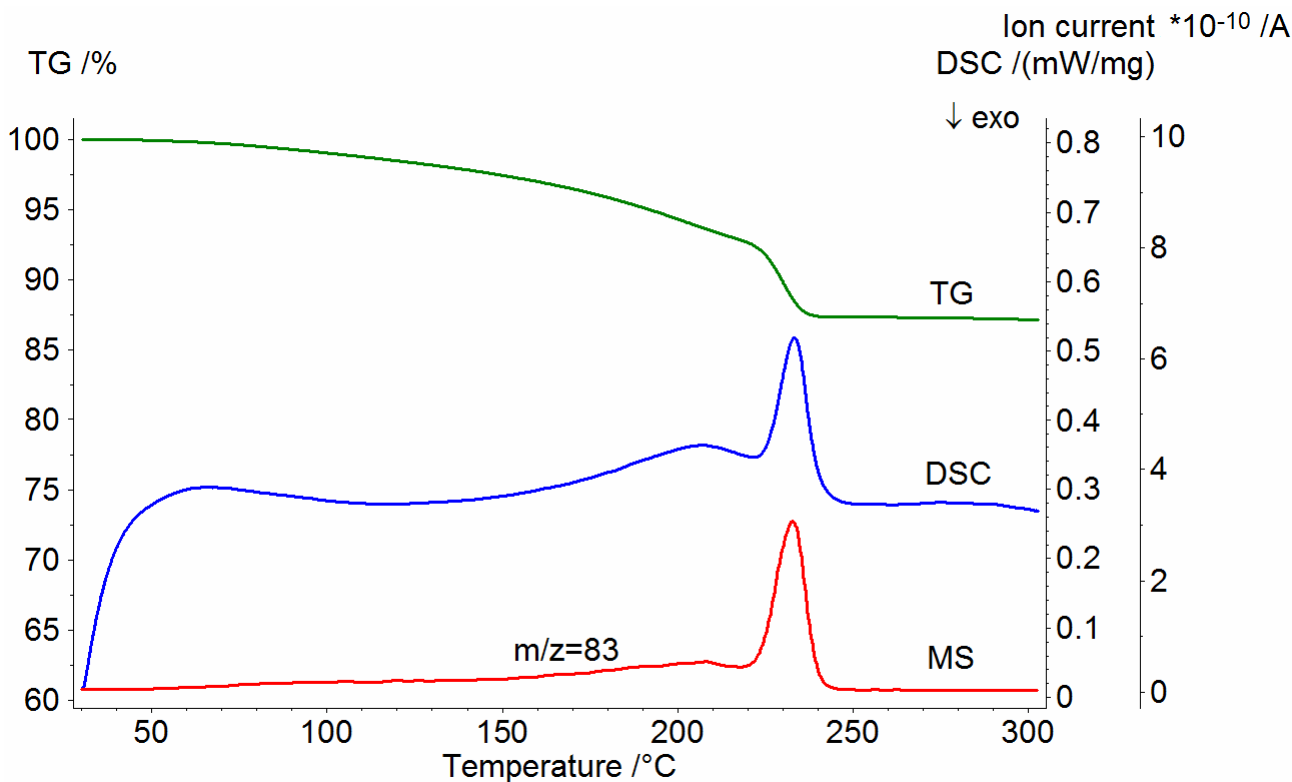


Fig. S2 The data of TG/DSC/MS analysis of *tert*-butylcalix[6]arene clathrate with CHCl_3 . Heating rate 10 K min^{-1} .

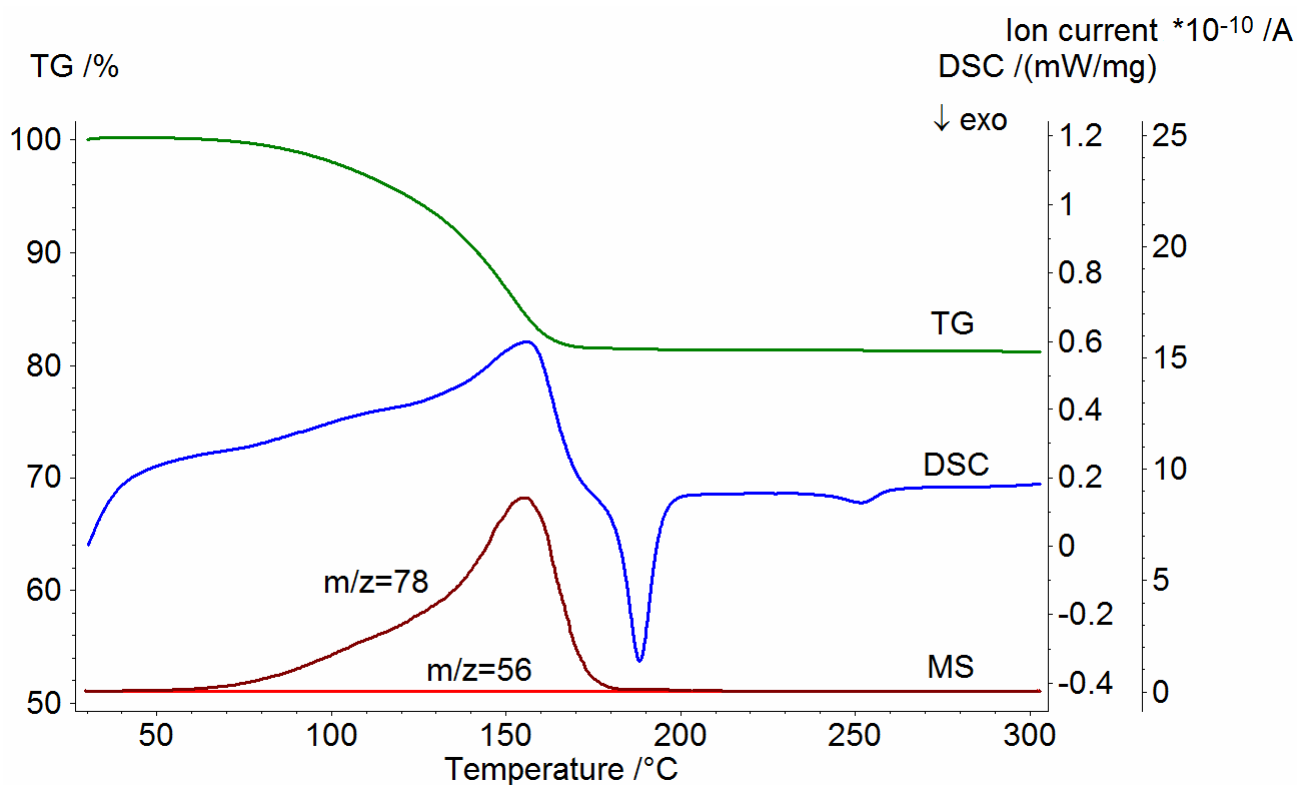


Fig. S3 The data of TG/DSC/MS analysis of *tert*-butylcalix[6]arene clathrate with C₆H₆. Heating rate 10 K min⁻¹.

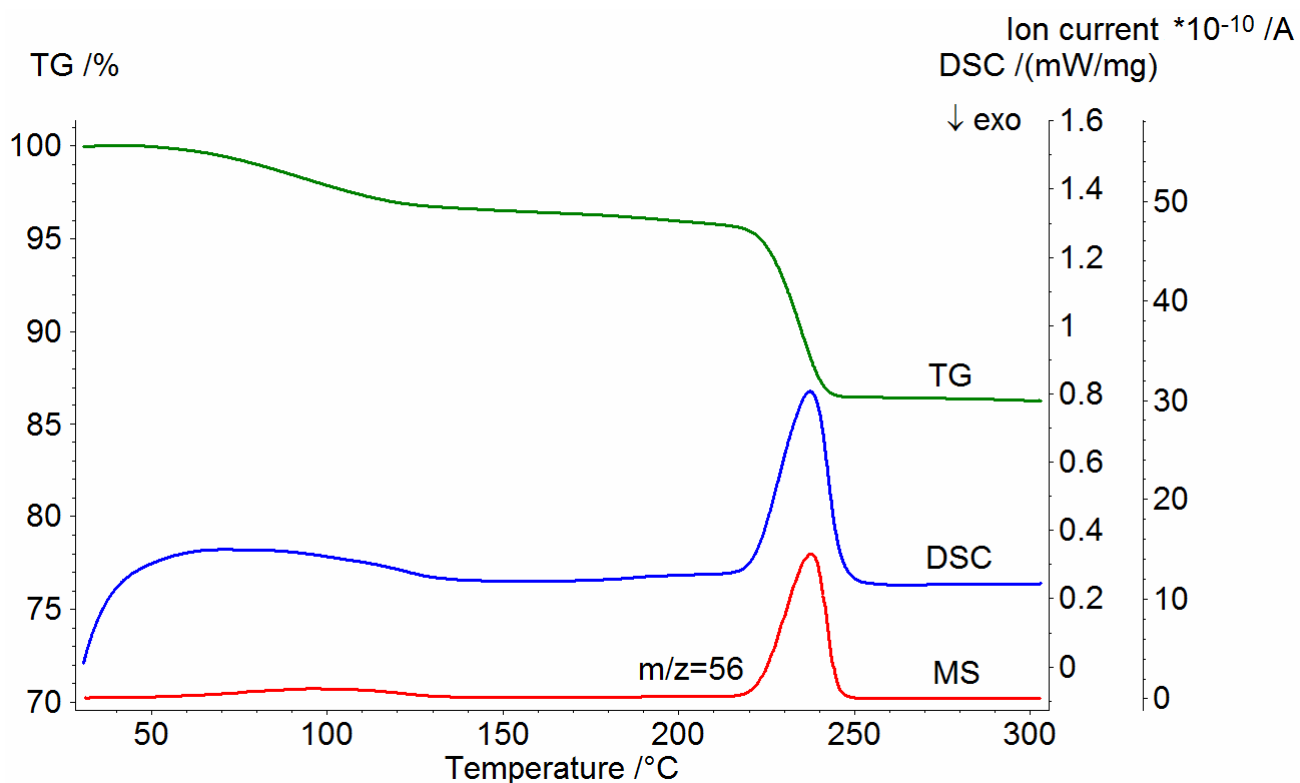


Fig. S4 The data of TG/DSC/MS analysis of *tert*-butylcalix[6]arene clathrate with *c*-C₆H₁₂. Heating rate 10 K min⁻¹.

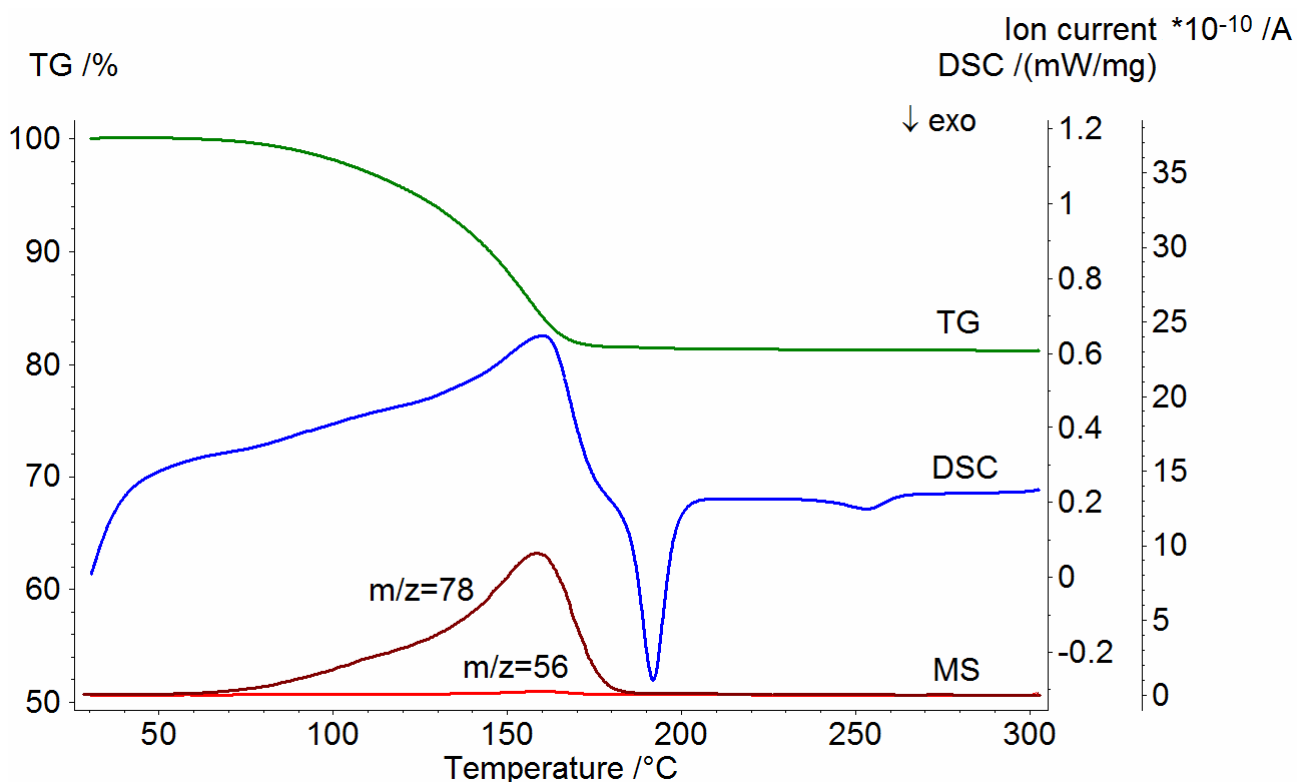


Fig. S5 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of cyclohexane solution (2 vol.%) in benzene at 25°C. Heating rate 10 K min⁻¹.

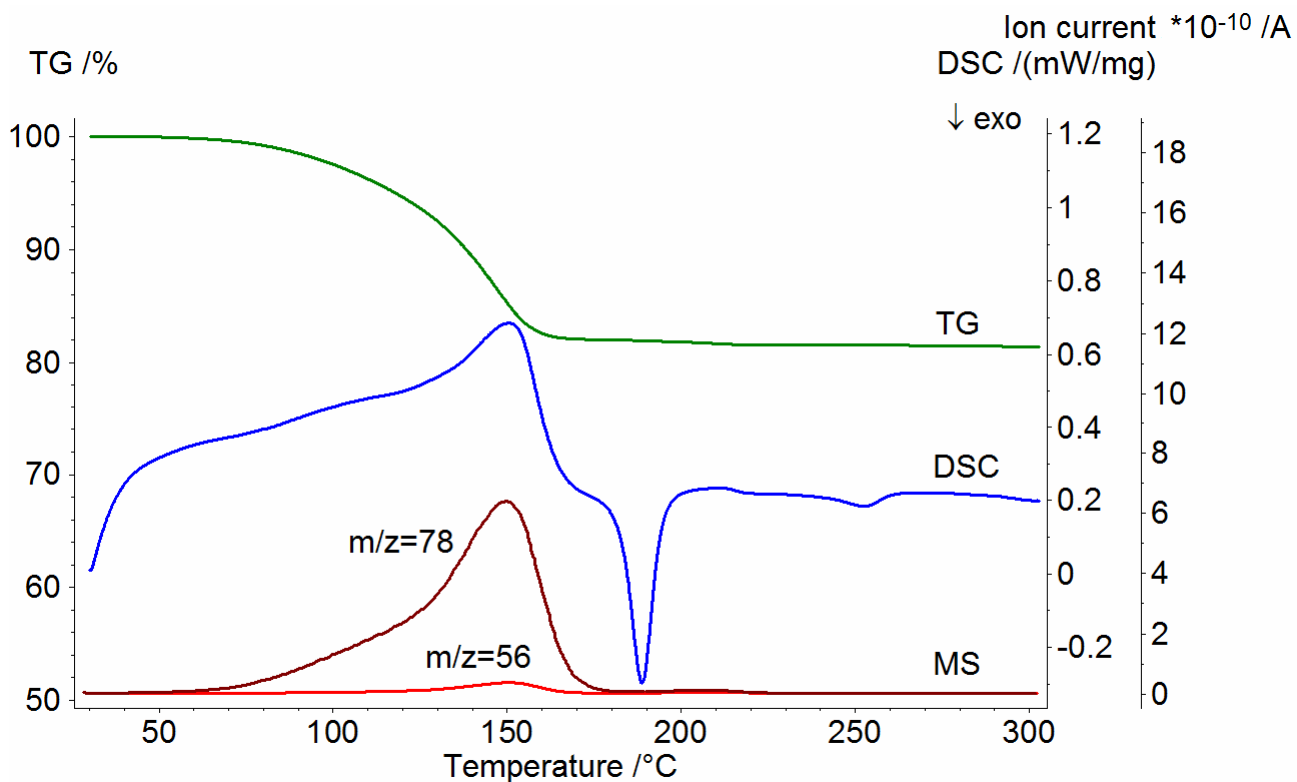


Fig. S6 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of cyclohexane solution (5 vol.%) in benzene at 25°C. Heating rate 10 K min⁻¹.

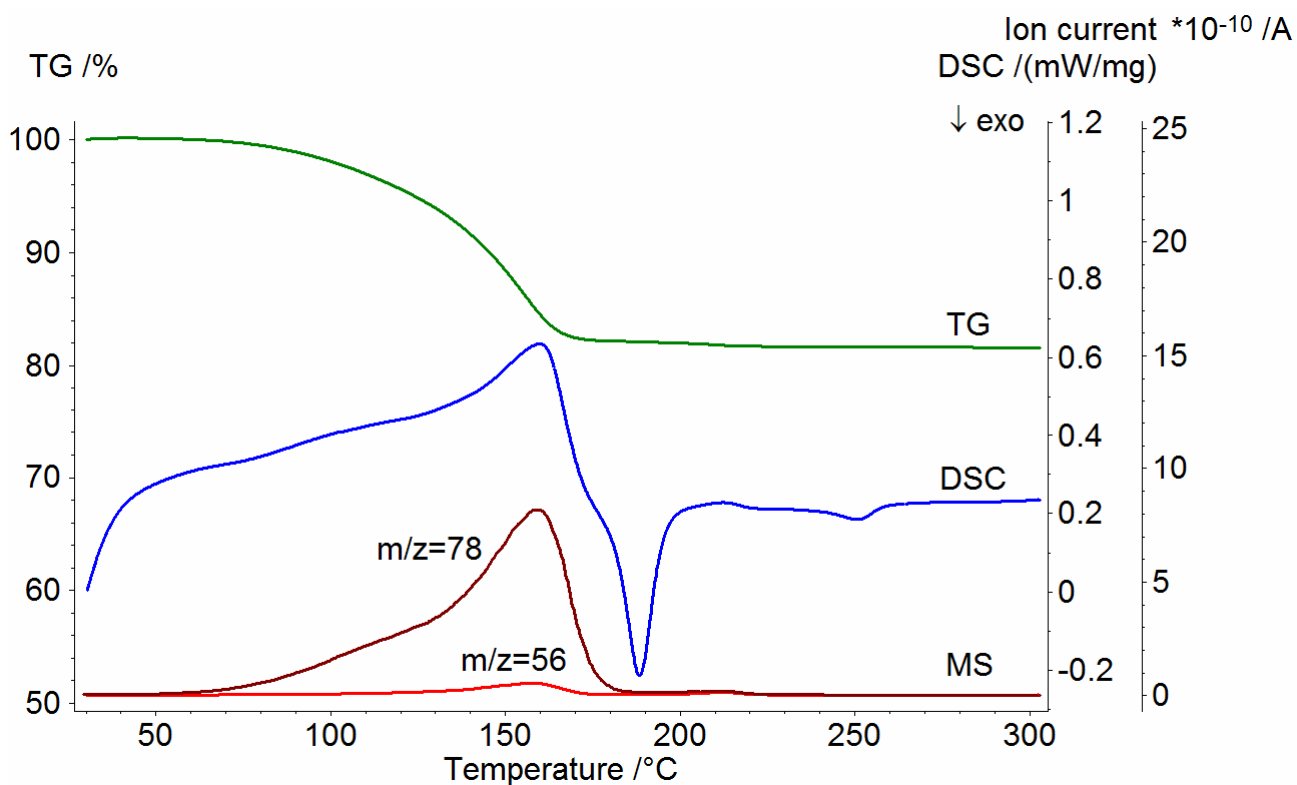


Fig. S7 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of cyclohexane solution (6 vol.%) in benzene at 25°C. Heating rate 10 K min⁻¹.

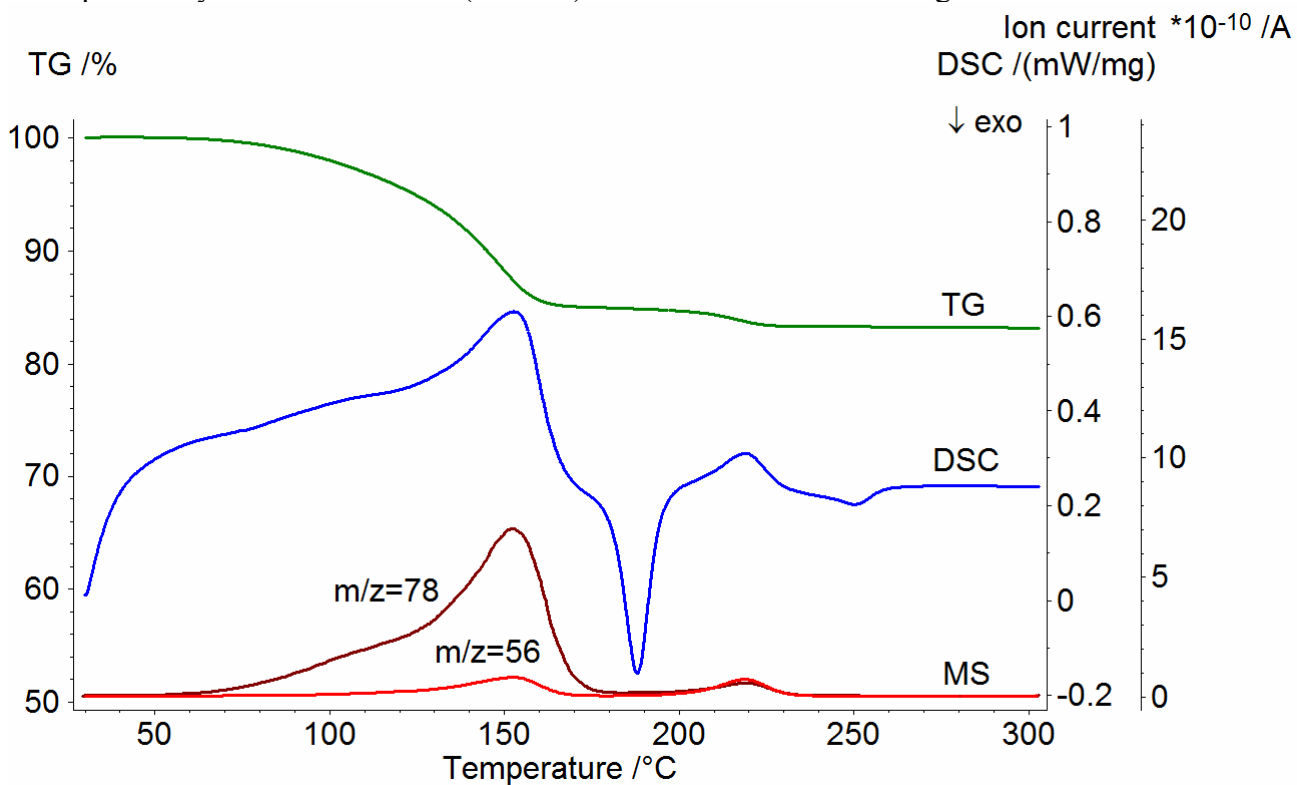


Fig. S8 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of cyclohexane solution (10 vol.%) in benzene at 25°C. Heating rate 10 K min⁻¹.

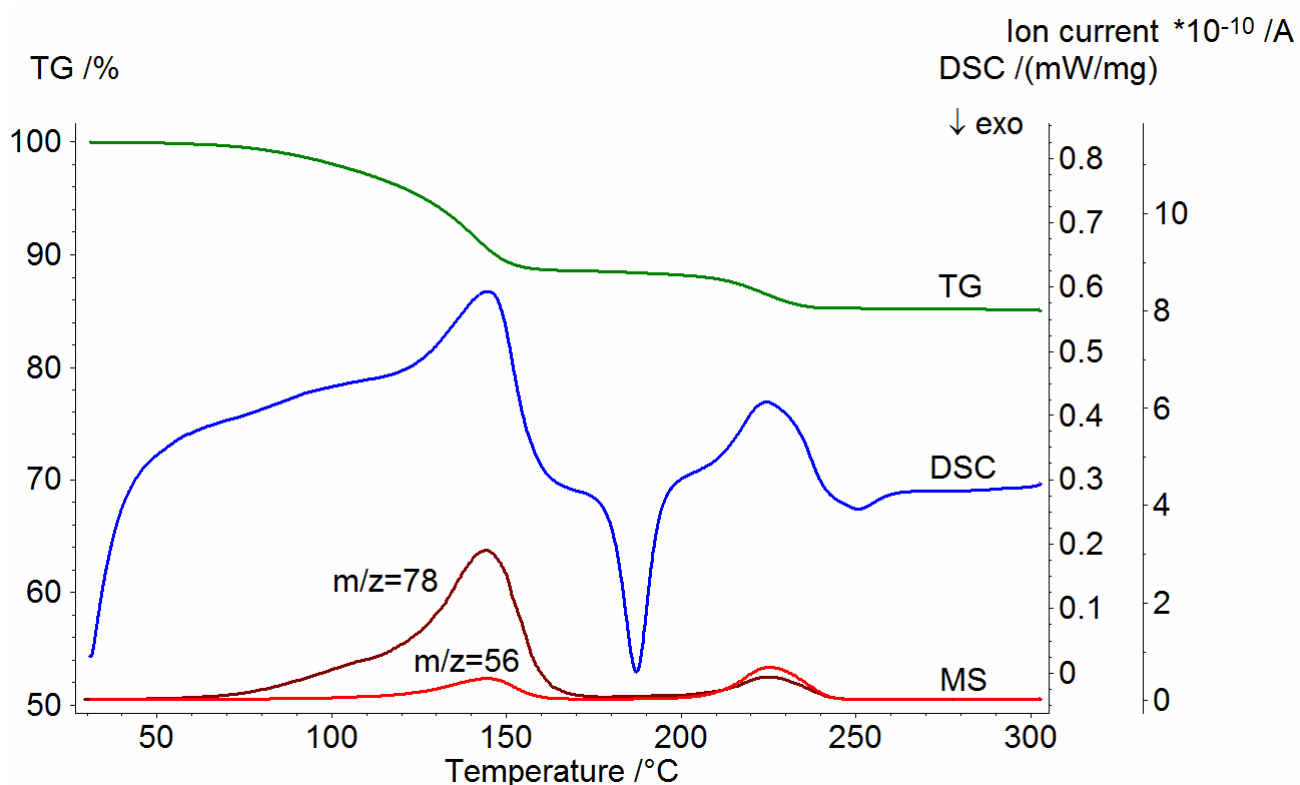


Fig. S9 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of cyclohexane solution (14 vol.%) in benzene at 25°C. Heating rate 10 K min⁻¹.

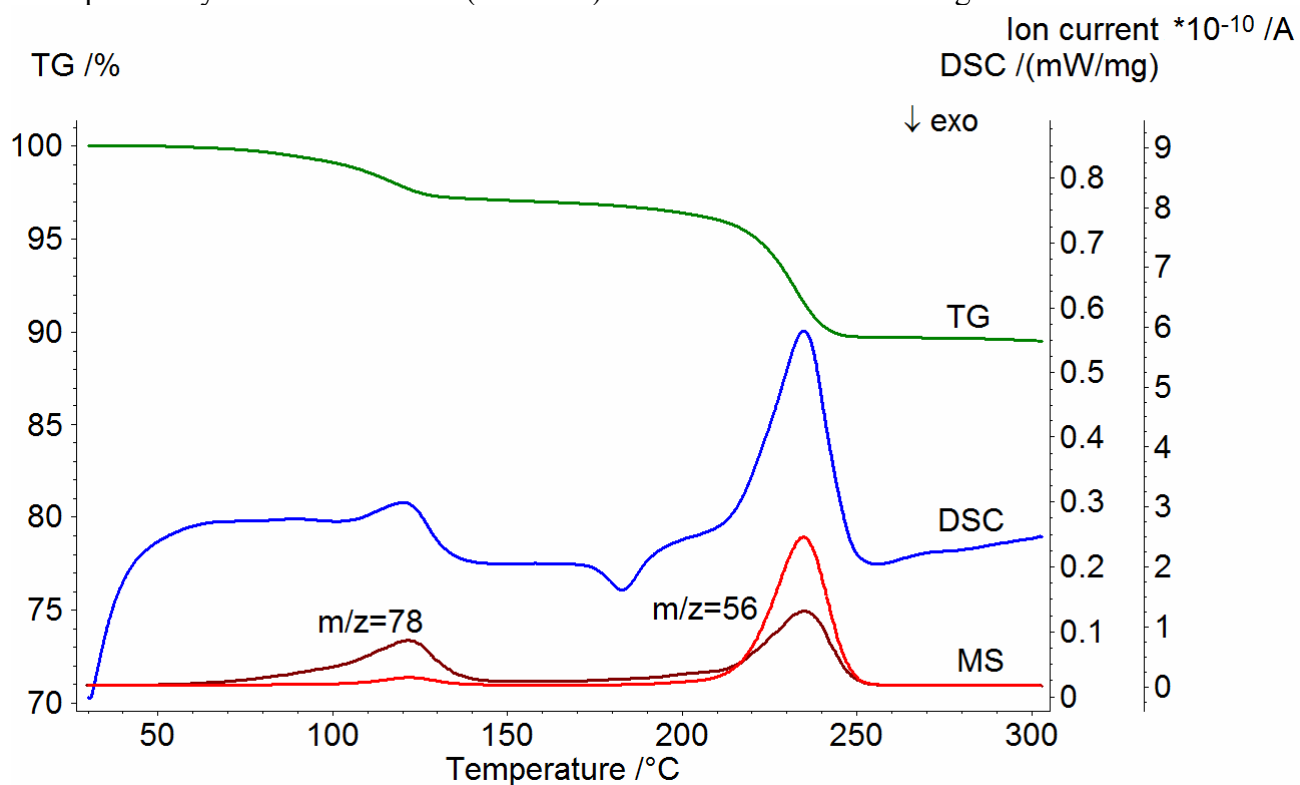


Fig. S10 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of cyclohexane solution (18 vol.%) in benzene at 25°C. Heating rate 10 K min⁻¹.

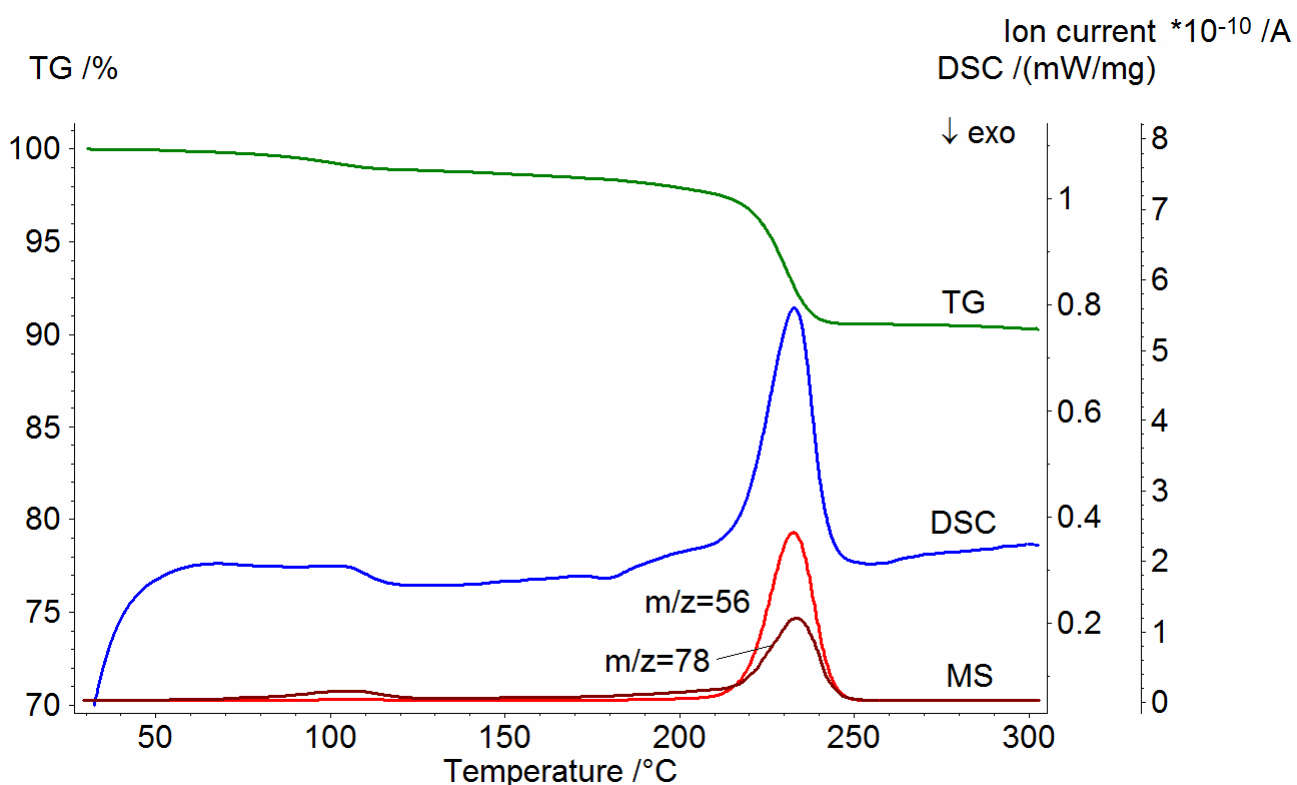


Fig. S11 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of cyclohexane solution (20 vol.%) in benzene at 25°C. Heating rate 10 K min⁻¹.

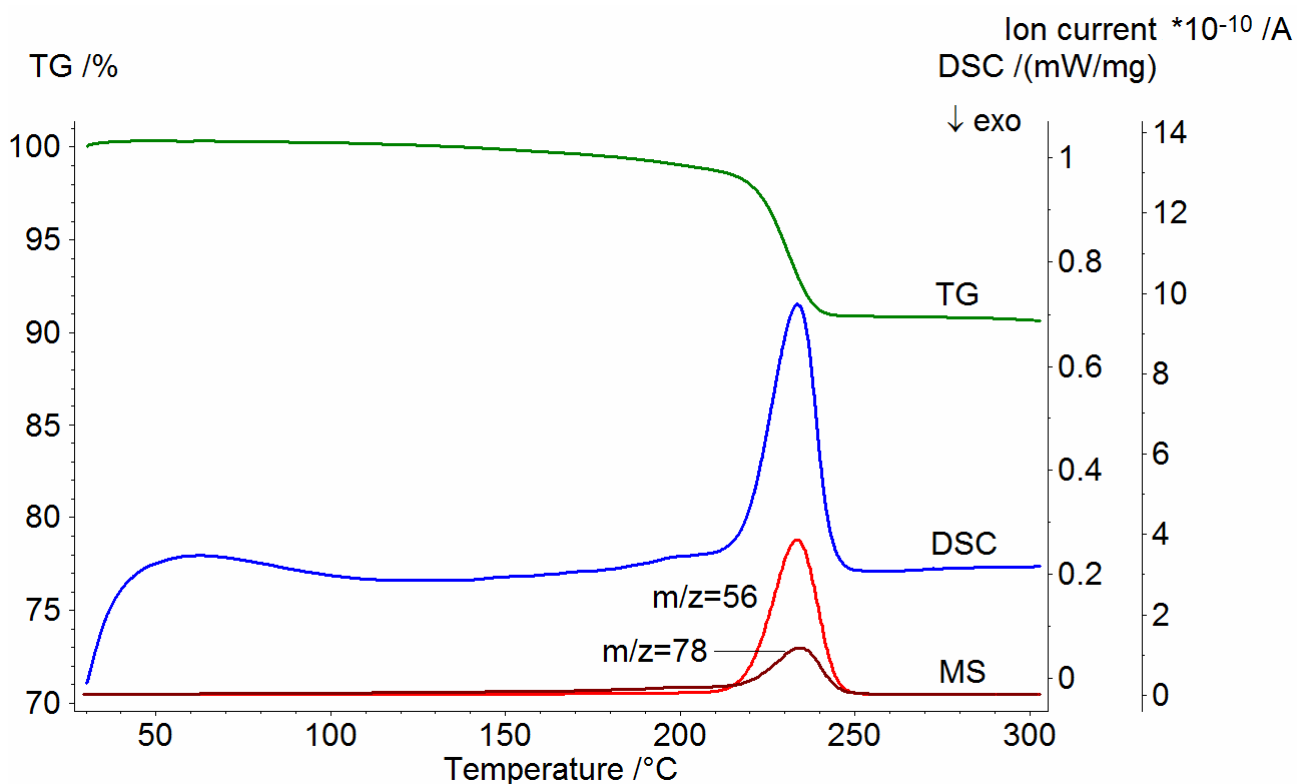


Fig. S12 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of cyclohexane solution (28 vol.%) in benzene at 25°C. Heating rate 10 K min⁻¹.

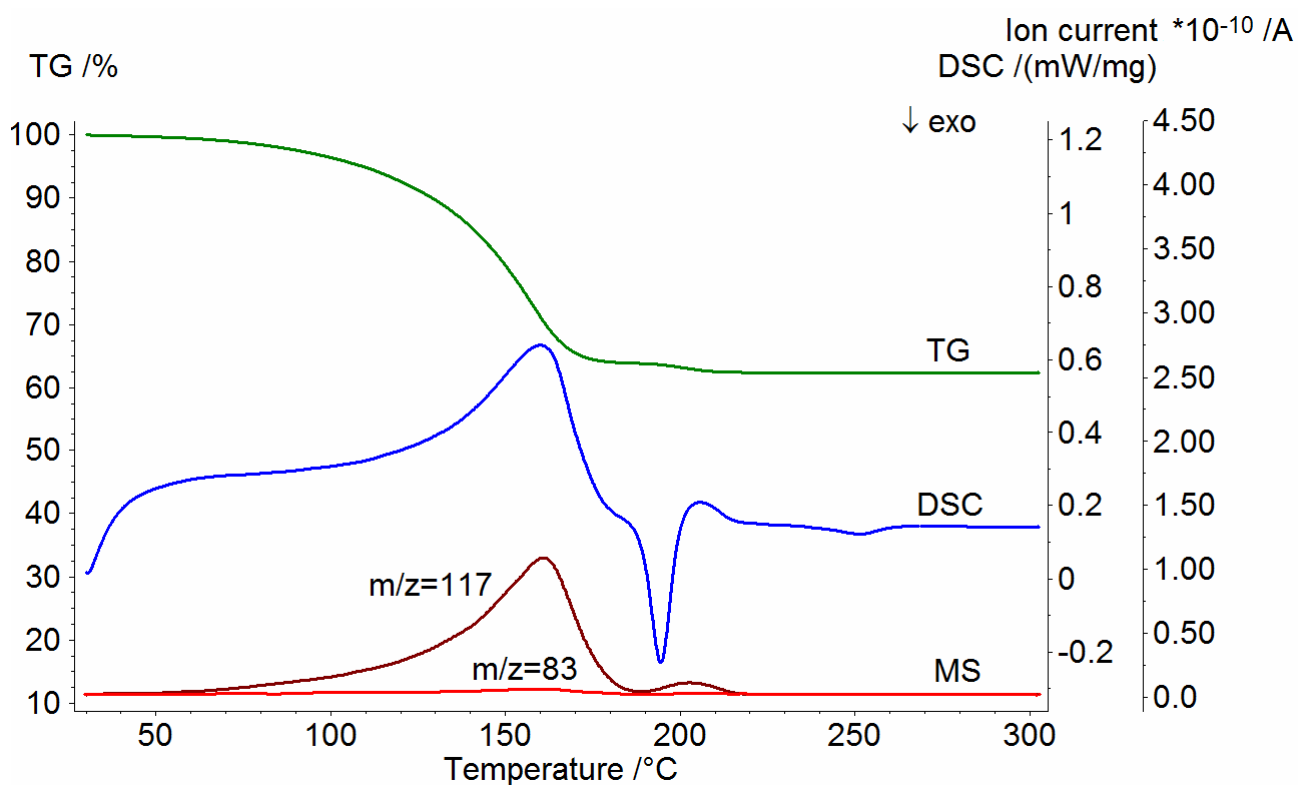


Fig. S13 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of chloroform solution (2 vol.%) in tetrachloromethane at 25°C. Heating rate 10 K min⁻¹.

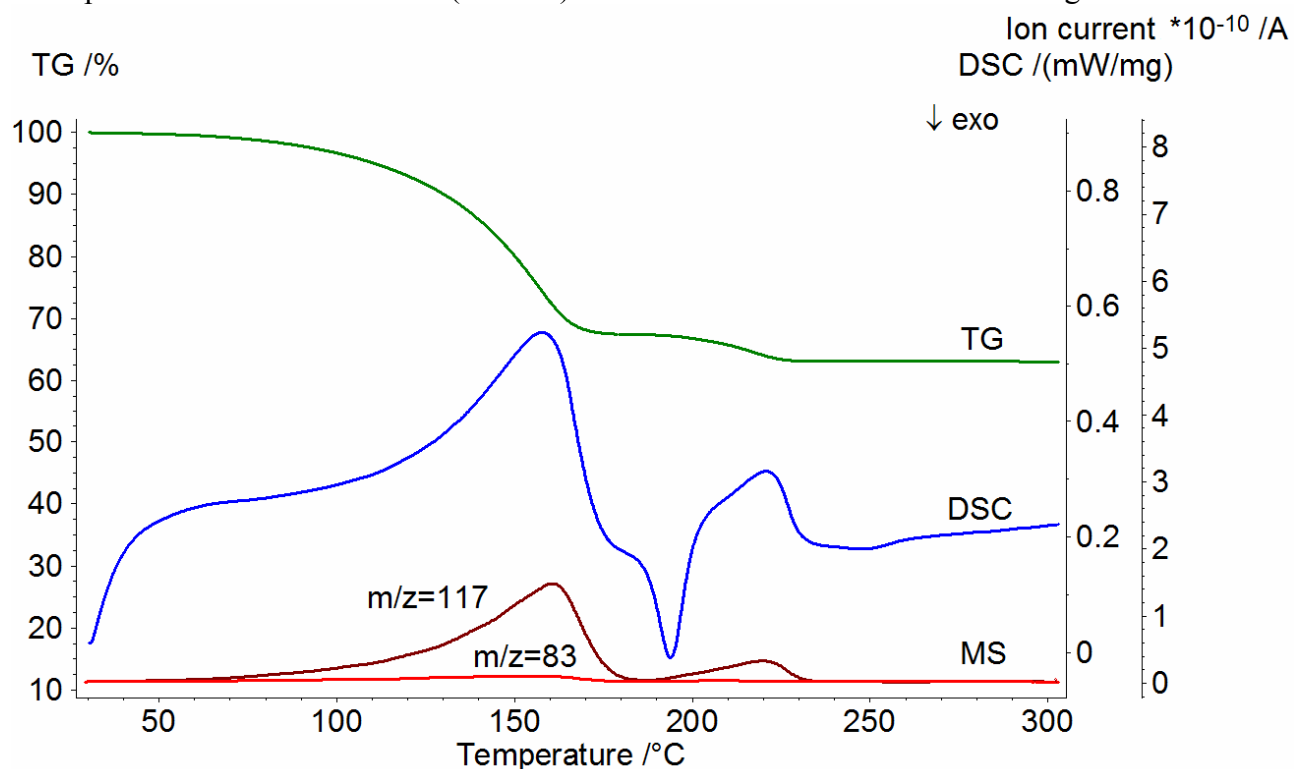


Fig. S14 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of chloroform solution (6 vol.%) in tetrachloromethane at 25°C. Heating rate 10 K min⁻¹.

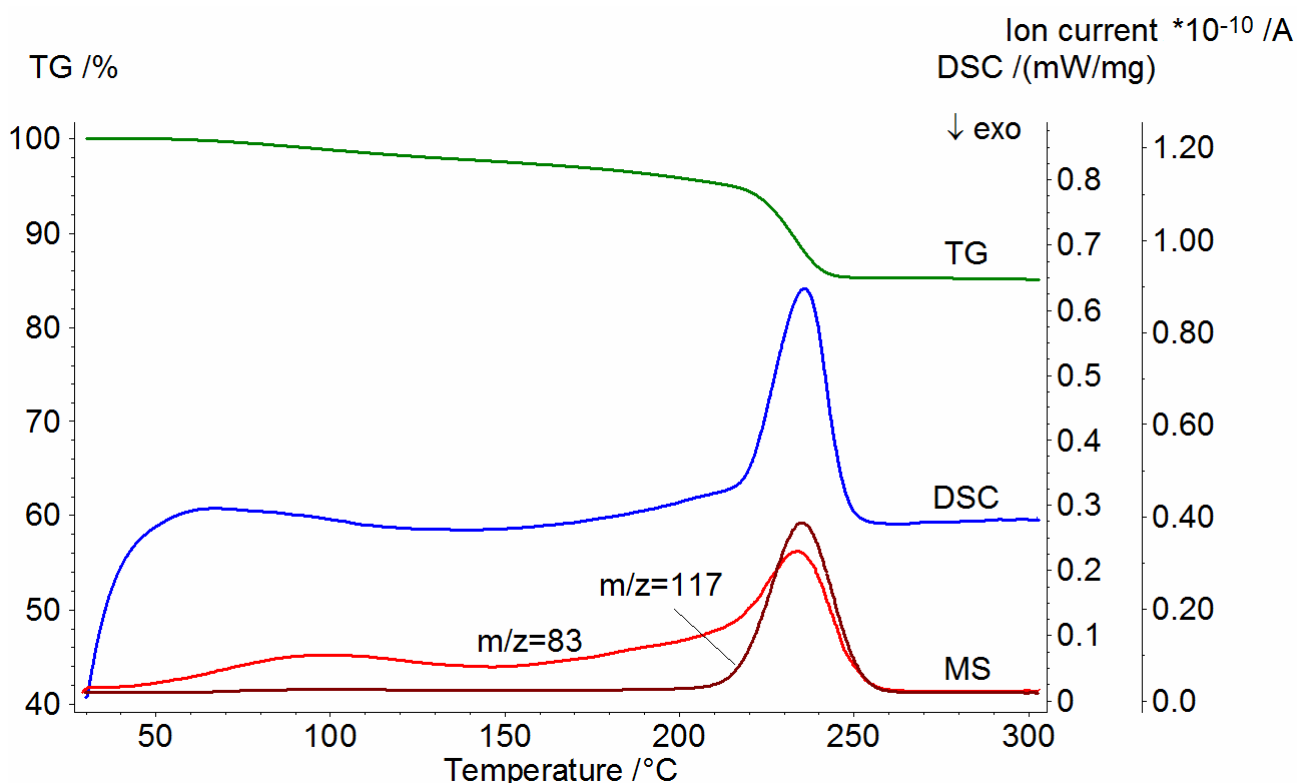


Fig. S15 The data of TG/DSC/MS analysis for product of *tert*-butylcalix[6]arene saturation by headspace of chloroform solution (70 vol.%) in tetrachloromethane at 25°C. Heating rate 10 K min⁻¹.

Table 1S. End points of the first decomposition step ($T_{end}^{(1)}$), temperatures of DTG-peaks ($T_{max}^{(1)}$, $T_{max}^{(2)}$) and the enthalpy of second exothermic peak ($\Delta H_{col}^{(2)}$) of decomposition of clathrates prepared by saturation of *tert*-butylcalix[6]arene with vapors of individual and mixed cyclohexane and benzene.

φ (<i>c</i> -C ₆ H ₁₂), vol.%	$T_{end}^{(1)}$, °C	$T_{max}^{(1)}$, °C	$T_{max}^{(2)}$, °C	$\Delta H_{col}^{(2)}$, kJ mol ⁻¹
0	163	152	-	-3
2	168	156	-	-4
5	158	147	210	-3
6	166	155	211	-3
8	158	148	213	-3
10	159	149	217	-3
12	154	143	220	-2
14	151	141	223	-2
16	139	129	228	-2
18	127	118	232	-2
20	111	101	230	-2
22	121	111	230	-1
24	-	0	230	-
28	-	0	231	-
30	-	0	230	-
50	-	0	231	-
100	118	91	235	-

Table 2S. End points of the first decomposition step ($T_{end}^{(1)}$), temperatures of DTG-peaks ($T_{max}^{(1)}$, $T_{max}^{(2)}$) and the second exothermic peak enthalpy ($\Delta H_{col}^{(2)}$) of decomposition of clathrates prepared by saturation of *tert*-butylcalix[6]arene with vapors of individual and mixed chloroform and tetrachloromethane.

φ (CHCl ₃), vol. %	$T_{end}^{(1)}$, °C	$T_{max}^{(1)}$, °C	$T_{max}^{(2)}$, °C	$\Delta H_{col}^{(2)}$, kJ mol ⁻¹
0	168	157	200	-2
2	168	158	201	-3
4	167	158	215	-2
6	166	157	217	-2
8	153	144	222	-2
12	153	143	224	-
15	143	130	225	-
20	132	105	237	-
56	-	81	235	-
70	-	90	233	-
100	-	202	230	-

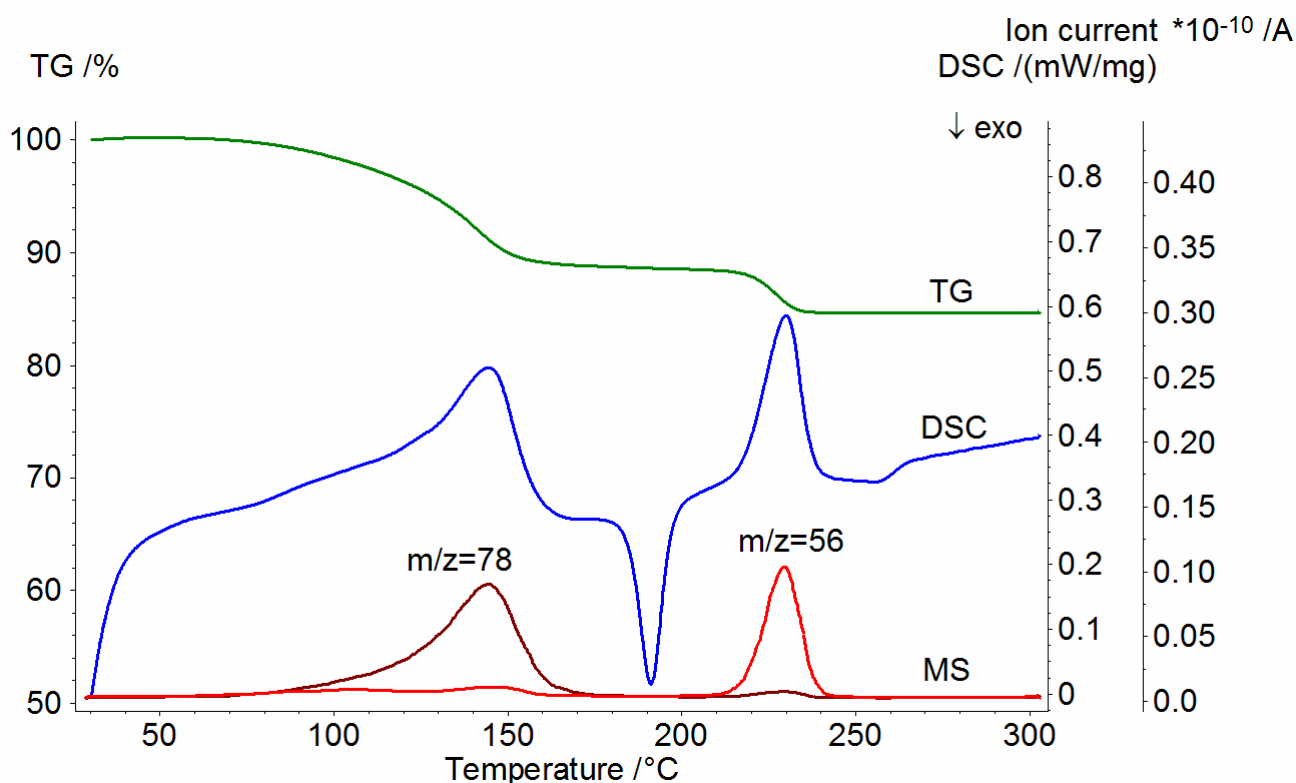


Fig. S16 The data of TG/DSC/MS analysis for mechanical mixture of host **1** clathrates with C₆H₆ and *c*-C₆H₁₂. Ion thermograms of C₆H₆ ($m/z=78$) and *c*-C₆H₁₂ ($m/z=56$) are shown. Heating rate 10 K min⁻¹.

The mechanical mixture contains 54% (w/w) of **1**•2.86 C₆H₆ and 46% (w/w) of **1**•1.18 *c*-C₆H₁₂. This mixture gives nearly the same guest ratio in solid host phase as the product of host **1** saturation by headspace of cyclohexane solution (16 vol.%) in benzene at 25°C.